

# Molecular Dynamics Simulation of Bacteriorhodopsin's Photoisomerization Using Ab Initio Forces for the Excited Chromophore

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**ABSTRACT** Retinal proteins are photoreceptors found in many living organisms. They possess a common chromophore, retinal, that upon absorption of light isomerizes and thereby triggers biological functions ranging from light energy conversion to phototaxis and vision. The photoisomerization of retinal is extremely fast, highly selective inside the protein matrix, and characterized through optimal sensitivity to incoming light. This article describes the first report of an ab initio quantum mechanical description of the in situ isomerization dynamics of retinal in bacteriorhodopsin, a microbial retinal protein that functions as a light-driven proton pump. The description combines ab initio multi-electronic state molecular dynamics of a truncated retinal chromophore model (*N*-methyl- $\gamma$ -methylpenta-2,4-dieniminium cation fragment) with molecular mechanics of the protein motion and unveils in complete detail the photoisomerization process. The results illustrate the essential role of the protein for the characteristic kinetics and high selectivity of the photoisomerization: the protein arrests inhomogeneous photoisomerization paths and funnels them into a single path that initiates the functional process. Supported by comparison with dynamic spectral modulations observed in femtosecond spectroscopy, the results identify the principal molecular motion during photoisomerization.

## INTRODUCTION

Retinal proteins, often called rhodopsins, are involved in a variety of responses of living cells to light, such as conversion of light to chemical energy and phototaxis in bacteria, or vision in higher organisms. All rhodopsins contain the molecule retinal as their chromophore. The activation of rhodopsins is initiated by a photoinduced isomerization of retinal around a specific double bond, which represents one of the fastest chemical reactions in nature (Birge, 1990). The extremely fast (200–500 fs) process prevents undesired nonspecific relaxation of the chromophore from the electronically excited state and, hence, furnishes pigments of high sensitivity. Bacteriorhodopsin (bR), shown in Fig. 1, functions as a light-driven proton pump in the purple membrane of *Halobacterium salinarum* (Oesterhelt and Stoechenius, 1971) and is the most investigated member of the family. Photoisomerization of the chromophore from the all-*trans* to the 13-*cis* configuration triggers a photocycle during which one proton is pumped from the cytoplasmic side to the extracellular side of the membrane. Numerous femtosecond spectroscopic experiments (Mathies et al., 1988; Gai et al., 1998; Ruhman et al., 2002) have been performed to identify the kinetics of the photoisomerization, which is schematically illustrated in Fig. 1. Time-resolved vibrational spectroscopy has suggested that bond stretching motions are the primary conformational changes of retinal in the excited state (Song and El-Sayed, 1998). Recent progress in such measurements has made it also possible to observe in real time wavepacket

motions of molecular vibrations during the isomerization of retinal in bR (Ye et al., 1999; Kobayashi et al., 2001) and in the visual receptor rhodopsin (Wang et al., 1994) through dynamic spectral modulations of photoabsorption and emission spectra. These studies have revealed characteristic vibrational motions along with their time-dependent couplings in the transient processes and demand a detailed interpretation in terms of the related protein function.

One long-standing question has been the role of the protein environment in the isomerization process, since the efficiency, bond selectivity, and dynamics of retinal's photoisomerization are strongly influenced by its surrounding. As the chromophore is a polyene, there are several double bonds capable of isomerization. In the homogeneous environment of a solution, photoisomerization of all-*trans* retinal results in 9-, 11-, and 13-*cis* products with quantum yields of 0.02, 0.14, and 0.01, respectively (Koyama et al., 1991). In bR, however, the photoisomerization has a much higher quantum yield (~0.6) (Tittor and Oesterhelt, 1990) and selectively leads to the formation of the 13-*cis* conformer (Birge, 1990). The rate of retinal isomerization also strongly depends on the surrounding environment. For instance, the rate in solution (Logunov et al., 1996) and in bR mutants (Song et al., 1993) is significantly lower than that in wild-type bR.

To address theoretically the issues described above, three key methodological advances are required. First, since the photoisomerization process involves molecular motions in many degrees of freedom and proceeds on multi-electronic state energy surfaces, methodology is needed that is capable of describing the chromophore's electronic structure and potential surfaces without imposing assumptions. Recently, high level ab initio quantum mechanical (QM) calculations have become applicable to the chromophore models of

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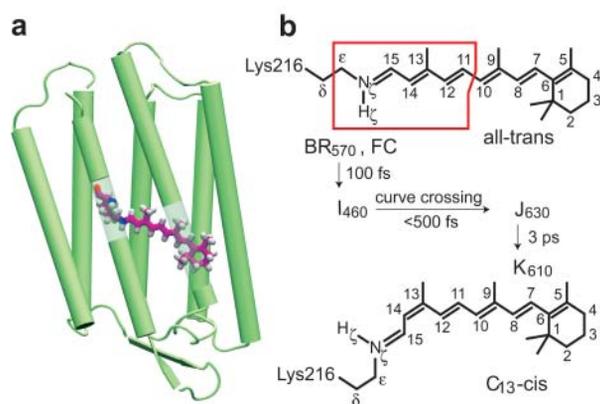


FIGURE 1 (a) Structure of bR. Seven transmembrane helices represented by green cylinders form a channel that embeds a retinal chromophore highlighted in purple. (b) Chemical structure of the retinal chromophore and its photoisomerization reaction in bR identified experimentally (Mathies et al., 1988; Gai et al., 1998; Ruhman et al., 2002). The retinal chromophore has a polyene chain with six double bonds and is bound to Lys-216 via a Schiff base. In the present simulation, a short chromophore model consisting of the Schiff base half, i.e., three conjugated double bonds from  $C_2$  to  $C_{11}$  enclosed by a red line, is employed as the quantum mechanically treated molecule. In the resting state of bR ( $BR_{570}$ ), retinal assumes an all-*trans* configuration. The isomerization process starts with a photoexcitation at 570 nm to the Frank-Condon region. 100 fs after the photoexcitation, a well-characterized excited state, called  $I_{460}$ , with absorption at 460 nm and emission in a longer wavelength region forms. The decay of  $I_{460}$  within the next 500 fs coincides with the isomerization process and results in a vibrationally hot 13-*cis* product in the ground state,  $J_{630}$ . Vibrational cooling in 3 ps leads to  $K_{610}$ , which is the first intermediate that can be trapped in cryogenic conditions.

increasing size (Garavelli et al., 1997, 1998; González-Luque et al., 2000; De Vico et al., 2002; Vreven and Morokuma, 2000; Molnar et al., 2002; Ben-Nun et al., 2002). The calculations have given reaction energy curves along minimum energy reaction paths (MEPs) or potential profiles around the conical intersection, revealing the complex nature of the potential energy surfaces of the photoisomerization process. Moreover, the description of reaction process requires not only accurate potential surfaces, but also the nonadiabatic electronic coupling between different electronic states that induces electronic transitions during the photoisomerization. Recent computational advance in this regard, amenable through sophisticated program packages, enables one to compute the nonadiabatic electronic coupling along all nuclear coordinates degrees of freedom.

