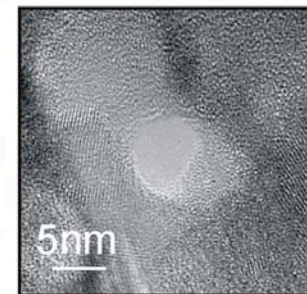
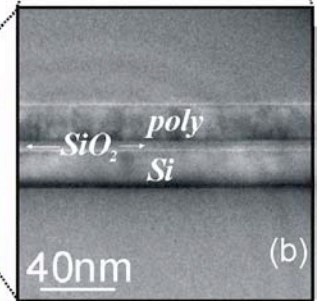
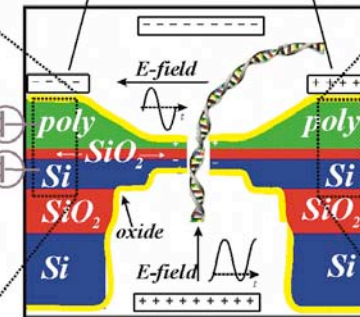
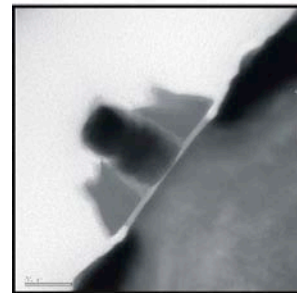
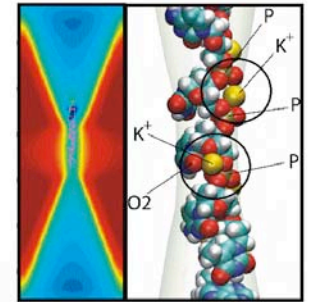
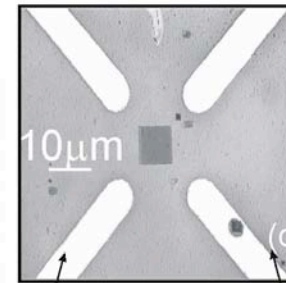


Introduction to:

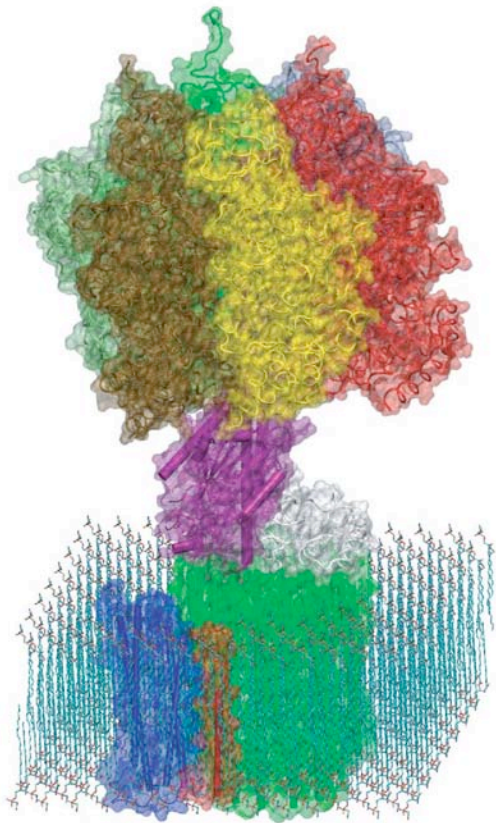
NAnoscale **M**olecular **D**ynamics

The Molecular Dynamics Method

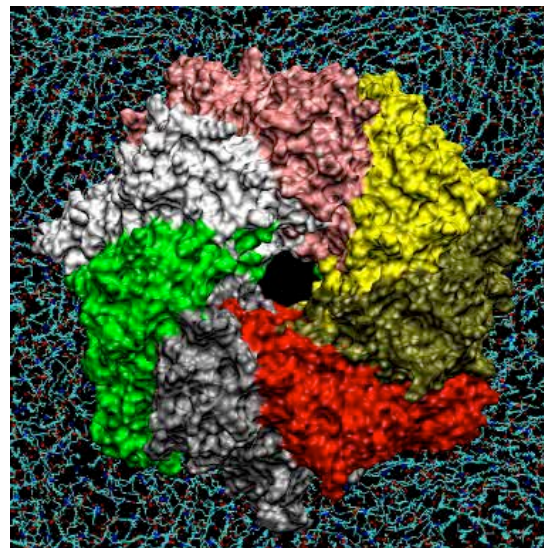
Integrated chip for DNA sequencing



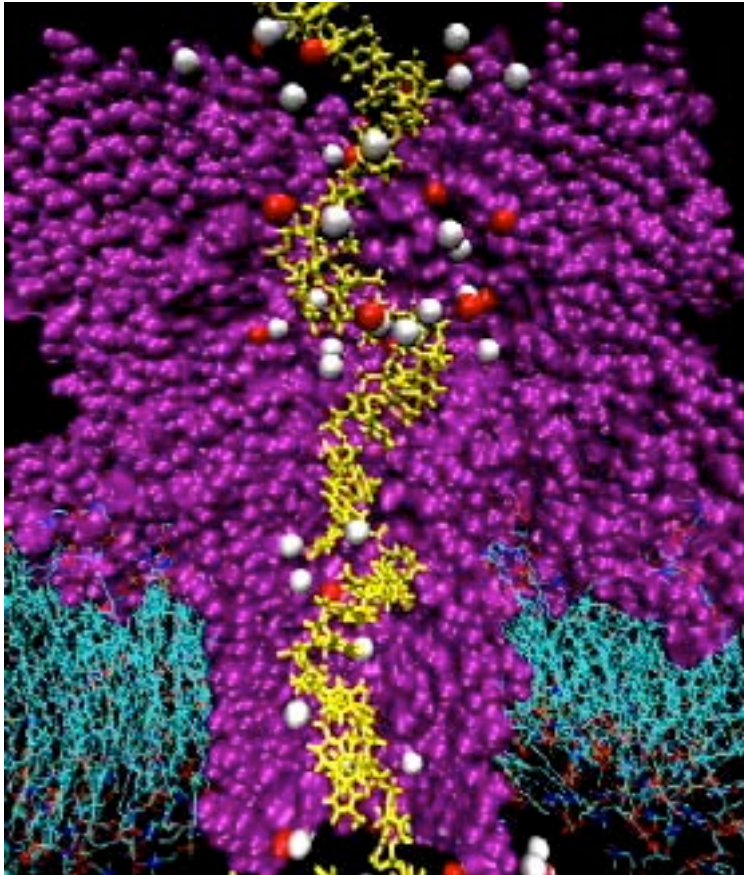
Bacterial toxin
alpha-hemolysin



ATPase, a molecular motor that synthesizes the body's weight of ATP a day



Molecular Dynamics Example



5 ns, 360,000 atoms
(water not shown)

MD simulation of ionic current
through membrane channel
 α -hemolysin blocked by DNA

What do you see?

Simulation conditions:

1.2 V electrostatic potential

NVT ensemble

298 K

PME 128x128x128

Classical Dynamics

F=ma at 300K

Energy function: $U(\vec{r}_1, \vec{r}_2, \dots, \vec{r}_N) = U(\vec{R})$

used to determine the force on each atom:

$$m_i \frac{d^2 \vec{r}_i}{dt^2} = \vec{F}_i = -\vec{\nabla} U(\vec{R})$$

yields a set of $3N$ coupled 2nd-order differential equations that can be propagated forward (or backward) in time.

Initial coordinates obtained from crystal structure, velocities taken at random from Boltzmann distribution.

Maintain appropriate temperature by adjusting velocities.

Langevin Dynamics

Langevin dynamics deals with each atom separately, balancing a small friction term with Gaussian noise to control temperature:

$$m \ddot{\vec{r}} = \vec{F}(\vec{r}) - \gamma m \dot{\vec{r}} + \vec{R}(t)$$

$$\langle \vec{R}(t) \cdot \vec{R}(t') \rangle = 6k_B T \gamma \delta(t - t')$$

Classical Dynamics

discretization in time for computing

$$m_i \frac{d^2 \vec{r}_i}{dt^2} = \vec{F}_i = -\vec{\nabla} U(\vec{R})$$

Use positions and accelerations at time t and the positions from time $t-\delta t$ to calculate new positions at time $t+\delta t$.

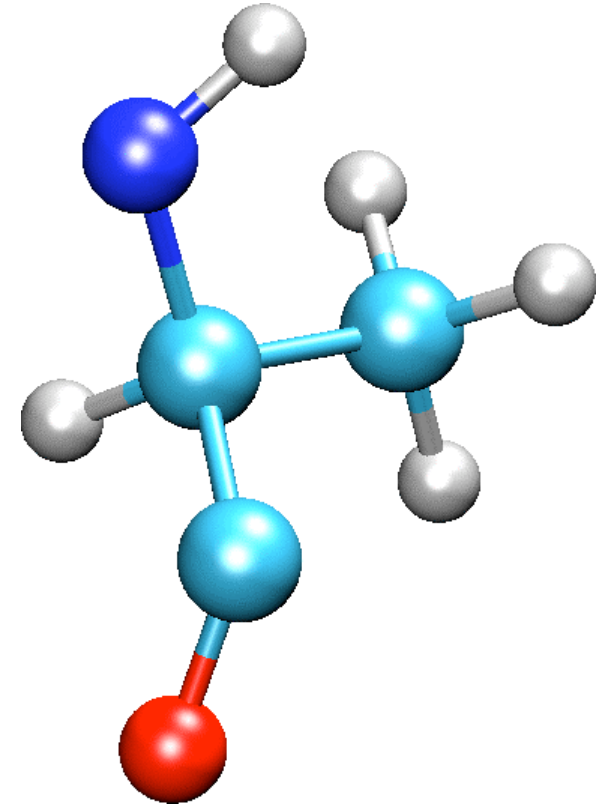
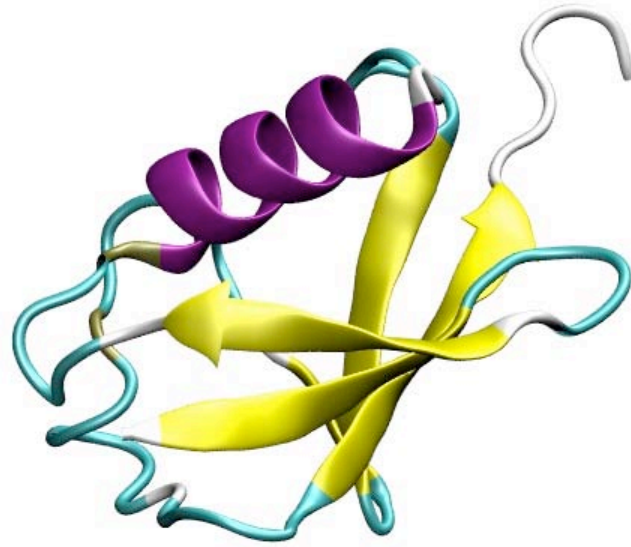
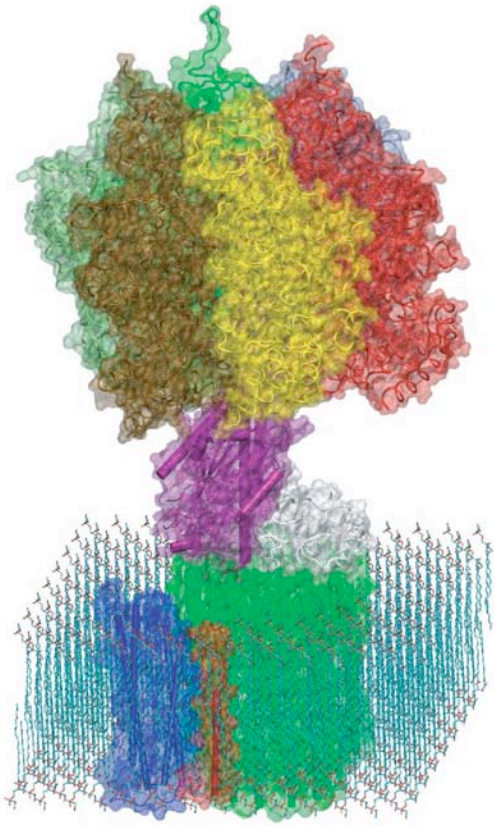
$$\begin{aligned} \mathbf{r}(t + \delta t) &\approx \mathbf{r}(t) + \mathbf{v}(t)\delta t + \frac{1}{2}\mathbf{a}(t)\delta t^2 \\ \mathbf{r}(t - \delta t) &\approx \mathbf{r}(t) - \mathbf{v}(t)\delta t + \frac{1}{2}\mathbf{a}(t)\delta t^2 \end{aligned} \quad +$$



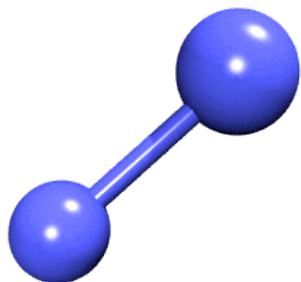
$$\mathbf{r}(t + \delta t) \approx 2\mathbf{r}(t) - \mathbf{r}(t - \delta t) + \mathbf{a}(t)\delta t^2$$

$$-\vec{\nabla} U(\vec{R}) / m_i$$

Protein Structure

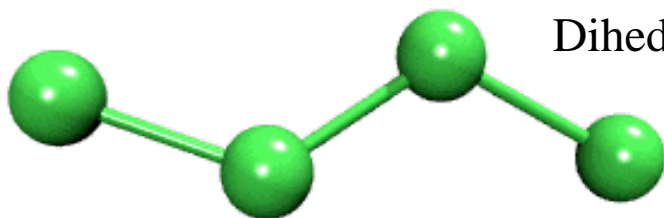
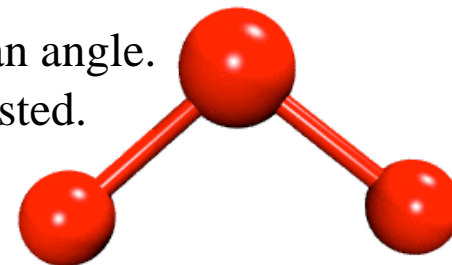


Molecular Structure (bonds, angles, etc.)



