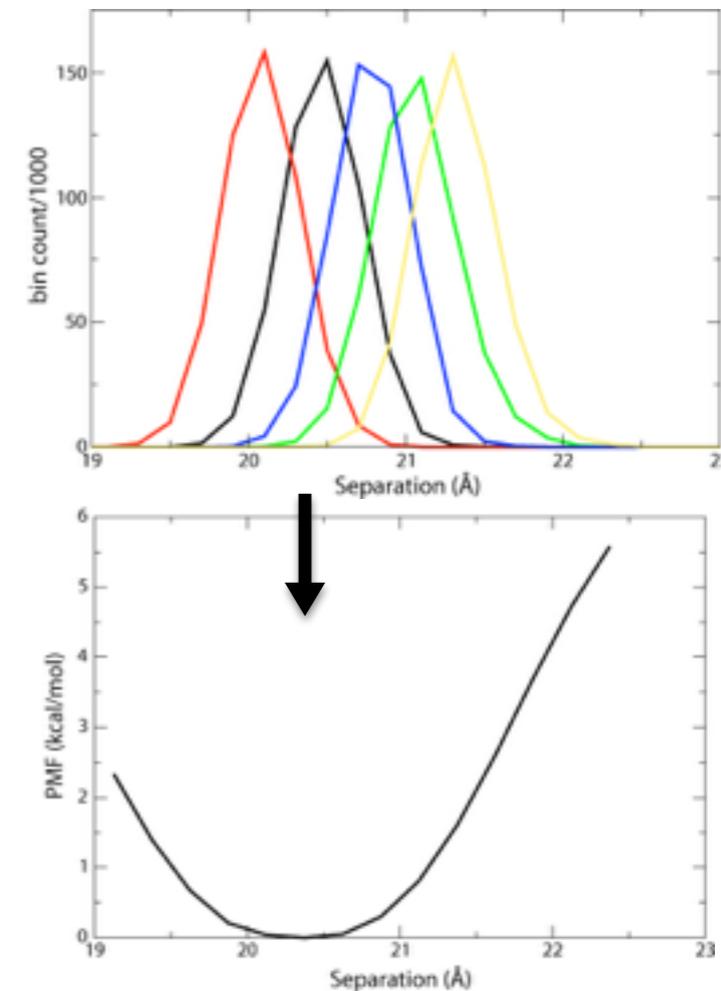
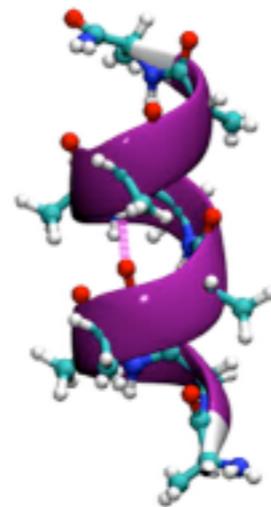
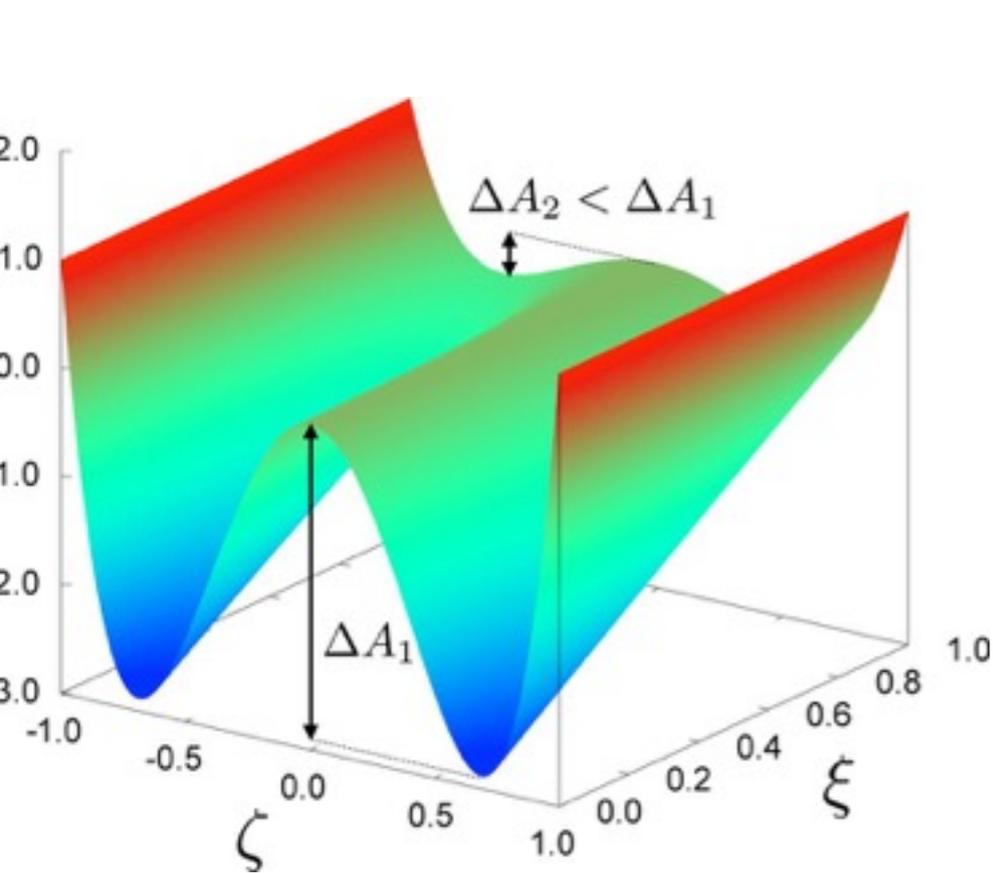


A general overview of Free energy methods



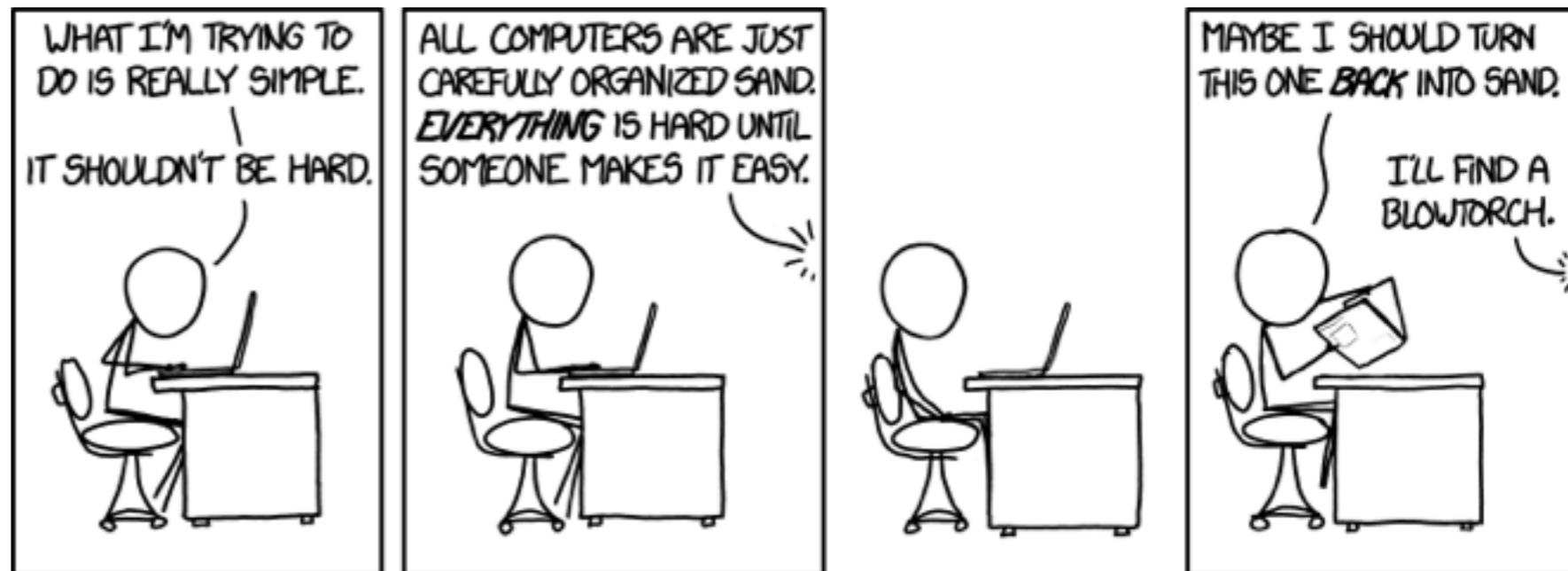
Comer et al. (2014)
JPCB 119:1129.

James C. (JC) Gumbart
Georgia Institute of Technology, Atlanta

Outline

- I. *Introduction to free energy and the need for biases*
- II. *Classes of common free-energy calculation methods*
- III. *Choosing a reaction coordinate and what is left behind*
- IV. *Example 1: deca-alanine folding*
- V. *Example 2: membrane permeation*

VI. *Final thoughts*



What is “free energy”?

the energy available to do work

If you're a physicist...

$$\Delta A = \boxed{\Delta U} - T \boxed{\Delta S}$$

*common view
of “energy”*

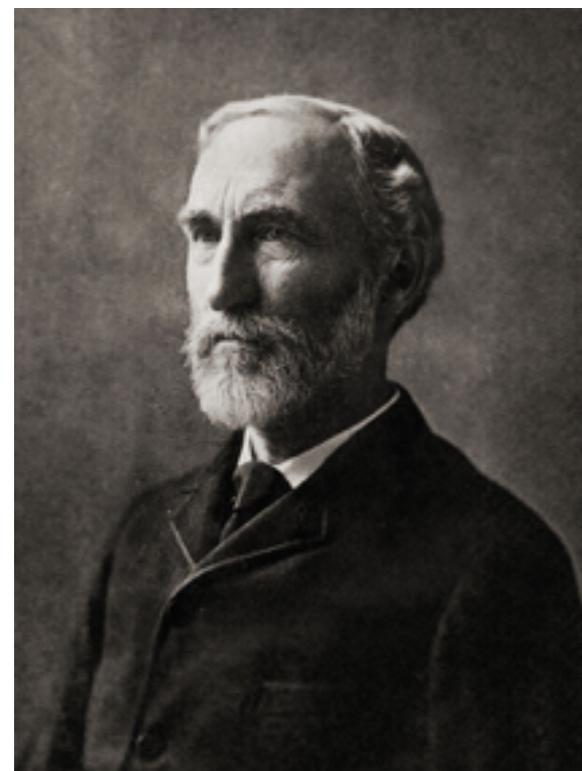
*entropy - an
effective energy*

constant volume/gas phase

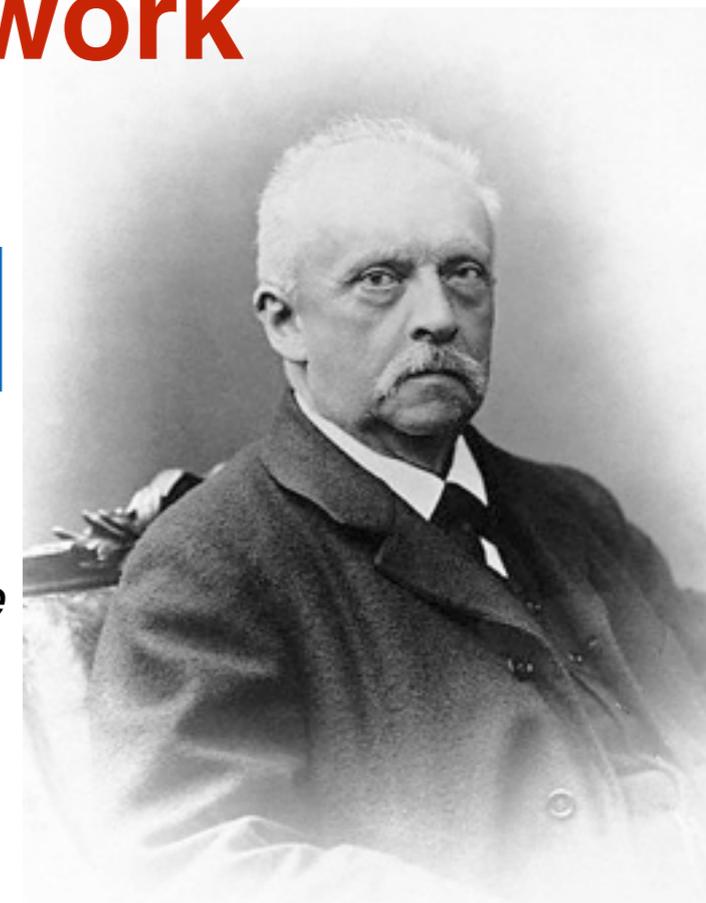
If you're a chemist...

$$\Delta G = \Delta U - T\Delta S + p\Delta V$$

constant pressure



Josiah Gibbs (technically, a mathematician)



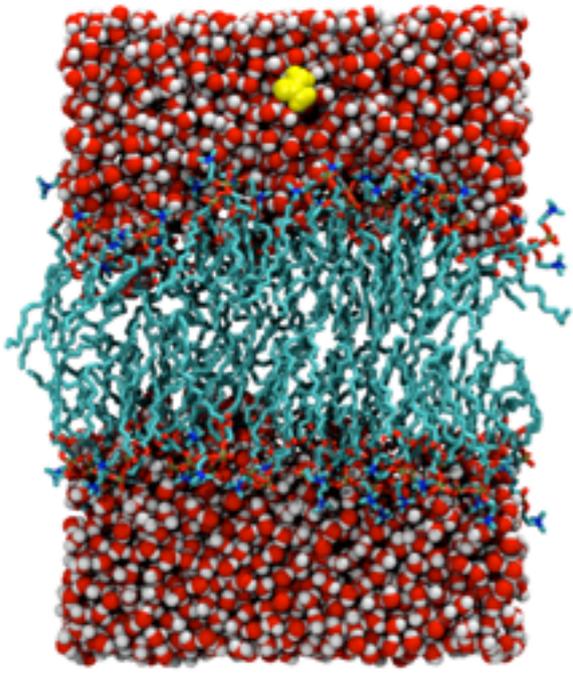
Hermann von Helmholtz

International Union of Pure and Applied Chemistry (IUPAC) has been trying to eliminate the term for years!

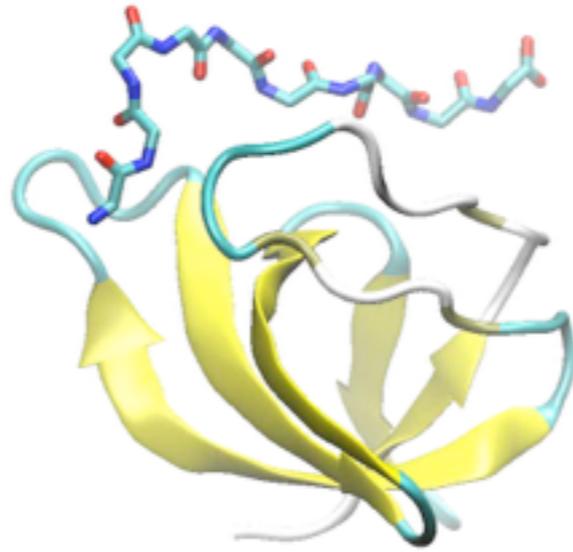
Fugacity: Fugacity is a thermodynamic quantity related to the Gibbs energy (older terminology of Gibbs free energy not recommended) of a gaseous molecule. For pressures of gases near 1 atm, the partial pressure of

from IUPAC Glossary of atmospheric chemistry terms 1990.

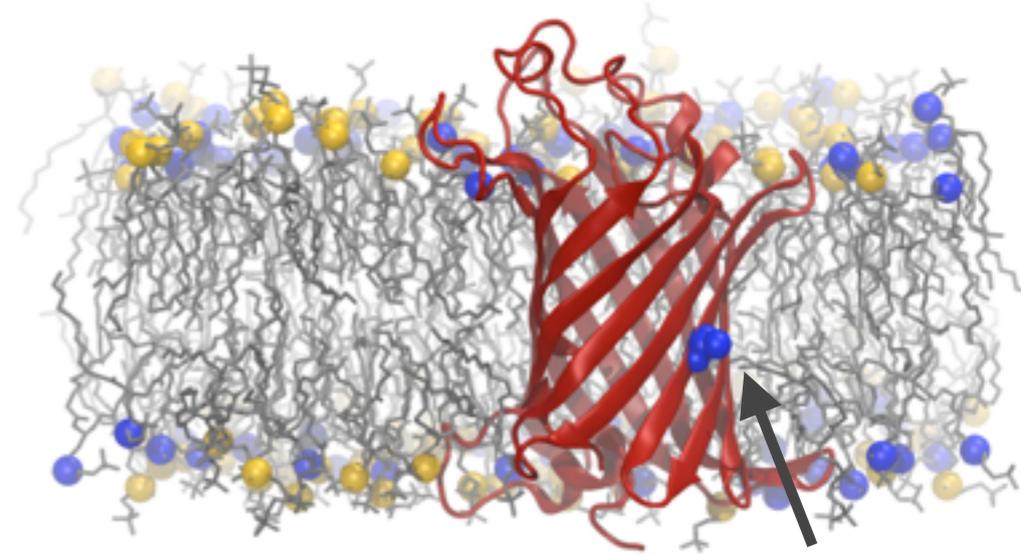
What can we measure with free energy?



**partition
coefficients**

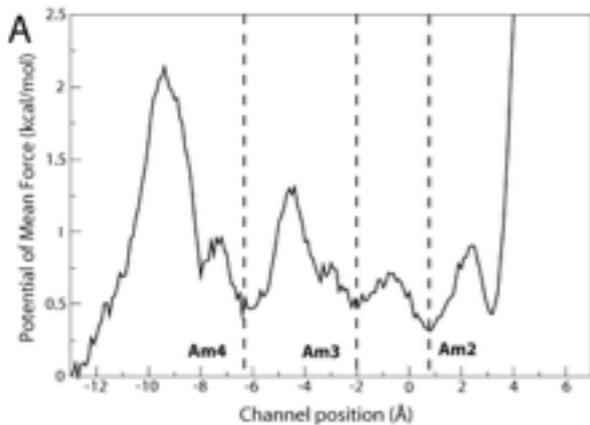


binding energies

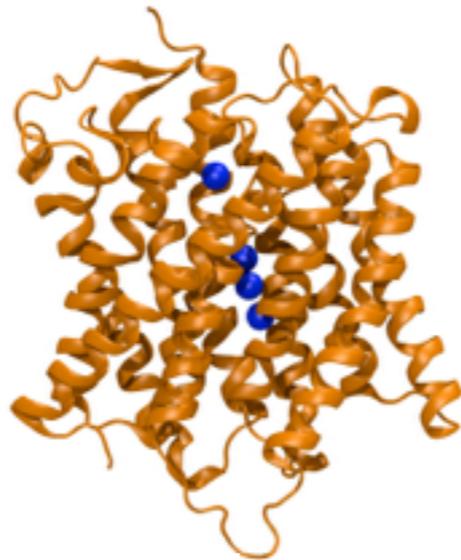


Ala → Arg

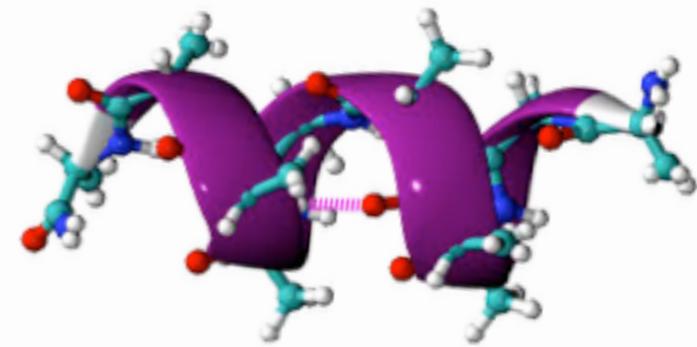
**mutagenesis
effects**



activation barriers



conformational change



How to measure free energy?

$$\beta = \frac{1}{kT}$$

Free energy tells us something about a system that is **general**, rather than specific to a single experiment or simulation

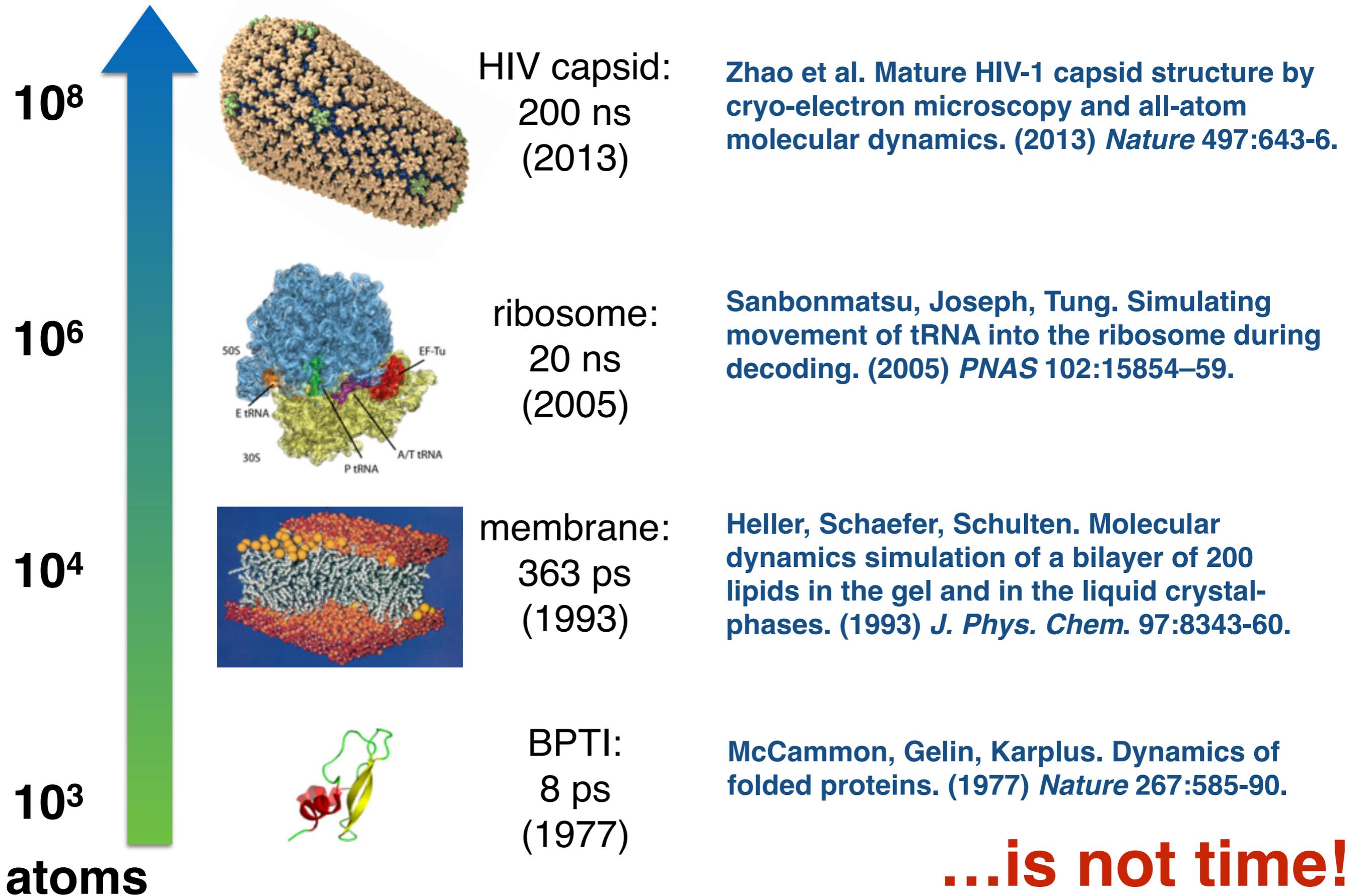
$$\Delta A = -\frac{1}{\beta} \ln\left(\frac{Z_1}{Z_0}\right) \quad \leftarrow \text{partition functions for two different states of the system}$$

$$p_i \propto Z_i \quad \longrightarrow \quad \Delta A = -\frac{1}{\beta} \ln\left(\frac{p_1}{p_2}\right) \quad \text{probability of observing a given state}$$

$$\lim_{t \rightarrow \infty} \frac{1}{t} \int q(t) dt = \frac{1}{M} \sum_M q_i = \langle q \rangle \quad \text{Ergodic hypothesis - running a simulation is the same as doing an experiment many times}$$

Therefore, we can determine free energies just by running the experiment/simulation for a **sufficiently long time and counting how often a given state appears!**

The value of brute-force simulations...