Second, the dynamics of the reaction process should be explicitly described. Although ab initio QM analyses of the MEP energy curves have provided a basic insight into the reaction process, actual dynamic paths can deviate from the MEP due to thermal fluctuation and reaction excess energy. Especially, the characteristic incoherent kinetics and dynamic spectral modulations found in the experimental studies can be directly captured only through molecular dynamics (MD) simulations. In this respect, it is fortunate that ab initio QM MD simulations have finally been feasible

to compute on the fly energies and forces of electronically excited molecules in isomerization processes (Vreven et al., 1997; Ben-Nun and Martínez, 1998; Doltsinis and Marx, 2002). These studies shed light on the complex photoisomerization dynamics that involves intramolecular vibrational energy transfers among multi-dimensional reaction coordinates.

Third, the photoisomerization simulation has to take into account the protein environment to elucidate the molecular mechanism of the protein environment catalytic effect essential for the function. The success of the ab initio QM MD simulation has been restricted so far to applications for small systems in isolated condition (in vacuo). The photoisomerization in protein environments (in situ) has been investigated only in the framework of less accurate semi-classical MD simulations using empirical potential energy functions obtained from prior QM calculations (Birge and Hubbard, 1980; Birge et al., 1989; Tallent et al., 1992; Ben-Nun et al., 1998; Humphrey et al., 1998). An alternative treatment applied a hybrid quantum mechanical/molecular mechanical (QM/MM) calculation (Warshel, 1976; Warshel et al., 1991; Warshel and Chu, 2001) that combined an MM description of the protein with a semiempirical (Pariser-Parr-Pople) QM calculation of retinal. Although the latter approach has successfully described multi-electronic processes in situ, the calculation is limited in its account of the excited-state electronic wave function and potential surface, neglecting to describe many degrees of freedom properly.

In this article, we present ab initio QM/MM MD simulations of the complete isomerization process inside bR, including the excited-state dynamics of the chromophore, the electronic transition from the excited state to the ground state, and the subsequent relaxation of the ground state. Combining an ab initio description of the Schiff base half of retinal, the *N*-methyl- $\gamma$ -methylpenta-2,4-dieniminium cation fragment depicted in Fig. 1, at the complete active space self-consistent field (CASSCF) level of theory with an MM description of the remaining part of the retinal chromophore and of the protein, enabled us to simulate for the first time the multi-electronic state dynamics of photoisomerization in a complex protein environment without assuming empirical potential functions for the core degrees of freedom.

Unfortunately, the very demanding computation of the CASSCF method for the QM part forced us to employ the short chromophore analog. However, this analog includes the essential retinal segment involved in the isomerization in bR, namely the three terminal double bonds, including the Schiff base. But it should be noted that the approximation of the short analog introduces several artifacts of the potential energy profiles and the electronic properties as investigated by ab initio QM studies (Garavelli et al., 1998; González-Luque et al., 2000). First, the shorter  $\pi$  conjugation results in an overestimated excitation energy, leading to a steeper potential profile along the isomerizing bond. The altered

photoisomerization potential surface also may severely affect vibrational behavior, in particular vibrational couplings between the stretching and torsional modes. Second, the three conjugated double bond system biases the isomerization around the center bond, i.e.,  $C_{13}=C_{14}$ , which is actually the bond undergoing the isomerization in bR, whereas a longer chromophore analog in vacuo favors isomerization around either  $C_{13}=C_{14}$  or  $C_{11}=C_{12}$  (De Vico et al., 2002). Finally, the short polyene chain prevents delocalization of the positive charge in the  $\beta$ -ionone ring half in the excited state, altering the chromophore-protein interaction.

Nevertheless, the present novel in situ simulation and its comparison with in vacuo simulation identify the essential role of the protein environment on dynamics and catalysis of the photoisomerization process in bR. The protein environment significantly affects the photoisomerization kinetics and dynamics. The use of the short analog permitted us to calculate 11 trajectories, and to elucidate the characteristic incoherent dynamics of isomerization in bR. The study also shows that the in situ photoisomerization attains high isomerization bond selectivity, as experimentally observed. Furthermore, since ab initio excited-state MD simulations can naturally provide time evolution of the emission spectrum, we were able to directly examine the dynamic spectral modulation phenomena observed in femtosecond spectroscopy (Ye et al., 1999; Kobayashi et al., 2001). Despite the limitations due to the short chromophore model, ab initio description of the vibrational motions successfully reproduced major features of the dynamic spectra observed experimentally, proposing molecular origins underlying the spectral phenomena. Although the approximation of the short chromophore model used in our investigation has to be reassessed in future studies, the results obtained reveal already now surprising aspects of the multi-dimensional and multi-electronic state dynamics of retinal's in situ photoisomerization.

