

Statistical Mechanics of Proteins

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- ▶ **Equilibrium and non-equilibrium properties of proteins**
 - ▶ **Free diffusion of proteins**
- ▶ Coherent motion in proteins: temperature echoes
- ▶ Simulated cooling of proteins

Molecular Modeling

1. Model building
2. Molecular Dynamics Simulation
3. **Analysis** of the
 - model
 - results of the simulation

Collection of MD Data

- DCD trajectory file
 - coordinates for each atom
 - velocities for each atom
- Output file
 - global energies
 - temperature, pressure, ...

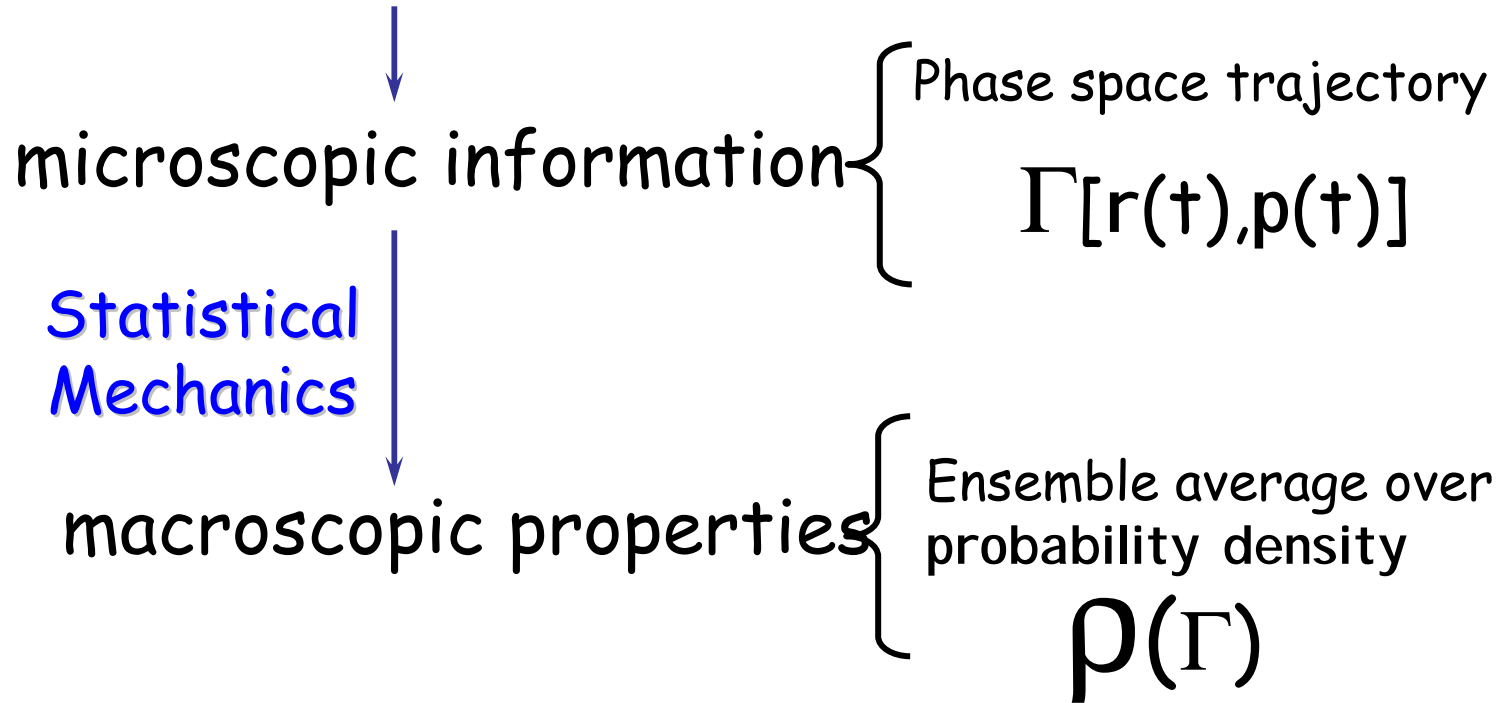
Analysis of MD Data

1. Structural properties
2. Equilibrium properties
3. Non-equilibrium properties

Can be studied via both equilibrium and non-equilibrium MD simulations

Equilibrium (Thermodynamic) Properties

MD simulation



Statistical Ensemble

Collection of *large* number of replicas (on a macroscopic level) of the system

Each replica is characterized by the same macroscopic parameters (e.g., NVT, NPT)

The microscopic state of each replica (at a given time) is determined by Γ in phase space

Time vs Ensemble Average

For $t \rightarrow \infty$, $\Gamma(t)$ generates an ensemble with

$$\rho(\Gamma)d\Gamma = \lim_{t \rightarrow \infty} d\tau / t$$

Ergodic Hypothesis: Time and Ensemble averages are

equivalent, i.e., $\langle A(r, p) \rangle_t = \langle A(\Gamma) \rangle_\rho$

Time average: $\langle A \rangle_t = \frac{1}{T} \int_0^T dt A[\mathbf{r}(t), \mathbf{p}(t)]$

Ensemble average: $\langle A \rangle = \int d\Gamma \rho(\Gamma) A(\Gamma)$

Thermodynamic Properties from MD Simulations

Thermodynamic (equilibrium) averages can be calculated via time averaging of MD simulation time series

$$\langle A \rangle \approx \frac{1}{N} \sum_{i=1}^N A(t_i)$$

Thermodynamic average

MD simulation time series

Finite simulation time
means incomplete sampling!