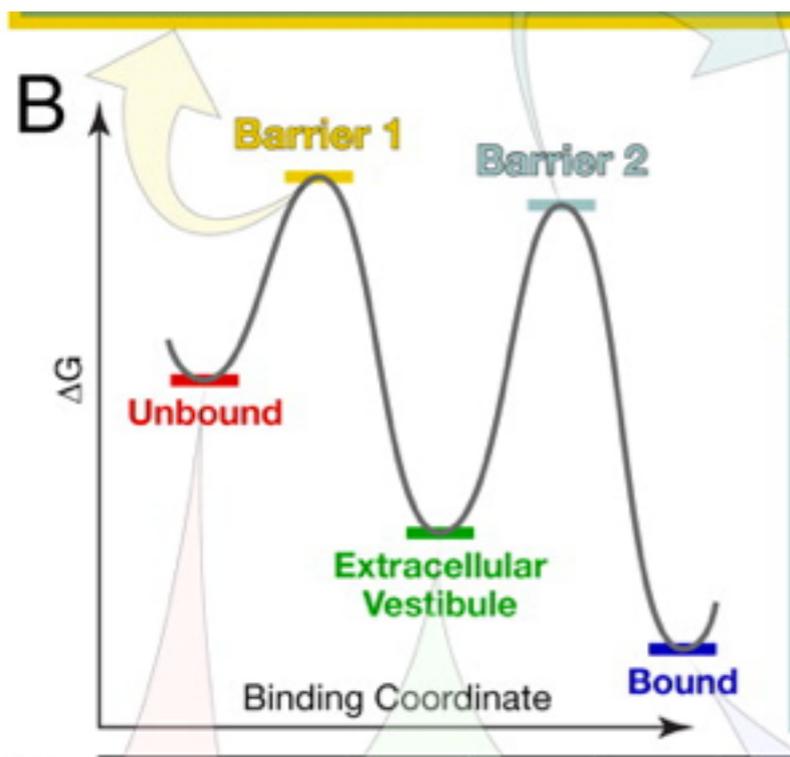
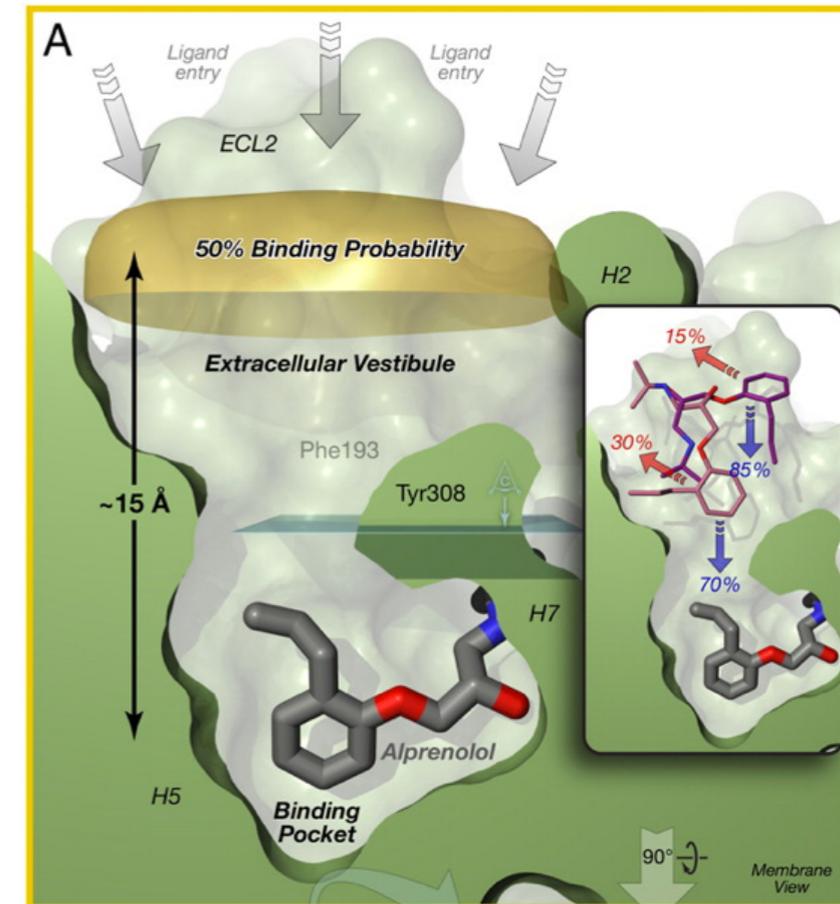
What is the lifetime of a graduate student?

Or, in other words, how long would you like to wait?

We also simulated the binding of dihydroalprenolol to the β_1 -adrenergic receptor (β_1 AR). We performed 82 simulations lasting from 1 to 19 μ s each, resulting in a total of 21 spontaneous binding events (Tables S1 and S2).

Using **Anton**, over 200 μ s of simulation were needed to observe multiple binding events

On a normal supercomputer at even 100 ns/day, would need 5.5 years! (So one grad student per ligand...)



BUT!!! Notice that there are no units?

To get free energy, must observe **UNbinding** as well as **binding**, a process orders of magnitude slower!

Dror,...,Shaw. Pathway and mechanism of drug binding to G-protein-coupled receptors. (2011) *PNAS* 108:13118-123.

How can we speed up the process?

Why don't we just **heat*** up the system?

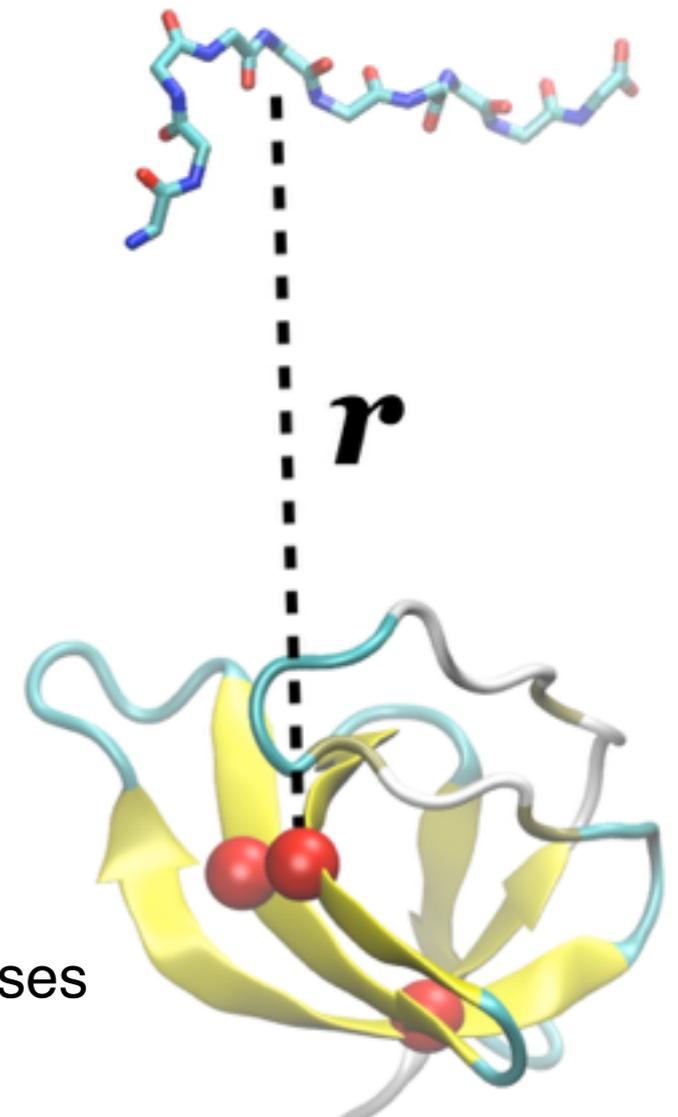


≠



We are not sampling the correct thermodynamic ensemble if we just unnaturally perturb the system!

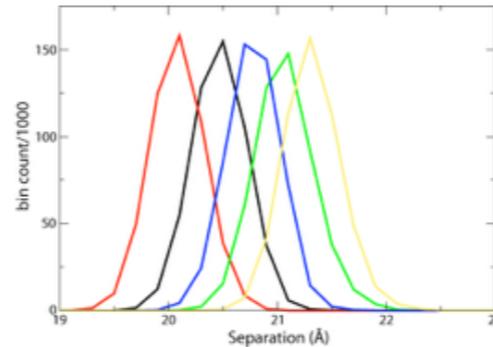
We must **enhance** the sampling along a particular coordinate or coordinates (*can be collective*) in such a way that we can **recover** the correct thermodynamics



*temperature replica exchange MD works on this principle, but one either uses only a low temperature replica for analysis *OR* unbiases higher *T* replicas

Methods for calculating free energies

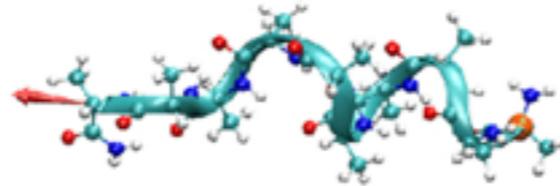
(1) histogram-based methods
(e.g., umbrella sampling,
metadynamics)



$$\Delta A_i = -\frac{1}{\beta} \ln P'_i(\xi) - w_i(\xi) + F_i$$

biased potentials

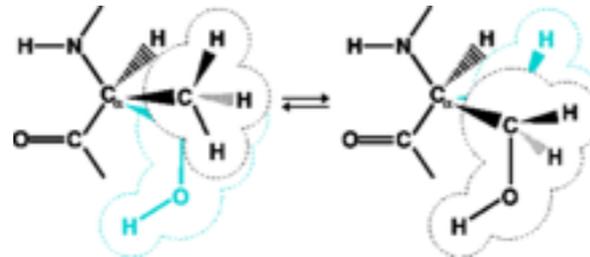
(2) non-equilibrium work (e.g.,
steered MD)



$$\exp(-\beta \Delta A) = \langle \exp(-\beta w) \rangle$$

biased coordinates

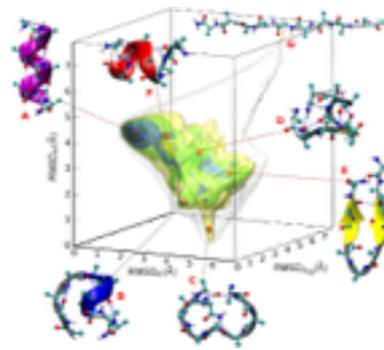
(3) alchemical transformations
(e.g., free energy perturbation)



$$\exp(-\beta \Delta A) = \langle \exp(-\beta \Delta U) \rangle_0$$

biased paths

(4) gradient-based methods (e.g.,
Adaptive Biasing
Forces, thermodynamic integration)



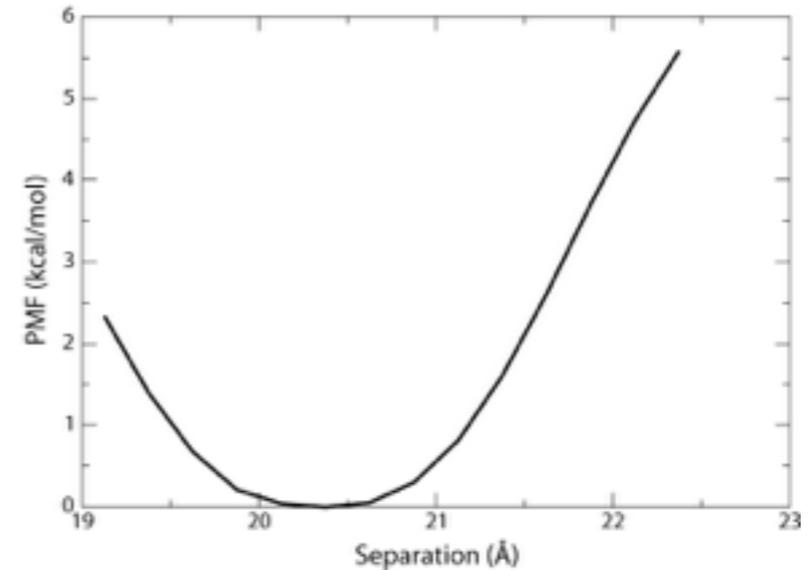
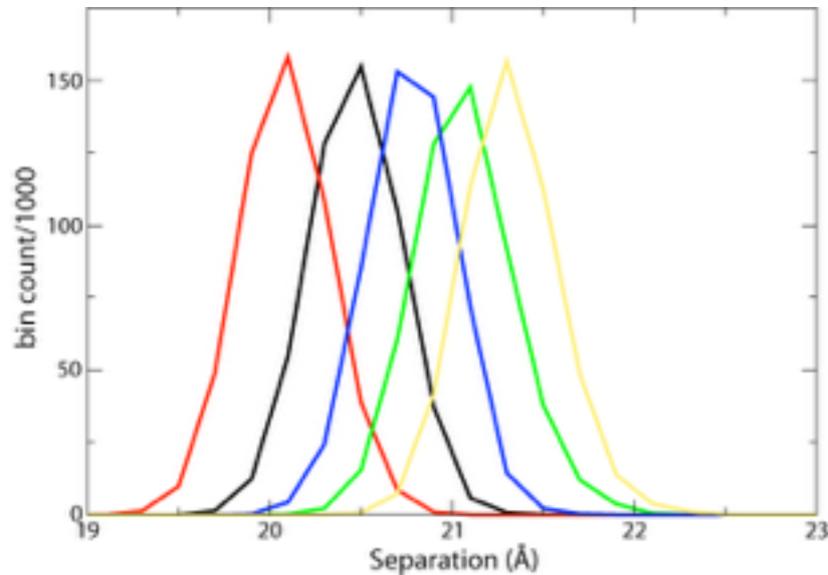
$$\frac{dA(\xi)}{d\xi} = \left\langle \frac{\partial U}{\partial \xi} - \frac{1}{\beta} \frac{\partial \ln |J|}{\partial \xi} \right\rangle_{\xi}$$

biased forces

Method 1: Umbrella sampling (US)

Apply restraining potential (bias) on reaction coordinate (RC) for a series of closely spaced windows

$$w_i(\xi) = \frac{1}{2}K(\xi - \xi_i)^2$$

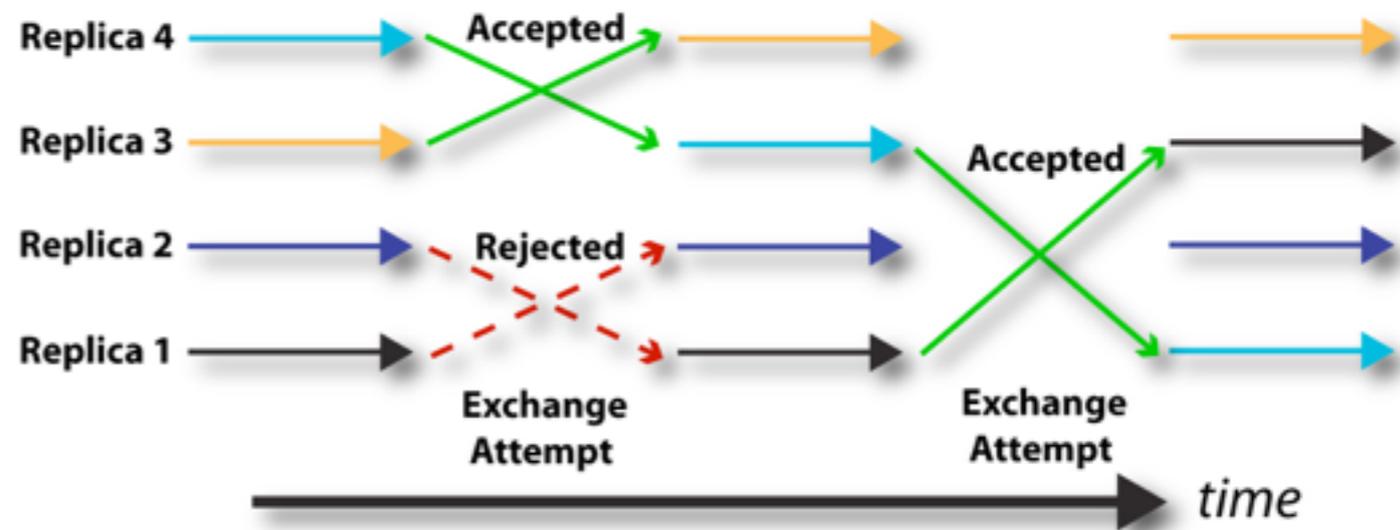


track fluctuations, compute histograms, and unbias/
combine with **WHAM**

$$\Delta A_i = -\frac{1}{\beta} \ln P'_i(\xi) - w_i(\xi) + F_i$$

Grossfield. "WHAM: the weighted histogram analysis method", <http://membrane.urmc.rochester.edu/content/wham>

replica exchange US: exchanging coordinates periodically between different windows to get around barriers in the RC



Method 2: Steered molecular dynamics (SMD)

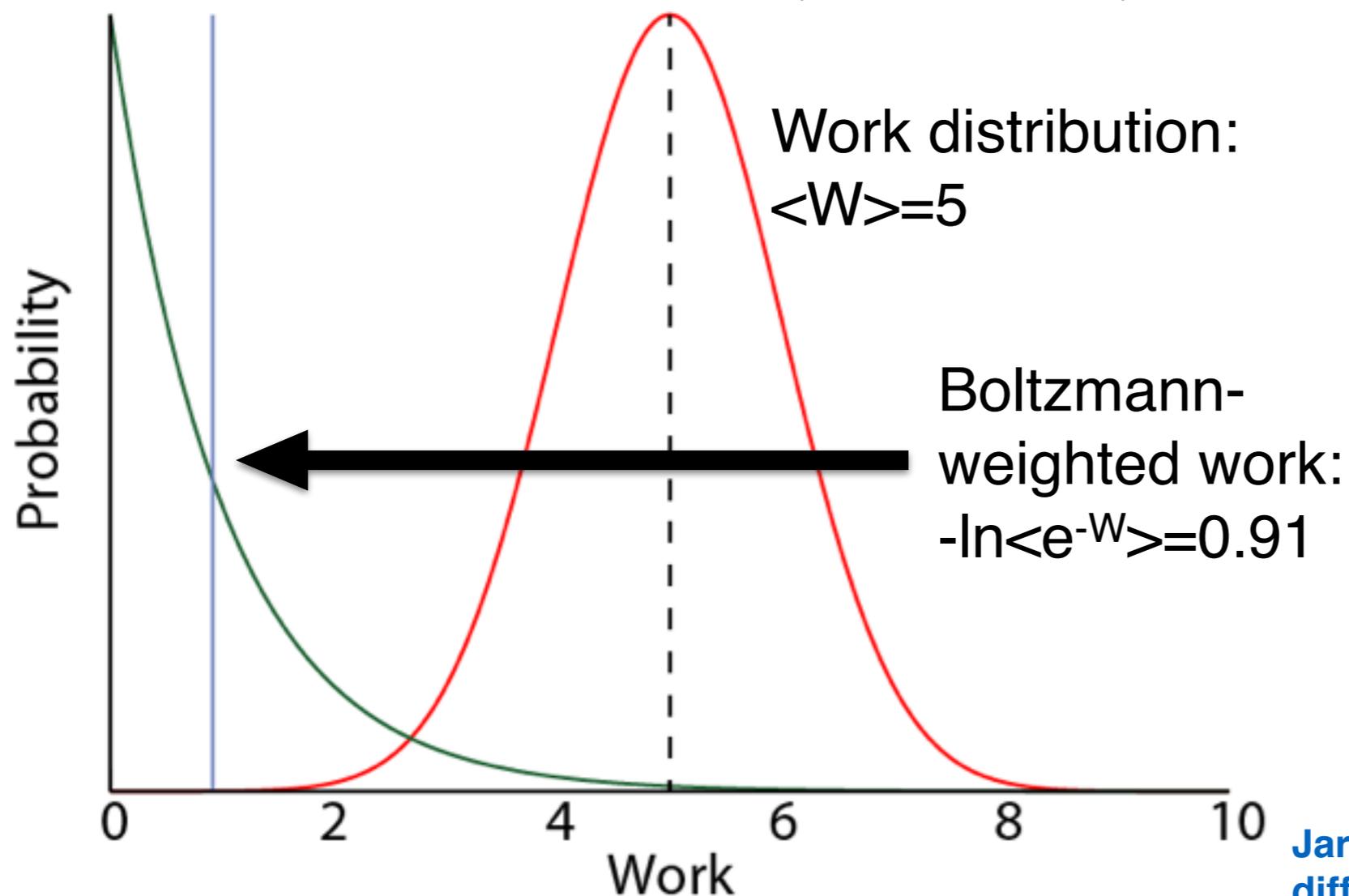
Apply a force to our coordinate of interest to **steer** it in the direction we want it to go

$$\Delta A \leq \langle W \rangle$$

*there is almost always some amount of work wasted**

**except when there is not!*

$$e^{-\beta \Delta A} = \langle e^{-\beta W} \rangle \quad \text{Jarzynski equality}$$



Non-equilibrium methods for free energies are often not practical as they require **significant sampling** to capture **rare events** with spontaneous $W < \Delta A$

Method 3: Free-energy perturbation (FEP)

Each state (e.g., bound ligand vs. unbound) is represented by its own Hamiltonian

Change in free-energy

now expressed as an average in **state 0**

Assume kinetic energy components of H s cancel (no change in mass!)

the perturbation

$$H_1(\mathbf{r}, \mathbf{p}) = H_0(\mathbf{r}, \mathbf{p}) + \Delta H(\mathbf{r}, \mathbf{p})$$

$$\Delta A = -\frac{1}{\beta} \ln\left(\frac{Z_1}{Z_0}\right)$$



an exercise for the reader!

$$\Delta A = -\frac{1}{\beta} \ln \langle \exp[-\beta \Delta H(\mathbf{r}, \mathbf{p})] \rangle_0$$



$$\Delta A = -\frac{1}{\beta} \ln \langle \exp[-\beta \Delta U(\mathbf{r})] \rangle_0$$

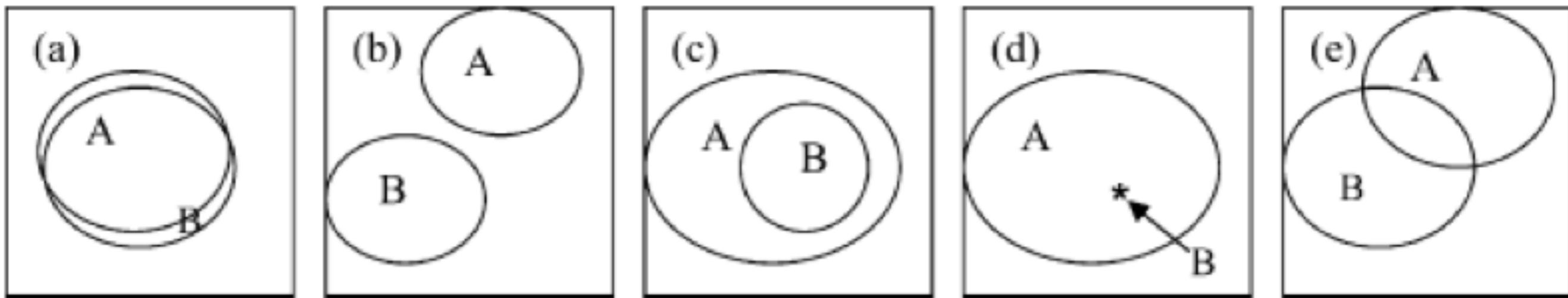
perturbations need to be small; requires large number of intermediate states

Method 3: Free-energy perturbation (FEP)

$$\Delta A = -\frac{1}{\beta} \ln \langle \exp[-\beta \Delta U(\mathbf{r})] \rangle_0$$

Formally valid for all cases,
but not **practically** so

perturbation approach only converges to the correct answer if there is a strong overlap of phase space $((\mathbf{x}, \mathbf{p})$ values accessible with given H) in states **0** and **1**, i.e, if the difference between **0** and **1** is small.



efficient

inefficient

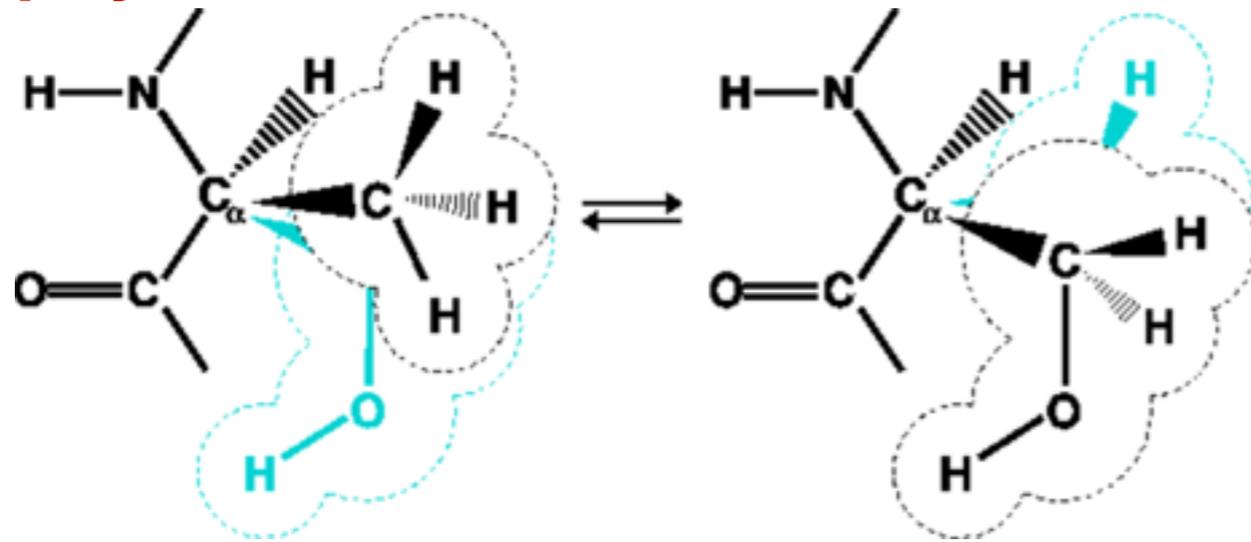
efficient

inefficient

inefficient

Method 3: Free-energy perturbation (FEP)

in practice, FEP obtains the free energy change by going through a series of **unphysical states**