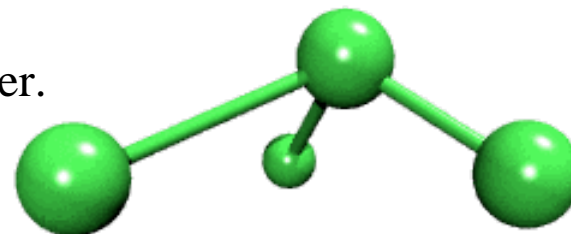
Bonds: Every pair of covalently bonded atoms is listed.

Angles: Two bonds that share a common atom form an angle.
Every such set of three atoms in the molecule is listed.



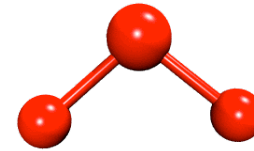
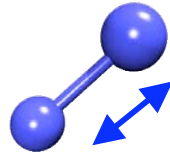
Dihedrals: Two angles that share a common bond form a dihedral.
Every such set of four atoms in the molecule is listed.

Improper: Any *planar* group of four atoms forms an improper.
Every such set of four atoms in the molecule is listed.

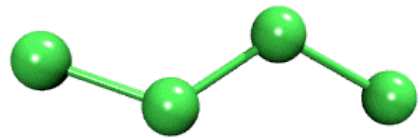


Potential Energy Function of Biopolymer

- Simple, fixed algebraic form for every type of interaction.
- Variable parameters depend on types of atoms involved.



$$U(\vec{R}) = \underbrace{\sum_{bonds} k_i^{bond} (r_i - r_0)^2}_{U_{bond}} + \underbrace{\sum_{angles} k_i^{angle} (\theta_i - \theta_0)^2}_{U_{angle}} +$$

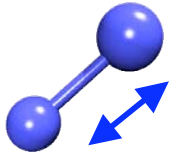
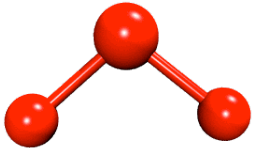
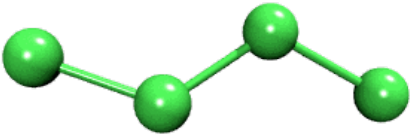


$$\underbrace{\sum_{dihedrals} k_i^{dihe} [1 + \cos(n_i \phi_i + \delta_i)]}_{U_{dihedral}} +$$

$$\underbrace{\sum_i \sum_{j \neq i} 4 \epsilon_{ij} \left[\left(\frac{\sigma_{ij}}{r_{ij}} \right)^{12} - \left(\frac{\sigma_{ij}}{r_{ij}} \right)^6 \right] + \sum_i \sum_{j \neq i} \frac{q_i q_j}{\epsilon r_{ij}}}_{U_{nonbond}}$$

Potential Energy Function of Biopolymer

- Simple, fixed algebraic form for every type of interaction.
- Variable parameters depend on types of atoms involved.

heuristic

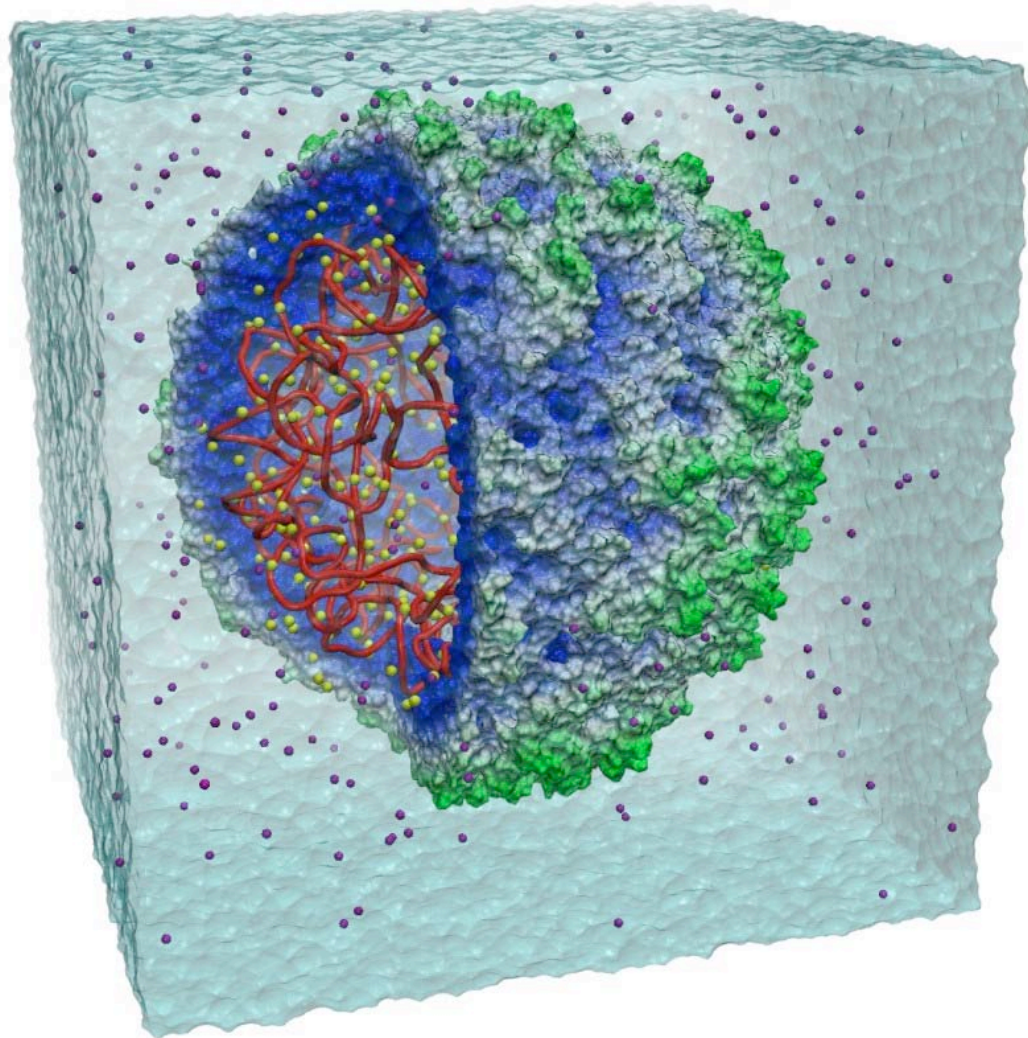
$$U(\vec{R}) = \underbrace{\sum_{\text{bonds}} k_i^{\text{bond}} (r_i - r_0)^2}_{U_{\text{bond}}} + \underbrace{\sum_{\text{angles}} k_i^{\text{angle}} (\theta_i - \theta_0)^2}_{U_{\text{angle}}} +$$

$$\underbrace{\sum_{\text{dihedrals}} k_i^{\text{dihe}} [1 + \cos(n_i \phi_i + \delta_i)]}_{U_{\text{dihedral}}} +$$

from physics

$$\underbrace{\sum_i \sum_{j \neq i} 4 \epsilon_{ij} \left[\left(\frac{\sigma_{ij}}{r_{ij}} \right)^{12} - \left(\frac{\sigma_{ij}}{r_{ij}} \right)^6 \right] + \sum_i \sum_{j \neq i} \frac{q_i q_j}{\epsilon r_{ij}}}_{U_{\text{nonbond}}}$$

Large is no problem. But ...



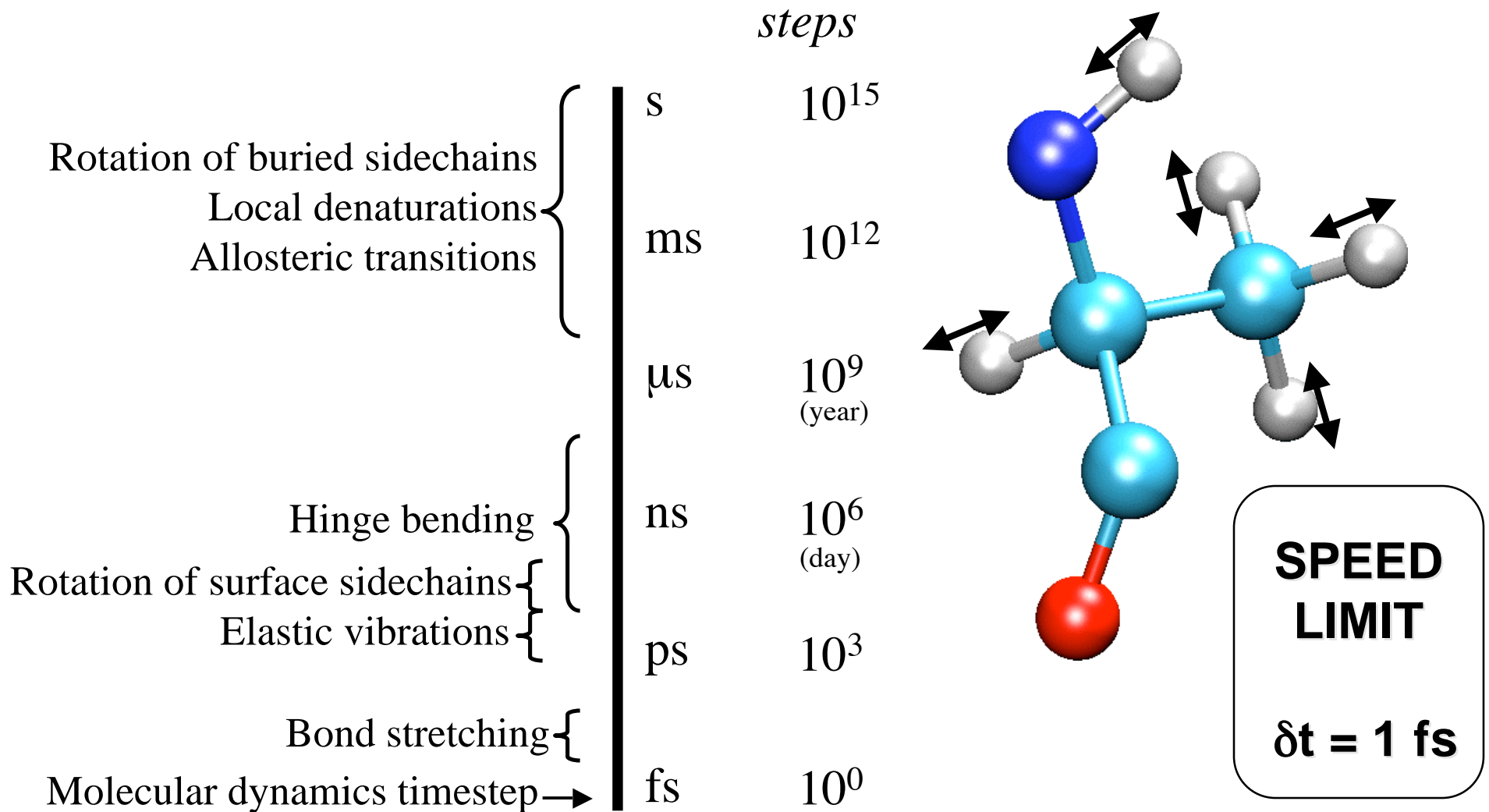
Molecular dynamics
simulation of satellite
tobacco mosaic virus
with over 1,000,000
atoms