Common Statistical Ensembles

1. Microcanonical (N,V,E):

$$\rho_{NVE}(\Gamma) \propto \delta[H(\Gamma) - E] \quad \leftarrow \text{Newton's eq. of motion}$$

2. Canonical (N,V,T):

$$\rho_{NVT}(\Gamma) = \exp\{[F - H(\Gamma)]/k_B T\} \quad \leftarrow \text{Langevin dynamics}$$

3. Isothermal-isobaric (N,p,T)

$$\rho_{NPT}(\Gamma) = \exp\{[G - H(\Gamma)]/k_B T\} \quad \leftarrow \text{Nose-Hoover method}$$

Different simulation protocols $[\Gamma(t) \rightarrow \Gamma(t + \delta t)]$ sample different statistical ensembles

Examples of Thermodynamic Observables

- Energies (kinetic, potential, internal,...)
- Temperature [*equipartition theorem*]
- Pressure [*virial theorem*]

Thermodynamic derivatives are related to mean square fluctuations of thermodynamic quantities

- Specific heat capacity C_V and C_P
- Thermal expansion coefficient α_P
- Isothermal compressibility β_T
- Thermal pressure coefficient γ_V

Mean Energies

Total (internal) energy:	$E = \frac{1}{N} \sum_{i=1}^N E(t_i)$	TOTAL
Kinetic energy:	$K = \frac{1}{N} \sum_{i=1}^N \sum_{j=1}^M \frac{\mathbf{p}_j^2(t_i)}{2m_j}$	KINETIC
Potential energy:	$U = E - K$	BOND ANGLE DIHED IMPRP ELECT VDW

Note: You can conveniently use `namdplot` to graph the time evolution of different energy terms (as well as T, P, V) during simulation

Temperature

From the equipartition theorem $\langle p_k \partial H / \partial p_k \rangle = k_B T$

$$T = \frac{2}{3Nk_B} \langle K \rangle$$

Instantaneous *kinetic temperature*

$$T = \frac{2K}{3Nk_B}$$

`namdplot TEMP vs TS ..`

Note: in the NVTP ensemble $N \rightarrow N - N_c$, with $N_c = 3$

Pressure

From the virial theorem $\langle r_k \partial H / \partial r_k \rangle = k_B T$

$$PV = Nk_B T + \langle W \rangle$$

The *virial* is defined as

$$W = \frac{1}{3} \sum_{j=1}^M \mathbf{r}_j \cdot \mathbf{f}_j = -\frac{1}{3} \sum_{i,j>i} w(r_{ij})$$

pairwise
interaction

with $w(r) = r dV(r) / dr$

Instantaneous *pressure* function (not unique!)

$$P = \rho k_B T + W / V$$

Thermodynamic Fluctuations (TF)

$$\langle \delta A \rangle \approx \frac{1}{N} \sum_{i=1}^N [A(t_i) - \langle A \rangle]$$

Mean Square Fluctuations (MSF)

$$\langle \delta A^2 \rangle = \langle (A - \langle A \rangle)^2 \rangle = \langle A^2 \rangle - \langle A \rangle^2$$

According to *Statistical Mechanics*, the probability distribution of thermodynamic fluctuations is

$$\rho_{fluct} \propto \exp\left(\frac{\delta P \cdot \delta V - \delta T \cdot \delta S}{2k_B T}\right)$$

TF in NVT Ensemble

In MD simulations distinction must be made between properly defined mechanical quantities (e.g., energy E , kinetic temperature T , instantaneous pressure P) and thermodynamic quantities, e.g., T , P , ...

For example: $\langle \delta E^2 \rangle = \langle \delta H^2 \rangle = k_B T^2 C_V$ ✓

But: $\langle \delta P^2 \rangle \neq \langle \delta P^2 \rangle = k_B T / V \beta_T$ ✗

Other useful formulas: $\langle \delta K^2 \rangle = \frac{3N}{2} (k_B T)^2$

$$\langle \delta U^2 \rangle = k_B T^2 (C_V - 3Nk_B / 2)$$

$$\langle \delta U \delta P \rangle = k_B T^2 (\gamma_V - \rho k_B)$$

$$C_V = (\partial E / \partial T)_V$$

$$\gamma_V = (\partial P / \partial T)_V$$

How to Calculate C_V ?

1. From definition

$$C_V = (\partial E / \partial T)_V$$

Perform multiple simulations to determine $E \equiv \langle E \rangle$ as a function of T , then calculate the derivative of $E(T)$ with respect to T