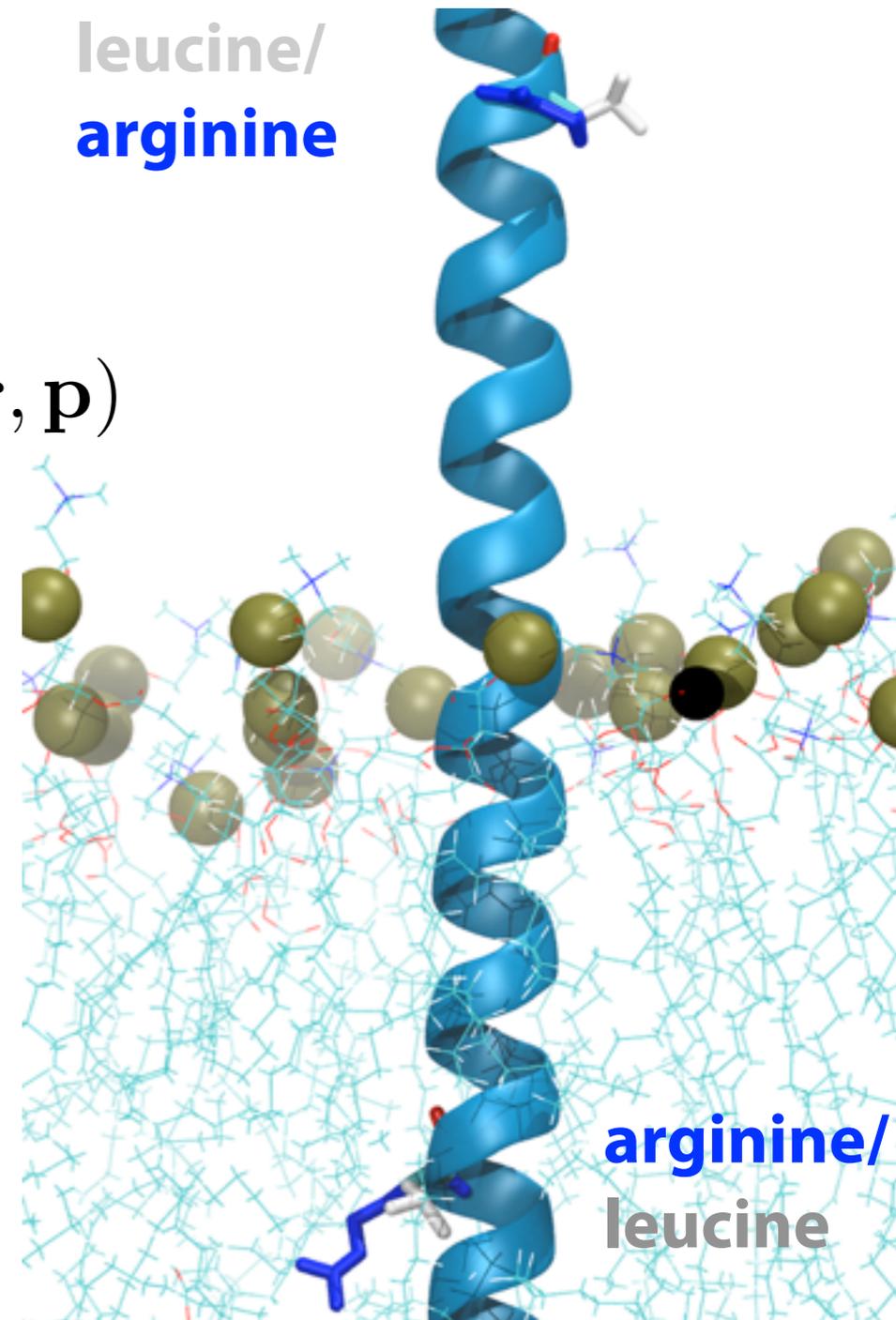


$$H(\mathbf{r}, \mathbf{p}) = H_0(\mathbf{r}, \mathbf{p}) + \lambda H_b(\mathbf{r}, \mathbf{p}) + (1 - \lambda) H_a(\mathbf{r}, \mathbf{p})$$

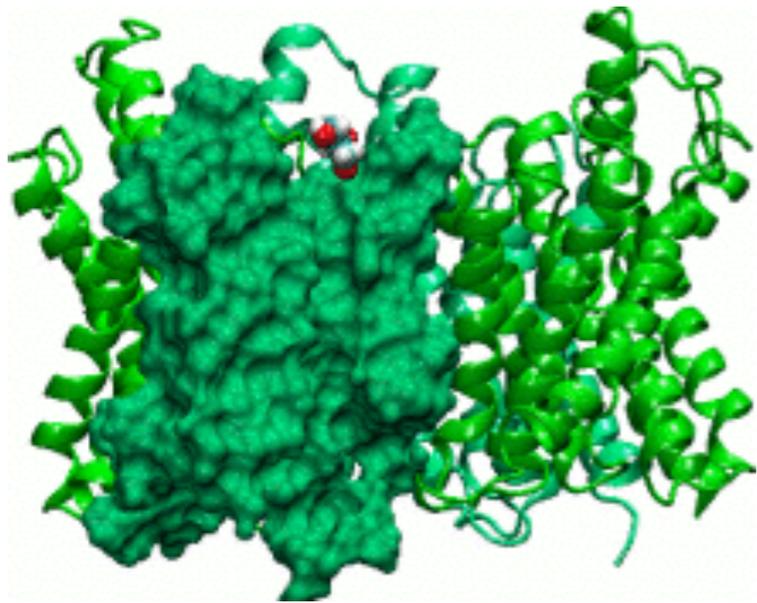
-interactions with the environment are scaled by λ (a coupling parameter) and $(1 - \lambda)$, respectively

Example: both **arginine** and **leucine** sidechains are at the same position on a **polyLeu helix** but don't interact (*dual topology approach*)

leucine/
arginine



Method 4: Adaptive biasing forces (ABF)



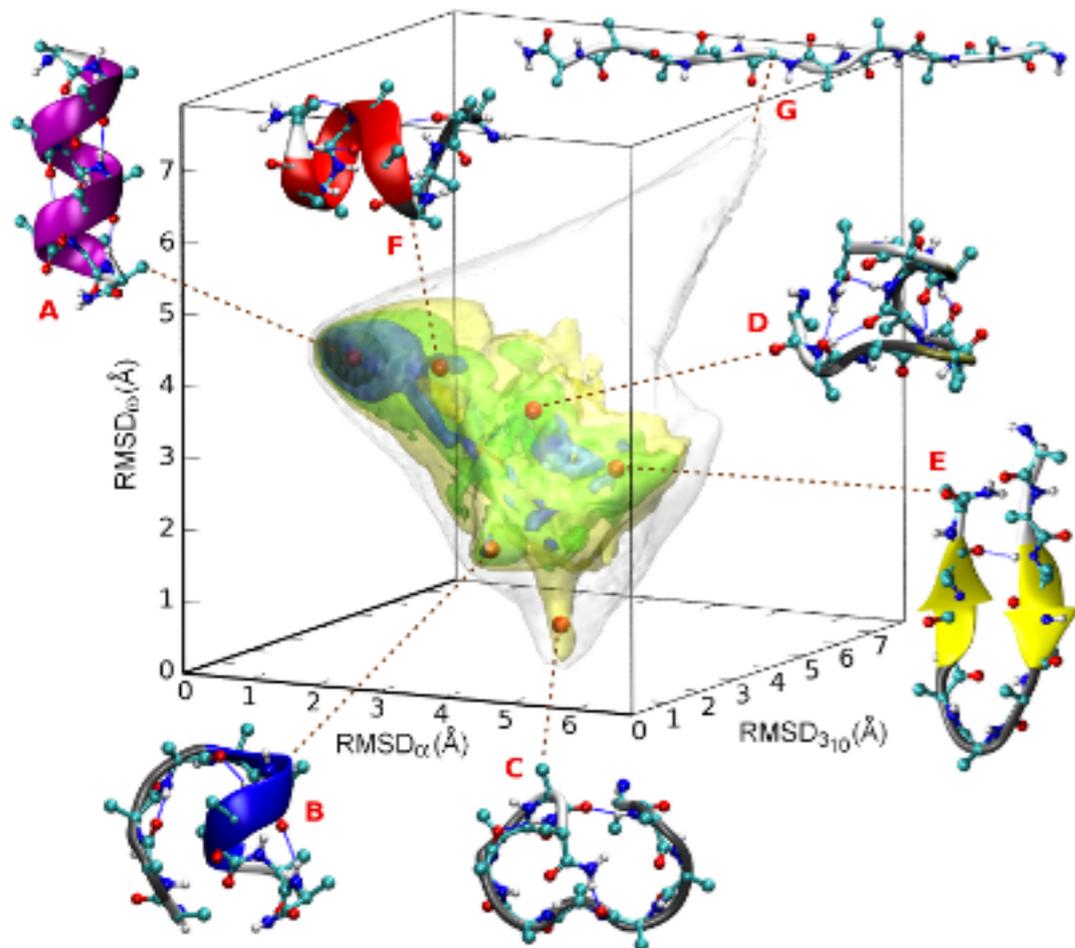
$$\Delta A(\xi) = -\frac{1}{\beta} \ln P(\xi)$$

Free energy as
function of ξ

$$\nabla_{\xi} A(\xi) = \langle -F_{\xi} \rangle_{\xi}$$
 Relation to average force

<http://www.edam.uhp-nancy.fr/ABF/>

$$\mathbf{F}^{\text{ABF}} = \nabla_x \tilde{A} = -\langle F_{\xi} \rangle_{\xi} \nabla_x \xi$$
 Compute average force
adaptively and apply
biasing force to cancel it



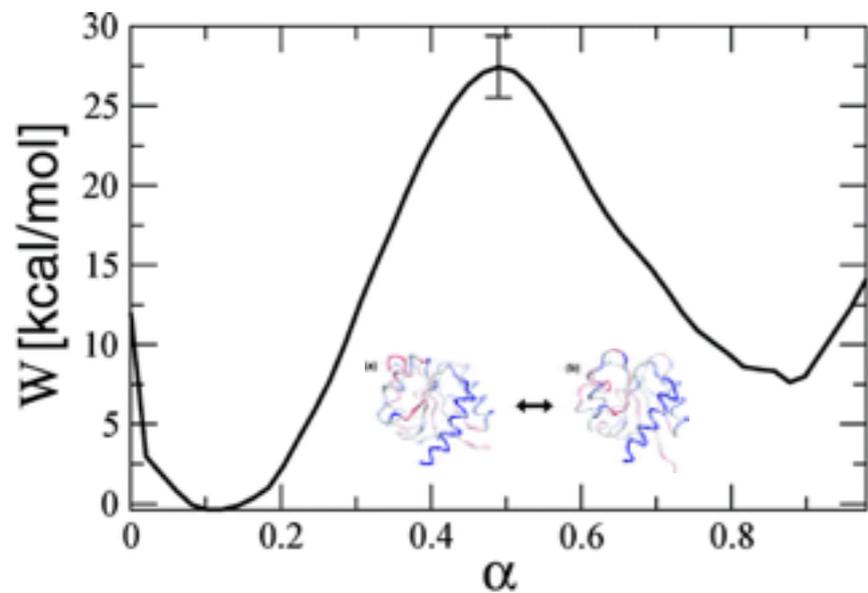
As the estimate of the PMF improves, the biasing forces *should* effectively cancel it, permitting the reaction coordinate to diffuse more easily

Hénin, Fiorin, Chipot, Klein, Exploring Multidimensional Free Energy Landscapes Using Time-Dependent Biases on Collective Variables. (2010) *J. Chem. Theory Comput.*, 6:35-47.

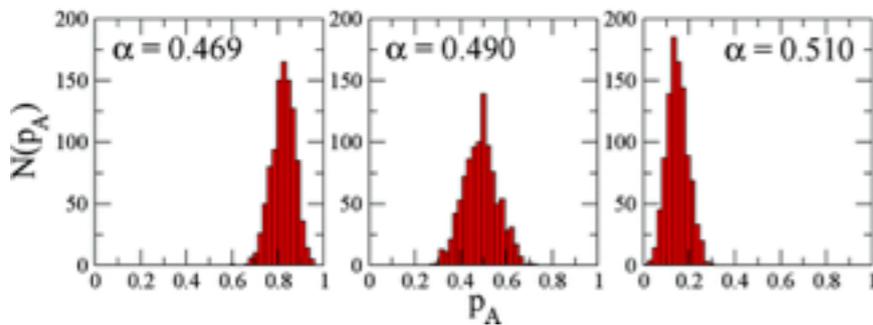
Choosing a *good* Reaction Coordinate

Attempting to reduce a curvilinear path in $3N$ -dimensional space to a **1D path**

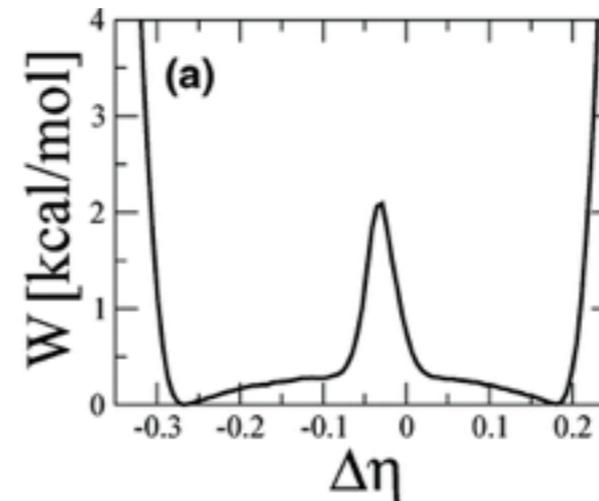
this 1D path is known as the **committer probability**; while it exists, it is typically completely non-intuitive — prefer to choose a physically meaningful coordinate that is close to the “true” one



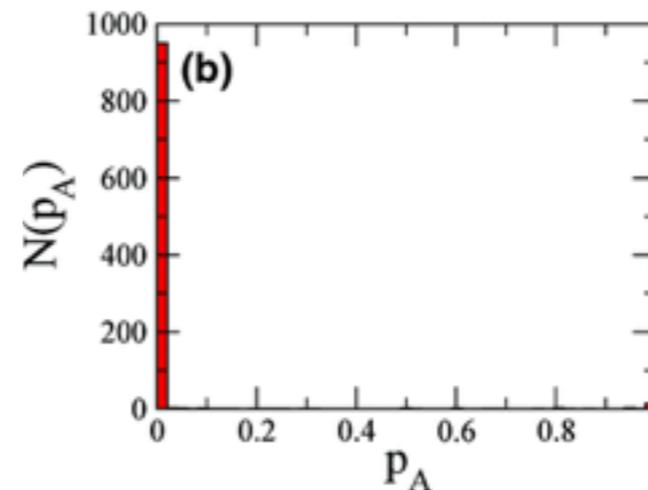
Free energy along a **good** transition coordinate (determined via *string method*)



committer probability is max around **0.5** for the transition state



Free energy along a **bad** transition coordinate (fraction of native contacts)



committer probability is all at 0, meaning at the free-energy peak, all configurations fall back to only one min.

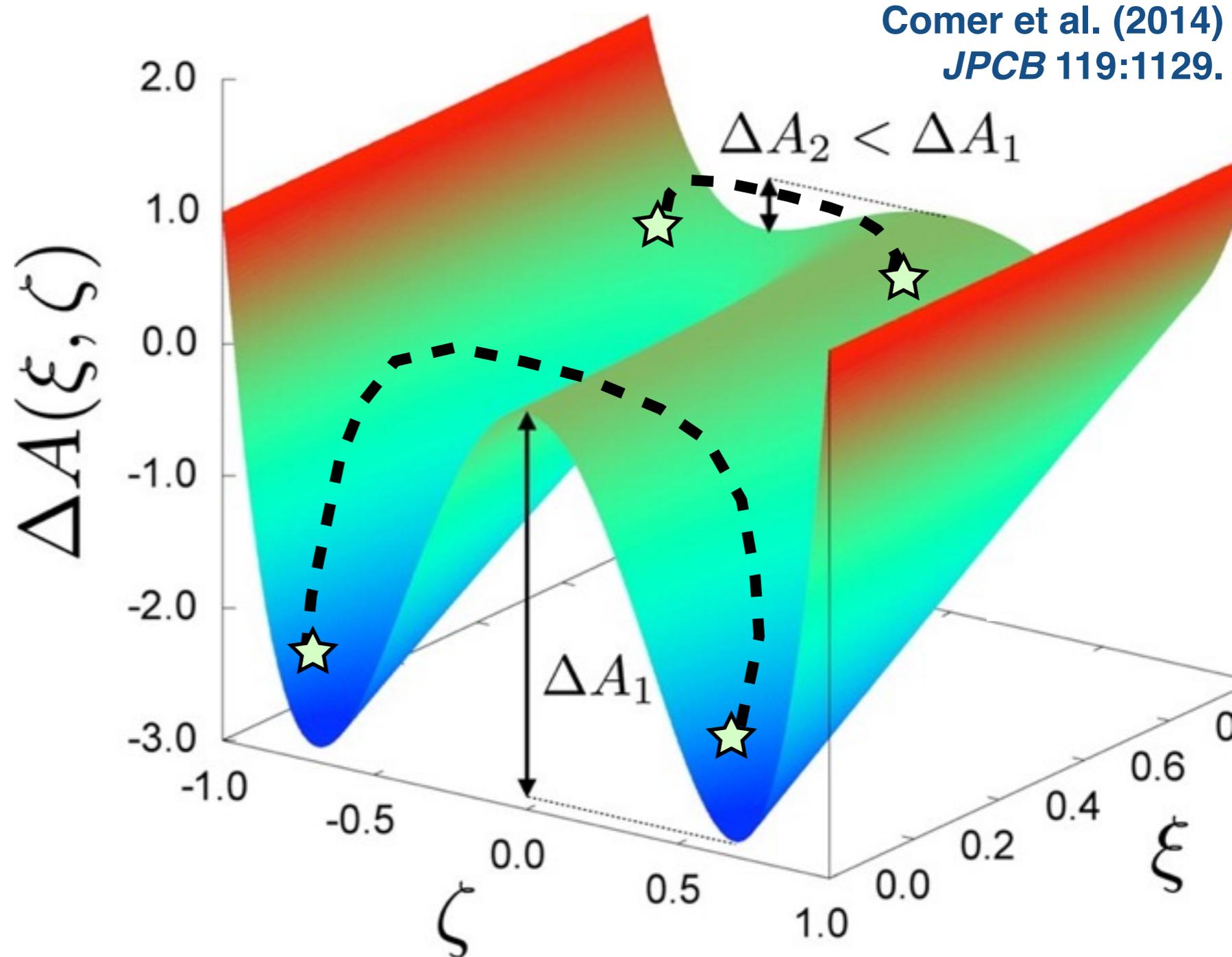
Orthogonal degrees of freedom

Because only one (or a few) coordinate(s) is biased in the enhanced sampling simulation, others may yet be slow to evolve

Example:

if ξ is restrained at **0.0** using **US**, a large barrier prevents it from fully sampling the orthogonal ζ

however, if ξ were free to diffuse, the system could take an alternate, lower energy path to reach **state 2** from **state 1** faster



ALL methods suffer from a version of this sampling difficulty!

Two forms of error and its estimation

1) statistical error (the **known** unknowns)

every free-energy method has ways of estimating this, although it is typically small and can be reduced through increased sampling

Ex: variance in free energy as a function of variance in each window for Umbrella Sampling simulation

$$\text{var}[G(\xi)] \approx (K \Delta\xi)^2 \cdot \sum_{i=1}^{(\xi - \xi_0) / \Delta\xi} \text{var}(\bar{\xi}_i)$$

Zhu, Hummer. Convergence and error estimation in free energy calculations using the weighted histogram analysis method. (2012) *J. Comput. Chem.* 33:453–465.

$$\sigma_{\Delta A}^2 = \frac{1}{N_0 \beta^2} \left[\frac{\langle f^2(x) \rangle_0}{\langle f(x) \rangle_0^2} - 1 \right] + \frac{1}{N_1 \beta^2} \left[\frac{\langle f^2(-x) \rangle_1}{\langle f(-x) \rangle_1^2} - 1 \right] \quad \begin{aligned} x &= \beta(\Delta U - C) \\ f(x) &= \frac{1}{1 + \exp(x)} \end{aligned}$$

*Ex: variance in stratified FEP simulations using the Bennett Acceptance Ratio (**BAR**) estimator*

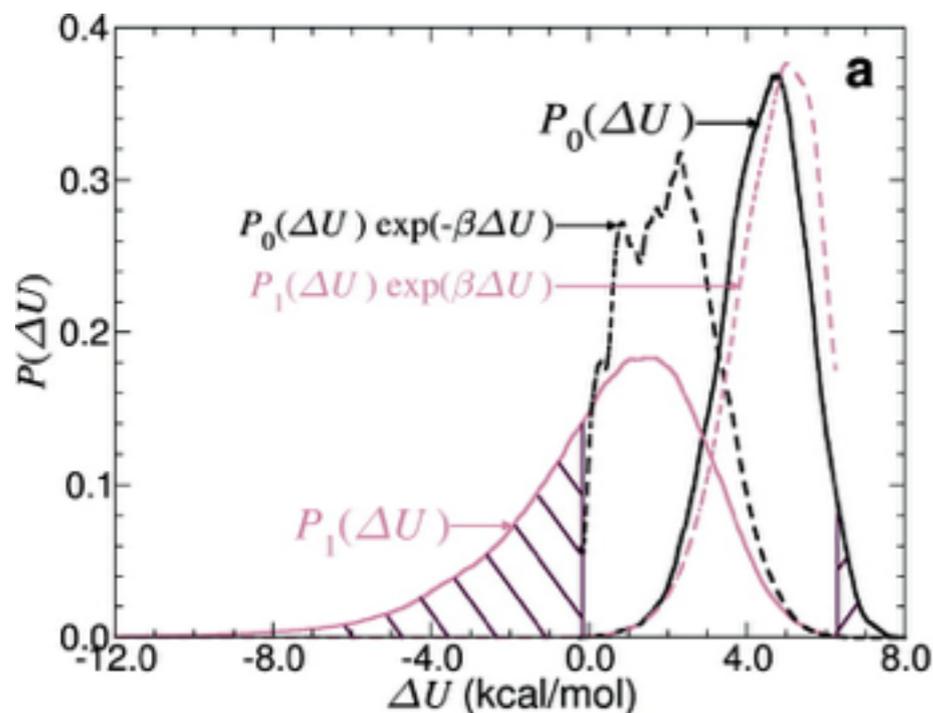
Pohorille, Jarzynski, Chipot. Good practices in free-energy calculations. (2010) *J. Phys. Chem. B.* 114:10235-53.

a common way of estimating statistical error is to use **block averaging**, where the simulation is divided into uncorrelated blocks, the free energy determined for each, and then the **standard error of means** is calculated