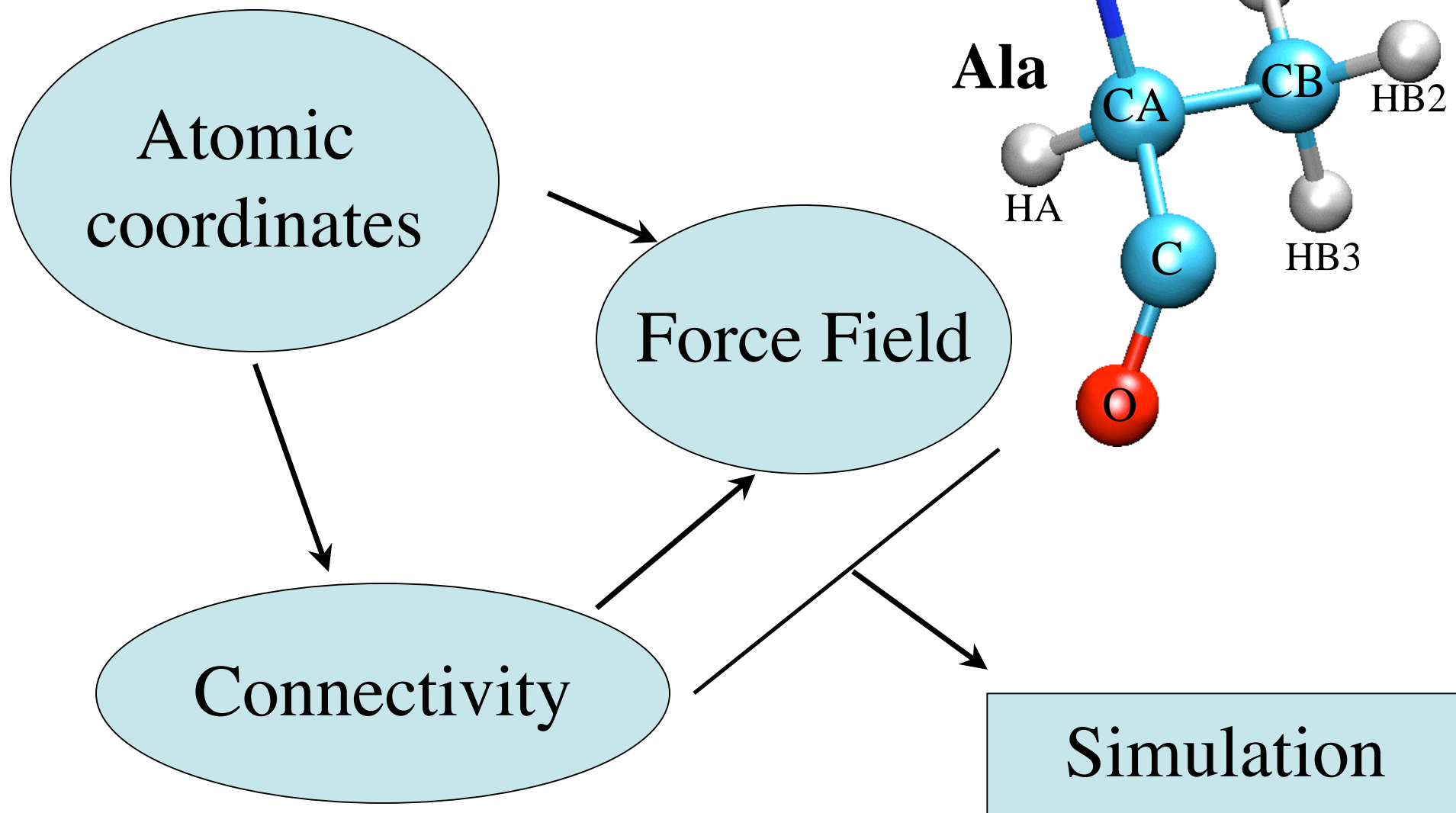
Massive parallel computer
(PSC Lemieux)

But long is!

biomolecular timescale and timestep limits



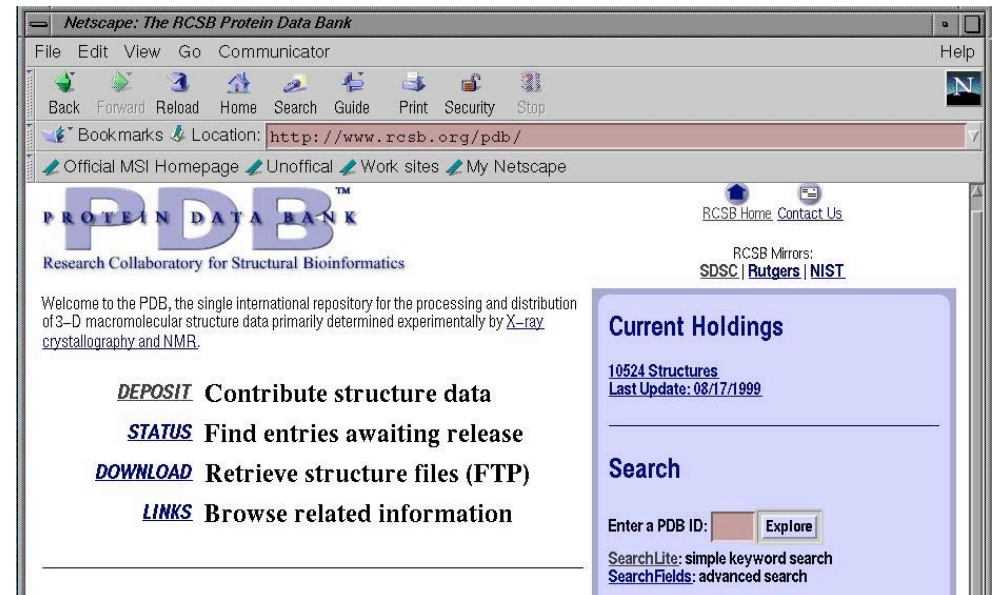
MD simulation basics



PDB Files

a little information

- Simulations start with a crystal structure from the Protein Data Bank, in the standard PDB file format.
- PDB files contain standard records for species, tissue, authorship, citations, sequence, secondary structure, etc.
- We only care about the atom records...
 - atom name (N, C, CA)
 - residue name (ALA, HIS)
 - residue id (integer)
 - coordinates (x, y, z)
 - occupancy (0.0 to 1.0)
 - temp. factor (a.k.a. beta)
 - segment id (6PTI)
- No hydrogen atoms!
(We must add them ourselves.)



PDB File

(available from www.rcsb.org if structure of biopolymer solved)

REMARK FILENAME="bpti19.pdb"

REMARK PROTEINASE INHIBITOR (TRYPSIN) 13-MAY-87 6PTI

REMARK BOVINE PANCREATIC TRYPSIN INHIBITOR

REMARK BOVINE (BOS TAURUS) PANCREAS

REMARK A.WLODAWER

REMARK DATE:26-Jun-00 21:34:42 created by user:

ATOM	1	HT1	ARG	1	13.150	-7.331	10.849	1.00	0.00	BPTI
------	---	-----	-----	---	--------	--------	--------	------	------	------

ATOM	2	HT2	ARG	1	11.747	-7.115	11.780	1.00	0.00	BPTI
------	---	-----	-----	---	--------	--------	--------	------	------	------

etc etc etc

ATOM	554	CA	GLY	56	15.319	0.828	11.790	1.00	17.33	BPTI
------	-----	----	-----	----	--------	-------	--------	------	-------	------

ATOM	555	C	GLY	56	16.029	-0.385	12.375	1.00	18.91	BPTI
------	-----	---	-----	----	--------	--------	--------	------	-------	------

ATOM	556	OT1	GLY	56	15.443	-1.332	12.929	1.00	21.00	BPTI
------	-----	-----	-----	----	--------	--------	--------	------	-------	------

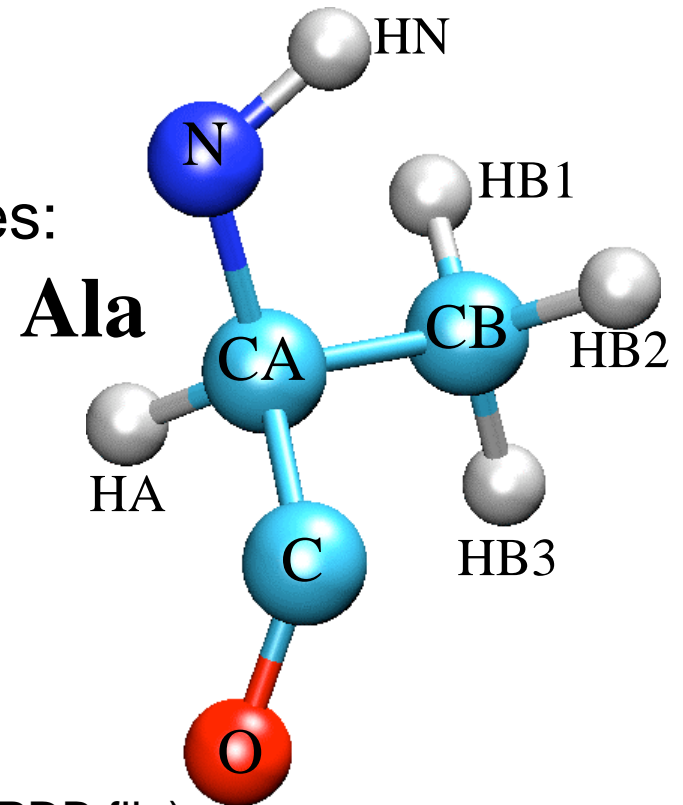
ATOM	557	OT2	GLY	56	17.308	-0.138	12.617	1.00	21.95	BPTI
------	-----	-----	-----	----	--------	--------	--------	------	-------	------

END

Topology Files

atomic properties (mass, charge, type)

- Every possible residue is listed.
- Provides all static atom-specific values:
 - atom name (N, C, CA)
 - atom type (NH1, CT1)
 - residue name (ALA, HIS)
 - residue id (integer)
 - segment id (6PTI)
 - atomic mass (in atomic mass units)
 - partial charge (in electronic charge units)
- What is not in the topology file?
 - coordinates (dynamic data, initially read from PDB file)
 - velocities (dynamic data, initially from Boltzmann distribution)
 - force field parameters (non-specific, used for many molecules)



Protein Primary Structure

The twenty amino acids

*Introduction to Protein
Structure, 2nd ed.*

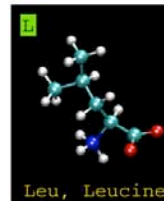
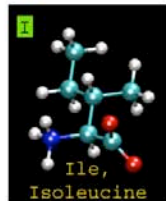
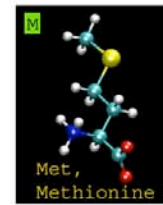
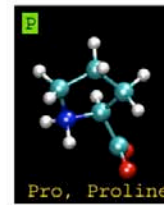
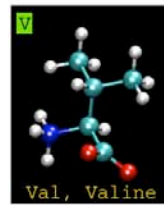
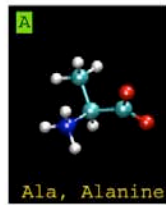
Carl Branden & John Tooze,
1999

*Protein Structure and
Function,*

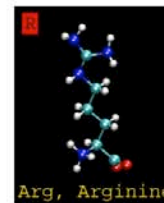
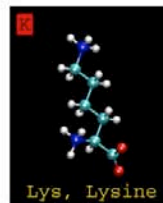
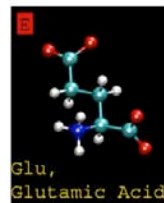
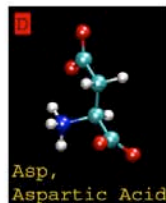
Greg Petsko & Dagmar Ringe,
2003

Molecular Biology of The Cell
Alberts, Johnson, Lewis, Raff,
Roberts, Walter, 2002

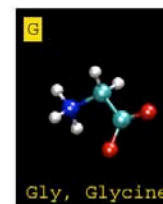
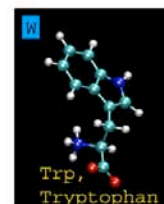
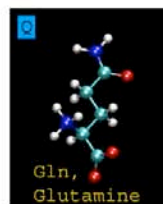
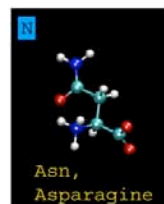
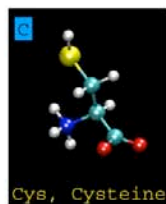
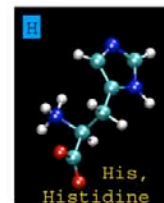
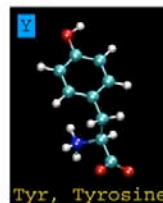
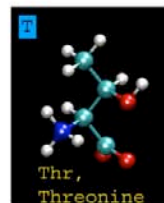
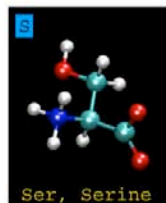
non-polar



charged



polar



URL: <http://lectures.molgen.mpg.de/ProteinStructure>

```

...
MASS 121 CTL1 12.011000 C ! sp3 carbon with 1 H (-CH1-)
MASS 122 CTL2 12.011000 C ! carbon of methylene group (-CH2-)
MASS 123 CTL3 12.011000 C ! carbon of methyl group (-CH3)
MASS 124 CTL5 12.011000 C ! carbon of methyl group (-CH3) for tetramethylammonium
MASS 125 CEL1 12.011000 C ! for alkene; RHC=CR
MASS 126 CEL2 12.011000 C ! for alkene; H2C=CR
MASS 140 NTL 14.007000 N ! ammonium nitrogen

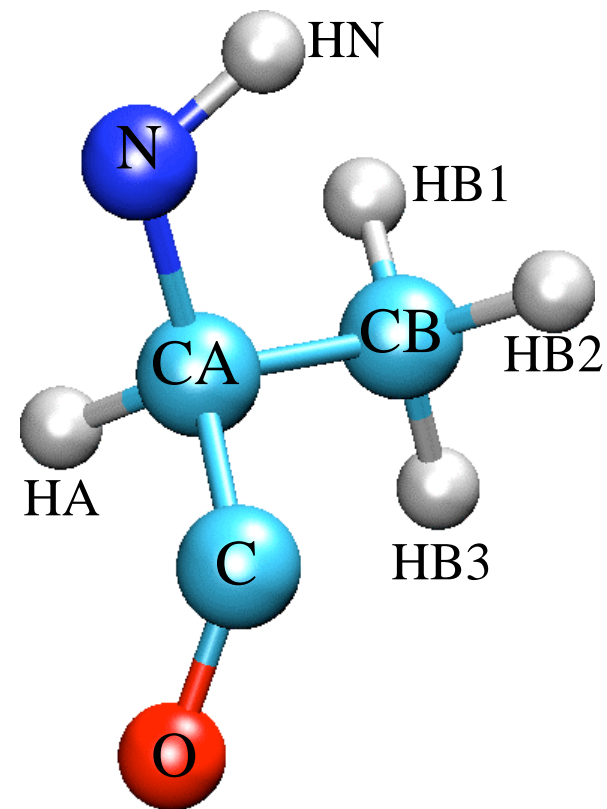
```