2. From the MSF of the total energy E

$$C_V = \langle \delta E^2 \rangle / k_B T^2$$

with $\langle \delta E^2 \rangle = \langle E^2 \rangle - \langle E \rangle^2$

Analysis of MD Data

1. Structural properties
2. Equilibrium properties
3. Non-equilibrium properties

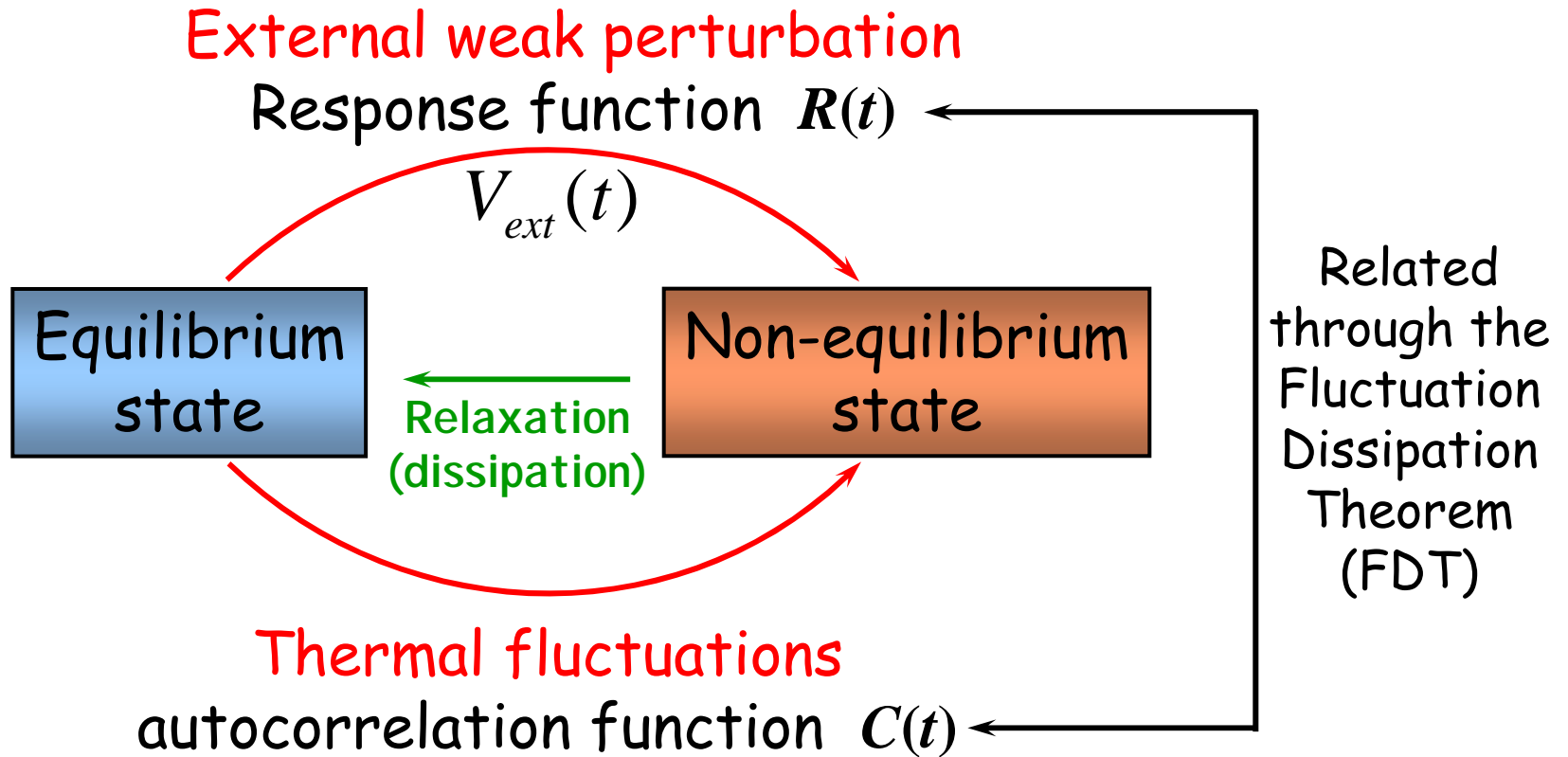
Can be studied via both equilibrium and/or non-equilibrium MD simulations

Non-equilibrium Properties

1. Transport properties
2. Spectral properties

Can be obtained from *equilibrium* MD simulations by employing *linear response theory*

Linear Response Theory



Time Correlation Functions

$$C_{AB}(t-t') = \langle A(t) B(t') \rangle = \langle A(t-t') B(0) \rangle$$

since ρ_{eq} is t independent!

$A \neq B$ cross-
 $A = B$ auto- } correlation function

Correlation time: $\tau_c = \int_0^{\infty} dt C_{AA}(t) / C_{AA}(0)$

Estimates how long the "memory" of the system lasts

In many cases (but not always): $C(t) = C(0) \exp(-t / \tau_c)$

Response Function

or generalized susceptibility

External perturbation: $V_{ext}(t) = -A \cdot f_{ext}(t)$

Response of the system: $\langle A(t) \rangle = \int_0^t dt' R(t-t') f_{ext}(t')$

Response function: $R(t) = \langle \{A(t), A\}_{PB} \rangle = -\beta \langle \partial_t A(t) A \rangle$

with $\beta = 1/k_B T$

Generalized susceptibility: $\chi(\omega) \equiv R(\omega) = \int_0^\infty dt e^{i\omega t} R(t)$

Rate of energy dissipation/absorption:

$$Q_\omega \equiv \overline{\langle A(t) \rangle \frac{df}{dt}} = \frac{1}{2} \omega \chi''(\omega) |f_0|^2, \quad f(t) = \text{Re } f_0 e^{-i\omega t}$$

Fluctuation-Dissipation Theorem

Relates $R(t)$ and $C(t)$, namely:

$$\chi''(\omega) = (\beta\omega/2) C(\omega)$$

In the static limit ($t \rightarrow \infty$): $C(0) = \langle A^2 \rangle = k_B T R(0)$

Note: quantum corrections are important when $k_B T \leq \hbar\omega$

$$\chi''(\omega) = \hbar^{-1} \tanh(\beta\hbar\omega/2) C(\omega)$$

Diffusion Coefficient

Generic transport coefficient: $\gamma = \int_0^{\infty} dt \langle \partial_t A(t) \partial_t A(0) \rangle$