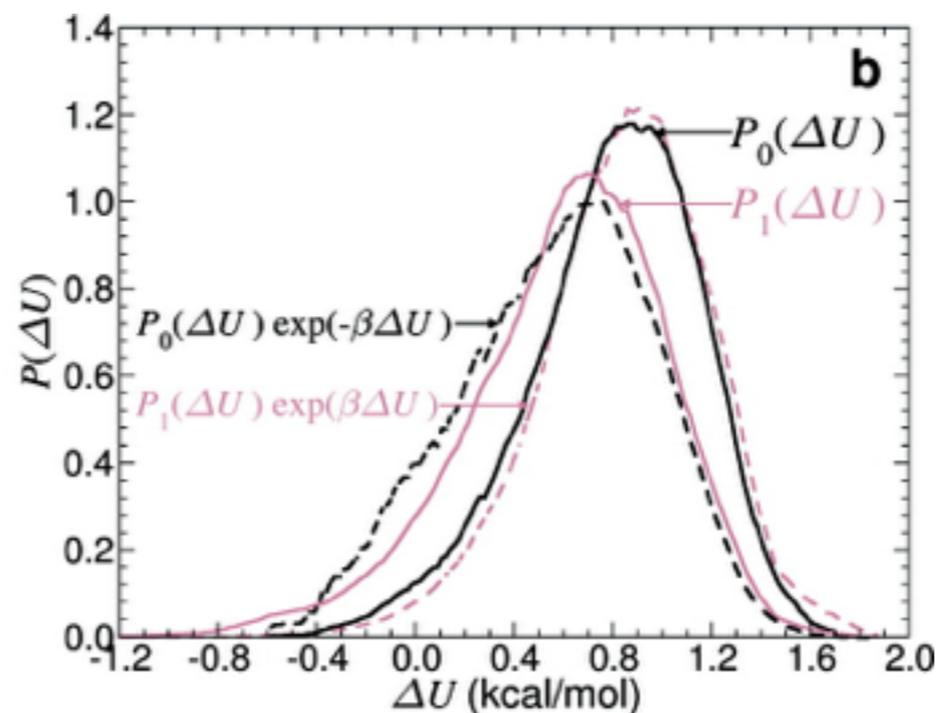
Two forms of error and its estimation

2) systematic error (the **unknown** unknowns)

much more difficult to estimate; sources include force field inaccuracies, poor overlap of neighboring windows, and quasi non-ergodic scenarios (e.g., sampling only one of two metastable states)



poor overlap



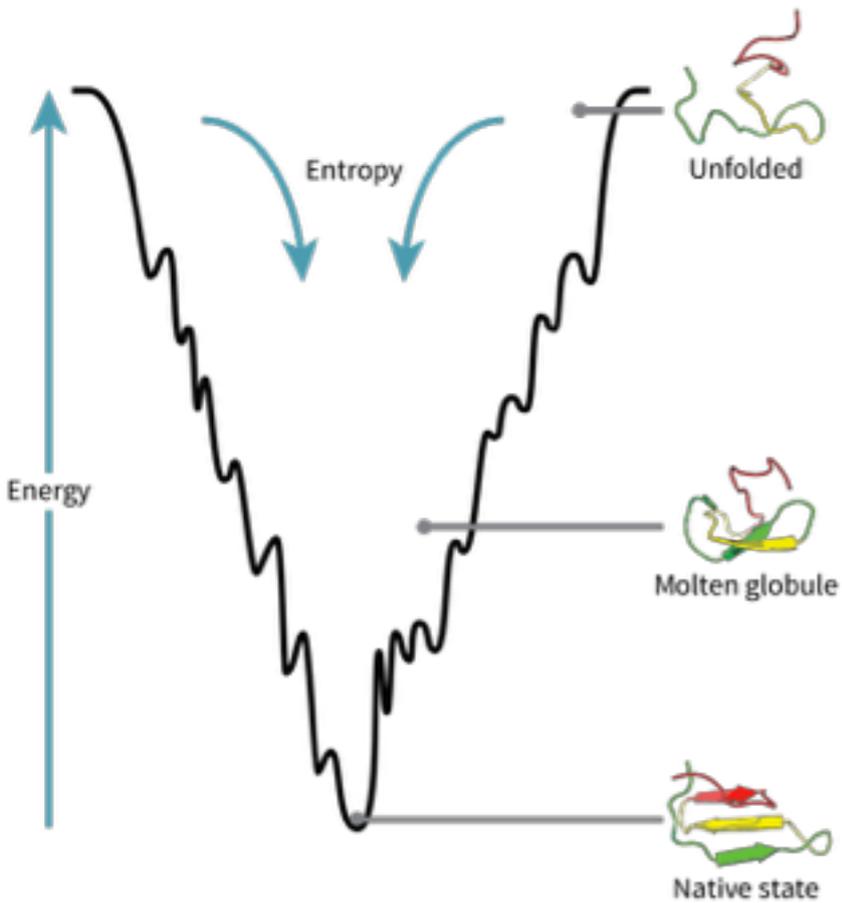
good overlap

*overlap between forward and backward **FEP** runs, **NOT** neighboring windows in one run*

typical ways of estimating this error are to look at overlap between windows (**umbrella sampling, FEP**) or continuity of forces between windows (**ABF**)

Example 1: Protein folding

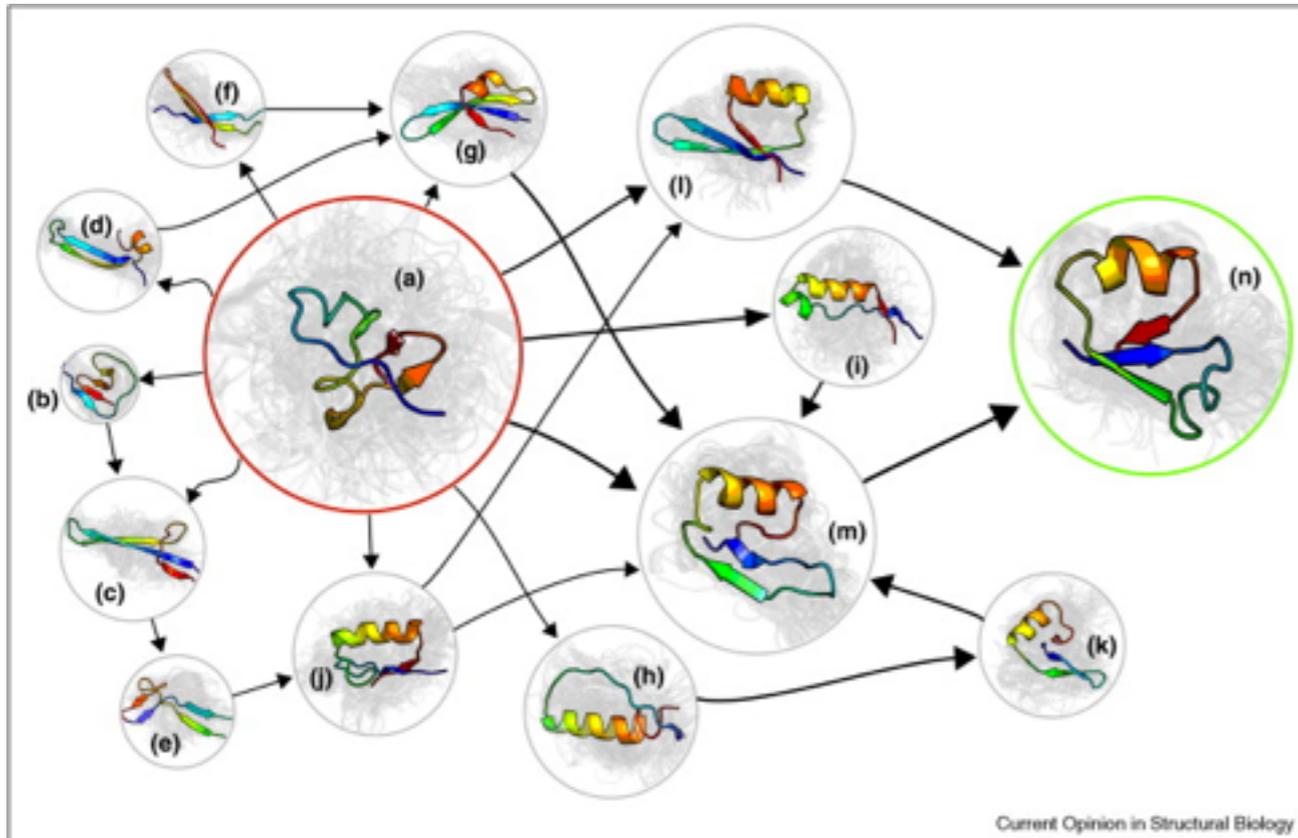
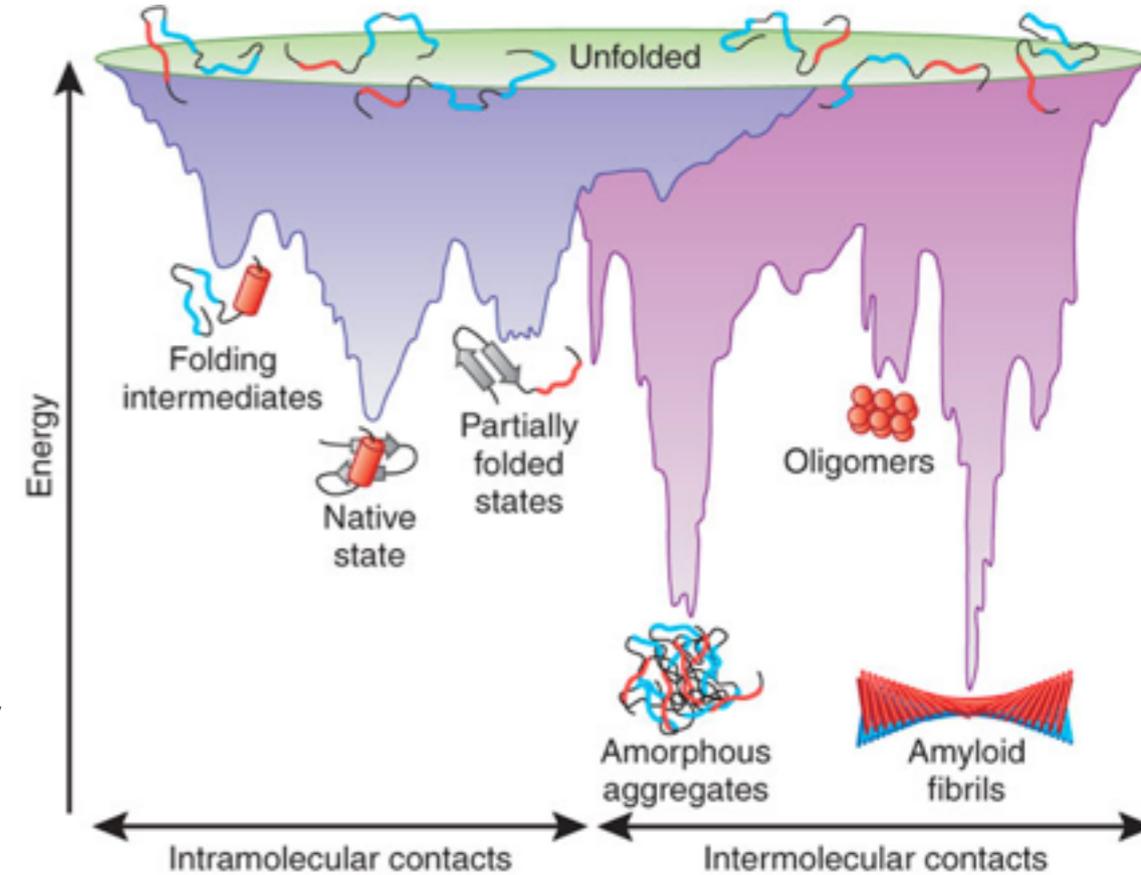
F Ulrich Hartl & Manajit Hayer-Hartl. Converging concepts of protein folding in vitro and in vivo
 Nat. Struct. Mol. Bio. 16, 574 - 581 (2009)



folding funnel

proceeds through a series of intermediate states

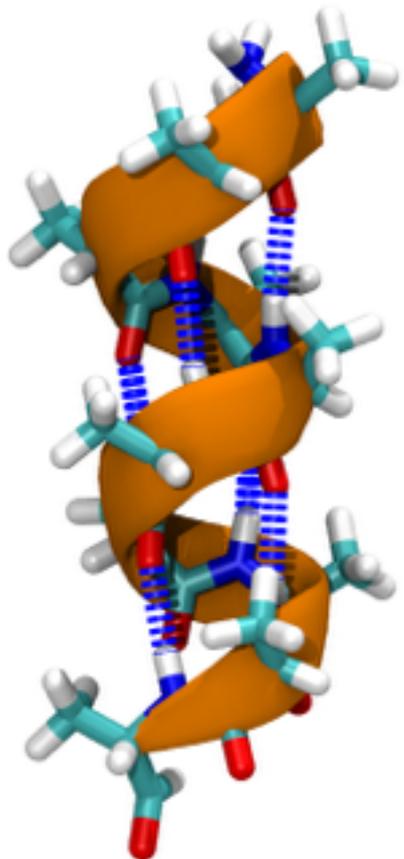
true free-energy landscape is typically much more complex



proteins may sample a number of intermediates *without native-like structure* on the folding pathway - *what is the reaction coordinate?*

Bowman, G. R.; Voelz, V. A.; Pande, V. S. Taming the complexity of protein folding
 Curr. Opin. Struct. Biol. 2011, 21, 4- 11

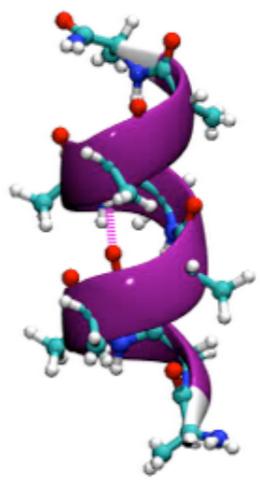
deca-alanine folding



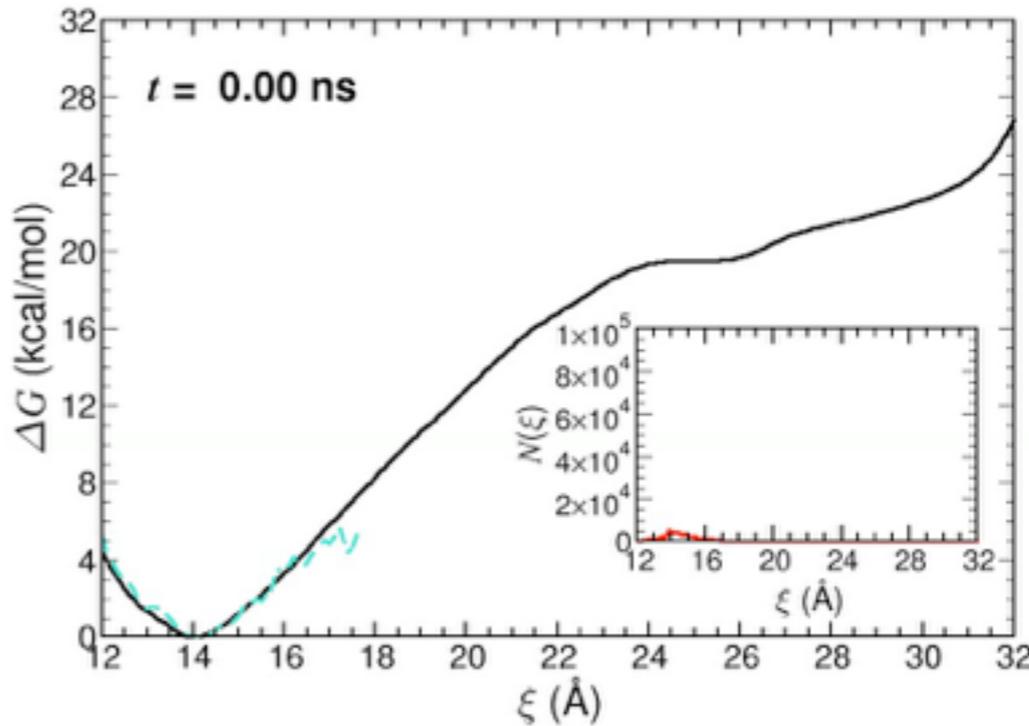
14 Å



32 Å

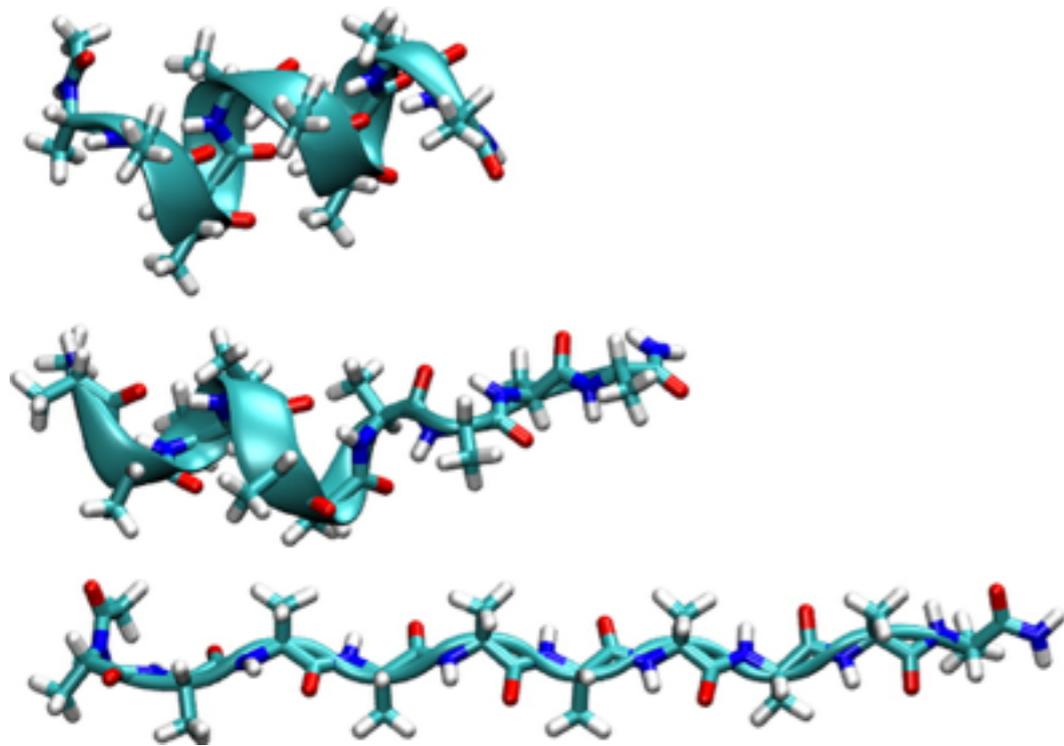


10-Ala helix (in vacuum)
end-to-end distance (ξ)
a common **RC**



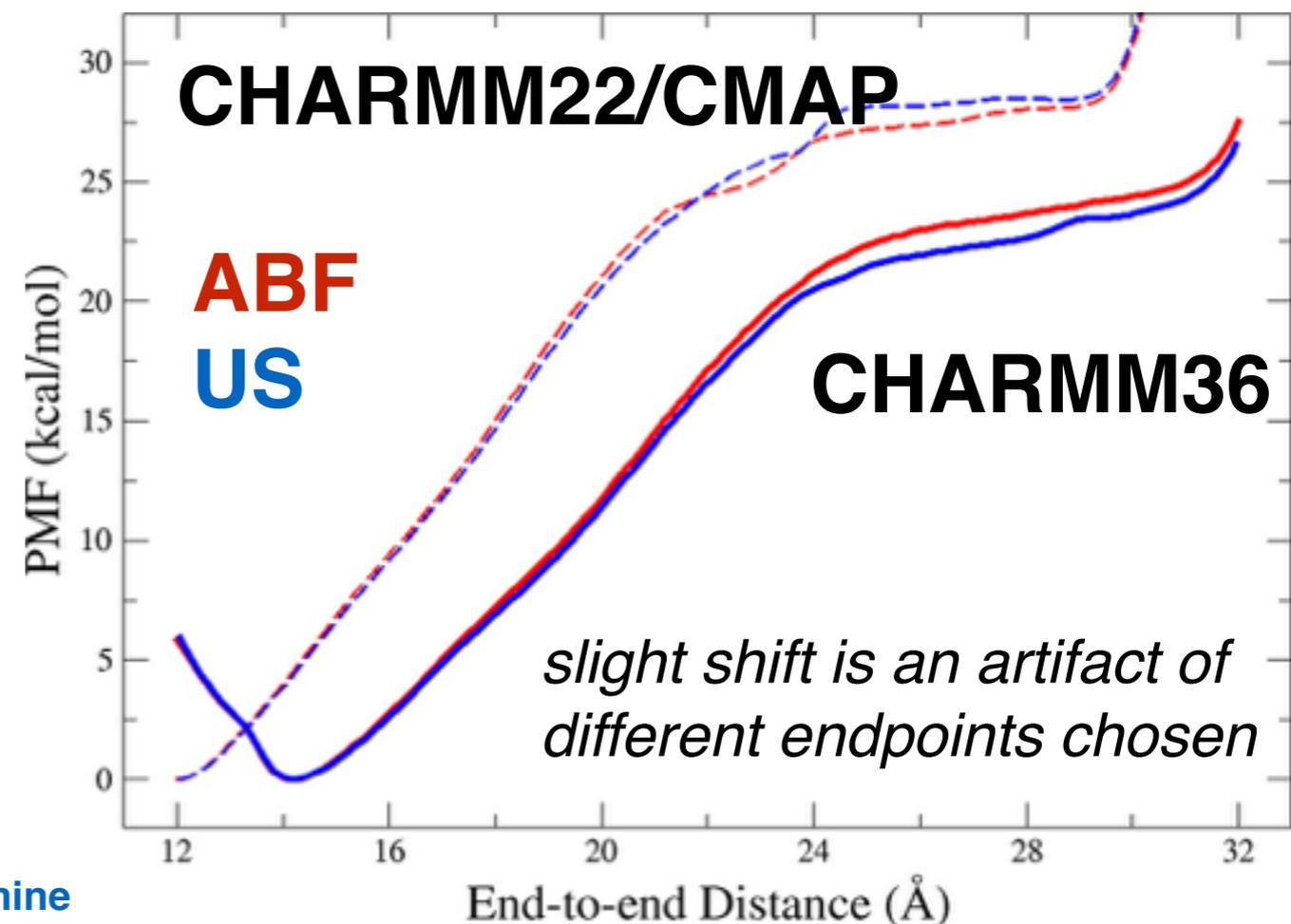
Chipot, Hémin. Exploring the free energy landscape of a short peptide using an average force. (2005) *J. Chem. Phys.* 123:244906.