## METHODS

Photoisomerization trajectories were computed by ab initio QM/MM MD simulations. The QM and the MM regions were described at the ab initio molecular orbital and molecular mechanics levels of theory, respectively. The Schiff base half of the chromophore with three conjugated double bonds was chosen as the QM region (Fig. 1). Electronic structure, isomerization potential profile, and dynamics of a similar system of the Schiff base with three conjugated double bonds have been examined previously (Garavelli et al., 1997; Vreven et al., 1997). The rest of the chromophore, the protein, and water molecules were treated as the MM region. The electronic wave functions of the QM region in the ground ( $S_0$ ) and the first  $\pi$ - $\pi^*$  excited ( $S_1$ ) states were determined by the state-average CASSCF method (Werner and Knowles, 1985; Knowles and Werner, 1985). All six valence  $\pi$  and  $\pi^*$  orbitals with six electrons were included in the CAS. Dunning's double  $\zeta$ -basis functions (Dunning, 1971) were used. The MM region was described by the AMBER94 force field (Cornell et al., 1995) for the protein, by the TIP3P force field (Jorgensen et al., 1983) for water molecules, and through previously determined parameters for the MM part of the chromophore

(Tajkhorshid et al., 2000; Hayashi et al., 2002). A conventional QM/MM scheme using the link atom approach (Field et al., 1990) was employed. Open valences of covalent bonds at the boundaries dividing the QM and MM regions were filled by dummy hydrogen atoms. Electrostatic interaction of electrons in the QM region with the MM effective point charges were taken into account through one-electron integrals. A QM/MM interface combined with our own MD routine was incorporated into the quantum chemistry package, MOLPRO (a package of ab initio programs written by H.-J. Werner and P. J. Knowles, with contributions from J. Almlöf and co-workers). The trajectories were computed by solving the Newtonian equations of motion with energies and forces obtained from the QM/MM calculations. We employed the velocity-Verlet integrator (Swope et al., 1982) with a 0.5 fs time step. Since energy changes rapidly around the electronic transition points, the time step was switched to 0.25 fs whenever the energy difference between the  $S_0$  and  $S_1$  states dropped below 0.01 a.u. The QM/MM MD simulation of 100 fs required  $\sim 40$  h on a Compaq Alpha21264 500 MHz processor.

A model of bR was constructed based on high resolution x-ray crystal structures (Luecke et al., 1999; Belrhali et al., 1999), as described elsewhere (Hayashi et al., 2002). To generate an ensemble of initial coordinates and velocities to be used in photoisomerization trajectories, a classical MD simulation of the resting state of bR (BR<sub>570</sub>) at constant temperature (300 K) was carried out (Hayashi et al., 2002). A 12-Å residue-based cutoff was used for nonbonding interactions. The system was equilibrated for 500 ps with 1 fs time step while constraining bond distances that included hydrogen atoms, followed by a 50 ps equilibration with 0.5 fs time step without any bond constraints. The 11 initial configurations used for photoisomerization simulations were then sampled from the following 200 ps equilibrium MD simulation. The on-the-fly QM/MM MD trajectories on the  $S_1$  excited state were started from the initial configurations simulated in an NVE ensemble. The total energy of the system was well conserved and its standard deviation was 0.3 kcal/mol. Each excited-state trajectory was continued for 600 fs or until the trajectory achieved 75% product (see below), resulting in simulation times of 90–600 fs. Along the excited-state trajectories, semiclassical nonadiabatic coupling elements between the  $S_0$  and  $S_1$  states,  $V_{12}$ , were computed using

$$V_{12} = \sum_{\alpha} \left\langle \Psi_{S_1}^{\text{el}} \left| \frac{d}{dR_{\alpha}} \right| \Psi_{S_0}^{\text{el}} \right\rangle \times v_{\alpha}, \quad (1)$$

where  $\Psi_{S_1}^{\text{el}}$  and  $\Psi_{S_0}^{\text{el}}$  are adiabatic (Born-Oppenheimer) electronic wave functions, and  $R_{\alpha}$  and  $v_{\alpha}$  are nuclear coordinates and velocities, respectively. At electronic transition points between  $S_1$  and  $S_0$  determined by a semi-classical calculation (see below), the  $S_0$  state trajectories were started to obtain the products. The total number of  $S_0$  state trajectories spawned from the 11 trajectories of  $S_1$  was 65. Excess energies upon the transitions from  $S_1$  to  $S_0$  were distributed to nuclear velocities along the nonadiabatic coupling vectors (Herman, 1984).

To evaluate the transition probabilities from  $S_1$  to  $S_0$  at the transition points, the quantum mechanical probabilities of  $S_1$  and  $S_0$ ,  $c_{S_1}$  and  $c_{S_0}$ , respectively, were computed by solving a semiclassical time-dependent Hamiltonian,

$$i\hbar \begin{pmatrix} \dot{c}_{S_1}(t) \\ \dot{c}_{S_0}(t) \end{pmatrix} = \begin{pmatrix} E_{S_1}(t) & i\hbar V_{12}(t) \\ -i\hbar V_{12}(t) & E_{S_0}(t) - i\hbar\Gamma \end{pmatrix} \begin{pmatrix} c_{S_1}(t) \\ c_{S_0}(t) \end{pmatrix}. \quad (2)$$

$E_{S_1}(t)$  and  $E_{S_0}(t)$  are potential energies of  $S_1$  and  $S_0$  along the trajectory on the  $S_1$  state, respectively, and  $V_{12}(t)$  is the nonadiabatic coupling element given in Eq. 1. Energy levels of  $S_1$  and  $S_0$  along the  $S_0$  state trajectories separate rapidly after the electronic transitions (see below), indicating fast decoherence between the  $S_0$  and  $S_1$  nuclear wave functions and prompt decay in the  $S_0$  state. To incorporate the decay effect in Eq. 2, an absorbing term in the  $S_0$  state,  $\Gamma = 1/\tau_{S_0}$ , was introduced.  $\tau_{S_0}$  was set to be the decoherence time constant, 3.8 fs, estimated based on a Gaussian wave-

packet approximation at short time and high temperature limits (Schwartz et al., 1996). The differential equation (Eq. 2) was solved numerically by the fourth order Runge-Kutta method with a time step of  $1.0 \times 10^{-3}$  fs. For the Hamiltonian matrix elements, spline-fitted functions of time were used to interpolate values from the trajectory calculations. The calculated population of the  $S_1$  state exhibits stepwise decays to either 13-*cis* or all-*trans* products at the transition points. The branching ratio between the *cis* and *trans* products was then estimated by accumulating the stepwise transition probabilities.

We also performed ab initio MD simulation of photoisomerization of the retinal analog in vacuo to compare the dynamics in vacuo with that in bR (in situ). The retinal analog (Fig. 1) and the method used were the same as in the in situ simulation, except for absence of the QM/MM coupling. An ab initio MD study of the photoisomerization of a similar retinal analog with three conjugated double bonds (Vreven et al., 1997) has shown the time constant of photoisomerization in vacuo to be  $\sim 70$  fs. However, as described below, six trajectories in situ exhibit remarkably longer lifetimes of the excited state ( $>180$  fs). We therefore computed the six trajectories in vacuo starting with the same initial coordinates and velocities as in situ to clarify the molecular origin of the difference. The resultant timescale of the photoisomerization in vacuo was consistent with the previous ab initio QM studies (Garavelli et al., 1997; Vreven et al., 1997).