Example of Topology File

```

...
RESI ALA      0.00
GROUP
ATOM N  NH1  -0.47  !  |
ATOM HN H     0.31  !  | HN-N
ATOM CA CT1   0.07  !  | HB1
ATOM HA HB    0.09  !  | /
GROUP        !  | HA-CA--CB-HB2
ATOM CB CT3  -0.27  !  | \
ATOM HB1 HA   0.09  !  | HB3
ATOM HB2 HA   0.09  !  | O=C
ATOM HB3 HA   0.09  !  |
GROUP        !
ATOM C   C    0.51
ATOM O   O   -0.51
BOND CB CA N HN N CA
BOND C  CA C +N CA HA CB HB1 CB HB2 CB HB3
DOUBLE O C
IMPR N -C CA HN C CA +N O

```



PDB FILE (sequence of residues)

+

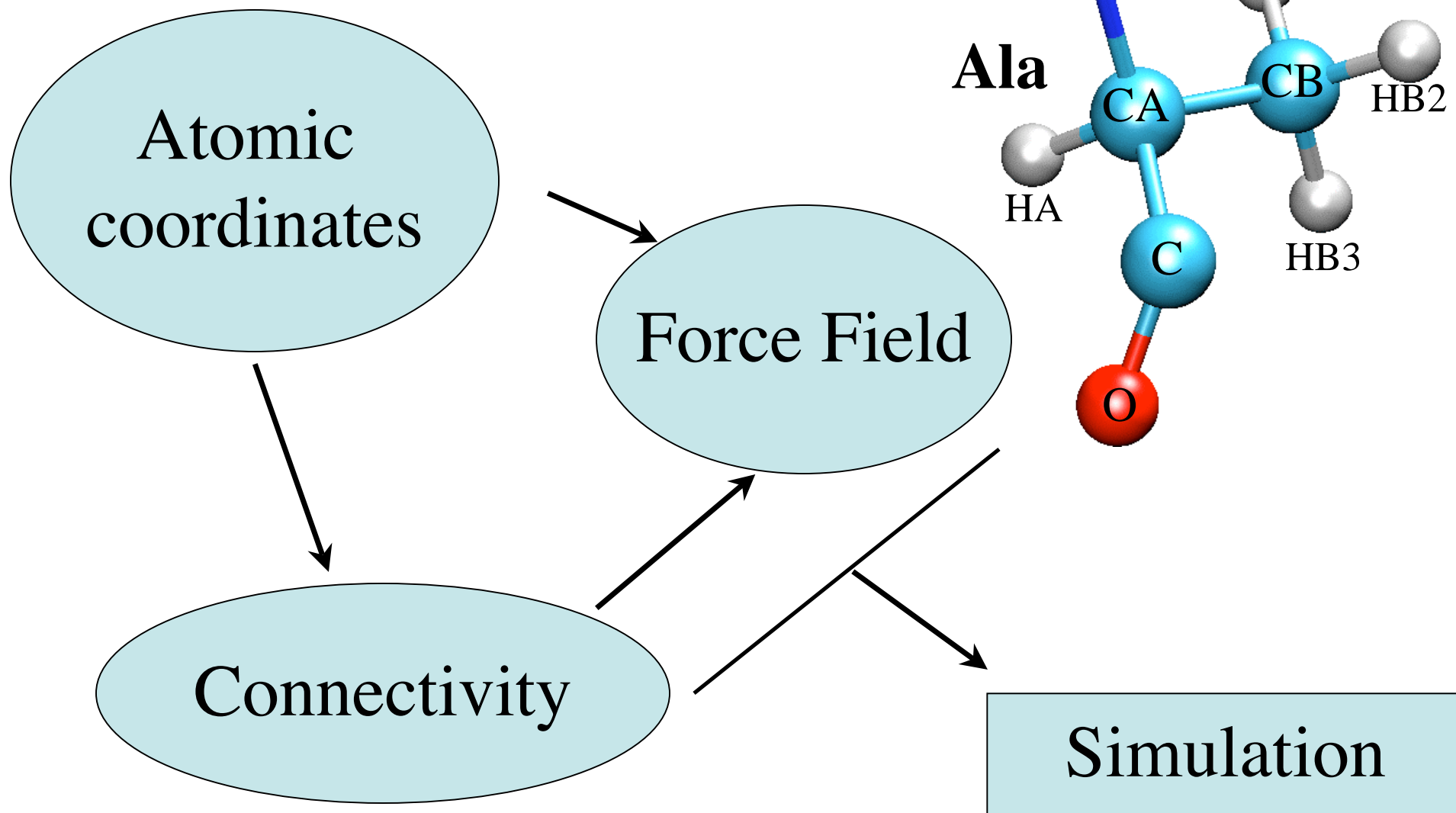
TOPOLOGY (of each residue)

=

CONNECTIVITY

(of entire protein)

MD simulation basics



Parameter Files

force constants for all types of interactions

BONDS

!V(bond) = Kb(b - b0)**2

!atom type Kb b0

!

CE1 CE1 440.000 1.3400 !

! for butene; from propene, yin/adm jr., 12/95

CE1 CE2 500.000 1.3420 !

! for propene, yin/adm jr., 12/95

ANGLES

!

!V(angle) = Ktheta(Theta - Theta0)**2

!

!V(Urey-Bradley) = Kub(S - S0)**2

!

!

!atom types Ktheta Theta0 Kub S0

!

CA CA CA 40.000 120.00 35.00 2.41620 !ALLOW ARO

! JES 8/25/89

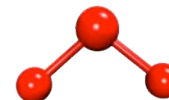
CE1 CE1 CT3 48.00 123.50 !

! for 2-butene, yin/adm jr., 12/95

CE1 CT2 CT3 32.00 112.20 !

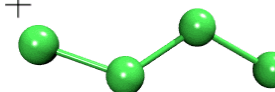
! for 1-butene; from propene, yin/adm jr., 12/95

CE2 CE1 CT2 48.00 126.00 !



$$U(\vec{R}) = \underbrace{\sum_{\text{bonds}} k_i^{\text{bond}} (r_i - r_0)^2}_{U_{\text{bond}}} + \underbrace{\sum_{\text{angles}} k_i^{\text{angle}} (\theta_i - \theta_0)^2}_{U_{\text{angle}}} +$$

$$\underbrace{\sum_{\text{dihedrals}} k_i^{\text{dihe}} [1 + \cos(n_i \phi_i + \delta_i)]}_{U_{\text{dihedral}}} +$$



$$\underbrace{\sum_i \sum_{j \neq i} 4\epsilon_{ij} \left[\left(\frac{\sigma_{ij}}{r_{ij}} \right)^{12} - \left(\frac{\sigma_{ij}}{r_{ij}} \right)^6 \right] + \sum_i \sum_{j \neq i} \frac{q_i q_j}{\epsilon r_{ij}}}_{U_{\text{nonbond}}}$$

Preparing Your System for MD Solvation

Biological activity is the result of interactions between molecules and occurs at the interfaces between molecules (protein-protein, protein-DNA, protein-solvent, DNA-solvent, etc).

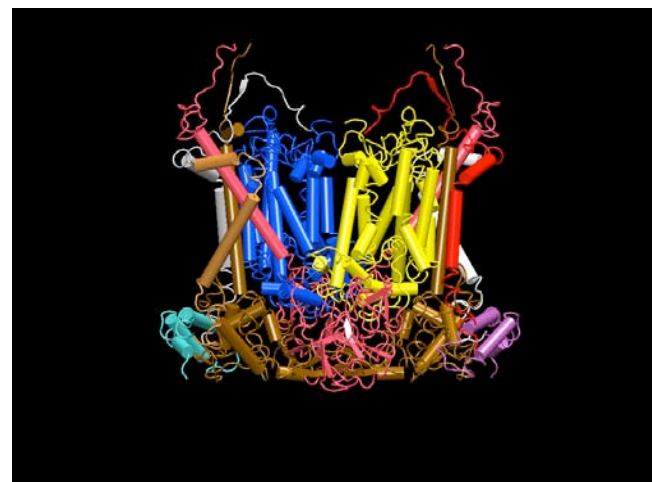
*mitochondrial
bc1 complex*

Why model solvation?

- many biological processes occur in aqueous solution
- solvation effects play a crucial role in determining molecular conformation, electronic properties, binding energies, etc

How to model solvation?

- explicit treatment: solvent molecules are added to the molecular system
- implicit treatment: solvent is modeled as a continuum dielectric



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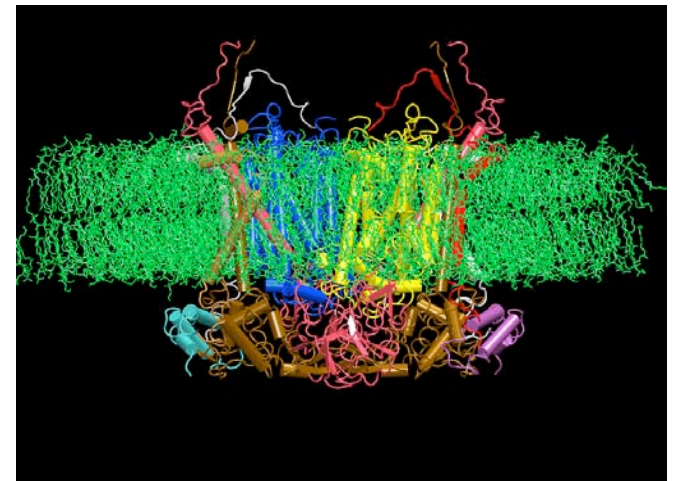
*mitochondrial
bc1 complex*

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Preparing Your System for MD Solvation

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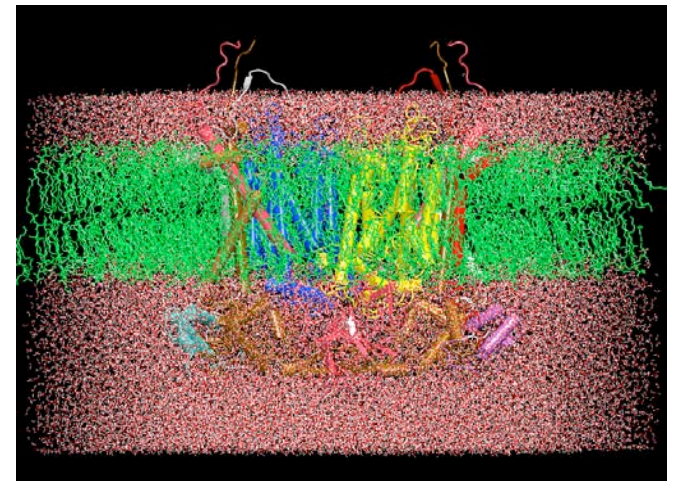
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- many biological processes occur in aqueous solution
- solvation effects play a crucial role in determining molecular conformation, electronic properties, binding energies, etc

How to model solvation?

- explicit treatment: solvent molecules are added to the molecular system
- implicit treatment: solvent is modeled as a continuum dielectric

*mitochondrial
bc1 complex*



From the Mountains to the Valleys

how to actually describe a protein

Initial coordinates have bad contacts, causing high energies and forces (due to averaging in observation, crystal packing, or due to difference between theoretical and actual forces)

Minimization finds a nearby local minimum.

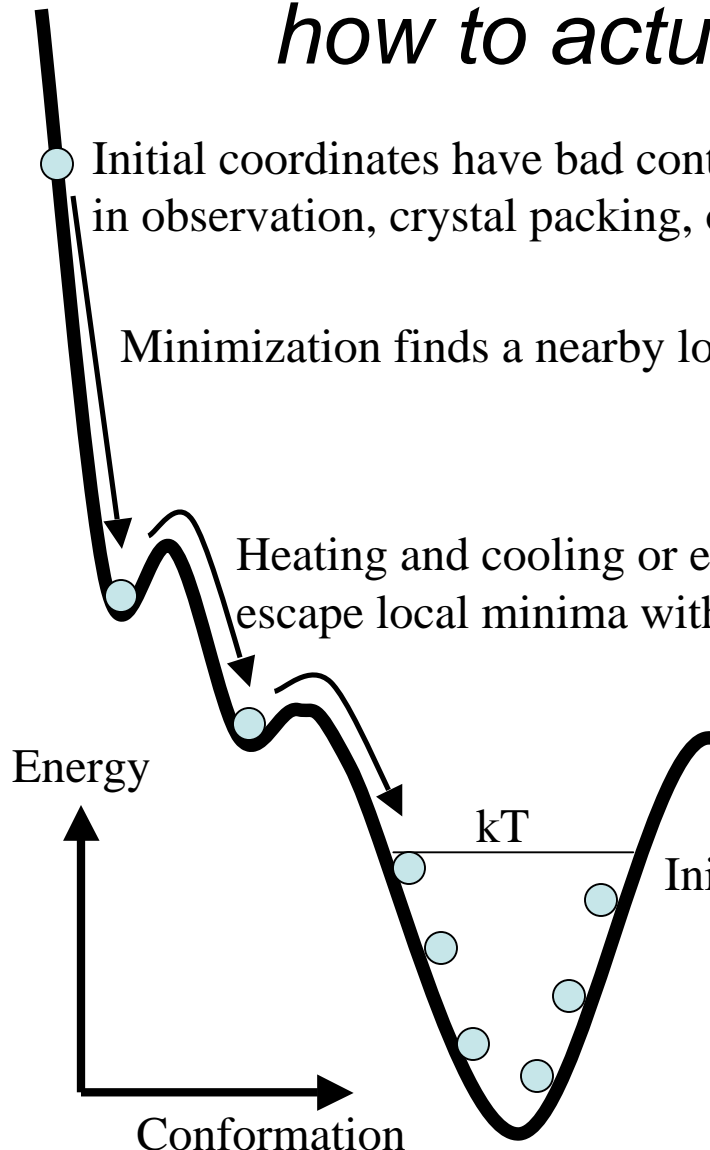
Heating and cooling or equilibration at fixed temperature permits biopolymer to escape local minima with