Einstein relation: $2\gamma t = \langle [A(t) - A(0)]^2 \rangle$

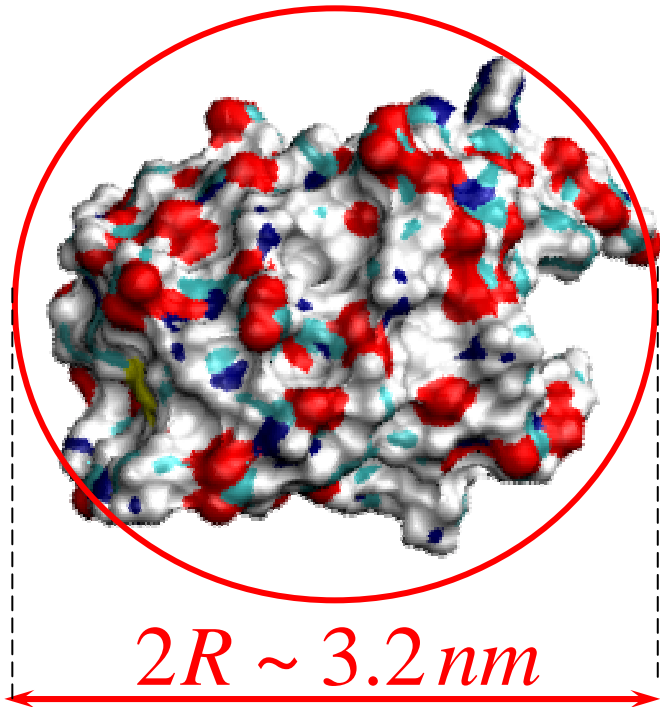
Example: *self-diffusion coefficient*

$$D = \frac{1}{3} \int_0^{\infty} dt \langle \mathbf{v}(t) \mathbf{v}(0) \rangle$$

$$6Dt = \langle [\mathbf{r}(t) - \mathbf{r}(0)]^2 \rangle$$

Free Diffusion (Brownian Motion) of Proteins

- ▶ in living organisms proteins exist and function in a viscous environment, subject to stochastic (random) thermal forces
- ▶ the motion of a globular protein in a viscous aqueous solution is diffusive



- ▶ e.g., *ubiquitin* can be modeled as a spherical particle of radius $R \sim 1.6 \text{ nm}$ and mass $M = 6.4 \text{ kDa} = 1.1 \times 10^{-23} \text{ kg}$

Free Diffusion of Ubiquitin in Water

► ubiquitin in water is subject to two forces:

- friction (viscous drag) force:

$$\mathbf{F}_f = -\gamma \mathbf{v} \quad \gamma = 6\pi\eta R \quad \text{(Stokes law)}$$

friction (damping) coeff

viscosity

- stochastic thermal (Langevin) force:

$$\mathbf{F}_L = \boldsymbol{\xi}(t) \quad \langle \boldsymbol{\xi}(t) \rangle = 0$$

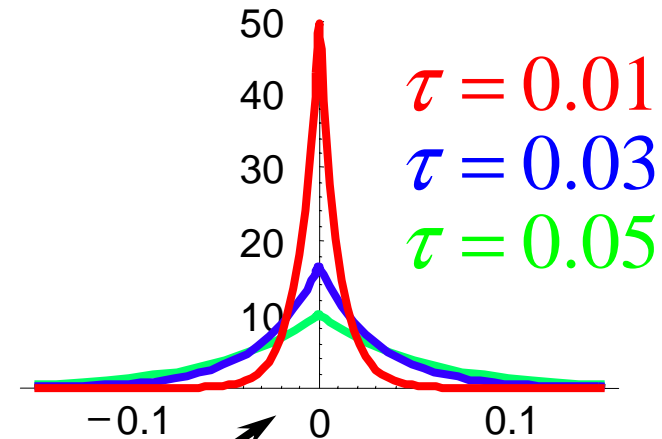
often modeled as a "Gaussian white noise"

$$\langle \boldsymbol{\xi}(t) \boldsymbol{\xi}(0) \rangle = 2\gamma k_B T \boldsymbol{\delta}(t)$$

$k_B = 1.38 \times 10^{-23} \text{ J/K}$ (Boltzmann constant), $T = \text{temperature}$

The Dirac delta function

$$\delta(t) = \begin{cases} \infty & \text{for } t = 0 \\ 0 & \text{for } t \neq 0 \end{cases}$$



In practice, it can be approximated as:

$$\delta(t) \approx \delta_\tau(t) = \frac{1}{2\tau} \exp(-|t|/\tau), \text{ as } \tau \rightarrow 0$$

$\Rightarrow \delta(t)$ describes $\tau=0$ correlation time ("white noise") stochastic processes

Useful formulas:

$$f(t) = \int_0^t f(t') \delta(t-t') dt'$$

$$\delta(at) = \delta(t) / |a|$$

Equation of Motion and Solution

Newton's 2nd law:

$$Ma = F_f + F_L \Rightarrow M \frac{dv}{dt} = -\gamma v + \xi(t)$$

Formal solution (using the variation of *const.* method):

$$v(t) = v_0 e^{-t/\tau} + \frac{1}{M} e^{-t/\tau} \int_0^t \xi(t') e^{t'/\tau} dt'$$

$$x(t) = x_0 + v_0 \tau (1 - e^{-t/\tau}) + \frac{\tau}{M} e^{-t/\tau} \int_0^t \xi(t') (1 - e^{(t'-t)/\tau}) dt'$$

$$\tau = \frac{M}{\gamma} = \text{velocity relaxation (persistence) time}$$

The motion is **stochastic** and requires statistical description formulated in terms of *averages & probability distributions*

Statistical Averages

$$\langle \xi(t) \rangle = 0 \quad \langle \xi(t) \xi(0) \rangle = 2\gamma k_B T \delta(t)$$

Exponential relaxation of x and v with characteristic time τ

$$\langle v(t) \rangle = v_0 e^{-t/\tau} \rightarrow 0 \quad \text{as } t \rightarrow \infty$$

$$\langle x(t) \rangle = x_0 + v_0 \tau (1 - e^{-t/\tau}) \rightarrow x_0 + v_0 \tau \quad \text{as } t \rightarrow \infty$$

$$\langle v^2(t) \rangle = \langle v(t) \rangle^2 + \frac{D}{\tau} (1 - e^{-t/\tau}) \rightarrow \frac{D}{\tau} \quad \text{as } t \rightarrow \infty$$

$$\langle x^2(t) \rangle = \langle x(t) \rangle^2 + 2Dt - 3D\tau + O(e^{-t/\tau})$$