## RESULTS AND DISCUSSION

Fig. 2 *a* shows the time evolution of potential energies of the first excited state ( $S_1$ ) and the ground state ( $S_0$ ) along a typical photoisomerization trajectory. Before photoexcitation from the resting BR<sub>570</sub> state to the Frank-Condon (FC) region at  $t = 0$ , the all-*trans* planar chromophore assumes a slight twist around the  $C_{13}=C_{14}$  bond, as suggested in previous MD (Tajkhorshid et al., 2000) and ab initio QM/MM (Hay-

ashi and Ohmine, 2000) simulations. Immediately after the photoexcitation, the in-plane  $C_{13}=C_{14}$  bond relaxation takes place in 50 fs, as suggested experimentally (Song and El-Sayed, 1998; Kobayashi et al., 2001) and theoretically (Vreven et al., 1997; Garavelli et al., 1997, 1999; Warshel and Chu, 2001), but the chromophore retains its all-*trans* configuration until  $t = 130$  fs, when the torsion around  $C_{13}=C_{14}$  starts to evolve. Once the dihedral angle reaches a value of about  $90^\circ$ , the twisted conformation remains stable as long as the trajectory proceeds on the  $S_1$  state surface. The overall energy slope in the  $S_1$  state from the FC region toward the  $90^\circ$  twisted conformation is rather gentle, but the energy shows large fluctuations due to the chromophore's intramolecular vibrations and its interaction with the protein environment. On the other hand, energy of the (unpopulated)  $S_0$  state increases steeply as the  $C_{13}=C_{14}$  bond is twisted and approaches the  $S_1$  state energy near the  $90^\circ$  twist, where accordingly frequent electronic transitions occur. The  $\pi$  orbitals on carbon atoms  $C_{13}$  and  $C_{14}$  are almost perpendicular at these transition points, even though the carbon center  $C_{14}$  assumes a highly pyramidalized structure (see below), and the nonadiabatic coupling element between  $S_1$  and  $S_0$  exhibits large peaks (Weiss and Warshel, 1979; Molnar et al., 2002), as seen in Fig. 2 *a*. A semiclassical analysis of the electronic transition process (see Methods) indicates that a significant fraction of the  $S_1$  population decays to  $S_0$  at each crossing point. After the transition to  $S_0$ , the highly twisted conformation relaxes quickly to a planar form, either to a 13-*cis* product (the J<sub>630</sub> state) or back to the all-*trans* one (the BR<sub>570</sub> state).

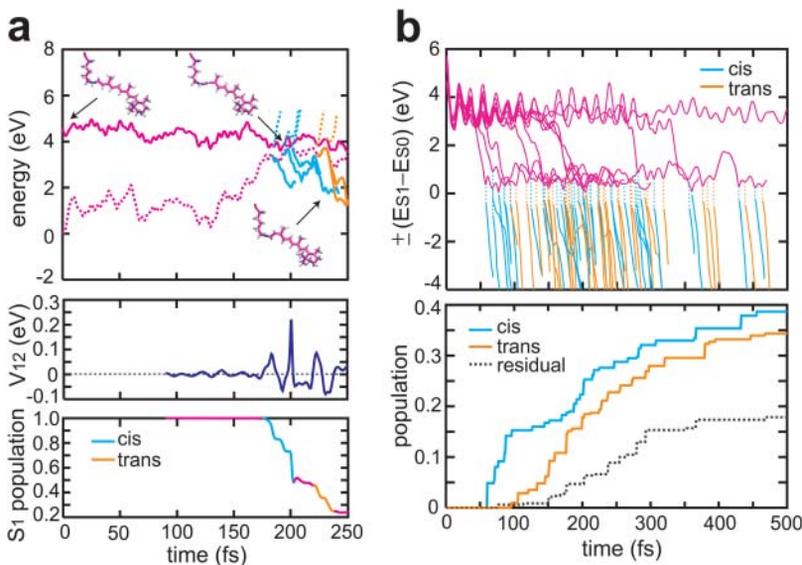


FIGURE 2 (a) Potential energies and populations of the  $S_1$  and  $S_0$  states along a typical photoisomerization trajectory. The top panel shows time evolution of the potential energies defined as the sum of the electronic energy of the QM region and interaction energy between the QM and the MM regions (see Fig. 1 and Methods). Solid and dashed purple lines indicate the potential energies of  $S_1$  and  $S_0$ , respectively, along the  $S_1$  state trajectory. The energies along the trajectories on the  $S_0$  state after electronic transitions from the  $S_1$  state are drawn by solid lines in blue and orange for the 13-*cis* (J<sub>630</sub>) and all-*trans* (BR<sub>570</sub>) branches, respectively; the dashed lines in blue and orange indicate the energies of  $S_1$  along the  $S_0$  trajectories. The chromophore structures shown in the top panel are snapshots at the FC region ( $t = 0$  fs), at a transition point ( $t = 190$  fs), and of a 13-*cis* product ( $t = 240$  fs). The middle panel shows the nonadiabatic coupling element between the  $S_1$  and the  $S_0$  states ( $V_{12}$ ) along the  $S_1$  state trajectory. Peaks of  $V_{12}$  indicate the transition points. The bottom panel displays the  $S_1$  population along the trajectory. The population decreases stepwise at the transition points, and five distinct transitions during the last 50 fs furnish 75% of

the overall transition.  $S_0$  trajectories obtained from the first three transitions result in 13-*cis* products with 52% yield, and the next two transitions lead to all-*trans* products with 23% yield. (b) Energy differences between  $S_1$  and  $S_0$  from 11 photoisomerization trajectories and the time evolution of population of the products. In the top panel, the energy differences along the  $S_1$  state trajectories (emission energies) are drawn by solid lines in purple. The negative energy differences along the  $S_0$  state trajectories (negatives of absorption energies) after the transitions are indicated by solid lines in blue and orange for the 13-*cis* and all-*trans* branches, respectively. The dashed lines connect the  $S_1$  and  $S_0$  states at the transition points. In the bottom panel, populations of the 13-*cis* and all-*trans* products resulting from the 11 trajectories are drawn in blue and orange, respectively. A dashed line in gray indicates the residual population of  $S_1$ , which did not yet decay to  $S_0$ .