folding Ala₁₀ in vacuum



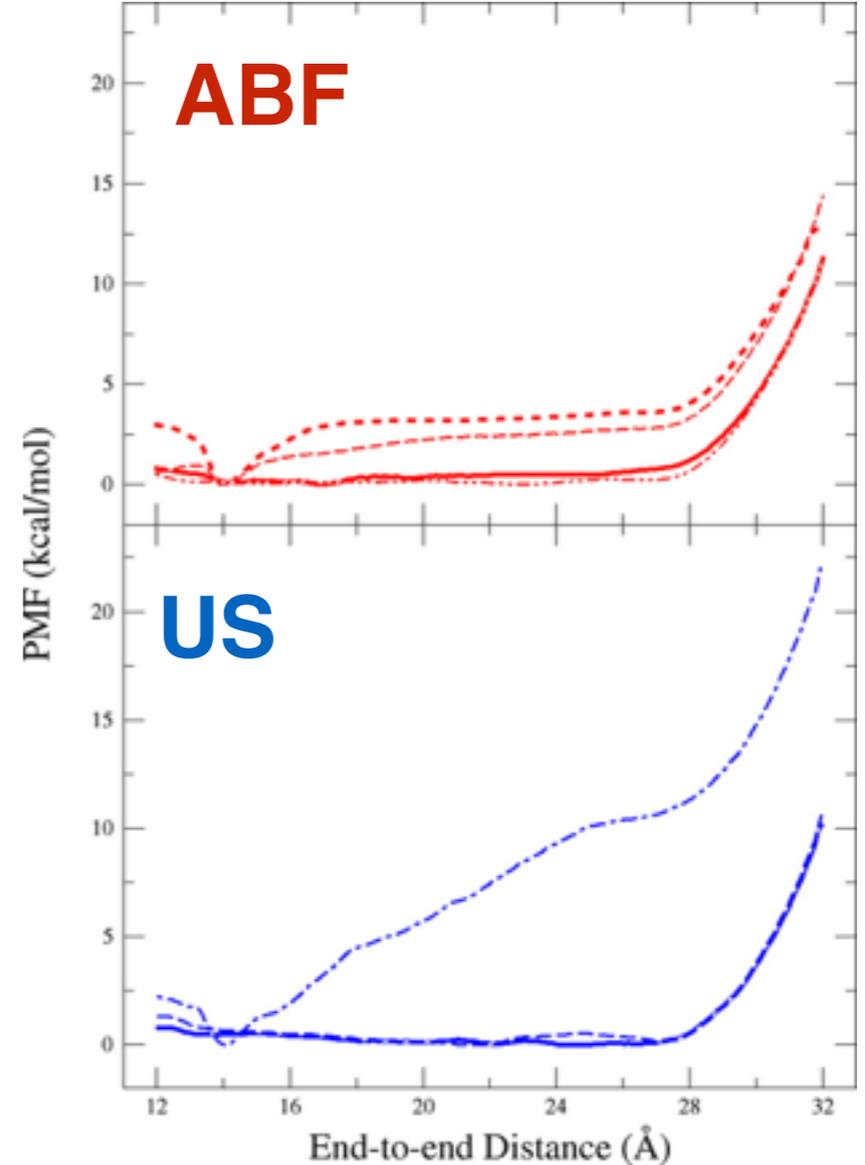
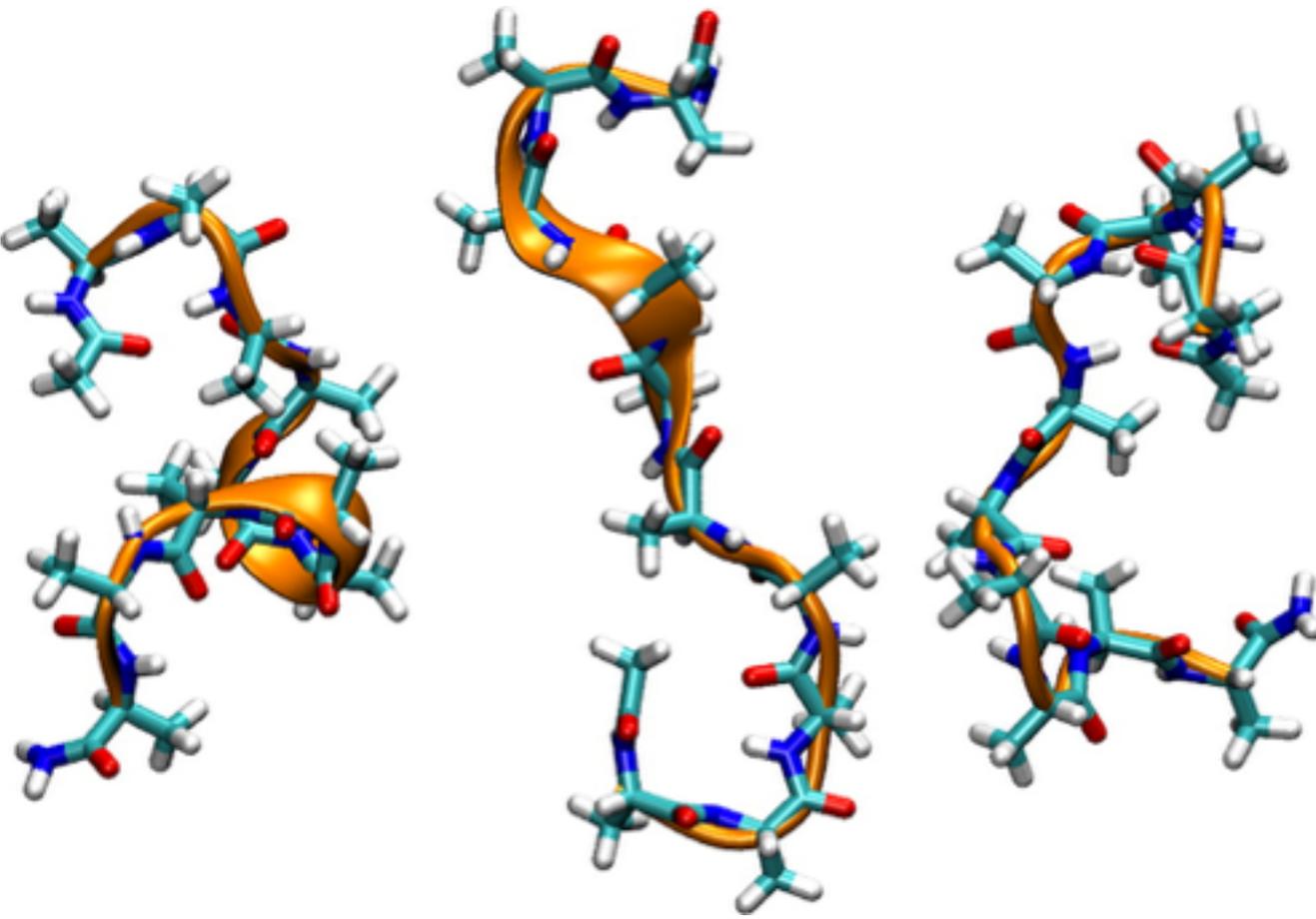
Ala₁₀ folds/unfolds in an accordion-like fashion in vacuum

end-to-end distance works well as an RC for both methods



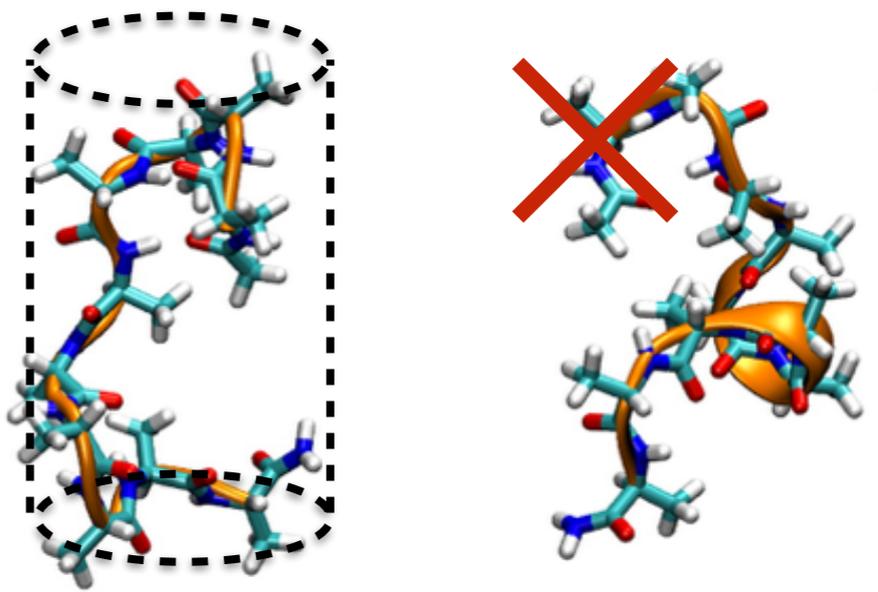
folding Ala₁₀ in water

end-to-end distance no longer works -
PMFs never converge!

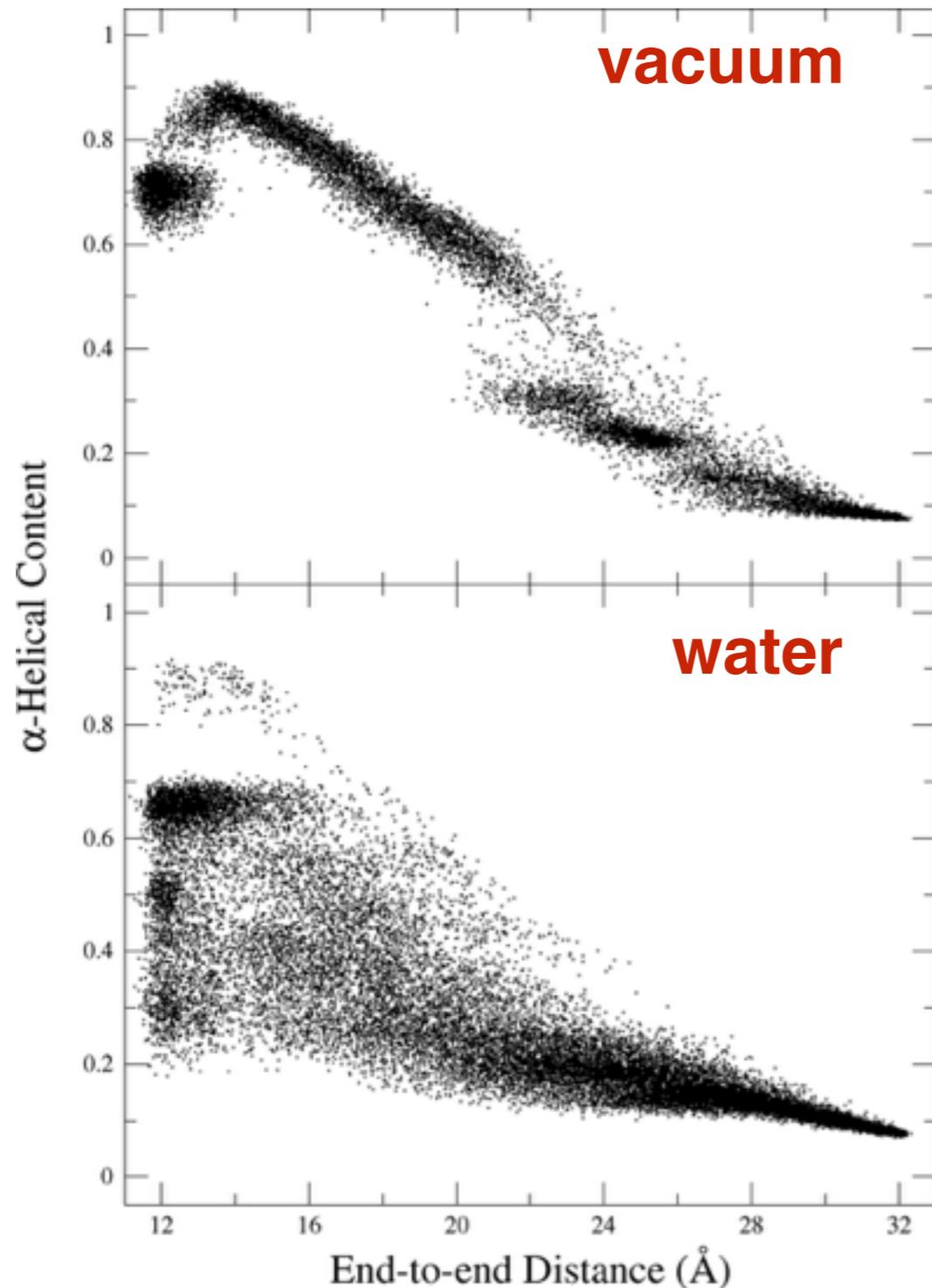


numerous states not observed in vacuum
appear

restraints restricting conformational
freedom tried to no avail



Going to a 2D description



in **vacuum**, alpha helical content and end-to-end distance are practically 1-1

in **water**, a number of compact, low-lying states appear that “contaminate” the 1D PMF (i.e., *are poorly sampled*)

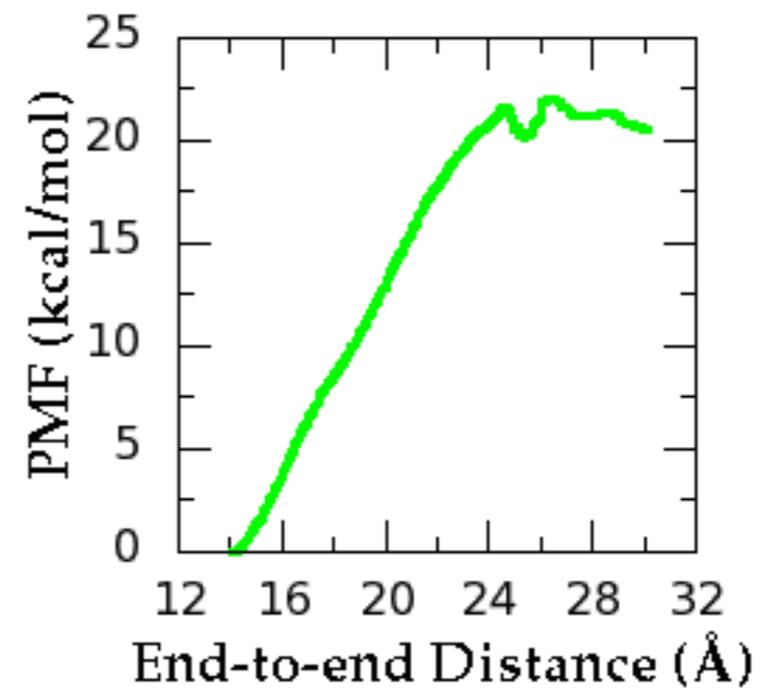
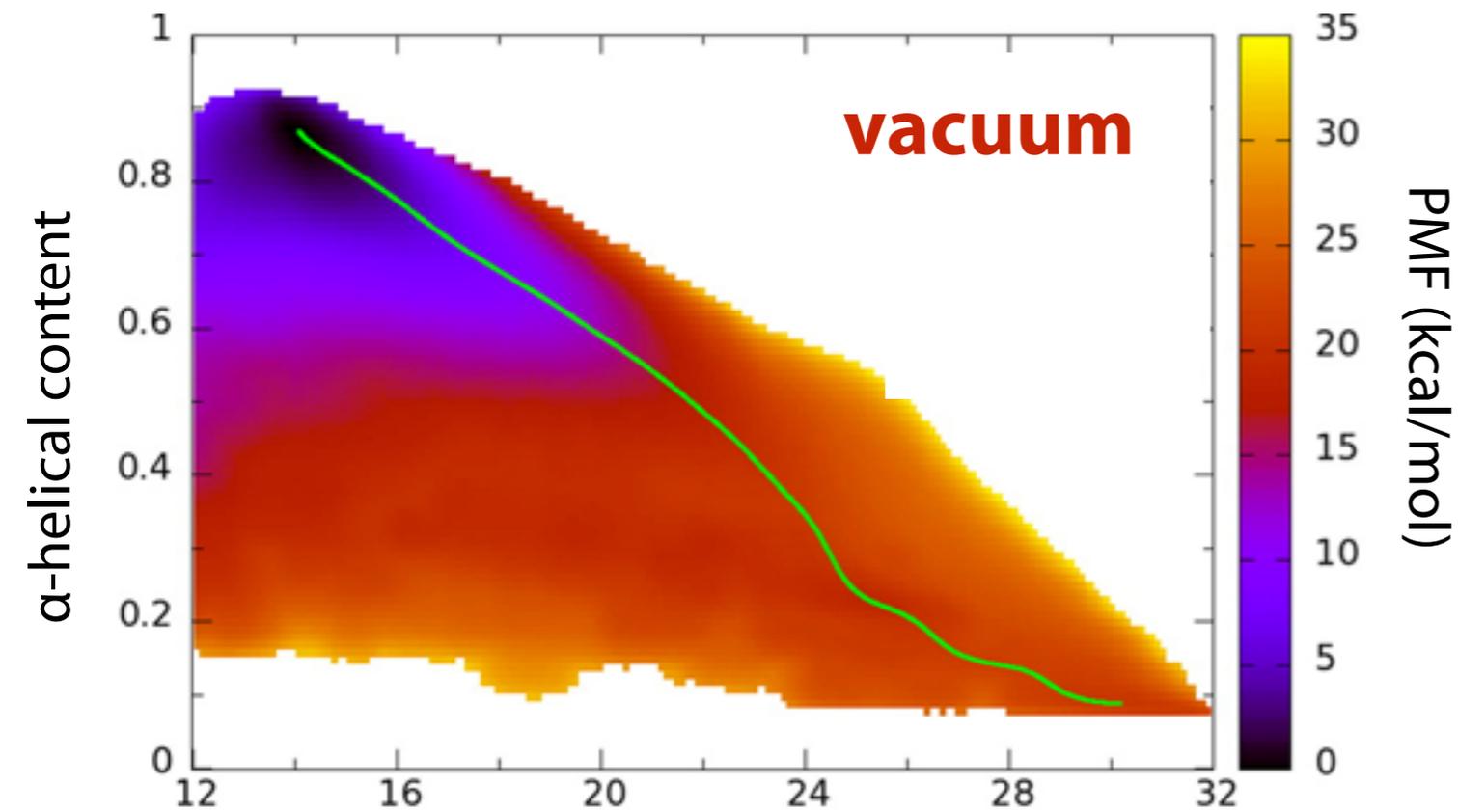
Solution: add a 2nd RC for alpha helicity (α), in addition to ξ

$$\alpha \left(C_{\alpha}^{(N_0)}, O^{(N_0)}, C_{\alpha}^{(N_0+1)}, O^{(N_0+1)}, \dots, N^{(N_0+5)}, C_{\alpha}^{(N_0+5)}, O^{(N_0+5)}, \dots, N^{(N_0+N)}, C_{\alpha}^{(N_0+N)} \right) = \frac{1}{2(N-2)} \sum_{n=N_0}^{N_0+N-2} \text{angf} \left(C_{\alpha}^{(n)}, C_{\alpha}^{(n+1)}, C_{\alpha}^{(n+2)} \right) + \frac{1}{2(N-4)} \sum_{n=N_0}^{N_0+N-4} \text{hbf} \left(O^{(n)}, N^{(n+4)} \right)$$

$$\text{angf} \left(C_{\alpha}^{(n)}, C_{\alpha}^{(n+1)}, C_{\alpha}^{(n+2)} \right) = \frac{1 - \left(\left| \theta \left(C_{\alpha}^{(n)}, C_{\alpha}^{(n+1)}, C_{\alpha}^{(n+2)} \right) - \theta_0 \right| / \Delta\theta_{tol} \right)^2}{1 - \left(\left| \theta \left(C_{\alpha}^{(n)}, C_{\alpha}^{(n+1)}, C_{\alpha}^{(n+2)} \right) - \theta_0 \right| / \Delta\theta_{tol} \right)^4}$$

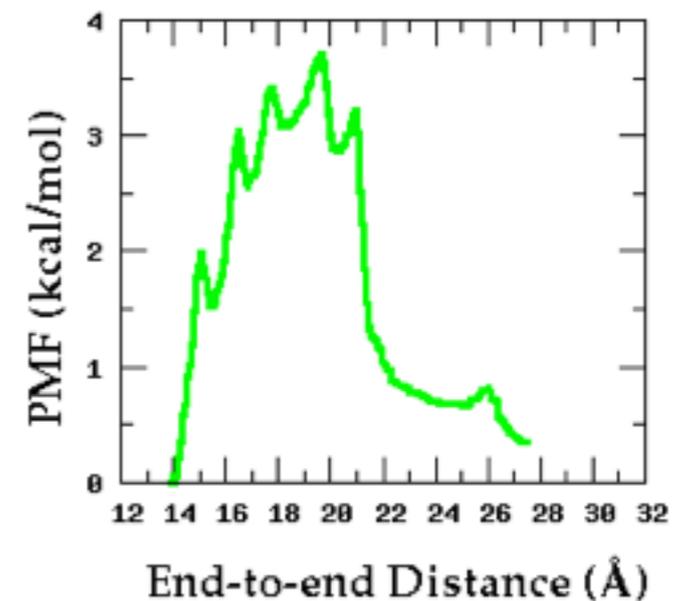
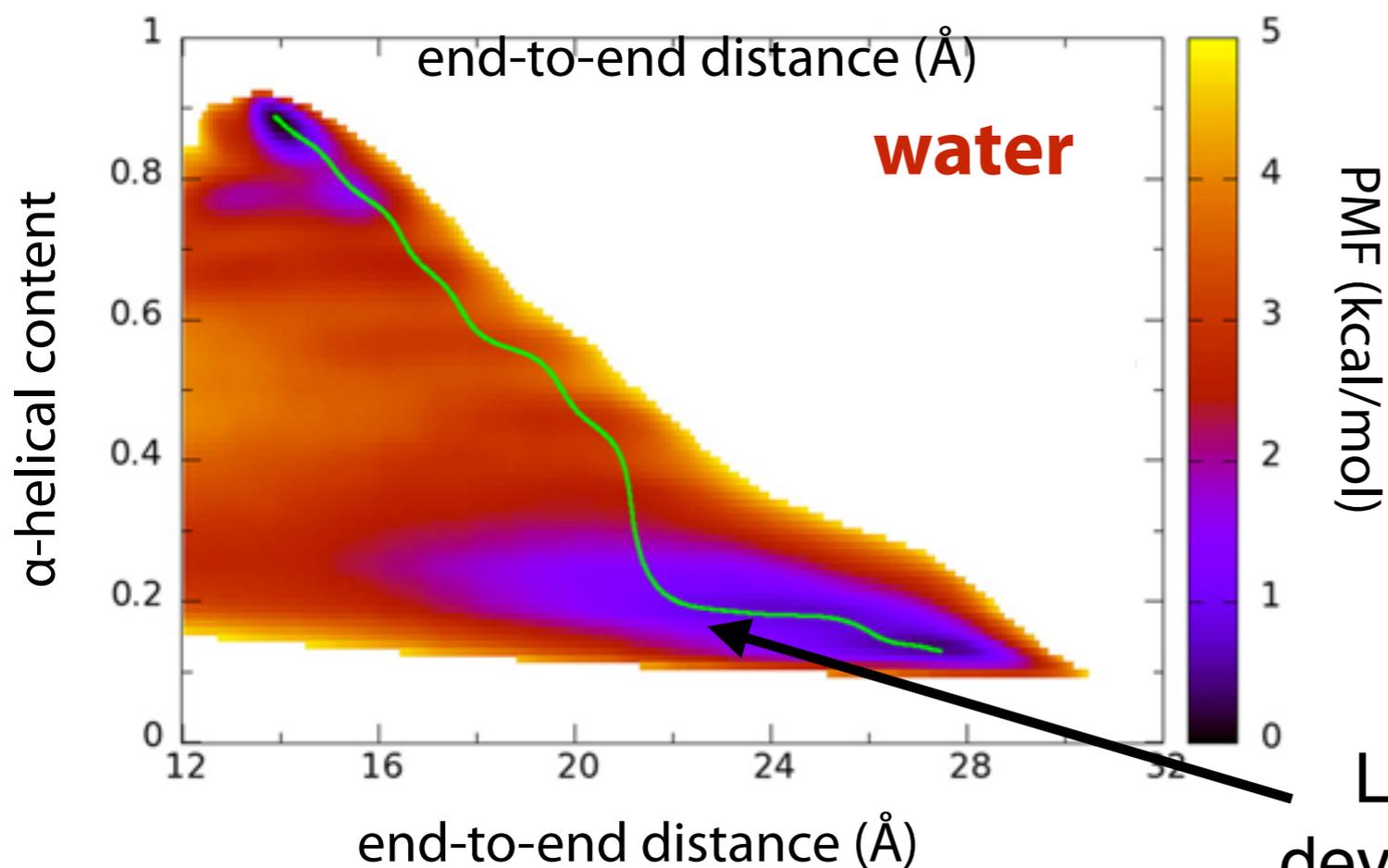
$$\text{hbf} \left(O^{(n)}, N^{(n+4)} \right) = \frac{1 - \left(\left| \mathbf{x}_{O^{(n)}} - \mathbf{x}_{N^{(n+4)}} \right| / d_0 \right)^n}{1 - \left(\left| \mathbf{x}_{O^{(n)}} - \mathbf{x}_{N^{(n+4)}} \right| / d_0 \right)^m}$$

Going to a 2D description



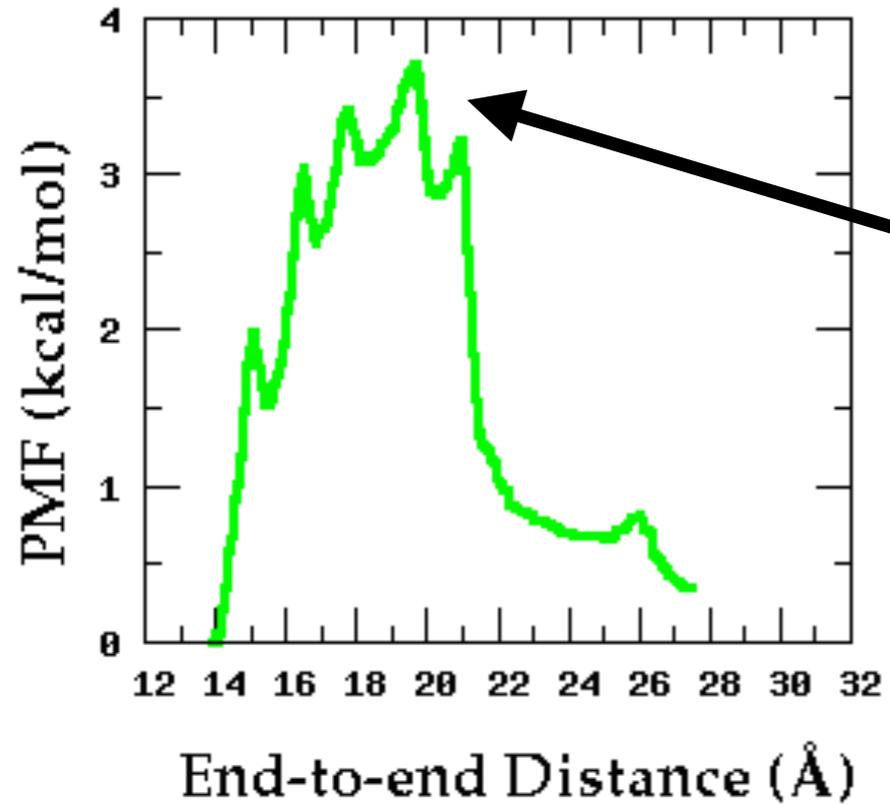
Least free-energy path (**LFEP***) agrees well with 1D vacuum PMF

*Ensing, Laio, Parrinello, Klein. (2005)
J. Phys. Chem. B. 109:6676– 6687.