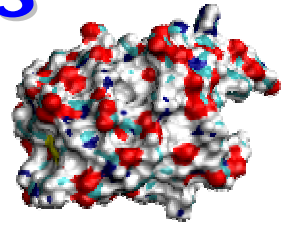
$$\Rightarrow \Delta x(t) = \sqrt{\langle x^2 \rangle - \langle x \rangle^2} = \sqrt{2Dt} \quad \text{as } t \rightarrow \infty$$

Diffusion coefficient:
(Einstein relation)

$$D = k_B T / \gamma$$

Typical Numerical Estimates

example: *ubiquitin* - small globular protein



mass: $M \approx 8.6 \text{ kDa} \approx 1.42 \times 10^{-23} \text{ kg}$, size: $R \approx 1.6 \text{ nm}$,

density: $\rho = M/V \approx 10^3 \text{ kg/m}^3$, temperature: $T = 310 \text{ K}$

Property	Theory	Simulation
$v_T = \sqrt{3k_B T / M}$	29.6 m/s	?
$\tau = \mathbf{M} / \gamma$	0.56 ps	?
$d = v_T \tau$	0.16 Å	?
γ	25.4 pN · s/m	?
$D = \frac{k_B T}{\gamma} = \frac{1}{3} v_T^2 \tau$	$1.6 \times 10^{-10} \text{ m}^2/\text{s}$?
$\eta = \gamma / 6\pi R$	0.9 mPa · s	?

Thermal and Friction Forces

- ▶ Friction force:

$$F_f = \gamma v_T \approx 4.5 \times 10^{-9} \text{ N} = 4.5 \text{ nN}$$

- ▶ Thermal force:

$$F_T = \sqrt{\frac{2\gamma k_B T}{\tau}} = \sqrt{\frac{2}{3}} \gamma v_T \sim F_f \approx 4.5 \text{ nN}$$

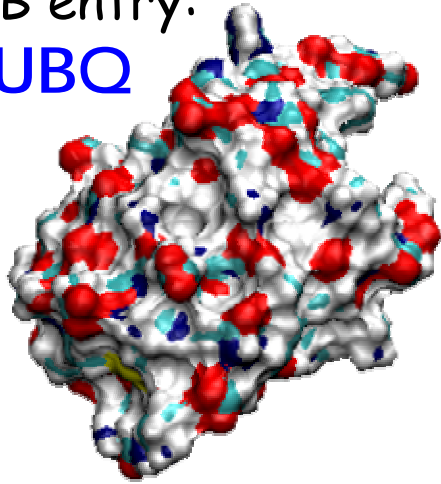
For comparison, the corresponding gravitational force:

$$F_g = Mg \sim 10^{-14} \text{ nN} \ll F_f \sim F_T$$

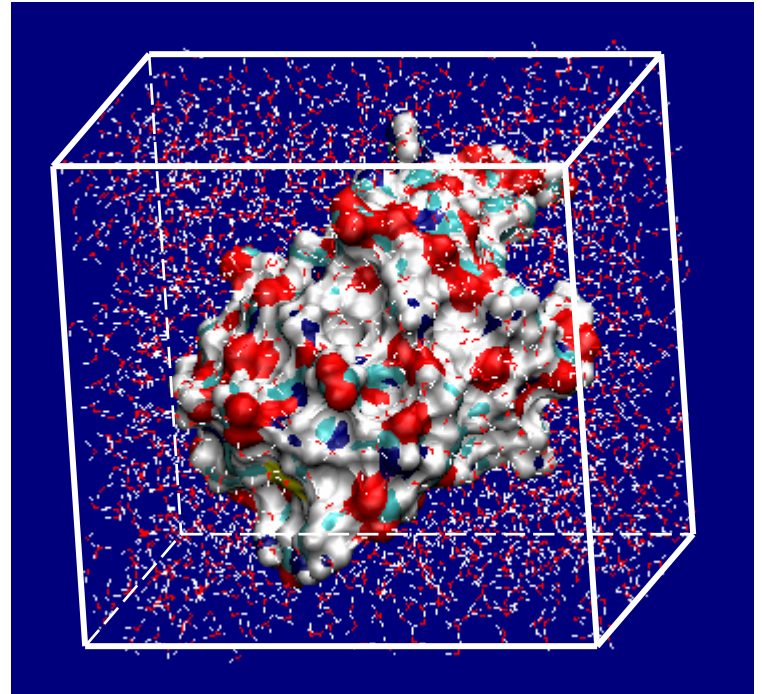
Diffusion can be Studied by MD Simulations!

ubiquitin in water

PDB entry:
1UBQ



solvate



total # of atoms: 7051 = 1231 (protein) + 5820 (water)

simulation conditions: NpT ensemble (T=310K, p=1atm),
periodic BC, full electrostatics, time-step 2fs (SHAKE)

simulation output: Cartesian coordinates and velocities of
all atoms saved at each time-step (10,000 frames = 40 ps)
in separate DCD files