An ensemble of 11 photoisomerization trajectories was simulated and is shown in Fig. 2 *b* in terms of the energy difference between  $S_1$  and  $S_0$ , i.e., emission or absorption energies. At  $t = 0$ , the  $S_0$  electronic state is excited to  $S_1$  with an average vertical excitation energy of 5.3 eV. The emission energies of all trajectories decrease to around 3.5 eV in 50 fs, and a well characterized transient state forms. This state corresponds closely to the fluorescent state  $I_{460}$  (Fig. 1), which exhibits a red-shifted emission. During the formation of  $I_{460}$ , the dominant conformational change is an in-plane bonding relaxation mainly at  $C_{13}=C_{14}$  (Song and El-Sayed, 1998; Kobayashi et al., 2001; Vreven et al., 1997; Garavelli et al., 1997, 1999; Warshel and Chu, 2001), but twist around  $C_{13}=C_{14}$  is not yet significant. After  $I_{460}$  formation, the  $S_1$  state trajectories undergo isomerization to the  $90^\circ$  twisted conformation, and the energy differences drop to around 0.5 eV.

Despite the presence of three isomerizable double bonds in the analog used in our calculations, isomerization proceeds in all trajectories only at the  $C_{13}=C_{14}$  bond, in agreement with experimental evidence (Birge, 1990). Moreover, the twist around this bond occurs in only one rotational sense, namely the one for which the Schiff base hydrogen approaches Asp-212. The observed unidirectional isomerization is consistent with the structures of the early intermediate  $K_{610}$  (Fig. 1) recently predicted theoretically (Hayashi et al., 2002) and determined by x-ray crystallography (Schobert et al., 2002; Matsui et al., 2002), where the Schiff base N-H bond points indeed toward Asp-212.

The bond selectivity and unidirectionality of the isomerization in bR are in striking contrast with those of the isolated chromophore (in vacuo). Fig. 3 shows the time evolution of emission energies along in vacuo trajectories and snapshots of the retinal analog with  $\sim 90^\circ$  twisted dihedral angles around the isomerizing bonds. Repeating the simulations for six trajectories in the absence of the protein environment, but using the same initial coordinates and velocities as in the in situ simulations, leads to two isomerizations around the  $C_{15}=N_\zeta$  bond in either rotational sense. The results clearly demonstrate the role of the protein environment in the high bond selectivity and directionality of the photoisomerization in bR. At room temperature the retinal chromophore undergoes large thermal fluctuations so that the initial conformation and motion at the moment of photoexcitation can lead to isomerization around the  $C_{15}=N_\zeta$  bond, and to a twist in any direction. The protein environment blocks alternative paths and funnels the trajectory to the isomerization that is coupled to the biological function of the system, i.e., to the  $C_{13}=C_{14}$  isomerization in the case of bR.

The present simulation reveals clearly a significant blockage of inhomogeneous paths by the protein, but one must note that the funneling into the isomerization around  $C_{13}=C_{14}$  may be enhanced also due to the short chromophore model employed (see Introduction). Indeed, earlier ab initio studies of the in vacuo photoisomerization around

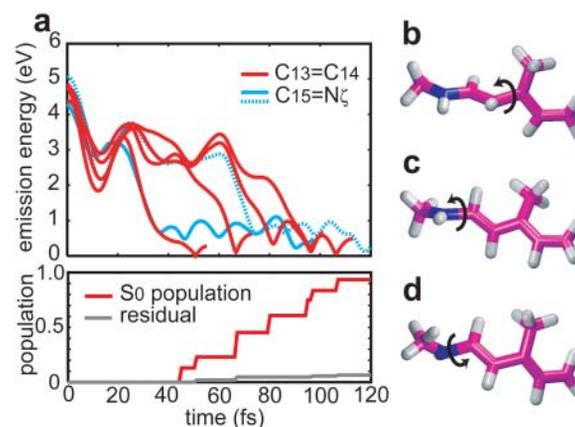


FIGURE 3 (a) Time evolution of emission energies along six trajectories in vacuo (*top*), and of the product formation (*bottom*). Solid lines in red and blue in the top panel indicate emission energies along isomerizations around  $C_{13}=C_{14}$  and  $C_{15}=N_\zeta$ , respectively. A blue dashed line shows the emission energy along an isomerization around  $C_{15}=N_\zeta$  in the direction opposite to that seen in the other trajectories. The product formation drawn by a red line in the bottom panel was computed with the semiclassical method (see Methods) for four trajectories that resulted in the isomerization around  $C_{13}=C_{14}$ . A line in gray indicates the residual population of  $S_1$  that did not yet decay to  $S_0$ . (b–d) Snapshots of molecular structures of the retinal analog with  $\sim 90^\circ$  twisted dihedral angles around the isomerizing bonds. (b) Isomerization around  $C_{13}=C_{14}$ , (c) around  $C_{15}=N_\zeta$ , and (d) around  $C_{15}=N_\zeta$  in the opposite direction.

either  $C_{13}=C_{14}$  or  $C_{11}=C_{12}$  of a longer chromophore model of retinal consisting of five conjugated double bonds exhibited almost barrierless energy profiles, i.e., activation energies of only 0.5 and 0.2 kcal/mol, respectively (De Vico et al., 2002). Taking into account the fact that the retinal chromophore possesses a polyene chain with six conjugated double bonds, the photoisomerization around  $C_{11}=C_{12}$  may be the favored one as experimentally observed in solution. The present simulation using the shorter chromophore model cannot describe the path around  $C_{11}=C_{12}$  properly due to lack of conjugation in the  $\beta$ -ionone ring half of retinal. To definitively elucidate the molecular origin of the selectivity, the QM/MM MD methodology developed in the study needs to be applied to a longer chromophore model.