LFEP from 2D PMF in water shows deviation from "accordion-like" unfolding

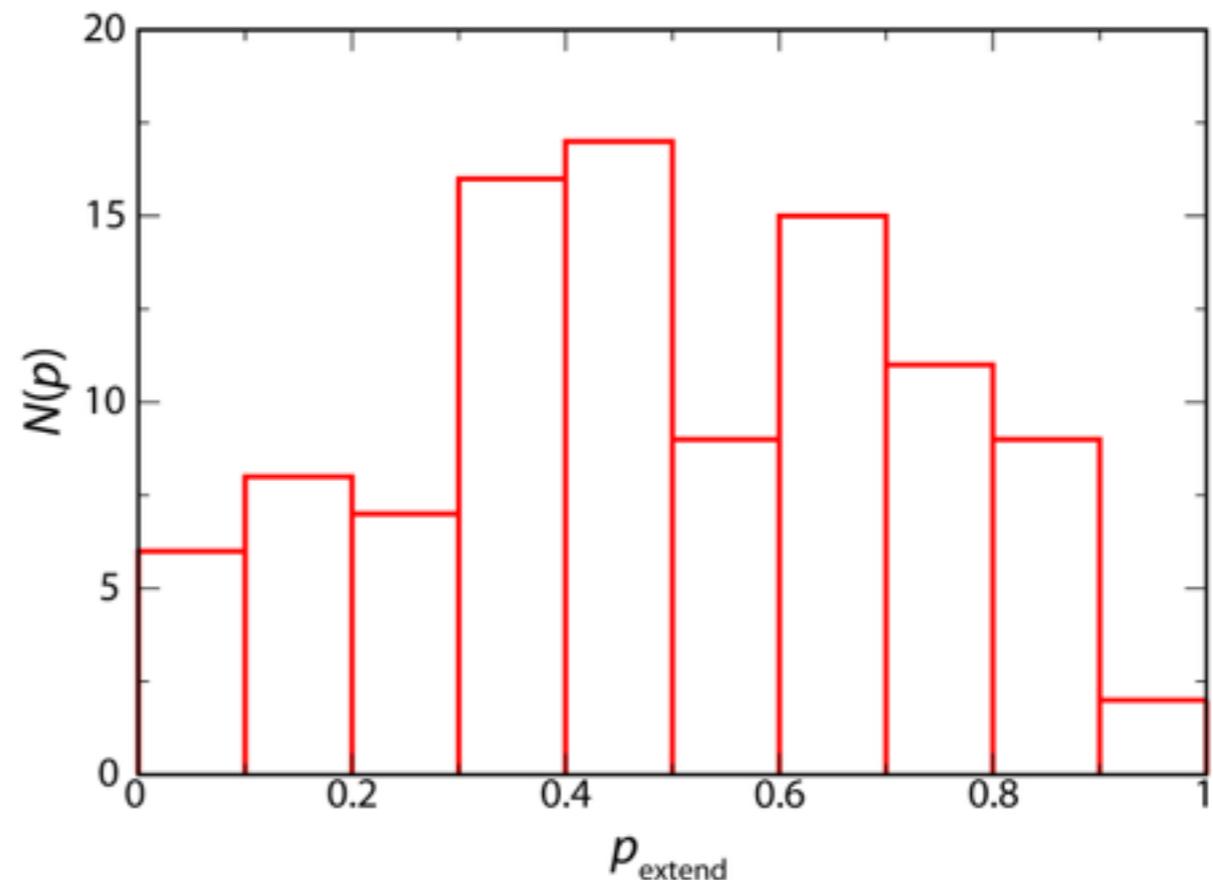
Is the LFEP the real pathway?



If the choice of RCs is reasonable, the peptide should have a **50/50 probability** of going either way at the free-energy maximum

estimate of the **committor probability distribution** is roughly peaked around 0.5 → a true transition state

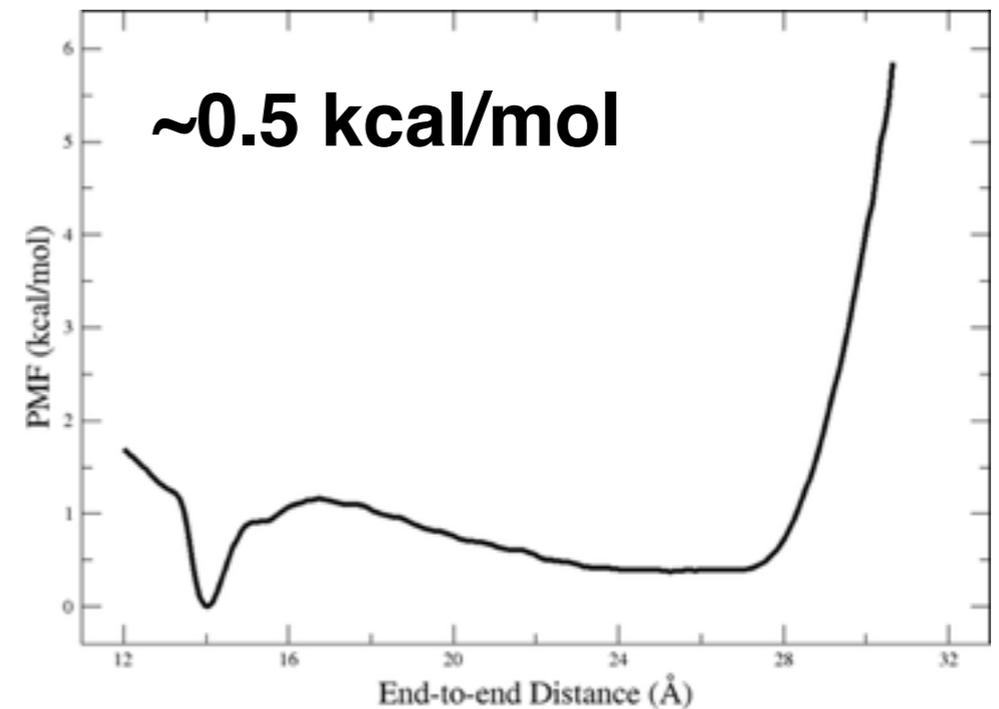
(~100 conformations * 50 10-ps simulations each)



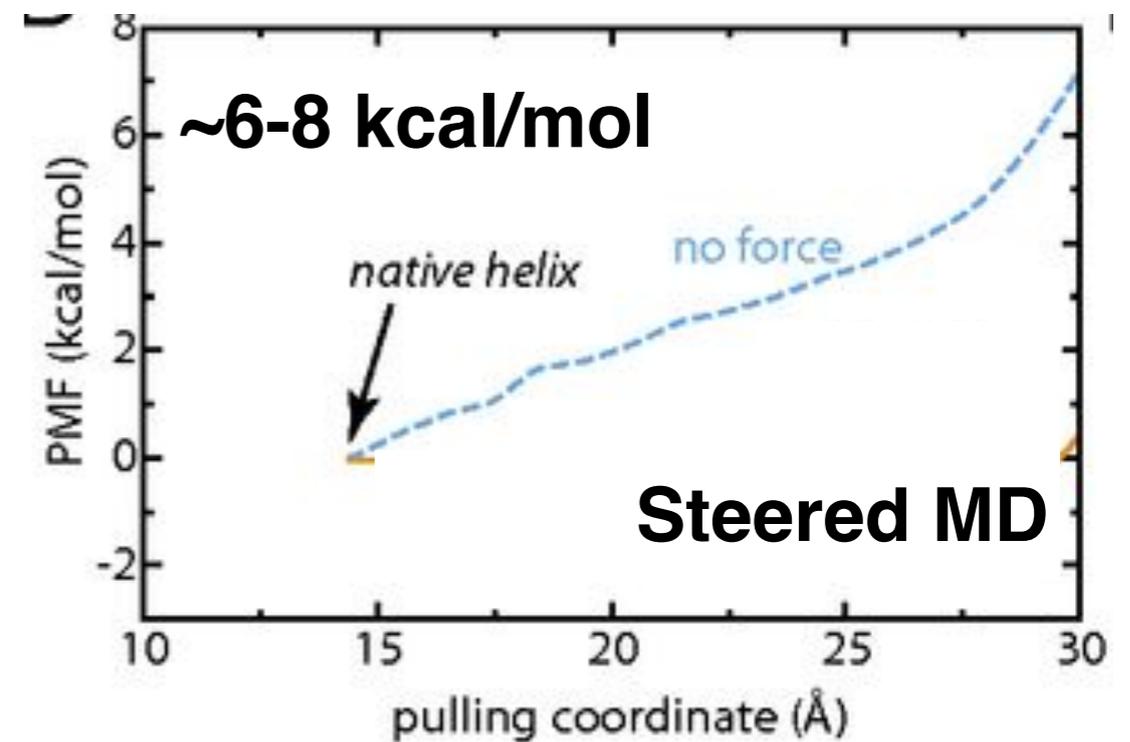
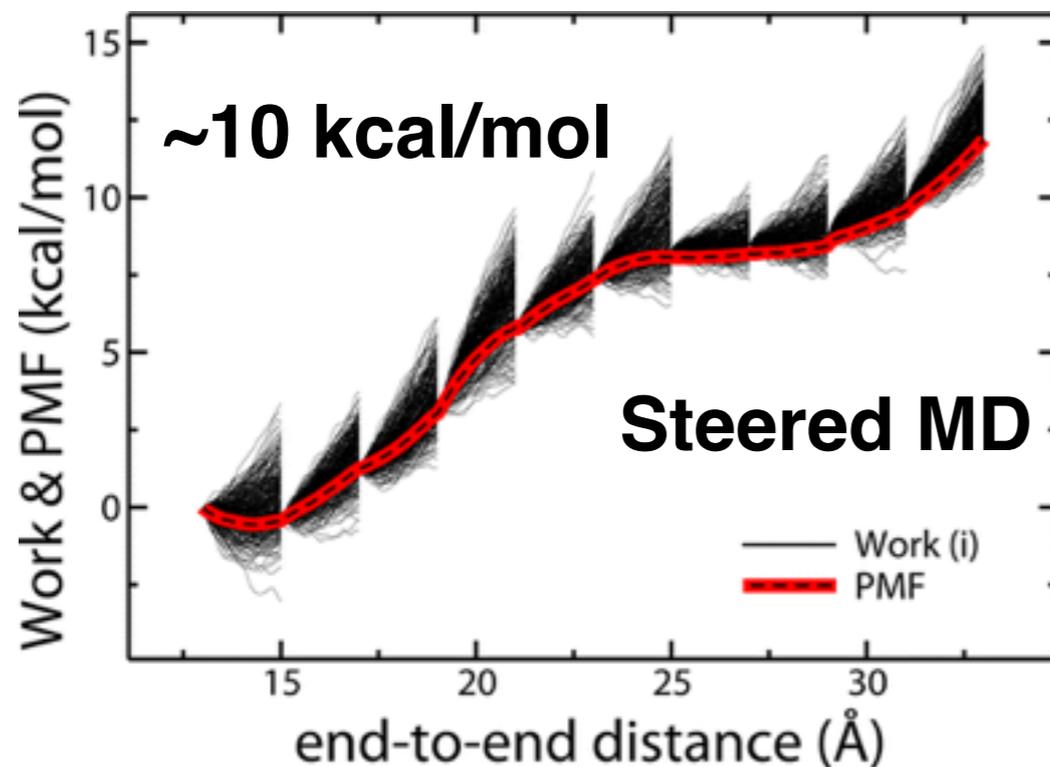
Revisiting the 1D PMF

Integrate out α RC to get 1D PMF as a function of distance ξ , as originally intended

compact states only *very slightly favored* over extended; barrier between them is $\sim 2kT$



Comparison to other published PMFs



Ozer, Quirk, Hernandez. Thermodynamics of decaalanine stretching in water obtained by adaptive steered molecular dynamics simulations. (2012) *J. Chem. Theory Comput.* 8:4837–4844.

Stirnemann, Kang, Zhou, Berne. How force unfolding differs from chemical denaturation. (2014) *PNAS* 111:3413-3418.

Do we trust our results?

Extended Starting State

α -Helical Starting State

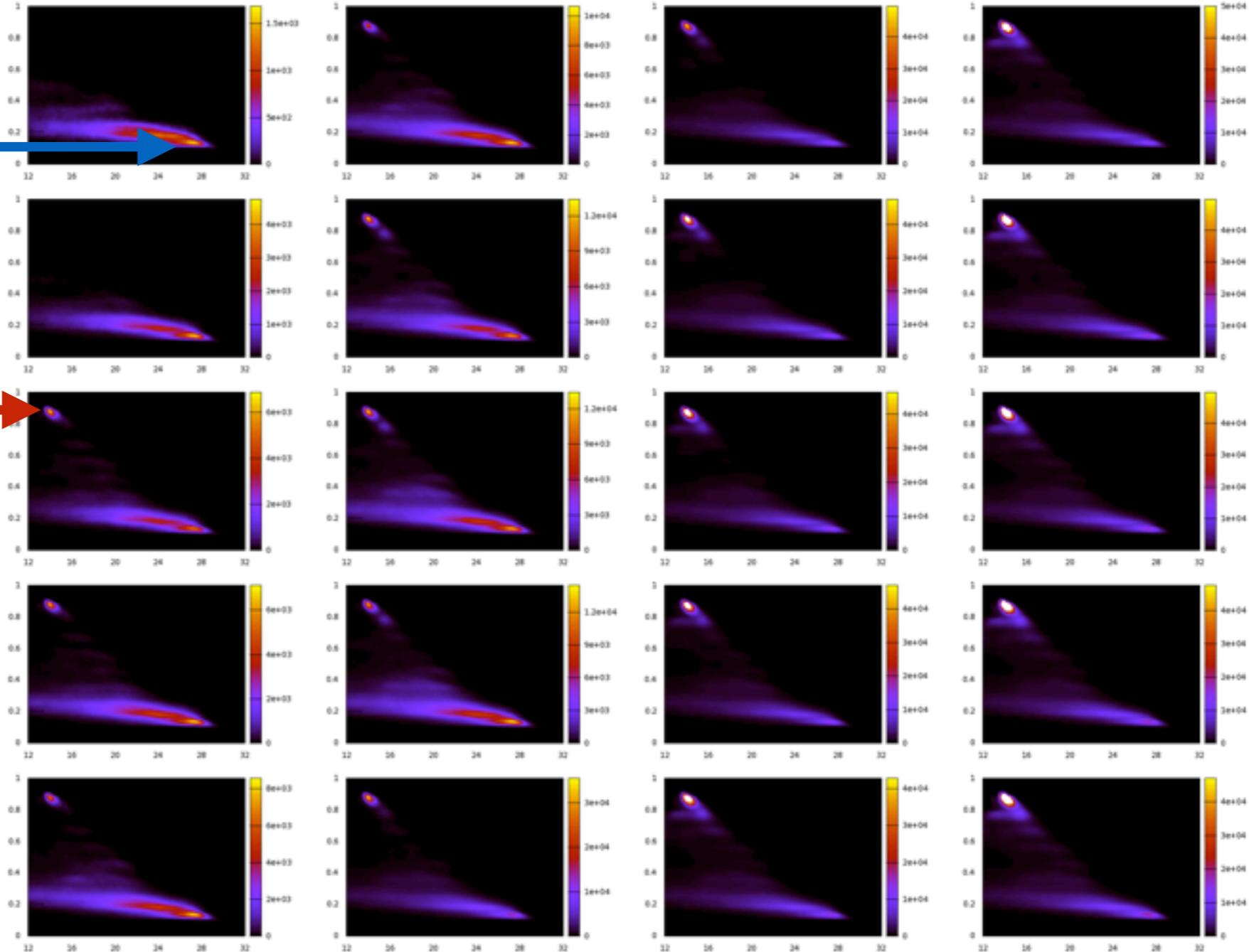
extended



helical



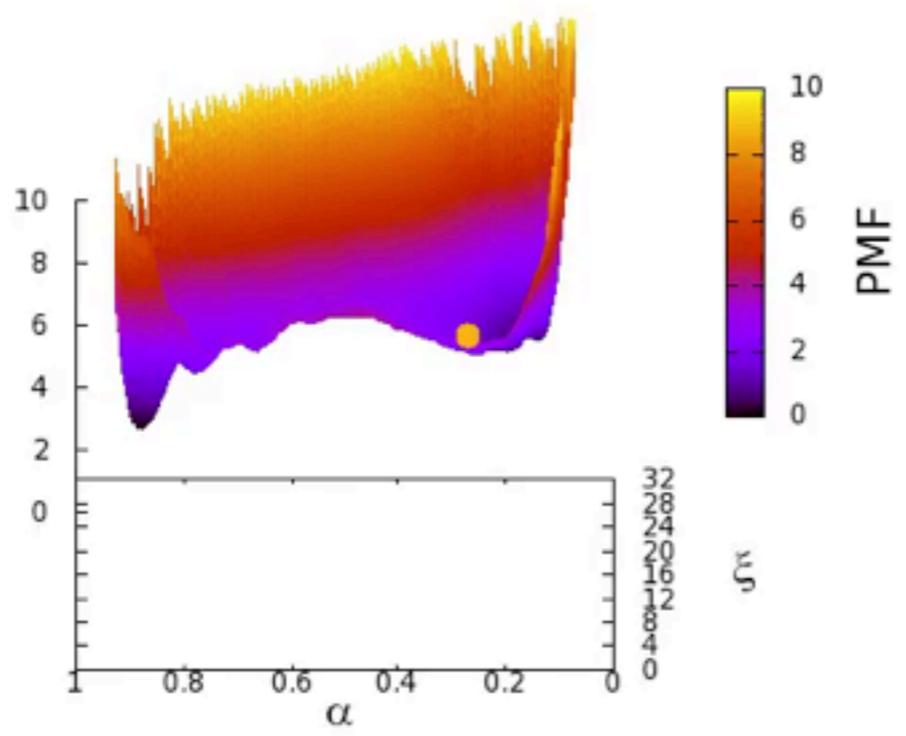
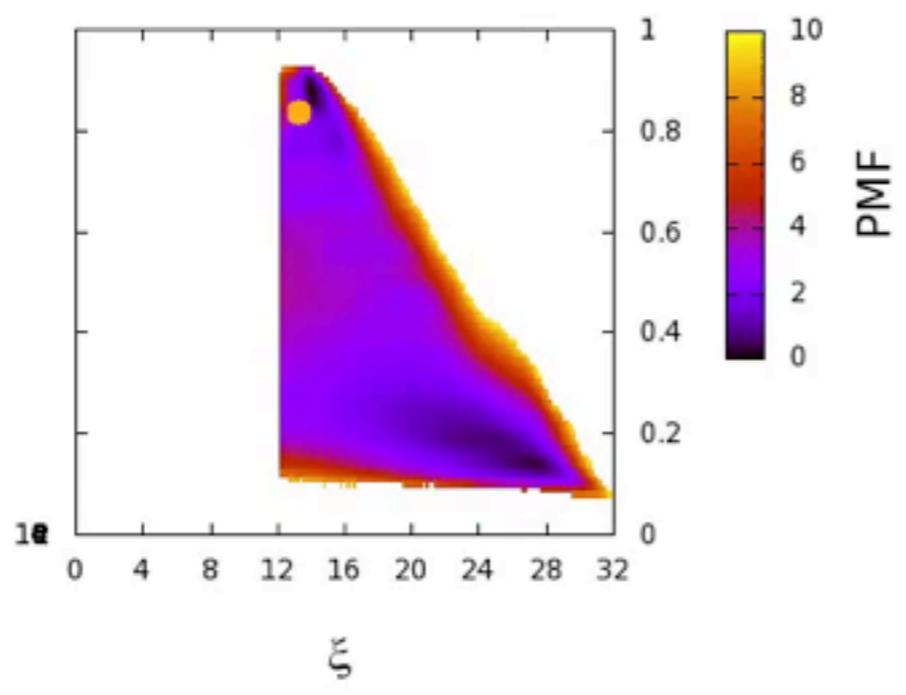
α -Helical Content



*20 simulations
50 ns each*

Counts

in almost every run, the peptide samples **both** helical and extended states

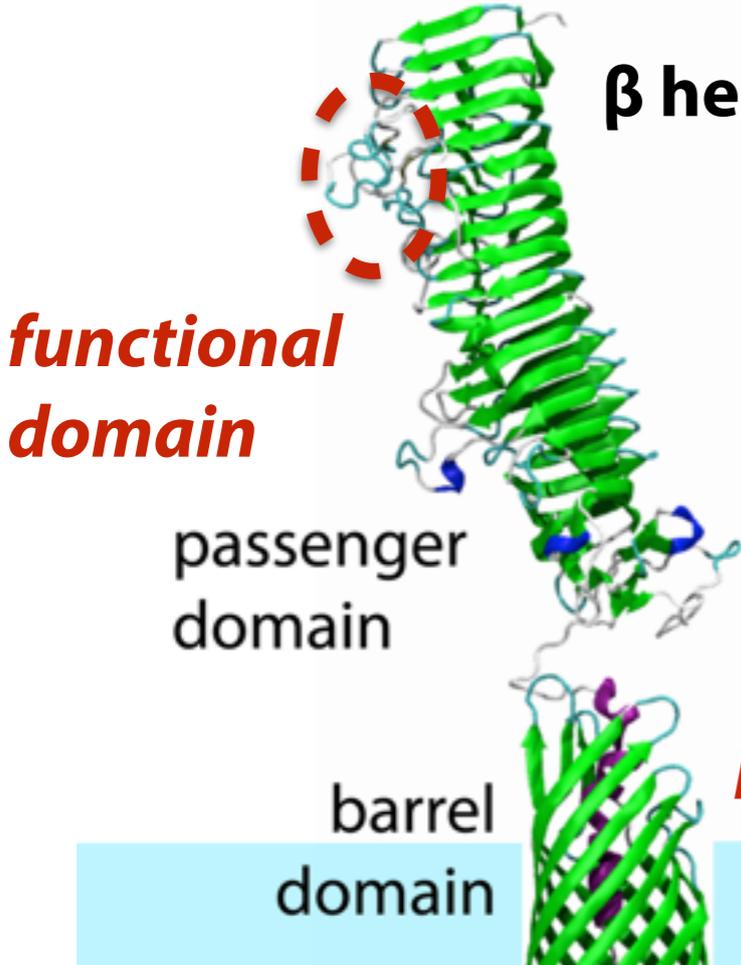


Example 2: OM protein development

insertion via BamA



hypothesis: folding and insertion are coupled via a strand-strand complement mechanism