Energy

kT

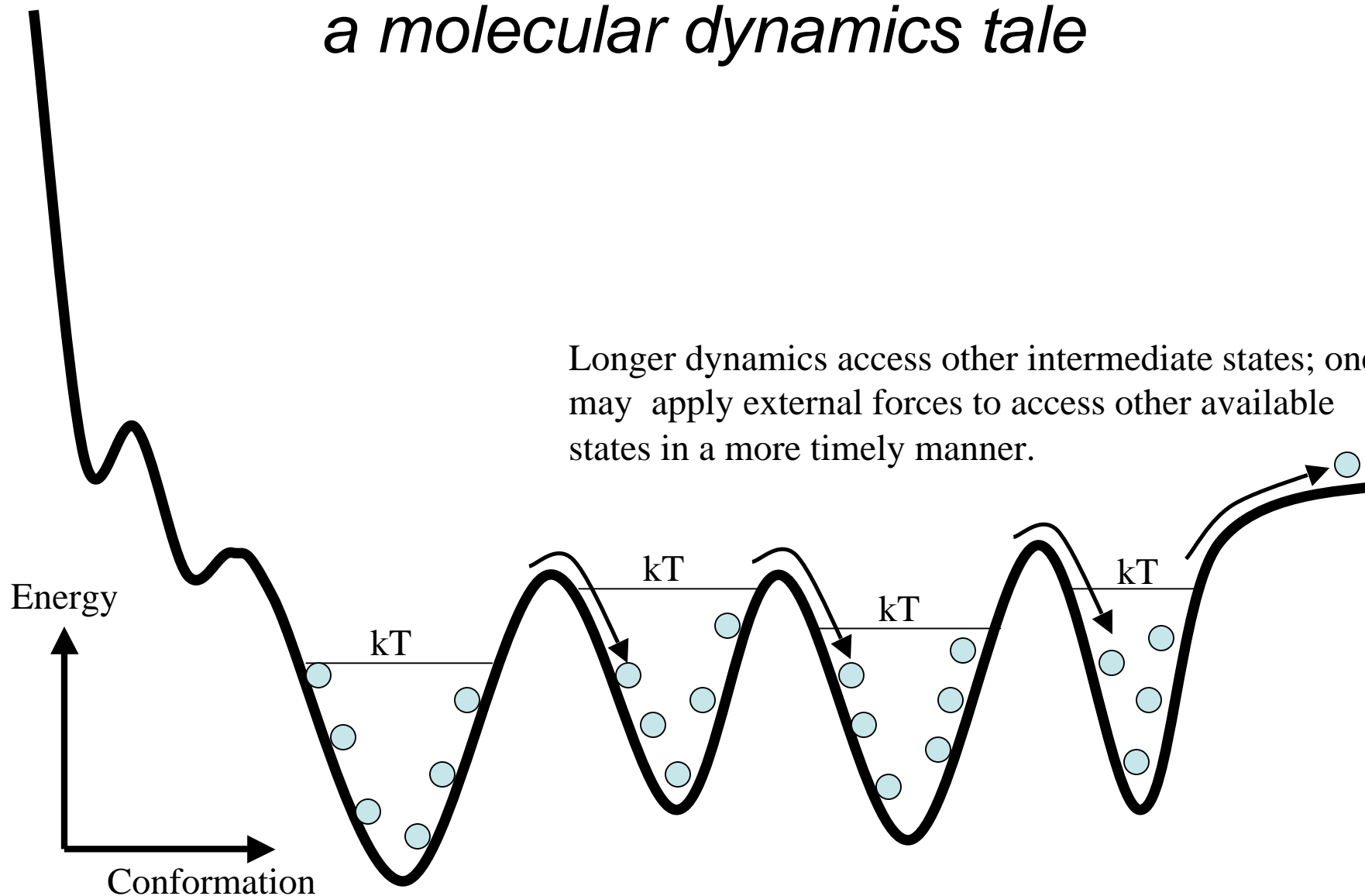
Initial dynamics samples thermally accessible states.

Conformation



From the Mountains to the Valleys

a molecular dynamics tale



Steps in a Typical MD Simulation

- 1. Prepare molecule
 - Read in pdb and psf file
- 2. Minimization
 - Reconcile observed structure with force field used ($T = 0$)
- 3. Heating
 - Raise temperature of the system
- 4. Equilibration
 - Ensure system is stable
- 5. Dynamics
 - Simulate under desired conditions (NVE, NpT, etc)
 - Collect your data
- 6. Analysis
 - Evaluate observables (macroscopic level properties)
 - Or relate to single molecule experiments

Molecular Dynamics Ensembles

Constant energy, constant number of particles (NE)

Constant energy, constant volume (NVE)

Constant temperature, constant volume (NVT)

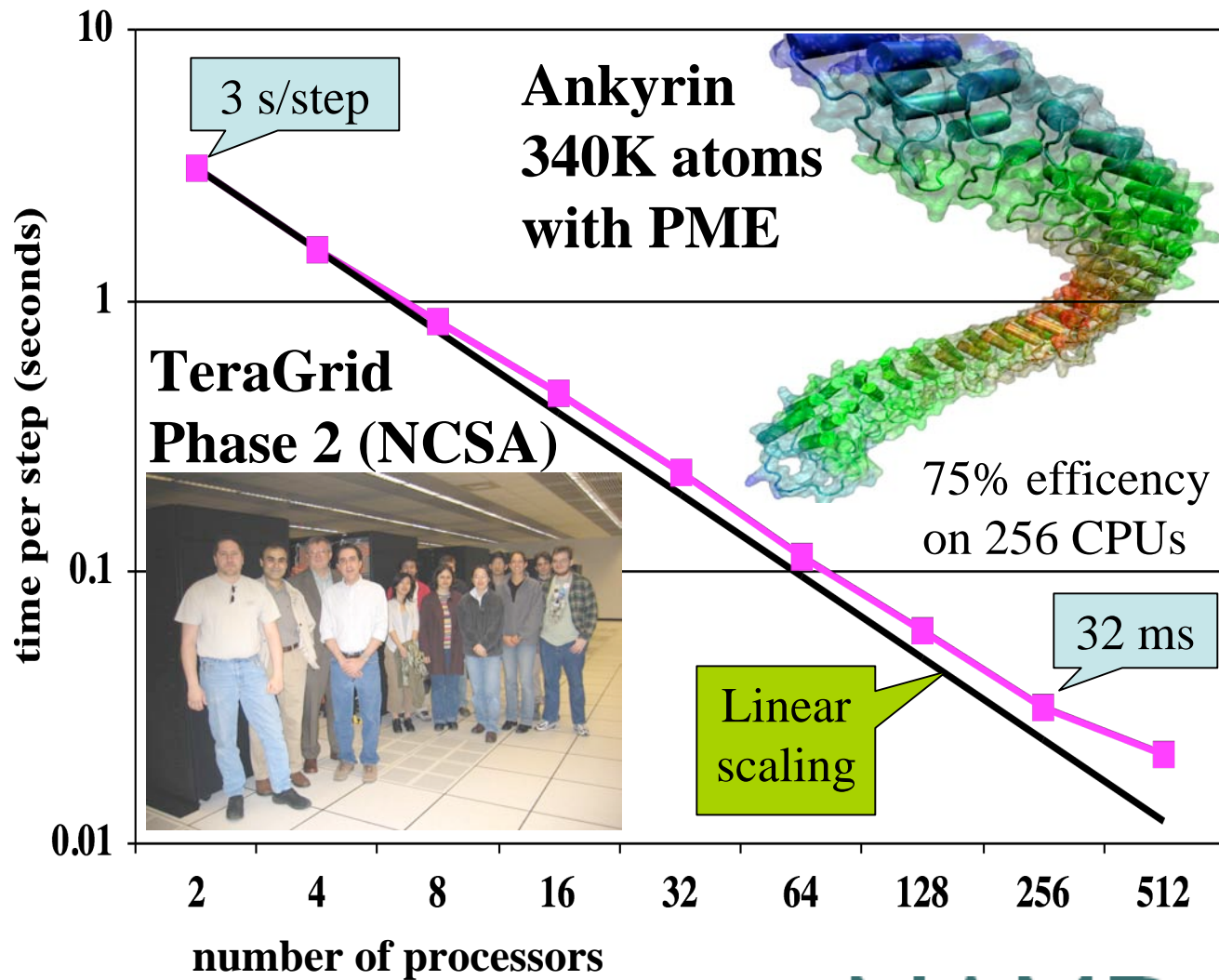
Constant temperature, constant pressure (NPT)

Choose the ensemble that best fits your system and start the simulations

NAMD: The Program we will Use



*NAMD
programmer
J. Phillips
Ph.D. UIUC
Physics*



Simulation of large biomolecular systems

2002 Gordon Bell Award for parallel scalability.

Runs at NSF centers, on clusters, and on desktop.

Available for **FREE** as precompiled binaries; includes source code.

10,000 registered users.



Linux Clusters 101

parallel computing on a professor's salary

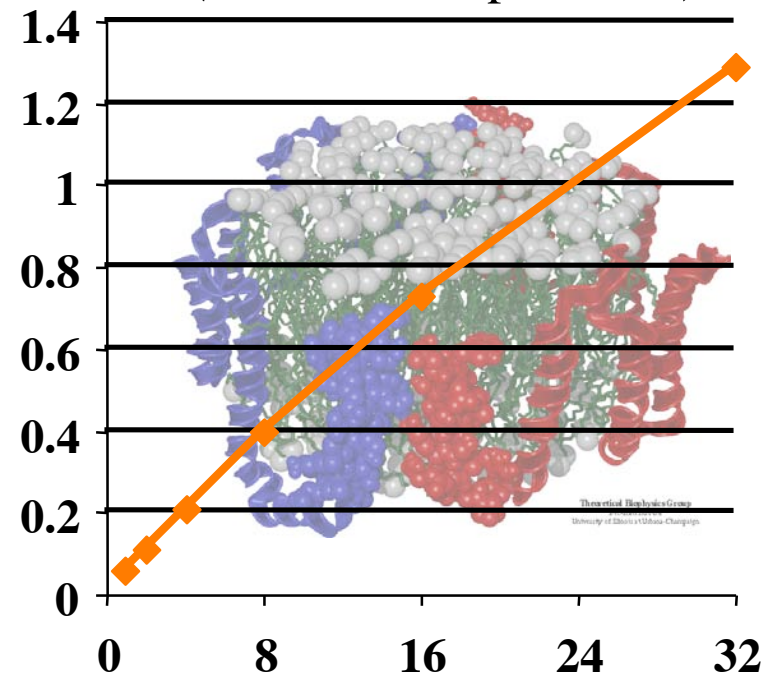
Learn to build your own Linux cluster!



\$1000 per processor



92K atoms with PME
(ns simulated per week)