On the other side, a previous MD simulation (Tajkhorshid et al., 2000) and an ab initio QM/MM calculation using the full retinal chromophore model as the QM region (Hayashi and Ohmine, 2000) have shown that the chromophore in resting bR assumes a significant twist in the  $C_{13}=C_{14}$  bond toward the Asp-212 direction, which is indicative of protein forces favoring isomerization around  $C_{13}=C_{14}$ . As suggested in a previous study (Hayashi et al., 2002), the unidirectionality of the isomerization is also due to the protein forces. The chromophore undergoes the unidirectional torsion around  $C_{13}=C_{14}$  to accommodate itself in the binding pocket. Key interactions of the chromophore with Trp-86 and a proximate water molecule that induce this torsion have been identified (Tajkhorshid et al., 2000; Hayashi and Ohmine, 2000).

It is noteworthy that the lifetime of  $I_{460}$  ranges from 50 fs to more than 500 fs for the in situ trajectories; the isomerization in some trajectories takes place immediately after formation of  $I_{460}$ , whereas some trajectories stay much longer in this state. The wide variation of the lifetime of  $I_{460}$  in bR is another feature that is absent from the faster and more coherent isomerization dynamics of the retinal Schiff base in vacuo; the isomerization in vacuo completes within 50–100 fs as shown by Vreven et al. (1997) and confirmed by our simulations as shown in Fig. 3. In bR, some trajectories stay and wander in the  $I_{460}$  state for a significant amount of time to find an escape path toward the 13-*cis* isomer. The escape from  $I_{460}$  requires matching of the rotational motion around the  $C_{13}=C_{14}$  bond with protein thermal motions as well as other torsions of retinal to minimize the chromophore's conformational change necessary for isomerization in a narrow binding pocket. Key protein motions coupled to the isomerization involve Lys-216, which binds the chromophore. The incoherent behavior of the escape from  $I_{460}$  toward the product formation in bR can explain femtosecond spectroscopic observations, such as silent vibrational bands of pump-probe signals (Kobayashi et al., 2001) and the statistical characteristics of stimulated emission decay (Ruhman et al., 2002) after the formation of  $I_{460}$ .

Fig. 2 *b* also shows the time evolution of the populations of 13-*cis* and all-*trans* products. The products form with a time constant of 200 fs, which is in good agreement with the experimental values of 240–500 fs (Gai et al., 1998); the deviation is most likely due to a narrow plateau of the potential energy curve along the  $C_{13}=C_{14}$  dihedral angle around the FC region and the steeper potential profile toward the 90° twisted region that arise in the short retinal analog used in the present simulations, but not in the complete chromophore (Garavelli et al., 1998; González-Luque et al., 2000). The 13-*cis* product formation is dominant over return to all-*trans* retinal, as a result of the first transitions that favor exclusively the *cis* form (see below). The *cis-trans* branching ratio calculated from the 11 trajectories is 0.54:0.46, as seen in Fig. 2 *b*, which agrees well with the experimental value, 0.6:0.4 (Tittor and Oesterhelt, 1990). The slightly underestimated quantum yield of the *cis* product is likely due to the altered potential profile caused by the use of the short chromophore model and lack of dynamic electronic correlation in the present treatment; there also remains ambiguity in the evaluation of the branching ratio due to poor statistical sampling in the long time region, a limit imposed by the extreme demand on computation time.

At a 90° twist of the  $C_{13}=C_{14}$  bond, each trajectory encounters many transition points (between 2 and 10) until achieving 75% product formation. After the electronic transition, the energy levels of  $S_0$  and  $S_1$ , calculated along the  $S_0$  state trajectory, separate rapidly, and recrossing to  $S_1$  does not occur. The first transition in each trajectory furnishes exclusively the *cis* product, because the transition

process involves the large momentum of the rotation around  $C_{13}=C_{14}$  from the *trans* conformation toward the *cis* one created by energy funneling from  $I_{460}$  to the transition region. In the calculated trajectories, the first transition event contributes on the average 17% *cis* product, i.e., nearly one-third of all successful isomerization events. Therefore the exclusive transition to the *cis* conformation at the first transition point enhances the quantum yield of the *cis* product, as already proposed by Warshel and co-workers (Weiss and Warshel, 1979; Warshel and Chu, 2001). On the other hand, branching to the *cis* and *trans* products at transition points after the first one seems to be rather random, and a regular pattern such as alternate transitions to *cis* and *trans* as suggested by Warshel and co-workers (Weiss and Warshel, 1979; Warshel and Chu, 2001) was only observed in four trajectories. This random behavior is due to chaotic motions in the transition region originating from highly excited hydrogen-out-of-plane (HOOP) and bending vibrations around the  $C_{14}$  atom, as discussed below.

The electronic transition events were found to be strongly perturbed by the electrostatic field in the protein. The multiple transition in bR is different from transition in vacuo; in the latter case, most of the product is yielded through only one or two transition events (Vreven et al., 1997) (Fig. 3). The relatively small transition probability at each transition point in bR originates from strong interaction of the protonated Schiff base with the counterion groups (Asp-85 and Asp-212) and a proximate water molecule. Fig. 4 displays key molecular orbitals at a transition point, indicating the so-called sudden polarization (Salem, 1979; Bonačić-Koutecky et al., 1984) of the electronic structure, i.e., a complete localization of the positive charge in the  $C_{14}\sim N_\xi$  and  $C_{11}\sim C_{13}$  halves in  $S_0$  and  $S_1$ , respectively. Hence, the counterion and water molecules located in the vicinity of the Schiff base stabilize the  $S_0$  electronic state more than  $S_1$ , which leads to a shift of the energy surface crossing seam and a larger separation of the energy levels at the transition point, and decreases the transition probability.

The inefficiency of the transition is, however, partially compensated by electronic transition events along molecular motions other than the rotation around the isomerizing bond. Because of the ionic electronic character of  $S_1$ , the molecule at some transition points is strongly deformed into highly pyramidalized (Salem, 1979) or straightened (Ohmine, 1985) structures at the  $C_{14}$  atom, as shown in Fig. 4. Such deformed structures, which have also been found in photoisomerization of ethylene and formalimine in vacuo simulated by ab initio MD methods (Ben-Nun and Martínez, 1998; Doltsinis and Marx, 2002), are unstable in the  $S_0$  state, which possesses covalent electronic character. It is interesting to note that the retinal  $C_{14}$  pyramidalization in bR had been postulated already long ago (Schulten, 1978). Thus, the structural deformations lift the energy level of  $S_0$  closer to  $S_1$  and enhance the transition probability. In fact, a recent ab initio study searching intersection space between the