β helix

functional domain

97% of autotransporters have a β helical domain that is **NOT** related to their virulence function - **WHY???**

passenger domain

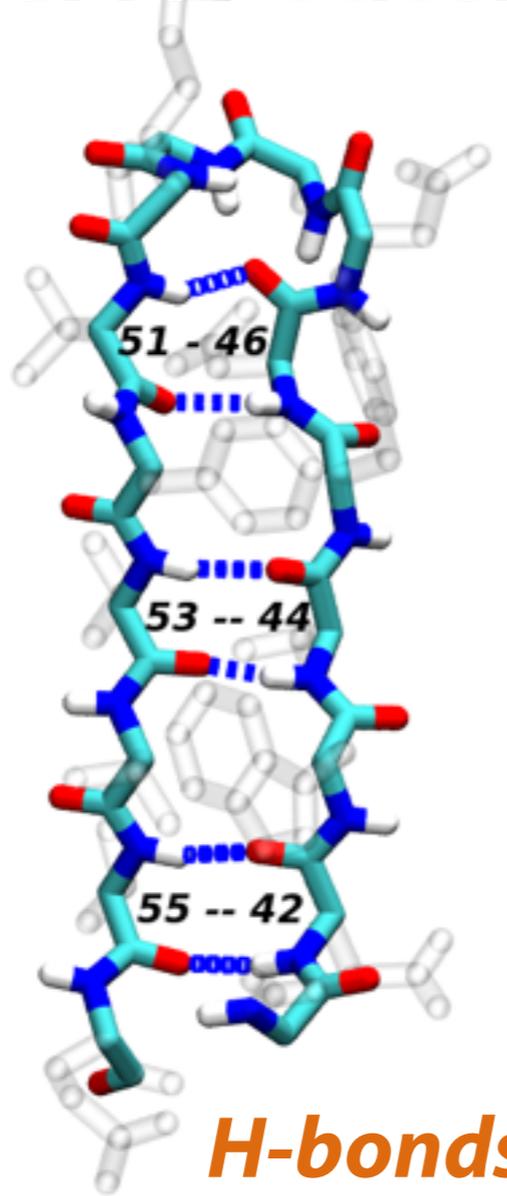
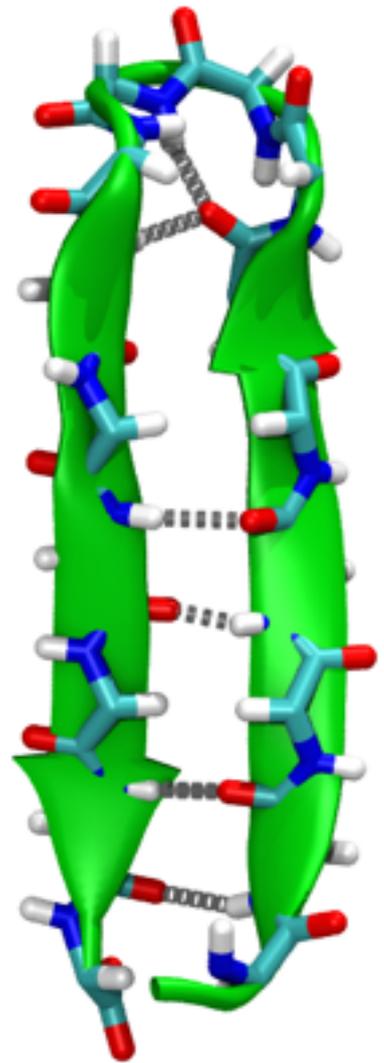
barrel domain

pertactin

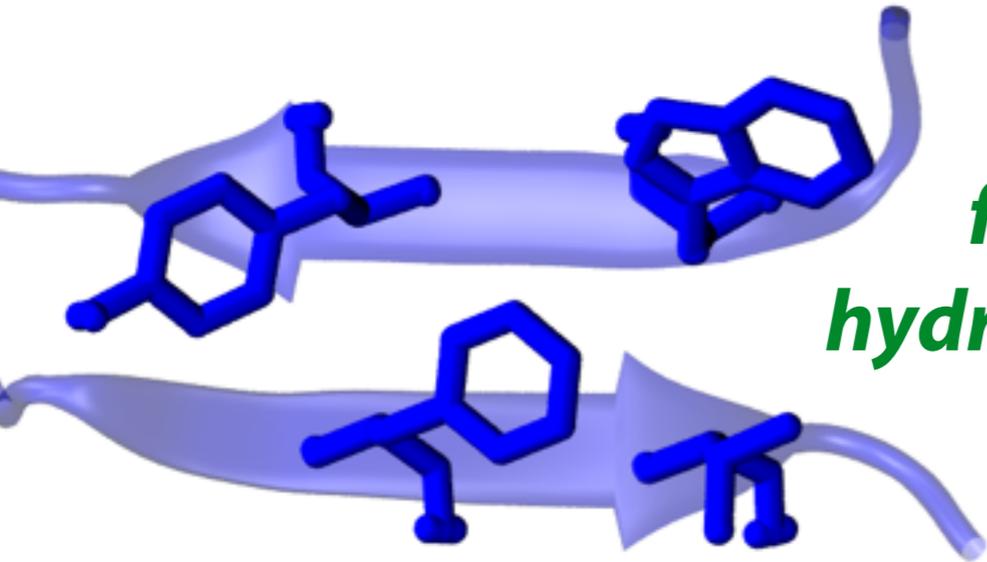
hypothesis: folding drives export

simplest model for outer-membrane protein folding

GB1 - a β hairpin



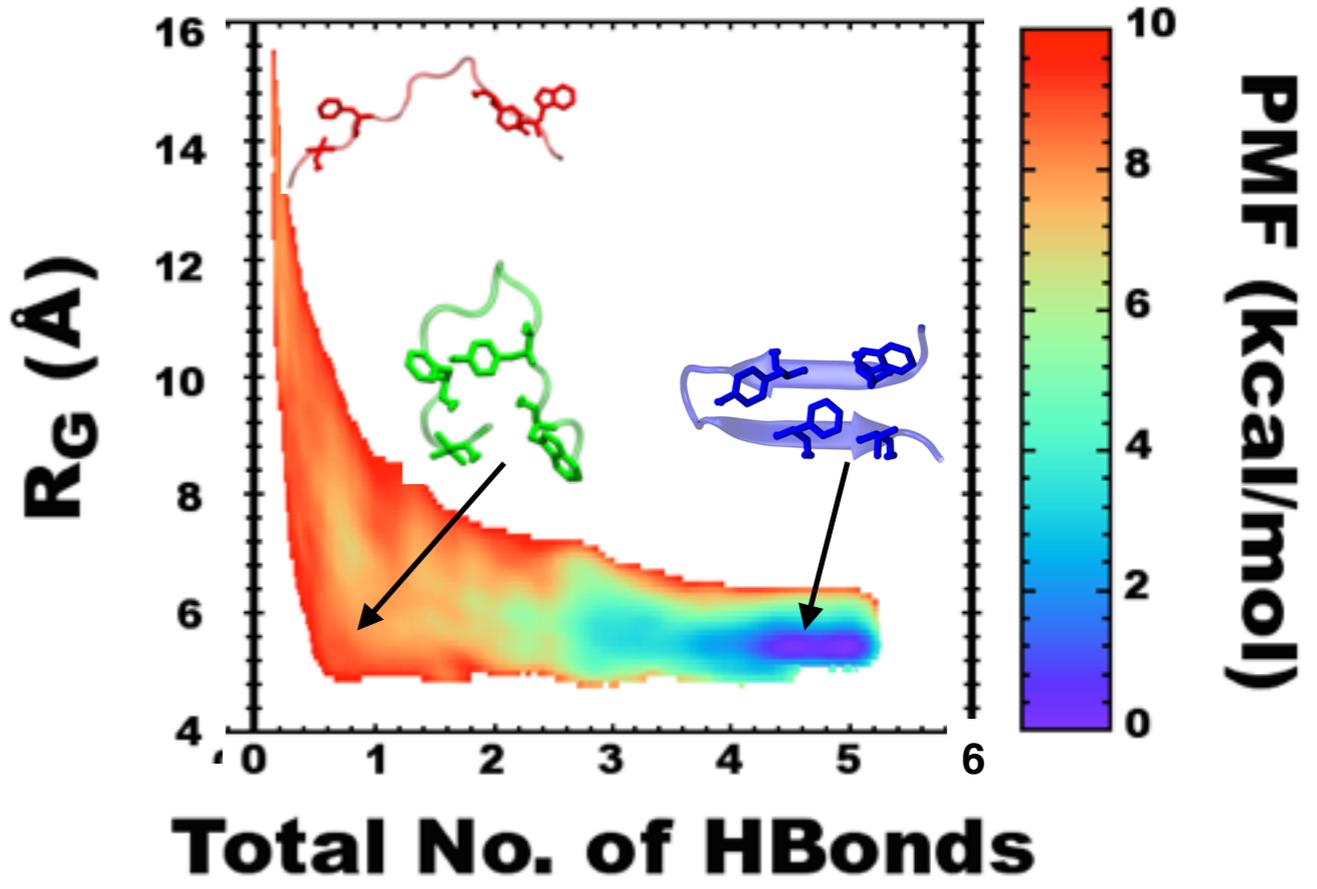
H-bonds



four core hydrophobic residues

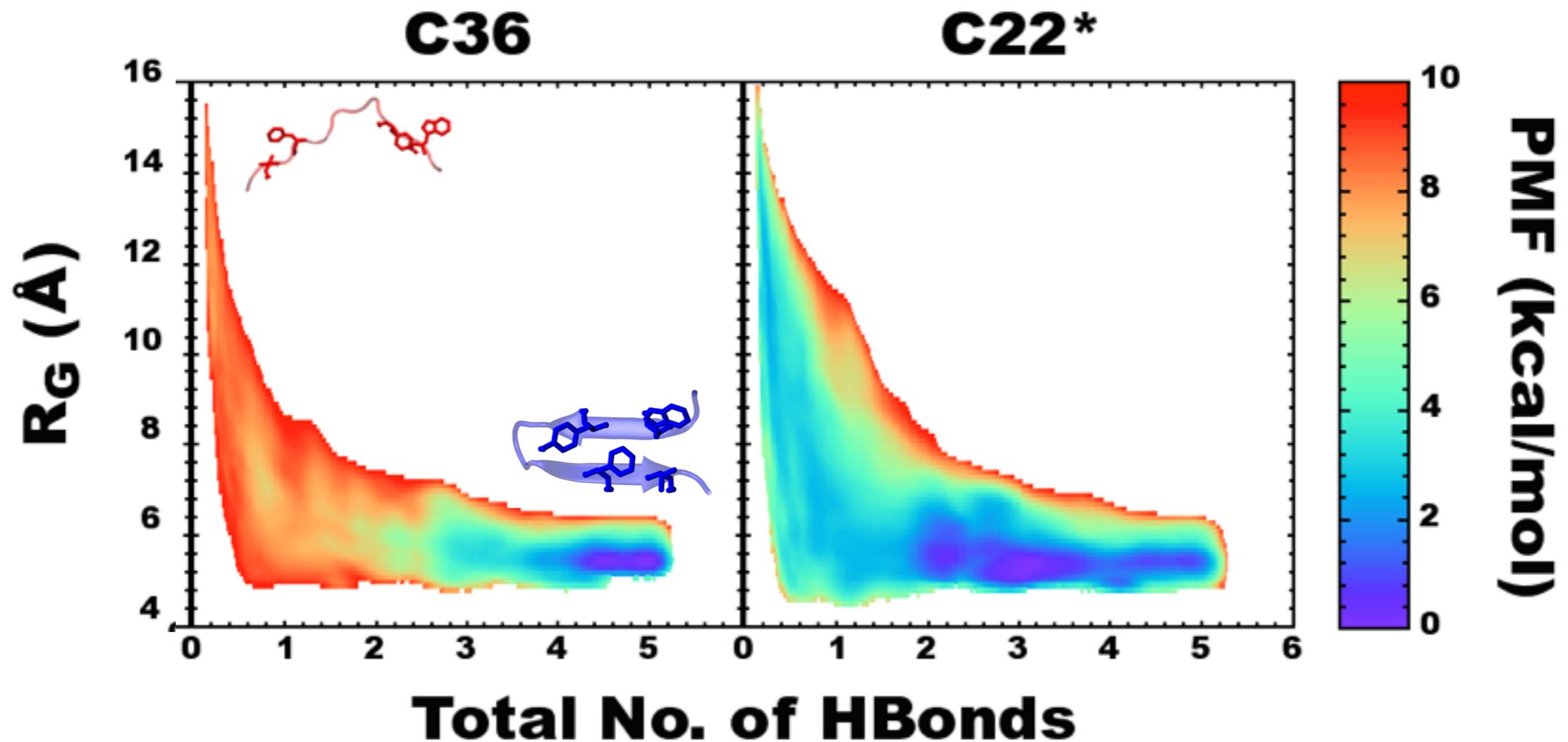
Again, use two reaction coordinates

- R_G for **hydrophobic core** (4 residues), a measure of size
- backbone **hydrogen bonds** (up to 6), a measure of structure



C36 favors hairpin by **8 kcal/mol!**

Comparison across force fields



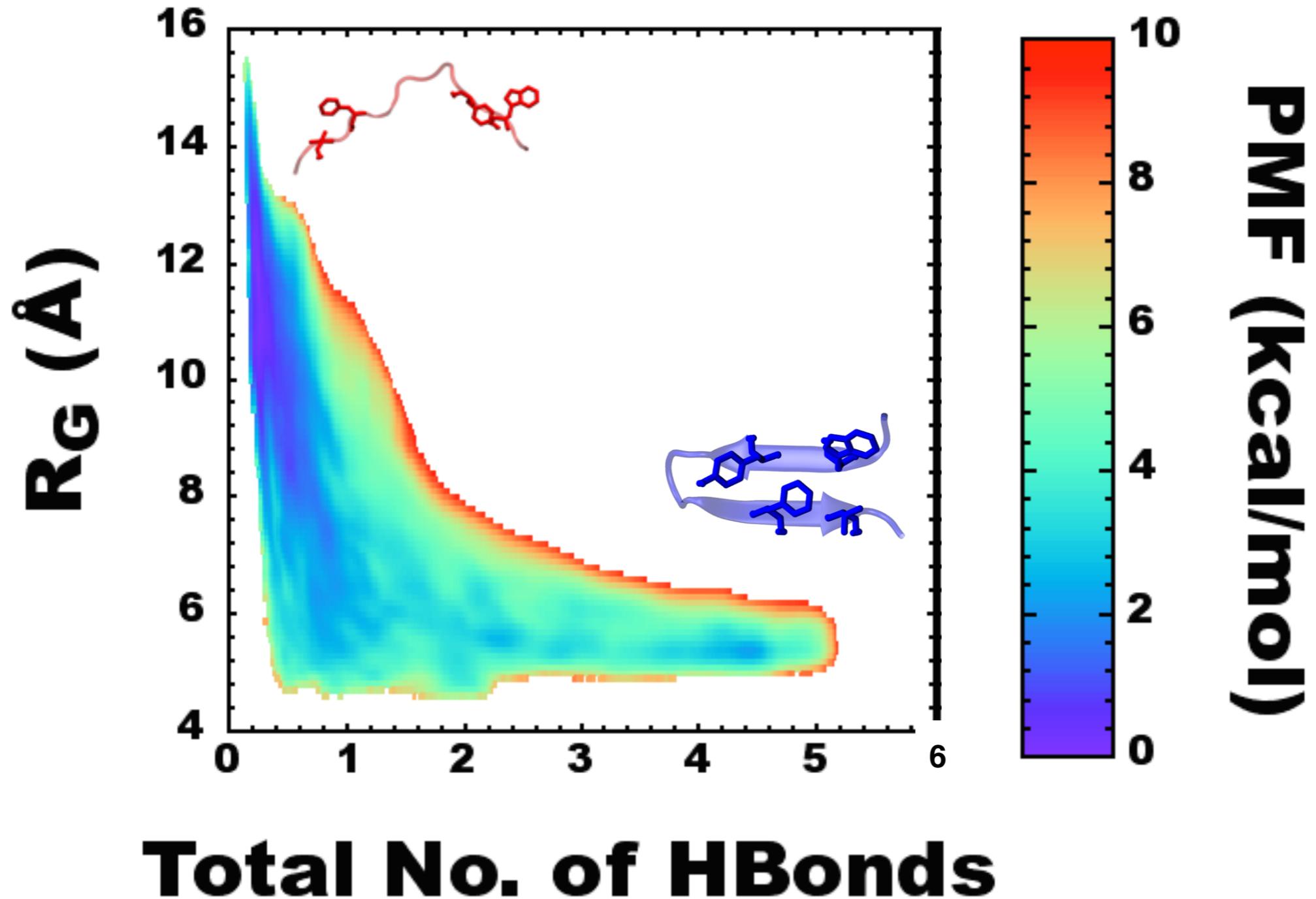
C22* favors hairpin by ~5 kcal/mol, a noticeable reduction

CHARMM36 known to **overstabilize** folded state (folded fraction of 77% vs. 60% experiment)

Best, ..., MacKerell. *JCTC*.
8:3257–3273. 2012.

Here, CHARMM22* may be best choice for observing folding transitions

What about Drude polarizable force field?

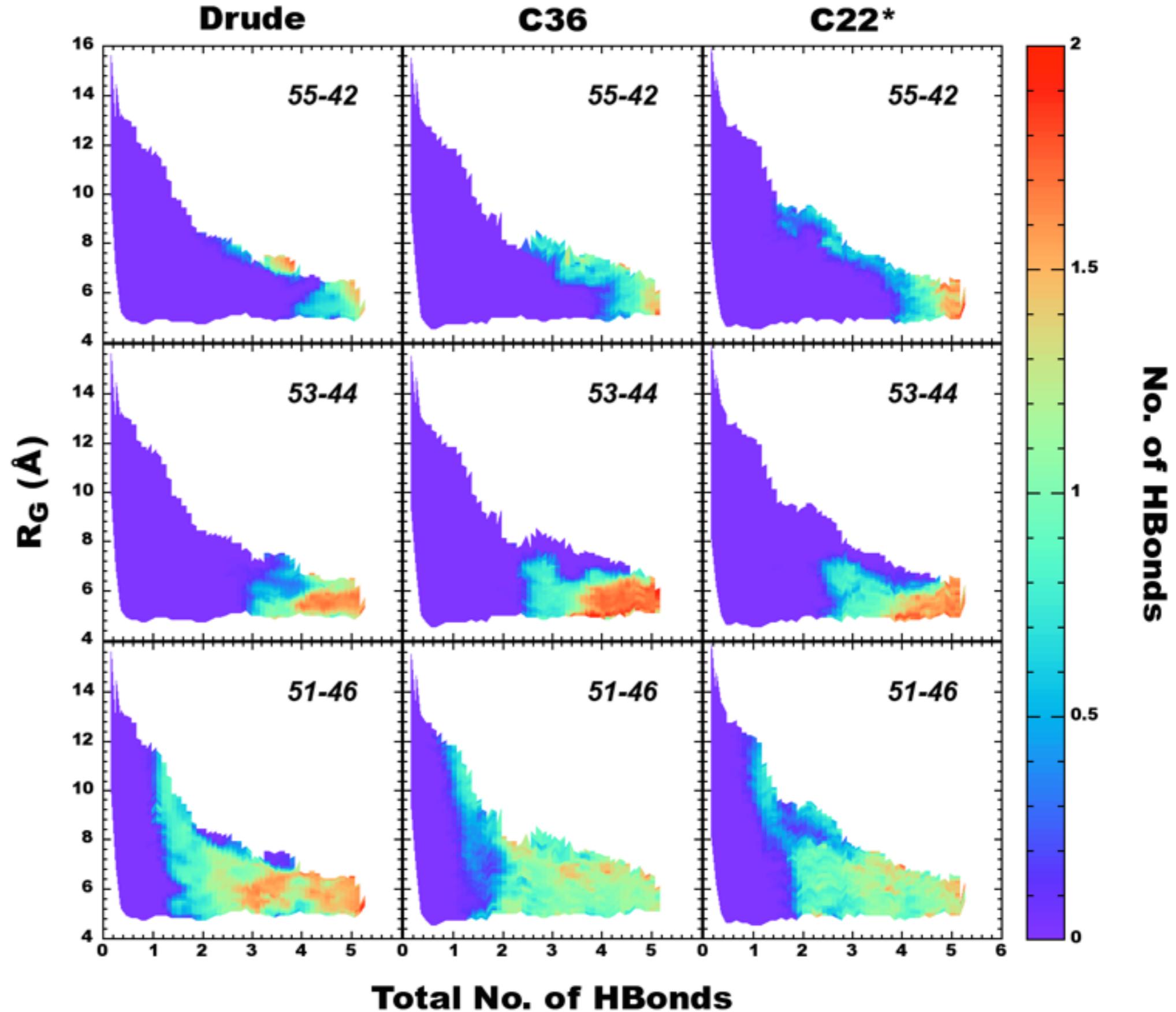


A **disaster**! Unfolded state is favored by 5-6 kcal/mol!

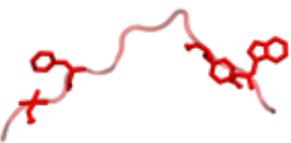
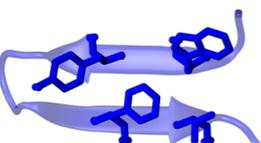
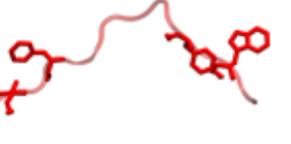
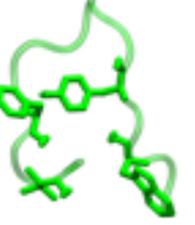
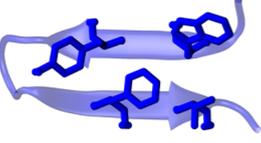
What is the source?

What about Drude?

pathway seems to be the same -
cannot appeal to
vastly different
underlying states



Energy breakdown

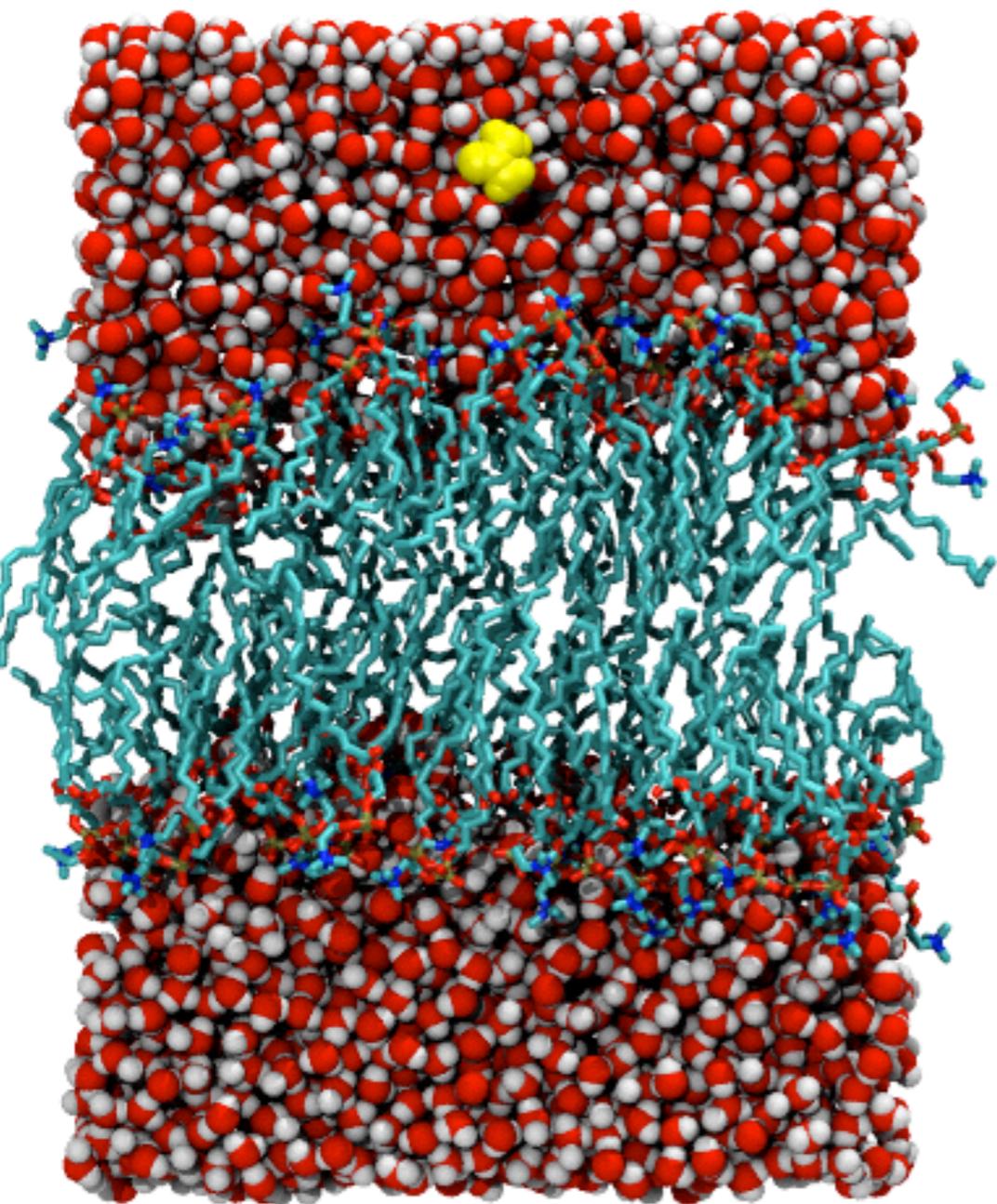
| | | Electrostatic | |
|--|-------------------|---------------|----------|
| | | Drude | C36 |
| <u>$\Delta H_{\text{Extended-Hairpin}}$</u> | | | |
|  —  | β - β | +22.2841 | +13.6653 |
| | β -Water | -34.1321 | -27.9719 |
| | β -Ions | +19.1761 | +19.9284 |
| | Core-Core | +0.1122 | -0.1911 |
| | Core-Water | -10.6076 | +0.8959 |
| | Core-Ions | +1.1662 | +0.8232 |
| | Sum | -2.0011 | +7.1498 |
| <u>$\Delta H_{\text{Extended-Compact}}$</u> | | | |
|  —  | β - β | -1.4706 | -1.3416 |
| | β -Water | +56.1550 | +56.3156 |
| | β -Ions | +3.8539 | +4.2264 |
| | Core-Core | -1.0638 | -0.8021 |
| | Core-Water | -40.6795 | -27.5980 |
| | Core-Ions | +1.5464 | +1.1326 |
| | Sum | +18.3414 | +31.9329 |
| <u>$\Delta H_{\text{Compact-Hairpin}}$</u> | | | |
|  —  | β - β | +23.7547 | +15.0069 |
| | β -Water | -90.2871 | -84.2875 |
| | β -Ions | +15.3222 | +15.7020 |
| | Core-Core | +1.1760 | +0.6110 |
| | Core-Water | +30.0719 | +28.4939 |
| | Core-Ions | -0.3802 | -0.3094 |
| | Sum | -20.3425 | -24.7831 |

hydrophobic core residues' interactions with water are **too strong** in the extended state

are they becoming over-polarized?

ongoing work...