How To: vel.dcd → vel.dat

- ▶ namd2 produces velocity trajectory (DCD) file if in the configuration file contains

```
velDCDfile  vel.dcd    ;# your file name
velDCDfreq  1          ;# write vel after 1 time-step
```
- ▶ load vel.dcd into VMD [e.g., `mol load psf ubq.psf dcd vel.dcd`]
note: run VMD in text mode, using the: `-dispdev text` option
- ▶ select only the protein (ubiquitin) with the VMD command

```
set ubq [atomselect top "protein"]
```
- ▶ source and run the following tcl procedure:

```
source v_com_traj.tcl
v_com_traj COM_vel.dat
```
- ▶ the file "COM_vel.dat" contains 4 columns:
time [fs], v_x , v_y and v_z [m/s]

```
70  12.6188434361 -18.6121653643 -34.7150913537
```

note: an ASCII data file with the trajectory of the COM coordinates can be obtained in a similar fashion

the v_COM_traj Tcl procedure

```
proc v_com_traj {filename {dt 2} {selection "protein"} {first_frame 0}
{frame_step 1} {mol top} args} {
    set outfile [open $filename "w"]
    set convFact 2035.4
    set sel [atomselect $mol $selection frame 0]
    set num_frames [molinfo $mol get numframes]
    for {set frame $first_frame} {$frame < $num_frames} {incr frame
    $frame_step} {
        $sel frame $frame
        set vcom [vecscale $convFact [measure center $sel weight mass]]
        puts $outfile "$frame\t $vcom"
    }
    close $outfile
}
```

Goal: calculate D and τ

by fitting the theoretically calculated center of mass (COM) velocity autocorrelation function to the one obtained from the simulation

► **theory:** $C_{vv}(t) = \langle v(t)v(0) \rangle = \langle v_0^2 \rangle e^{-t/\tau}$

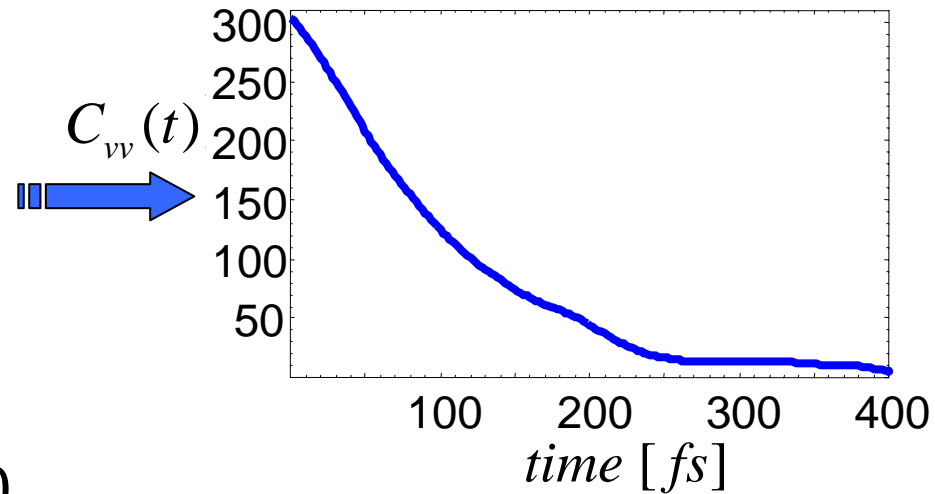
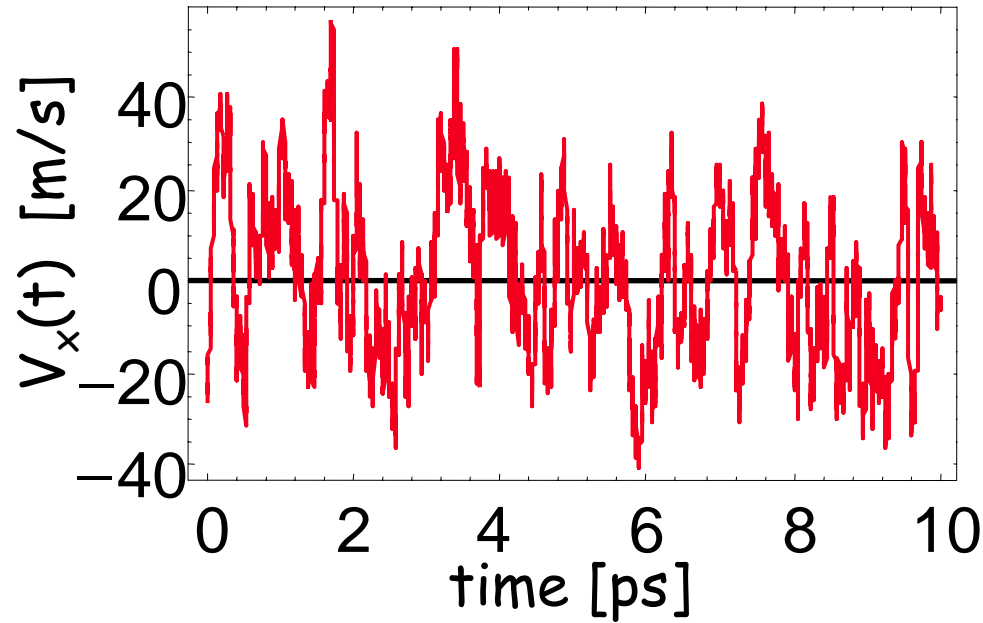
$$\langle v_0^2 \rangle = \frac{k_B T}{M} = \frac{D}{\tau} \quad (\text{equipartition theorem})$$

► **simulation:** consider only the x-component ($v_x \rightarrow v$)
replace ensemble average by time average