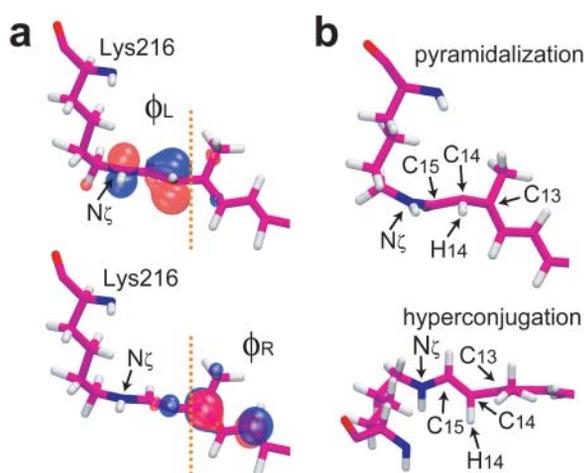


FIGURE 4 (a) Two  $\pi$  orbitals ( $\phi_L$  and  $\phi_R$ ) characterizing the  $S_0$  and  $S_1$  electronic structures. In the  $S_0$  state, each of the orbitals is occupied by one electron, whereas in the  $S_1$  state,  $\phi_L$  is doubly occupied and  $\phi_R$  is vacant. Orange dotted lines indicate the isomerizing  $C_{13}=C_{14}$  bond. The localized nature of the  $\pi$  orbitals gives rise to localization of the chromophore's positive charge in the  $C_{14}\sim N_\zeta$  ( $\phi_L$ ) and  $C_{11}\sim C_{13}$  ( $\phi_R$ ) halves for  $S_0$  and  $S_1$ , respectively (sudden polarization, see text). (b) Typical conformations of the chromophore for efficient transitions at the electronic transition points in situ. The top and bottom panels show views from side and top, respectively. In the top panel, a high degree of pyramidalization at  $C_{14}$  can be seen, and the wagging angle of  $C_{14}-H_{14}$  ( $124^\circ$ ) largely deviates from planarity ( $180^\circ$ ). The doubly occupied  $\phi_L$  orbital at  $C_{14}$  exhibits an  $sp^3$ -type hybridization (Salem, 1979) consistent with the pyramidalization frequently found at crossing points. The bottom panel reveals a straightening at  $C_{14}$ . The bond angles of  $C_{13}=C_{14}-C_{15}$  ( $135^\circ$ ) and  $C_{13}=C_{14}-H_{14}$  ( $98^\circ$ ) largely deviate from the  $sp^2$  angle ( $\sim 120^\circ$ ). This is due to a strong hyperconjugation interaction of the vacant  $\phi_R$  orbital with the  $C_{14}-H_{14}$   $\sigma$  orbital providing an  $sp$ -type mixing (Ohmine, 1985).

excited- and ground-state energy surfaces of a short chromophore model consisting of three double bonds in vacuo (Migani et al., 2003) has revealed such a pyramidalized deformation in the intersection space apart from the minimum conical intersection point. The pyramidalization of  $C_{14}$  originates from the lone pair nature of the doubly occupied molecular orbital completely localized in the  $N_\zeta\sim C_{14}$  moiety as shown in Fig. 4. Although the pyramidalization is therefore not directly affected by the truncation of the  $\beta$ -ionone ring half in the short chromophore model, a reduced delocalization of the positive charge in the  $\beta$ -ionone ring half may slightly attract more electron density from  $N_\zeta$  to  $C_{14}$ , leading to weak enhancement of the pyramidalization. After the transition, wagging and in-plane rocking motions of  $H_{14}$  are highly excited in the  $S_0$  state due to the strongly deformed structures at the transition points.

The very demanding computation for this study did not allow us to simulate the complete relaxation of bR to the  $K_{610}$  intermediate after retinal's isomerization. However, the observed photoisomerization dynamics corresponds closely to that obtained in a previous study using classical MD simulations (Hayashi et al., 2002), which examined the formation of  $K_{610}$ . Hence, we expect that the photoisome-

rization event in the present simulation leads also to the same  $K_{610}$  intermediate. In this regard it is noteworthy that Hayashi et al. (2002) included an ab initio QM/MM analysis of  $K_{610}$  for the complete retinal chromophore and several groups in the vicinity of the Schiff base, which reproduced the experimentally measured (Birge and Cooper, 1983; Birge et al., 1989) energy storage of 16 kcal/mol in  $K_{610}$ , and identified the molecular mechanisms of energy storage, namely conformational distortion of retinal and weakened interaction between the Schiff base and the surrounding polar groups. Similar mechanisms were proposed by other researchers (Birge and Cooper, 1983) to be responsible for storing energy after retinal's isomerization in bR.

A stimulated emission pumping measurement by Ruhman et al. (2002) has proved that  $I_{460}$  is an excited-state intermediate of the photocycle that is arrested in a specific configuration and decays kinetically to a next photocycle intermediate. The molecular structure and the kinetic decay of  $I_{460}$  are discussed above. On the other hand, a key observation of a dynamic spectral modulation in the stimulated emission signal by Ye et al. (1999) detected dynamics within the  $I_{460}$  state. Fig. 5 displays the time evolution of emission energy in a trajectory with a long-lived  $I_{460}$  state. The spectrum exhibits high frequency oscillations, superimposed on a slow dynamic modulation with a 200 fs period due to molecular vibrations. Indeed, the slow 200 fs component reproduces well a prominent feature in the femtosecond-stimulated emission spectrum observed by Ye et al. (1999). This slow oscillation is assigned to torsional rocking around  $C_{13}=C_{14}$ , during which the twist develops up to  $40^\circ$  away from planarity, but a mismatch with neighboring torsion and protein motion prevents the trajectory from proceeding to the  $90^\circ$  twisted conformation.