Example 3: membrane permeation



small molecules have a probability of breaching the membrane barrier related to their **potential of mean force $W(\mathbf{z})$** and **diffusivity $D(\mathbf{z})$** by the solubility-diffusion model:

$$\frac{1}{P} = \int_{\text{bulk}}^z \frac{e^{W(z')/kT}}{D(z')} dz'$$

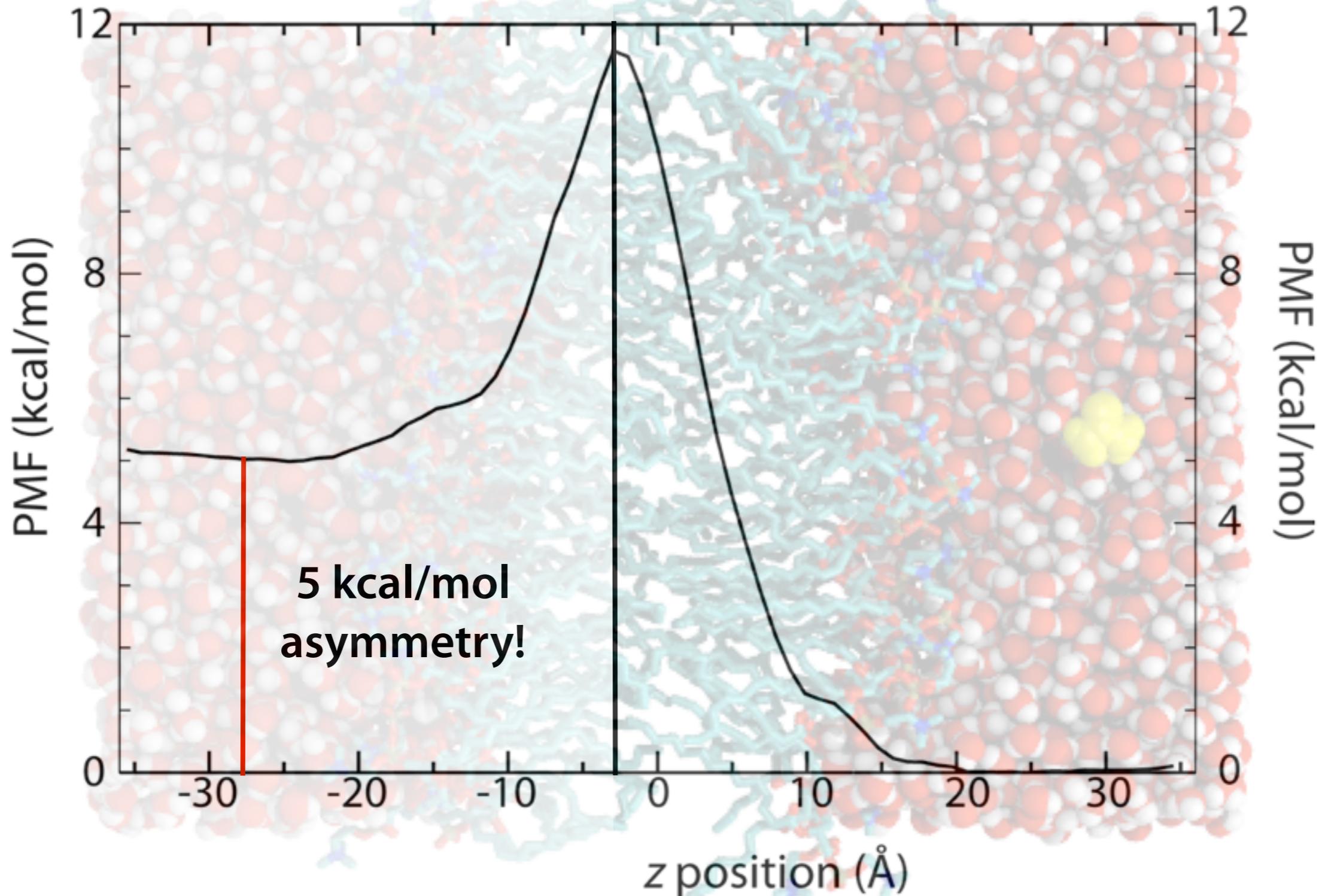
note: focus is on the PMF only here, not the permeability

using urea, etc. as simple test cases for this model and its constituent calculations

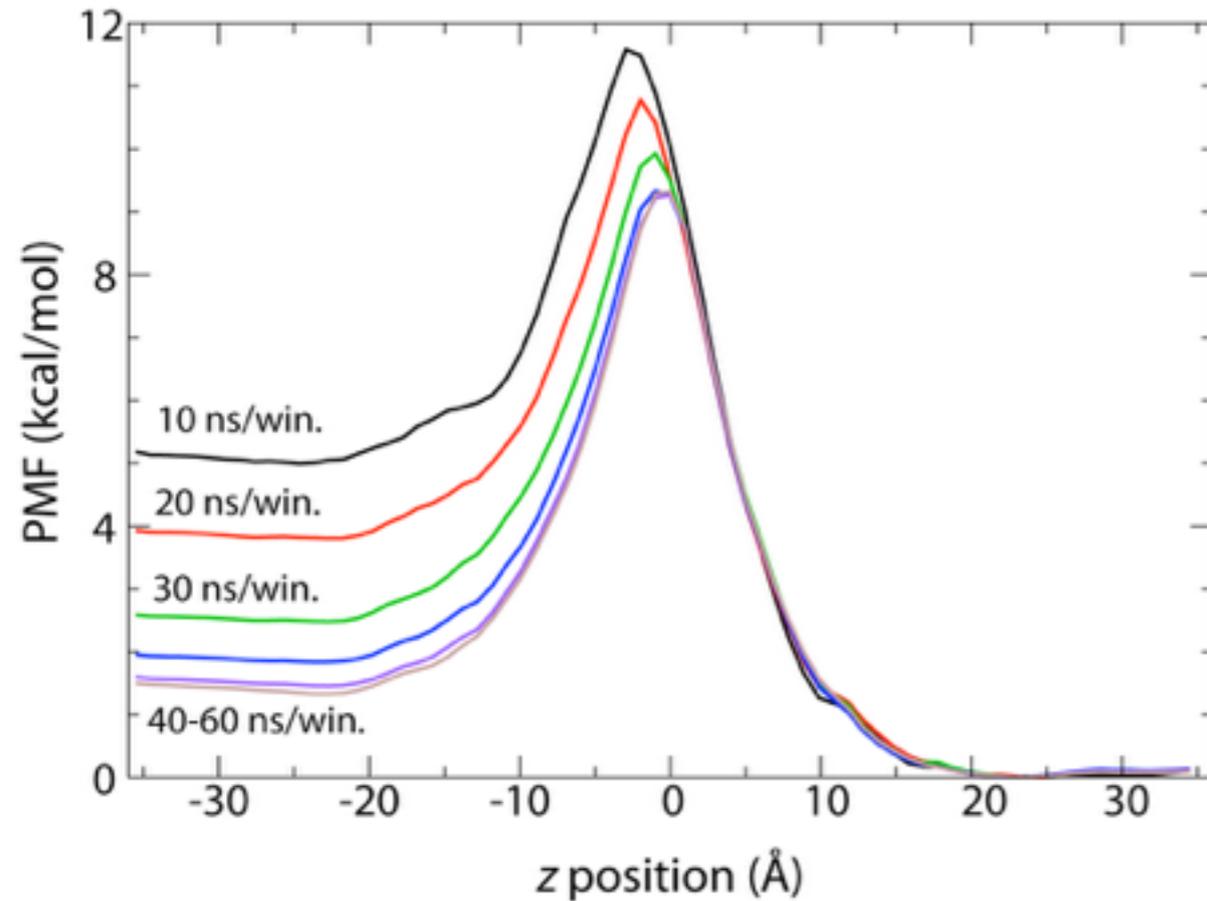
**collaboration with Chris Chipot, Yi Wang, Chris Rowley, and Rommie Amaro*

How much sampling is needed?

is 10 ns/window is enough? (360 ns total)



How much sampling is needed?



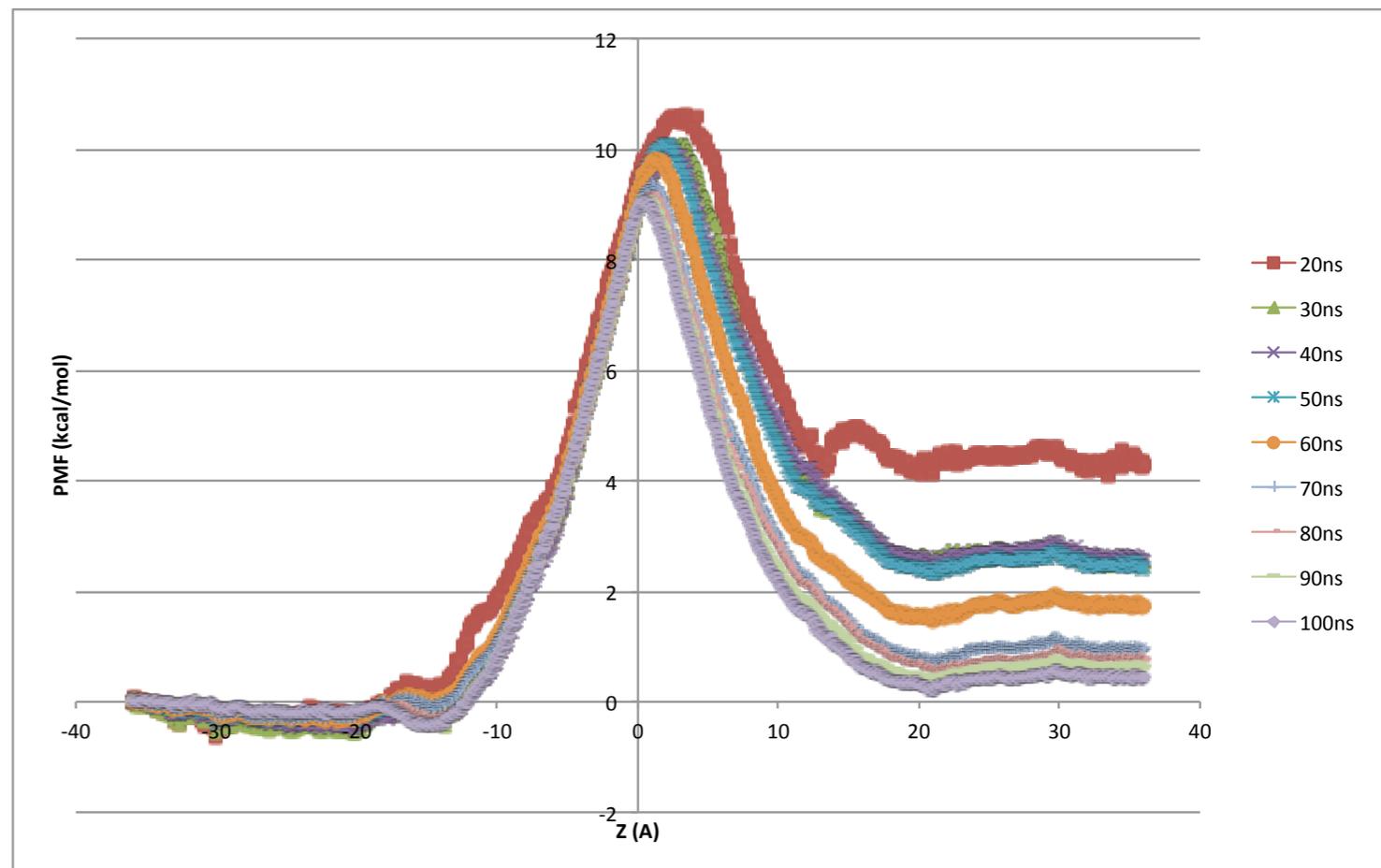
requires over $4 \mu\text{s}$ of REMD-US to bring asymmetry to ~ 1 kcal/mol but the results are good, $\log(P) = -5.83$

exp. ~ -5.4 (Finkelstein 1976), although from PAMPA, ~ -9.0 (not a membrane!)

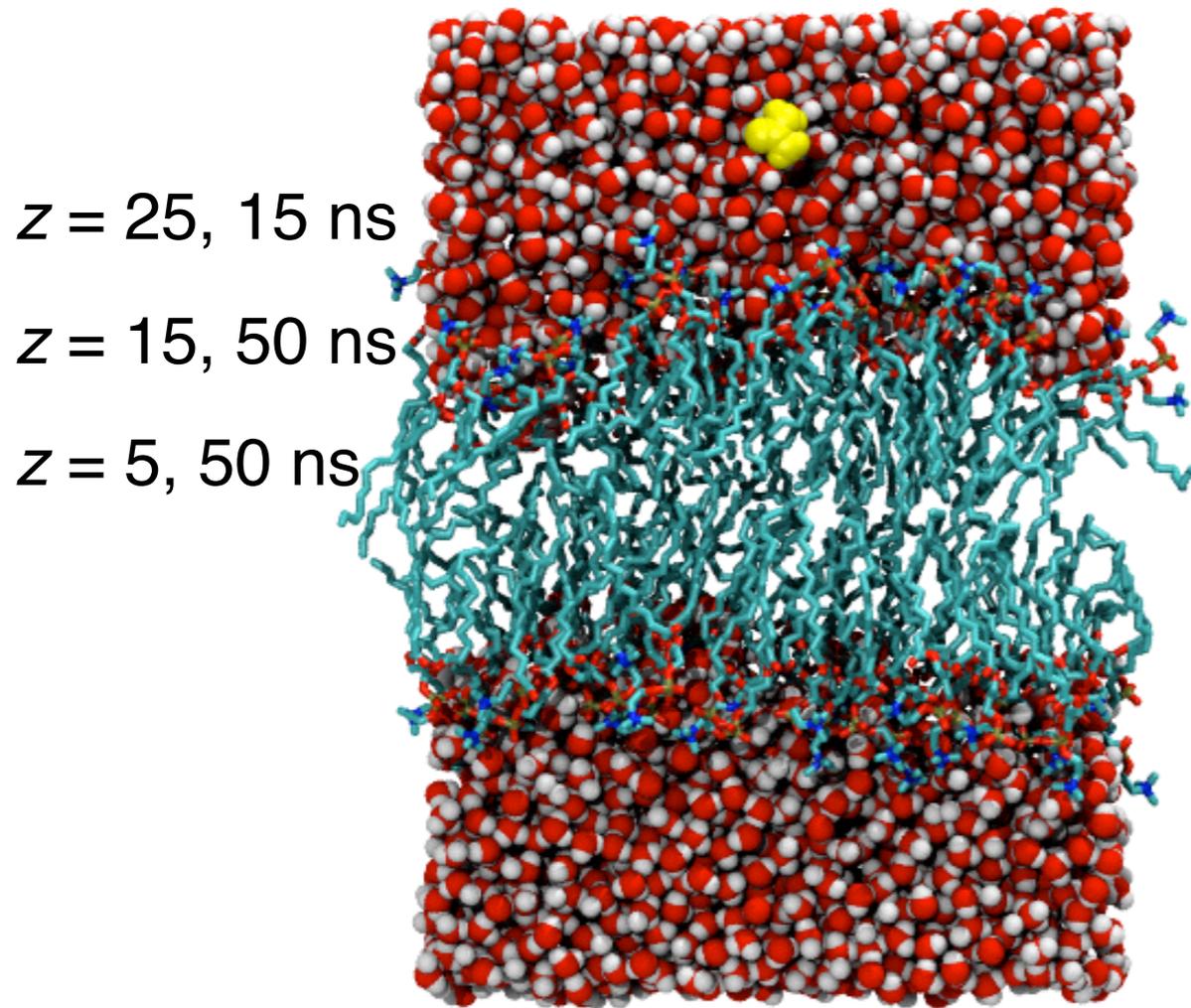
ABF, on the other hand, requires less than $1 \mu\text{s}$ for similar results - *why?*



ABF calculations run in the lab of Yi Wang, Chinese Univ. of Hong Kong



Can REMD-US be rescued here?



z = 30, 10 ns

z = 20, 25 ns

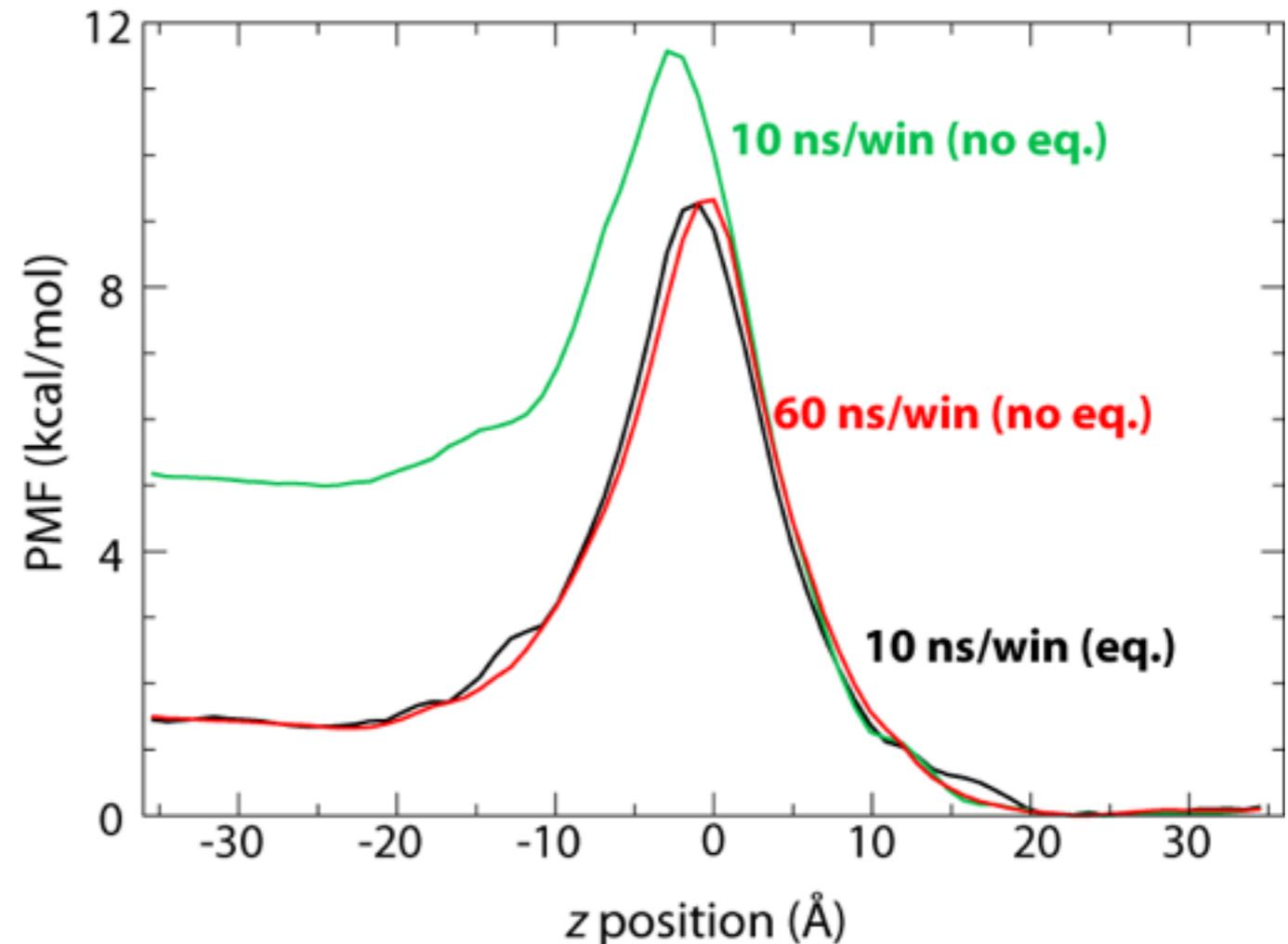
z = 10, 50 ns

z = 0, 100 ns

Equilibrate a few selected windows and use them to seed intermediate ones

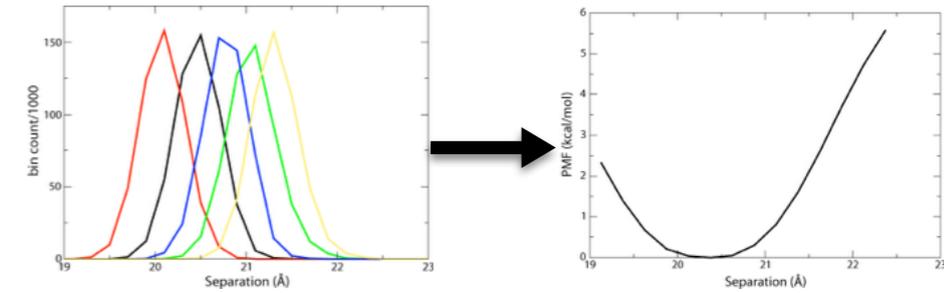
Every 5 Å, both above and below,
500 ns total

Using the newly **equilibrated** states, REMD-US produces an identical PMF in **~25%** of the time (500 ns + 720 ns = 1220 ns vs. 4320 ns)

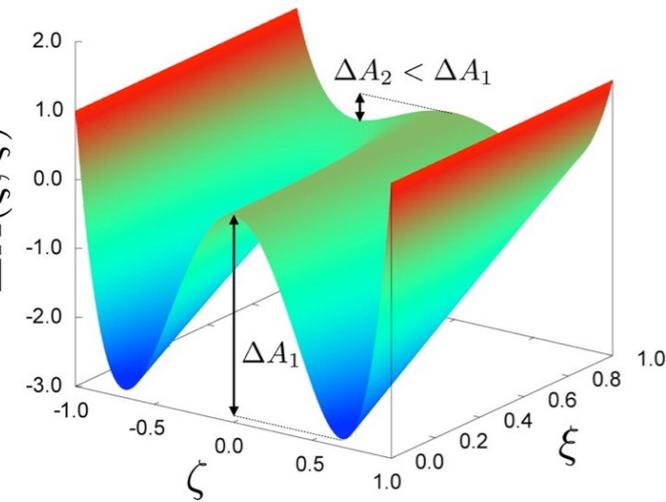


Final thoughts and reminders

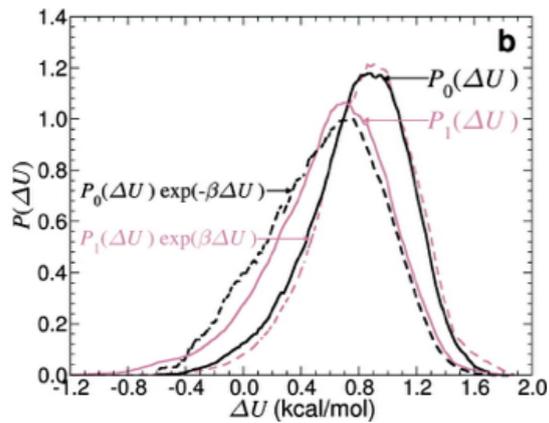
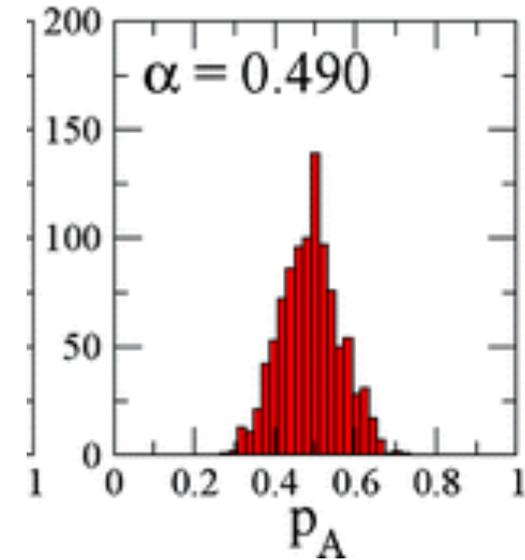
Biasing is fine as long as there is a method to **unbias** the results



Be aware of **orthogonal barriers** that may hamper sampling in some regions

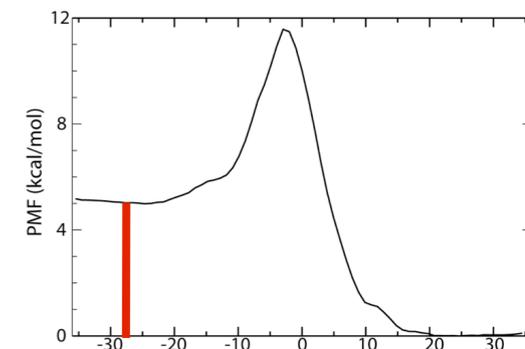


Choose **reaction coordinate(s)** that capture the transition of interest well and check it (e.g., with the **committor probability**)



For any method involving stratification of a path (US, FEP, etc.) confirm that the neighboring states have good **overlap**

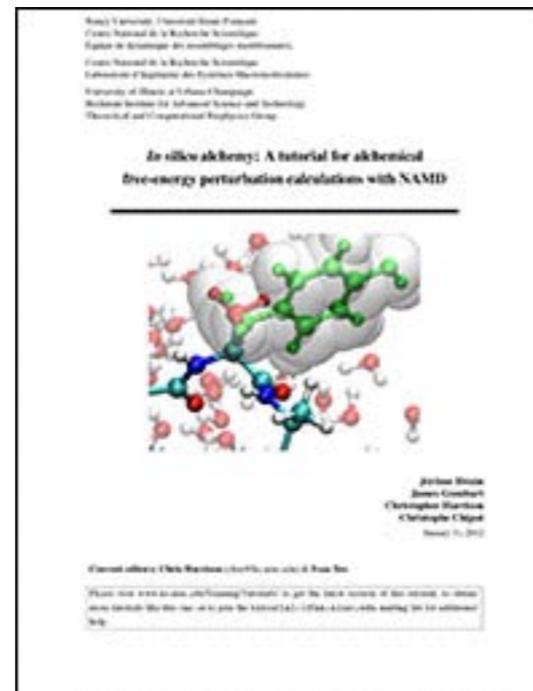
Be certain that your initial states are **well equilibrated**



Free energy tutorials available

<http://www.ks.uiuc.edu/Training/Tutorials>

**Free-energy
perturbation**



**Protein:ligand
binding**



**Umbrella
Sampling, SMD**



**Adaptive
Biasing Forces**