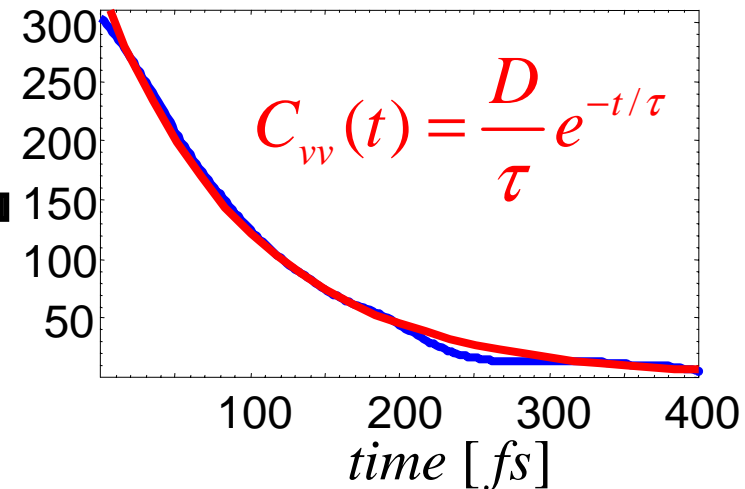
$$C_{vv}(t) \approx C_i = \frac{1}{N-i} \sum_{n=1}^{N-i} v_{n+i} v_n$$

$t \equiv t_i = i\Delta t$, $v_n = v(t_n)$, $N = \#$ of frames in vel.DCD

Velocity Autocorrelation Function

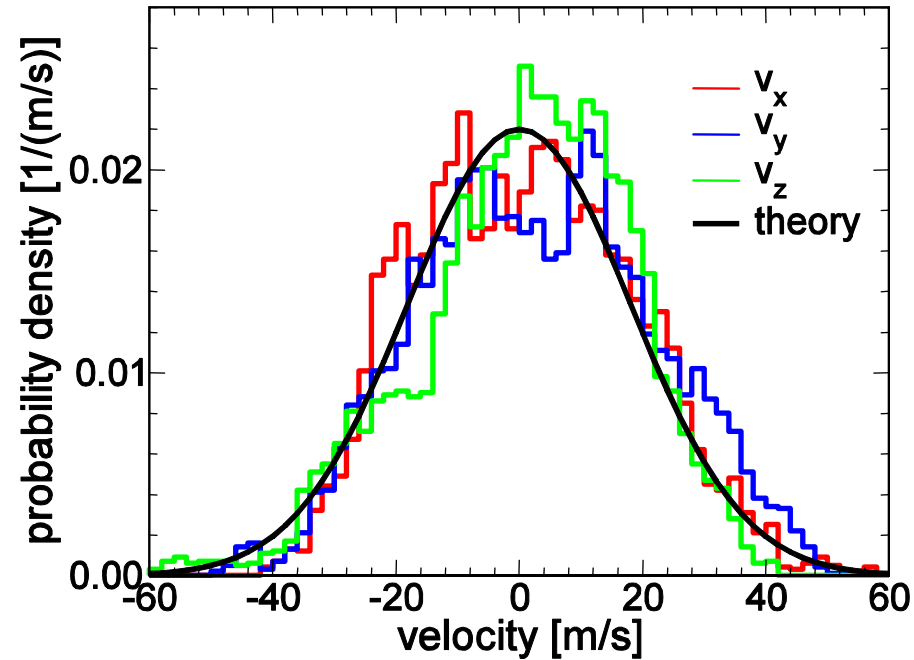
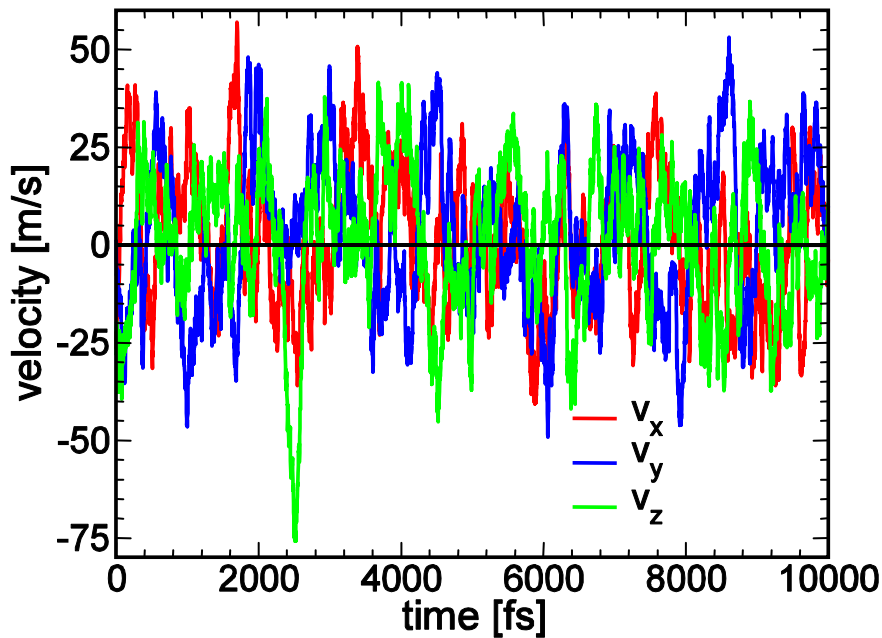


Fit



$$\tau \approx 0.1 \text{ ps}$$
$$D = k_B T / \gamma =$$
$$= \langle v_x^2 \rangle \tau \approx 3.3 \times 10^{-11} \text{ m}^2 \text{ s}^{-1}$$