It is noted that a similar oscillatory behavior has also been observed in stimulated emission signals of the retinal chromophore in solution (Hou et al., 2001) and even of a  $C_{13}=C_{14}$  locked analog of the chromophore in bR (Ye et al., 1999) and in solution (Hou et al., 2001). Since the locked analog is supposed to prevent torsional motion around the  $C_{13}=C_{14}$  bond, the observed oscillations are unlikely due to the same molecular motion seen in the present study. However, the close similarity of the oscillatory behavior and a slight difference of the frequencies between the locked and unlocked chromophores suggest that a similar torsional rocking around a different double bond, e.g.,  $C_{11}=C_{12}$ , is responsible for the oscillation in the locked chromophore. As mentioned above, although the  $C_{11}=C_{12}$  bond is also capable of undergoing isomerization in vacuo (De Vico et al., 2002), isomerization around this bond is inhibited by the protein environment; the  $C_{13}=C_{14}$  locked chromophore apparently reaches a situation resembling the long-lived  $I_{460}$  state of the native chromophore. Similarly, the chromophore in solution is tightly packed by solvent molecules and its isomerization is most likely disturbed by the confinement as indicated also by the fact that the

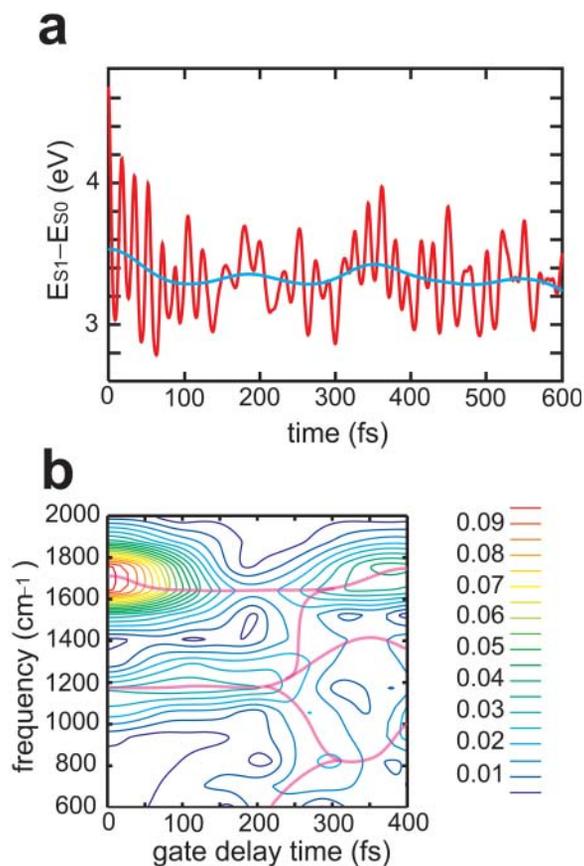


FIGURE 5 (a) Emission energy along a trajectory with a long-lived  $I_{460}$  that does not undergo isomerization in 600 fs. The emission energy (red line) exhibits oscillations due to molecular vibrations. The blue line indicates a slow oscillatory component with a 200 fs period obtained by smoothing with a time window function. A Hanning function with a halfwidth at half-maximum of 64 fs was used for the time window. (b) Spectrogram showing the time evolution of characteristic frequencies of the oscillations of the emission energy in (a). The spectrogram was calculated by Fourier transformation of the emission energy gated by the Hanning time window (Kobayashi et al., 2001). The time evolution of the frequencies was computed along the window gate delay time. The frequencies are uniformly scaled by a factor 0.9 (Garavelli et al., 1999). Purple lines indicate traces of the frequency peaks.

isomerization in solution (Logunov et al., 1996, Hou et al., 2001) is significantly slower than in bR.

Kobayashi et al. (2001) have analyzed the fast oscillatory component of their 5 fs pump-probe absorption spectrum through a spectrogram and revealed remarkable dynamic frequency shifts and vibrational mode couplings. Fig. 5 shows the respective spectrogram calculated here for the long-lived  $I_{460}$  trajectory. A strong peak at  $1700\text{ cm}^{-1}$  is due to the  $C_{13}=C_{14}$  and  $C_{15}=N_{\zeta}$  stretching vibrations excited during the formation of  $I_{460}$ . The signal decays quickly, and the peak frequency shifts downward in 100 fs, in good agreement with observation (Kobayashi et al., 2001). The downward shift is attributed to a decrease of the bond order of  $C_{13}=C_{14}$  due to its torsion. After 300 fs, the peak frequency up-shifts, accompanied by an increase of the

intensity, which is due to a less twisted conformation on this time domain. A broad peak in the  $1000\text{--}1250\text{ cm}^{-1}$  region includes the HOOP and the in-plane bending of  $H_{14}$  in its low- and high-frequency regions, respectively. Although the peak frequency is almost constant in the first 200 fs, the HOOP component exhibits an upward shift as indicated by a change of the spectral shape. This upward shift originates from a slightly pyramidalized structure at the  $C_{14}$  atom due to the increase of the ionic character of the electronic structure induced by the torsion around  $C_{13}=C_{14}$ . After 200 fs, the peak bifurcates to the corresponding HOOP and in-plane modes, as also clearly seen in the experimental spectrogram (Kobayashi et al., 2001). The pyramidalized structure formed in the first 200 fs excites the distinct HOOP motion upon the recovery of the planar conformation.

## CONCLUDING REMARKS

The novel computational technique based on the ab initio description of in situ isomerization dynamics of retinal unveiled in complete molecular detail the anomalously fast event in the primary process of bR's proton pump function. The simulation reproduced the characteristic kinetics and high selectivity of the photoisomerization in bR observed in experiments, i.e., reproduced that the excited-state process starts with in-plane bond relaxation corresponding to the formation of  $I_{460}$ , and then undergoes isomerization selectively around the  $C_{13}=C_{14}$  bond in an incoherent manner. Comparison of the isomerization dynamics in situ with that in vacuo proves that the high selectivity in bR is due to the protein environment that blocks inhomogeneous reaction paths and guides retinal to the 13-*cis* isomerization product. The incoherent dynamics originates from trajectories that stay in the  $I_{460}$  state for periods of several hundred femtoseconds until they find an escape pathway to the 13-*cis* product. The simulations also revealed the essential multi-dimensional molecular motions that enhance the electronic transition from the excited state to the ground one and permitted us to identify the molecular origin of the dynamic spectral modulation phenomena observed in femtosecond spectroscopy. Likewise, simulations based on the available x-ray crystallographic structure of rhodopsin (Palczewski et al., 2000) can describe in complete detail the primary process in vision, which in contrast to bR exhibits apparently highly coherent isomerization dynamics (Wang et al., 1994). Without doubt, simulations similar to the present ones, but incorporating a larger segment of retinal in the ab initio quantum chemical treatment, will be carried out sooner or later, but it is expected that these simulations will alter the detailed picture of the bR photoreaction only in respect to quantitative features such as the photoisomerization timescale.

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