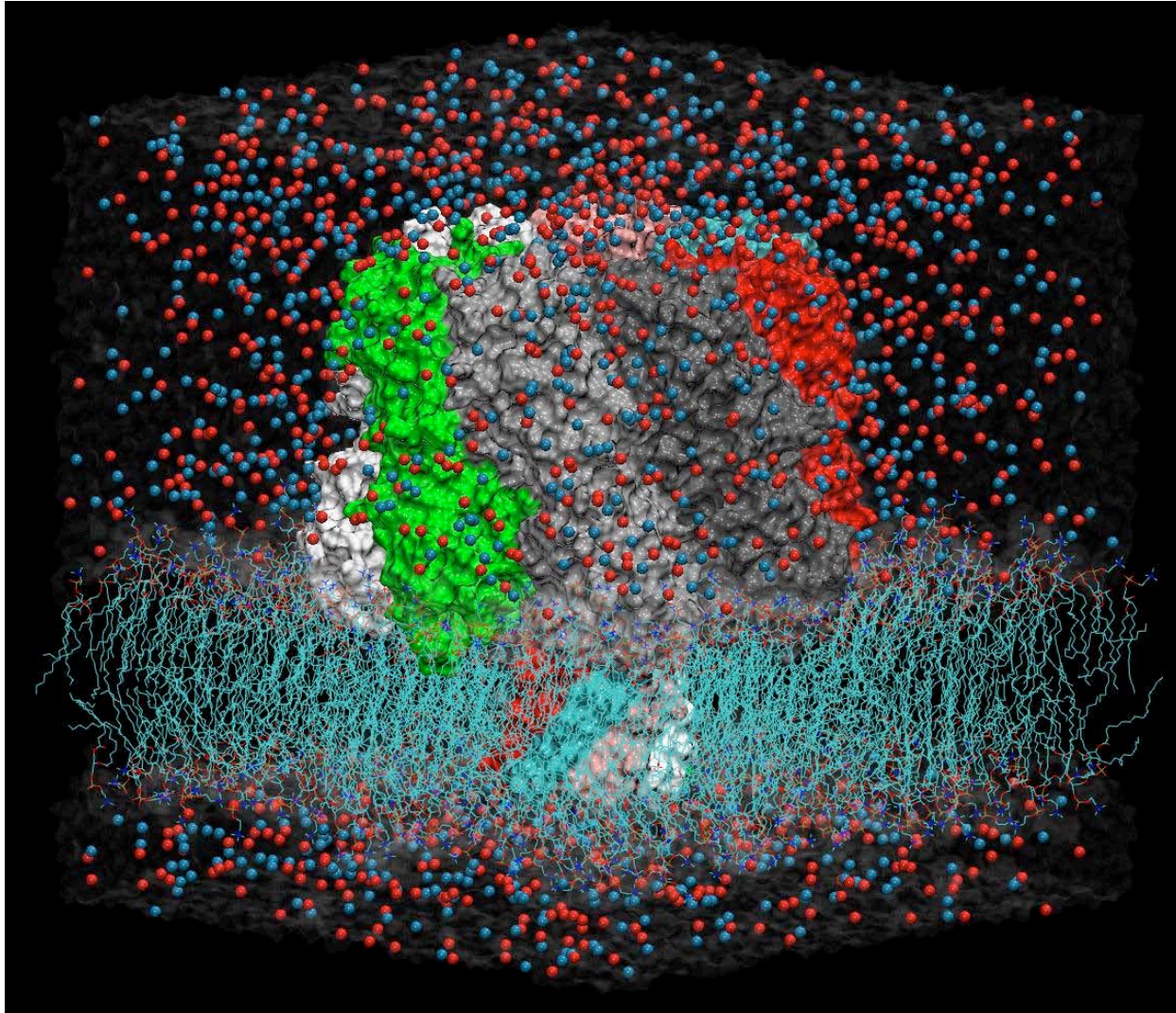


Cutting Corners

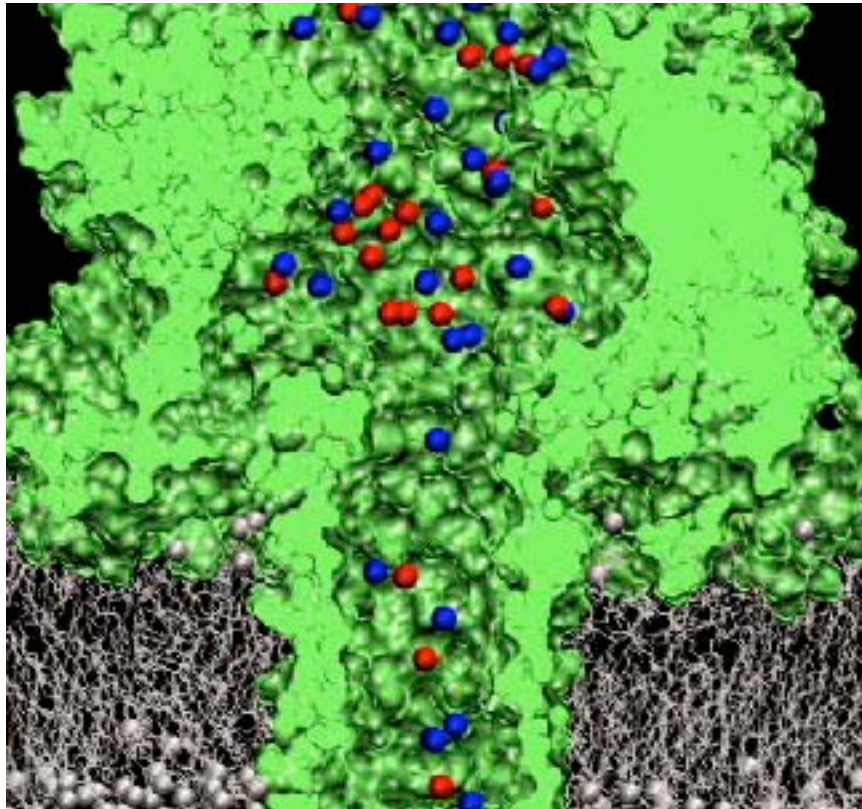
cutoffs, PME, rigid bonds, and multiple timesteps

- Nonbonded interactions require order N^2 computer time!
 - Truncating at R_{cutoff} reduces this to order $N R_{\text{cutoff}}^3$
 - Particle mesh Ewald (PME) method adds long range electrostatics at order $N \log N$, only minor cost compared to cutoff calculation.
- Can we extend the timestep, and do this work fewer times?
 - Bonds to hydrogen atoms, which require a 1fs timestep, can be held at their equilibrium lengths, allowing 2fs steps.
 - Long range electrostatics forces vary slowly, and may be evaluated less often, such as on every second or third step.

Setting up an α -hemolysin simulation



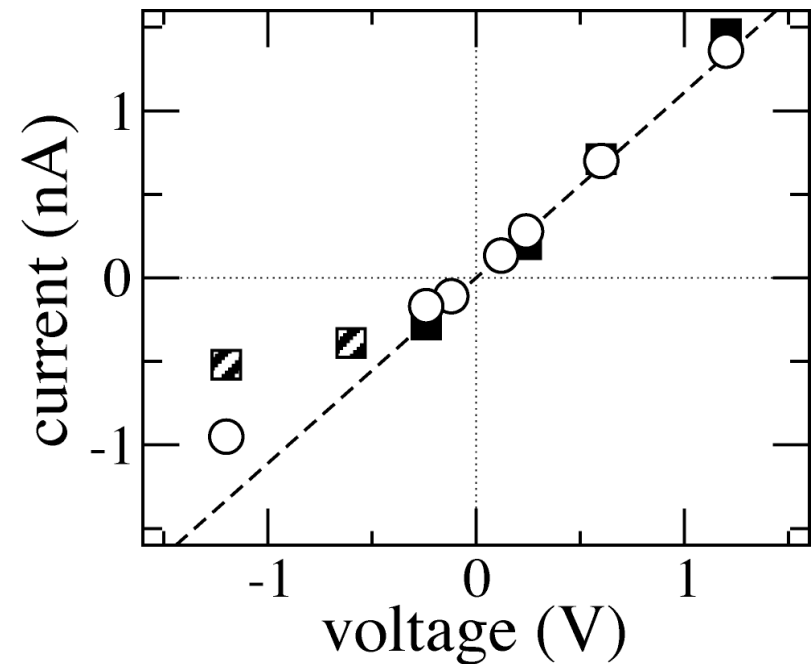
Computing current-voltage curve from MD



Ionic current at 120mV:

Experiment: 120 pA

Simulations: 130 pA

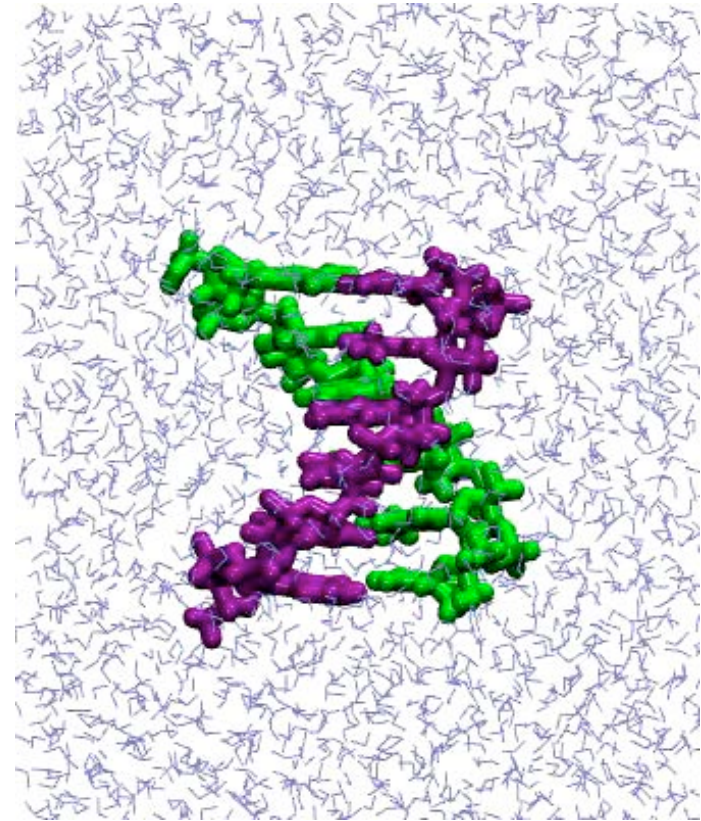
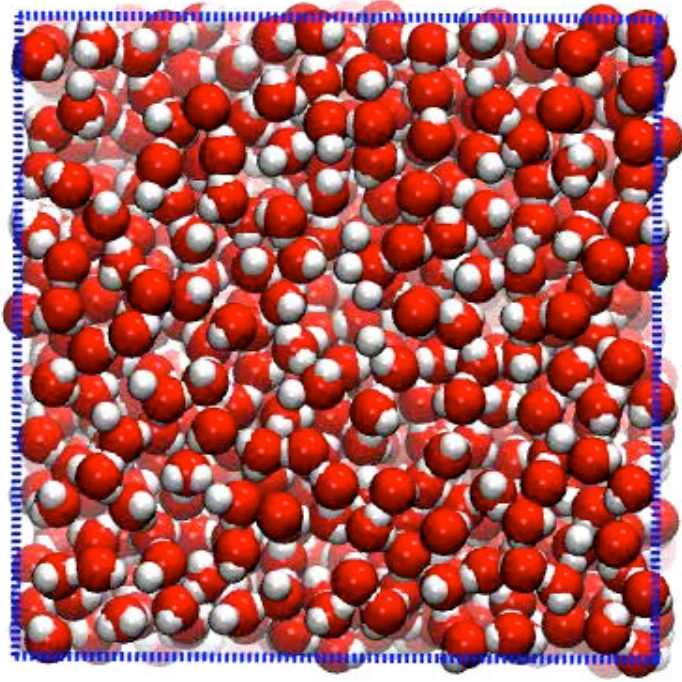


$$I(t) = \frac{1}{\Delta t L_z} \sum_{i=1}^N q_i (z_i(t + \Delta t) - z_i(t))$$

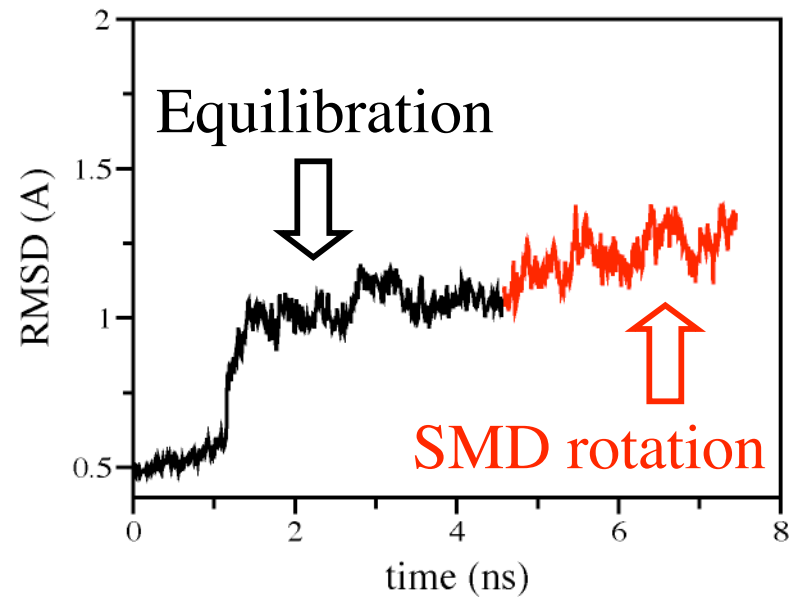
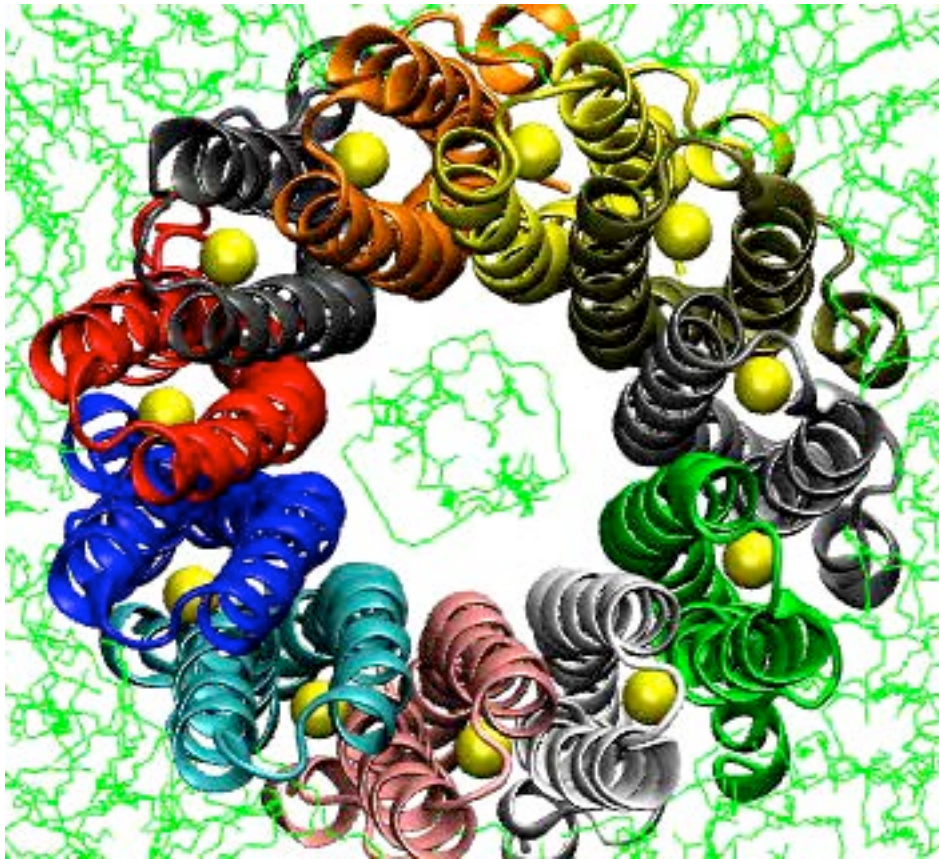
Instantaneous current