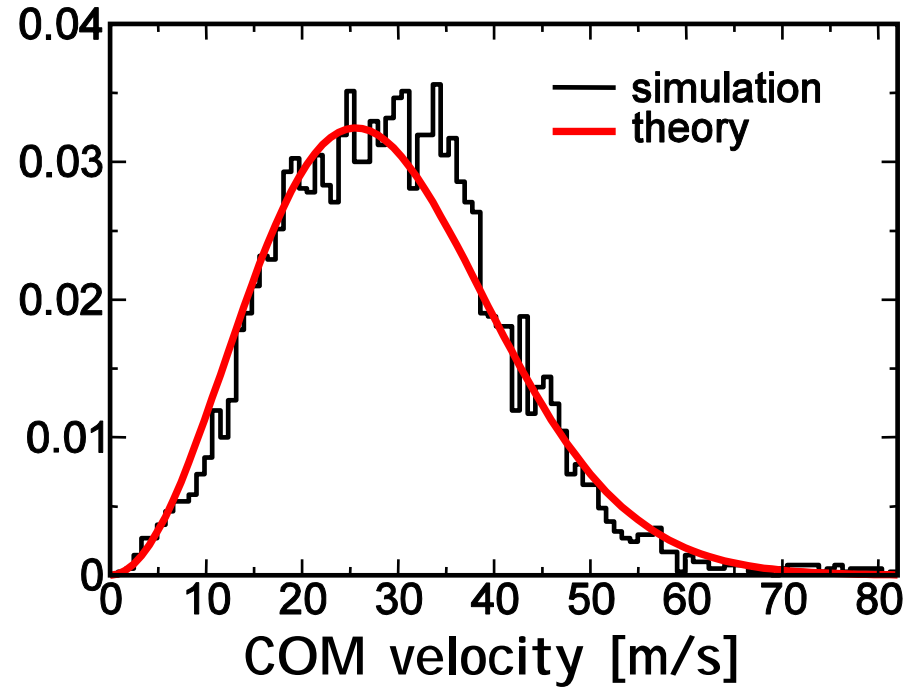
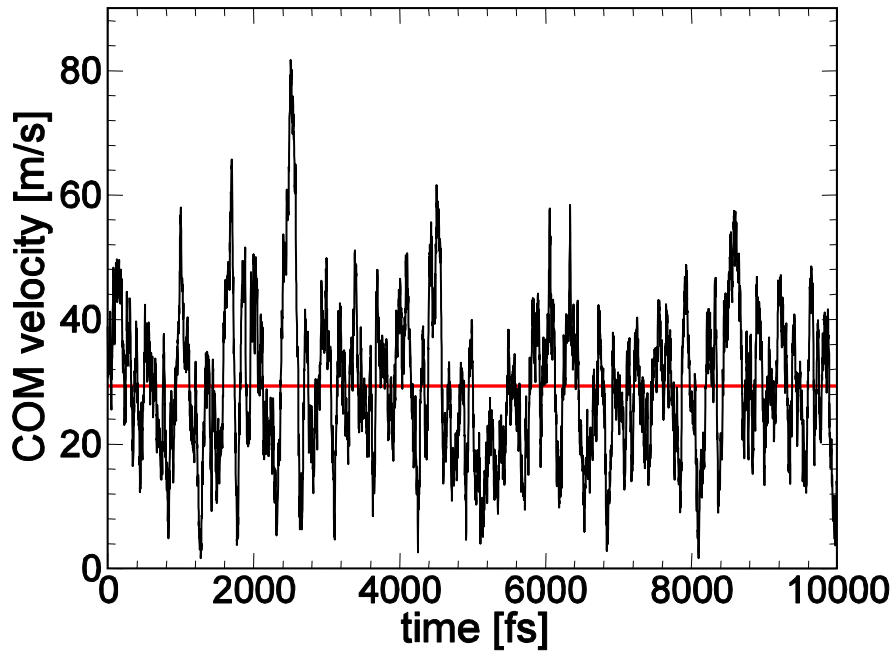
Probability distribution of $v_{x,y,z}$



$$p(v) = \left(2\pi \langle v^2 \rangle\right)^{-1/2} \exp\left(-v^2/2 \langle v^2 \rangle\right)$$
$$= \sqrt{\tau/2\pi D} \exp\left(-\tau v^2/2D\right)$$

with $v \equiv v_{x,y,z}$

Maxwell distribution of v_{COM}



$$\begin{aligned} P(\mathbf{v})d\mathbf{v} &= p(v_x)p(v_y)p(v_z)dv_xdv_ydv_z \\ &= (\pi 2\pi D)^{3/2} \exp(-\tau v^2/2D) 4\pi v^2 dv \\ &= \sqrt{\frac{2}{\pi}} \left(\frac{M}{k_B T} \right)^{3/2} v^2 \exp\left(-\frac{Mv^2}{2k_B T}\right) dv \end{aligned}$$

What have we learned ?

soluble, globular proteins in aqueous solution at physiological temperature execute free diffusion (Brownian motion with typical parameter values:

Property	Theory	Simulation
$v_T = \sqrt{3k_B T / M}$	29.6 m/s	31.6 m/s
$\tau = M / \gamma$	0.56 ps	0.1 ps
$d = v_T \tau$	0.16 Å	0.03 Å
γ	25.4 pN · s/m	141.6 pN · s/m
$D = \frac{k_B T}{\gamma} = \frac{1}{3} v_T^2 \tau$	$1.6 \times 10^{-10} \text{ m}^2/\text{s}$	$0.3 \times 10^{-10} \text{ m}^2/\text{s}$
$\eta = \gamma / 6\pi R$	0.9 mPa · s	4.7 mPa · s

How about the motion of parts of the protein ?

- ▶ parts of a protein (e.g., side groups, a group of amino acids, secondary structure elements, protein domains, ...), besides the viscous, thermal forces are also subject to a resultant force from the rest of the protein
- ▶ for an effective degree of freedom x (reaction coordinate) the equation of motion is

$$m\ddot{x} = -\gamma\dot{x} + f(x) + \xi(t)$$

In the harmonic approximation $f(x) \approx -kx$
and we have a 1D Brownian oscillator