Current-voltage dependence

Apply external forces

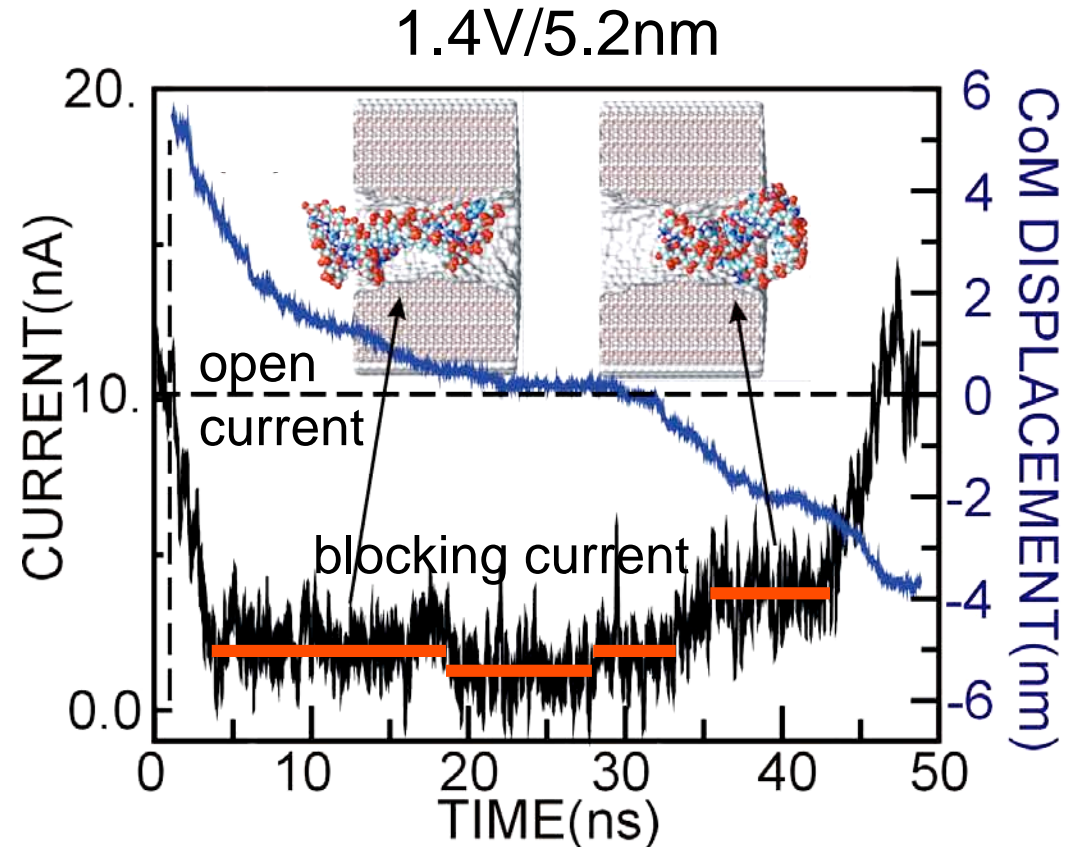
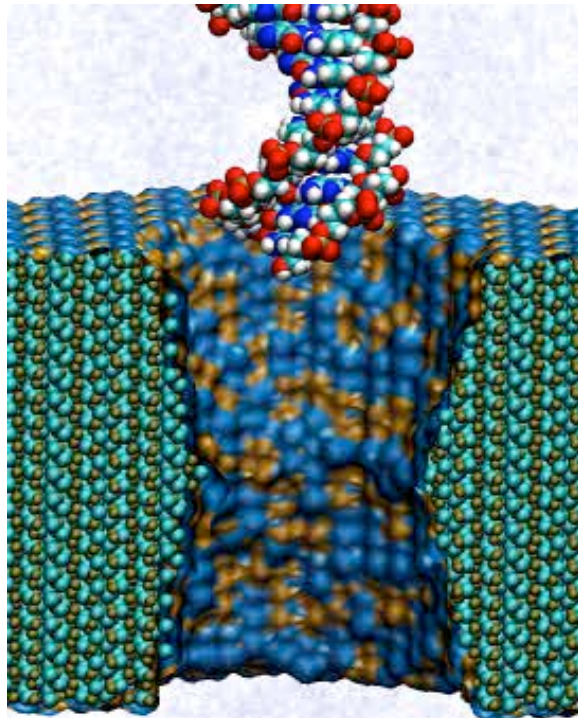


Forced Rotation of c_{11} in Na^+ -ATP Synthase



3-ns SMD simulation
torque: 1000 kcal/mol

DNA translocation through Si_3N_4 nanopore

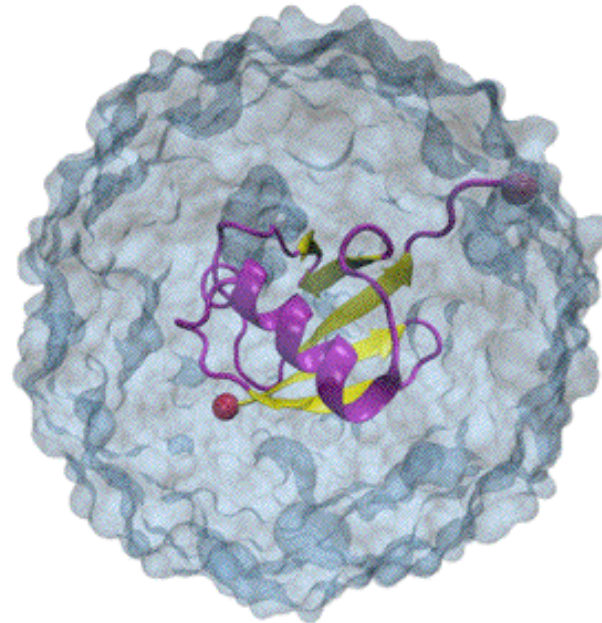


- at the end of the translocation *DNA* partially denatures
- blocking current correlates with molecular velocity
- translocation time: 10 ns - 3 μ s depending on the field

• Simulations: 1.4V/5.2nm @ $F \sim 400$ pN pore diameter @ $d = 2.5$ nm
• DNA sequence is CCCCCCCCCCCCCCCCCC

NAMD TUTORIAL

Unix/MacOSX Version



NAMD Developer: James Phillips

Timothy Isgro

James Phillips

Marcos Sotomayor

Elizabeth Villa

February 2006

The NAMD Configuration File / 1

Files needed:

```
structure          mypsf.psf
coordinates        mypdb.pdb
```

Define temperature

```
set temperature    310
    ;# target temperature used several times below
```

Starting simulation with random velocities

```
# starting from scratch
temperature        $temperature
    ;# initialize velocities randomly
```

The NAMD Output File / 1

Preamble

Info: NAMD 2.5b2ss03 for Linux-i686-Clustermatic

Info:

Info: Please visit <http://www.ks.uiuc.edu/Research/namd/>

Info: and send feedback or bug reports to namd@ks.uiuc.edu

Info:

Info: Please cite Phillips et al., J. Comp. Chem. 26: 1781-1802 (2005)

Info: in all publications reporting results obtained with NAMD.

Info:

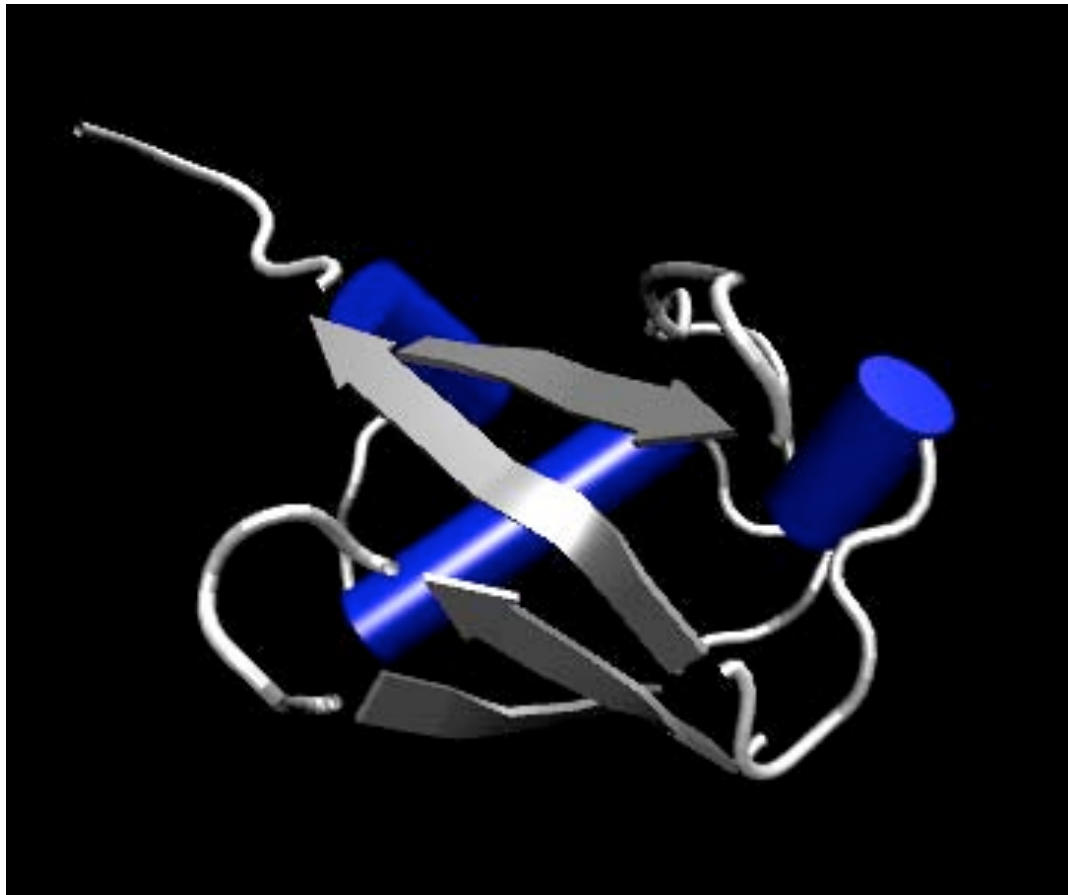
Info: Built Fri May 30 13:09:06 CDT 2003 by jim on umbriel

Info: Sending usage information to NAMD developers via UDP.

Info: Sent data is: 1 NAMD 2.5b2ss03 Linux-i686-Clustermatic 47 umbriel jim

Info: Running on 47 processors.

Ubiquitin

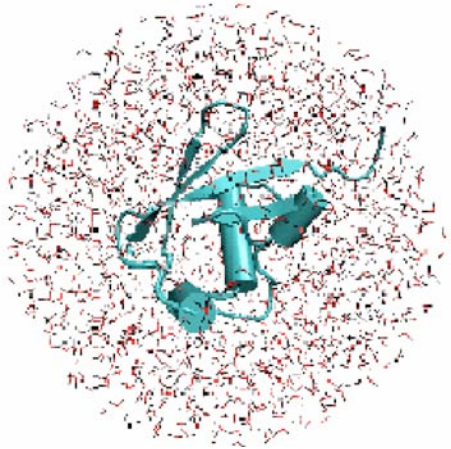


Fatemeh Araghi, Timothy Isgro, Marcos Sotomayor

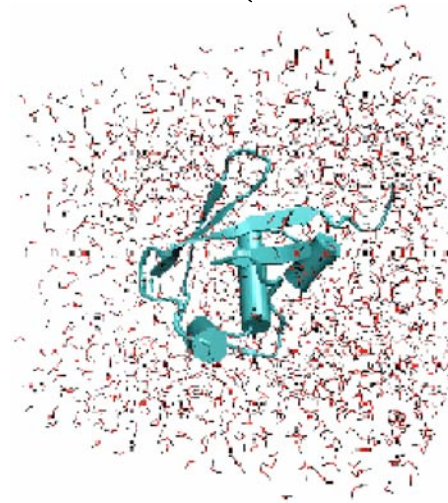
The NAMD Experience

You will first simulate ubiquitin in a water sphere and water box:

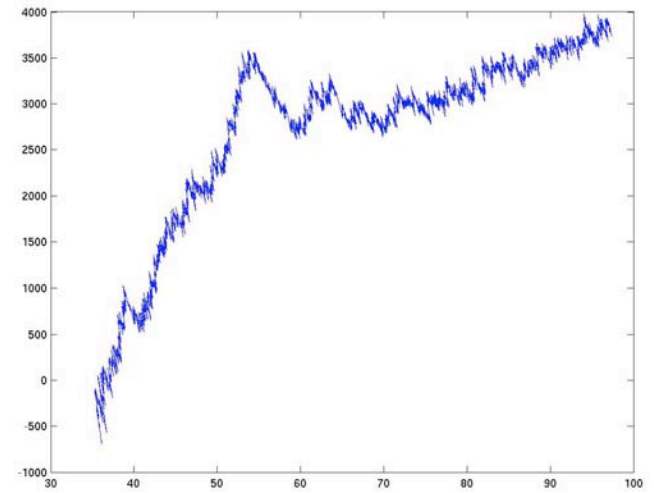
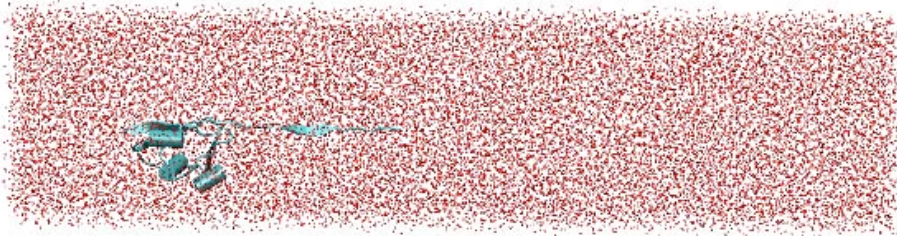
**Solvate the protein in a
water sphere (from VMD)**



**Solvate the protein in a
water box (from VMD)**



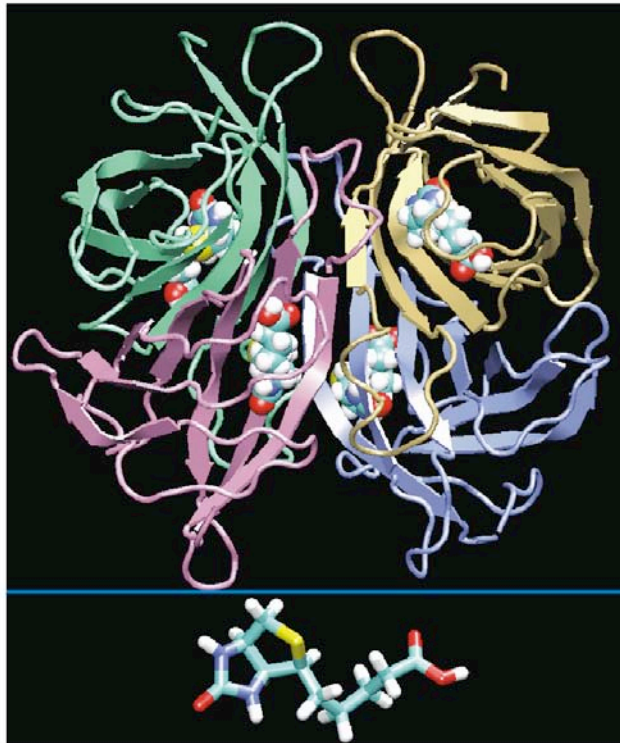
First SMD Simulation



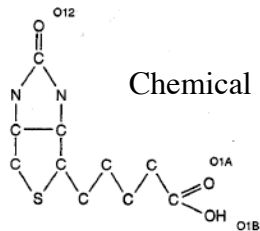
First peak when the first beta strand is stretched out

- SMD simulation, with constant velocity
- Box of water 70x240x70 Å **~81K atoms**
- smd velocity 0.4 Å/ps
- smd spring constant 7 kcal/mol Å²

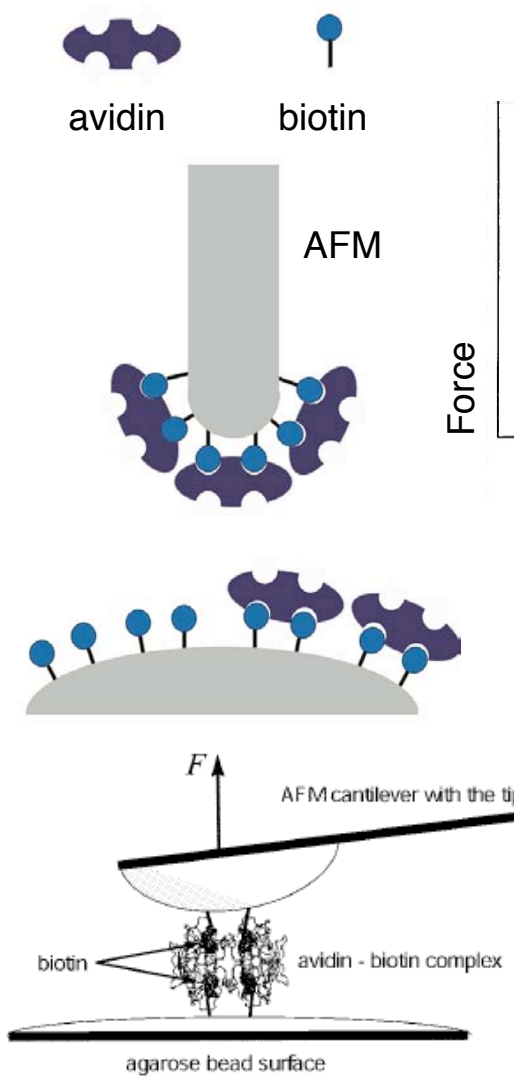
Atomic Force Microscopy Experiments of Ligand Unbinding



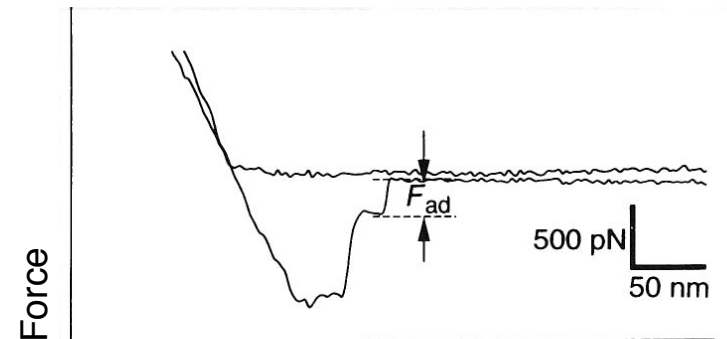
Biotin



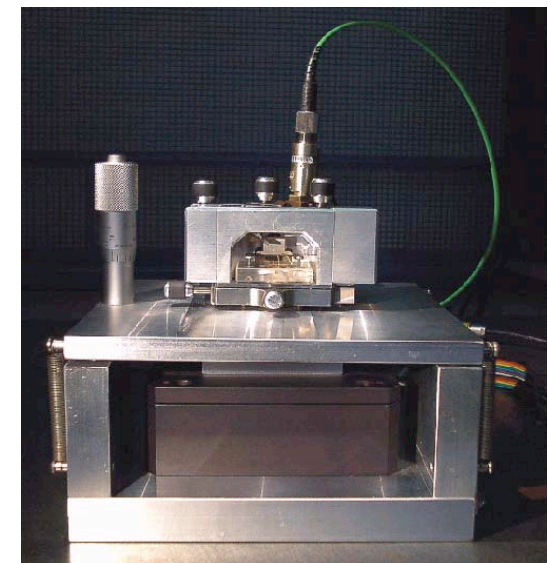
Chemical structure of biotin



Florin et al., Science 264:415 (1994)

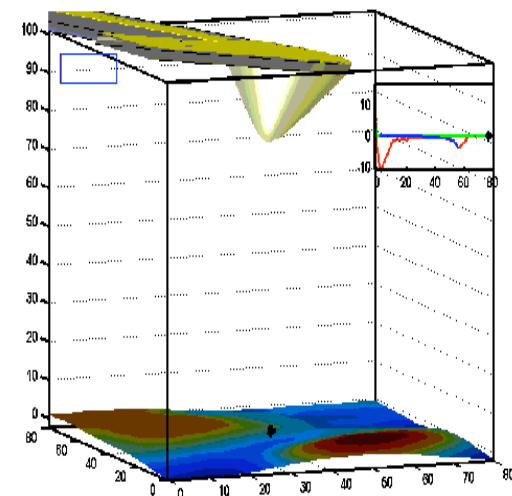
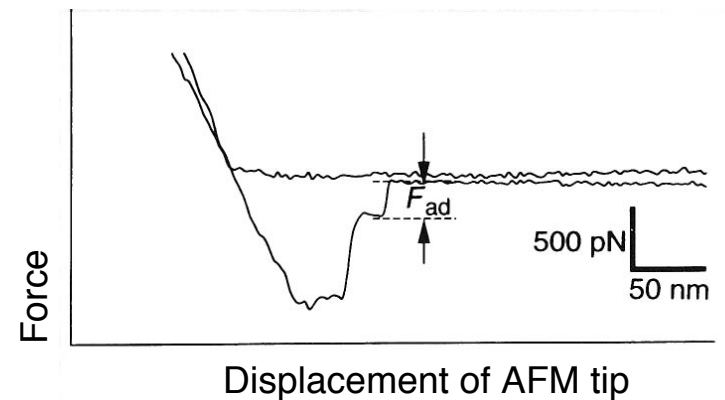
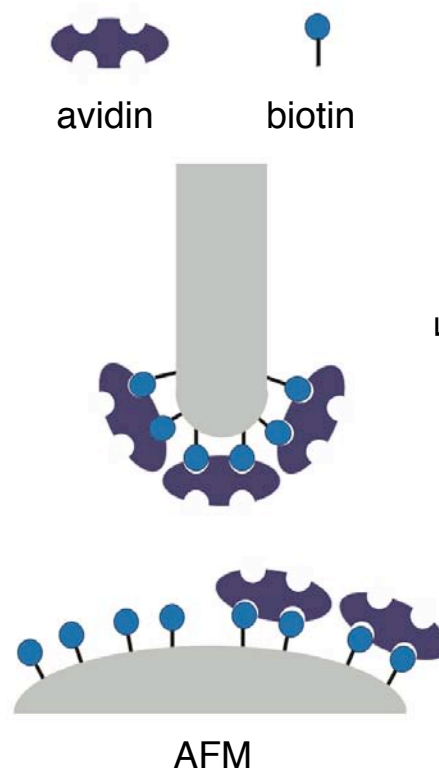
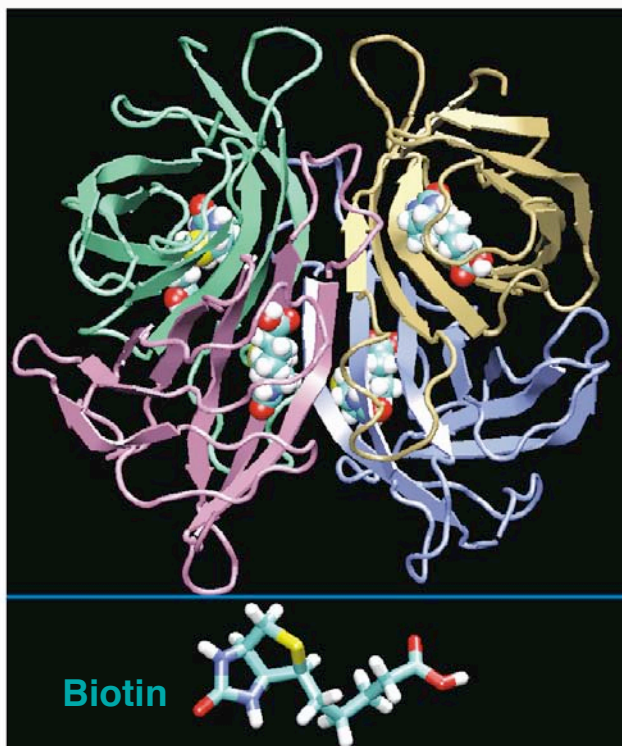


Displacement of AFM tip



Atomic Force Microscopy Experiments of Ligand Unbinding

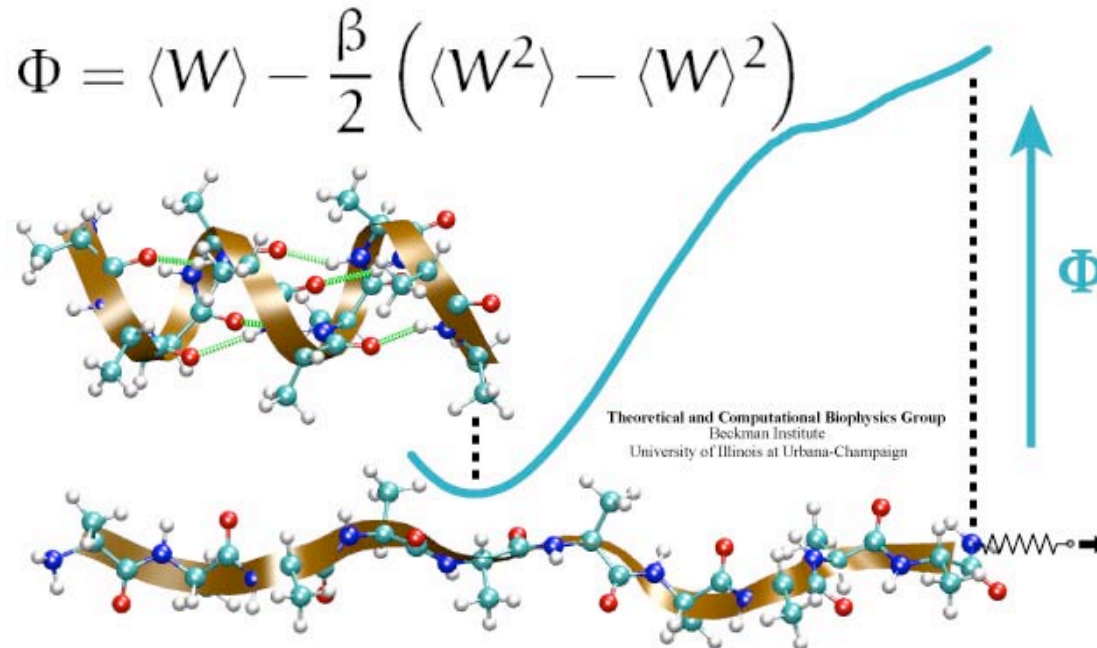
Florin et al., Science 264:415 (1994)



Free Energy of Stretched Alpha-Helix (Deca-alanine)

Thermodynamics: $\Delta G \leq \langle W \rangle$

Jarzynski (1997): $e^{-\Delta G/k_B T} = \langle e^{-W/k_B T} \rangle$



Free energy calculation from steered molecular dynamics simulations using Jarzynski's equality. S. Park, F. Khalili-Araghi, E. Tajkhorshid, and K. Schulten. *Journal of Chemical Physics*, 119:3559-3566, 2003

Calculating potentials of mean force from steered molecular dynamics simulations. S. Park and K. Schulten. *Journal of Chemical Physics*, 120: 5946-5961, 2004