

# Molecular Dynamics Simulation of Membrane Channels

## Part I. Overview and Examples

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# Molecular Dynamics Simulation of Membrane Channels

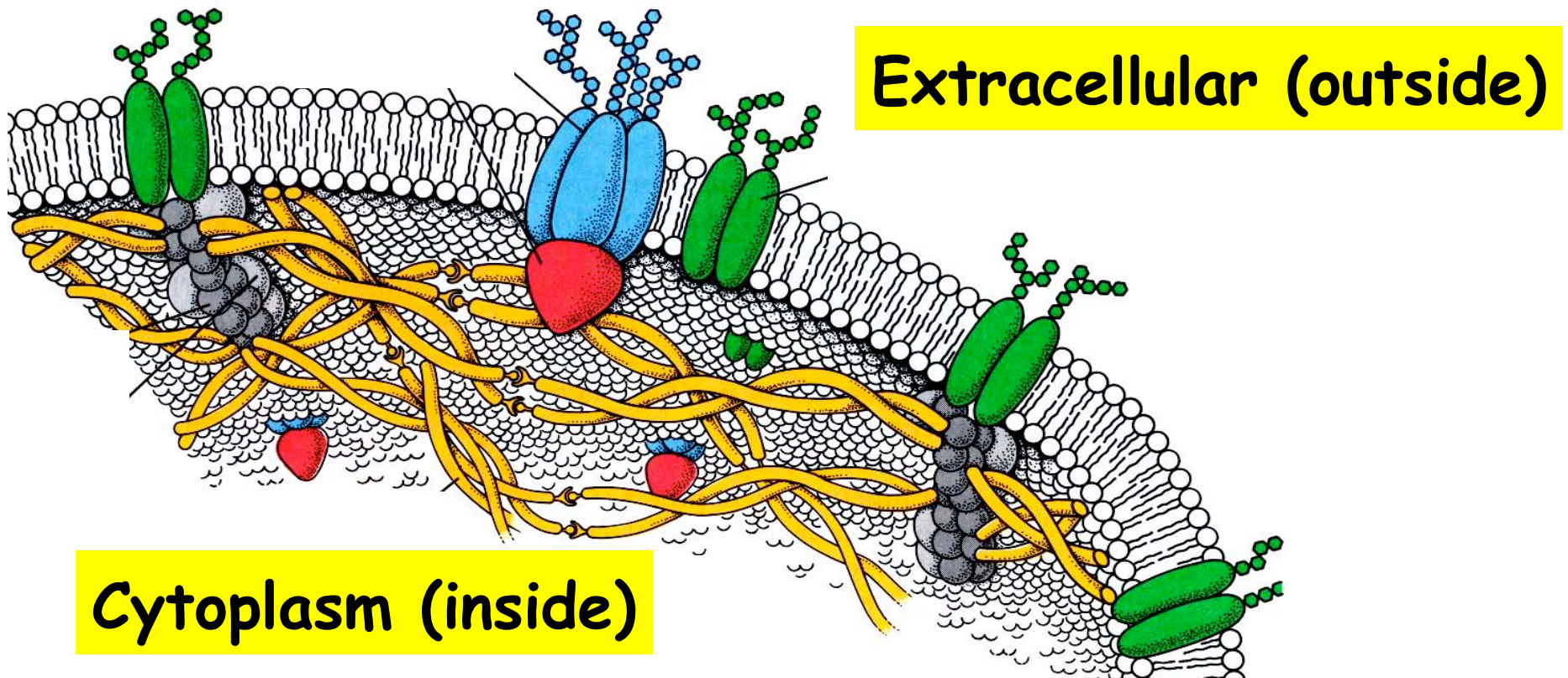
- Brief Introduction to Membrane and a few examples of Membrane Channels
- Aquaporin Water Channels
  - How to model membrane proteins in membrane
  - How to analyze the data? Where to look?
  - How much we can learn from simulations?
- Nanotubes and today's tutorial
  - Nanotubes as simple models for membrane water channels
  - Theory of water transport and its modeling using MD simulations

# Why Do Living Cells Need Membranes and Membrane Channels?

- **Water** is the medium of life: water is the most abundant compound in living cells/organisms, and all biochemical reactions take place in water.
- Living cells need to isolate their interior compartment from the environment, a task that, in a water-dominated medium, can be best done by fatty molecules.

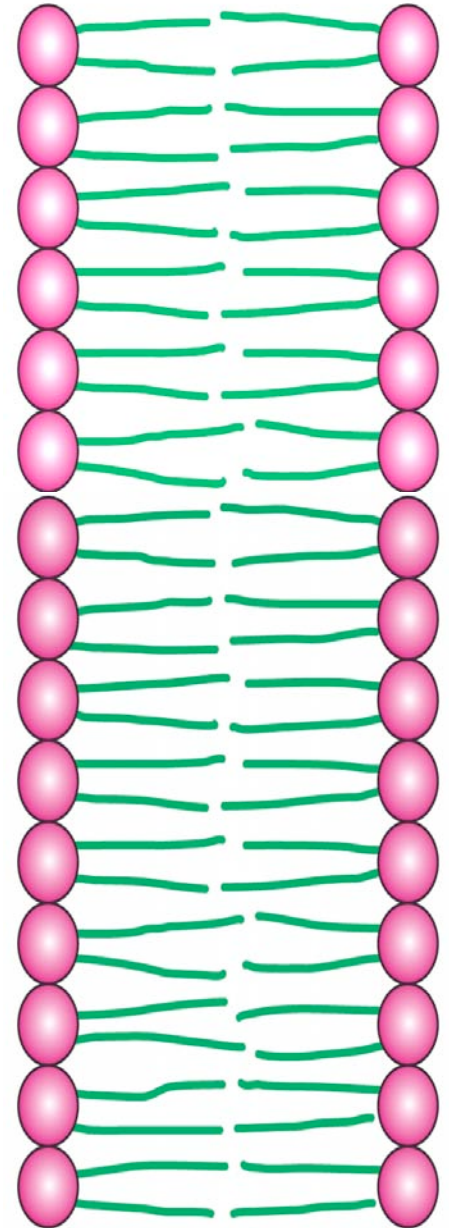
# The Roles of Cell Membranes

- Conservation of materials inside the cell
- Protection against undesired substances
- But Access for desired substances!



# Lipid Bilayers are Excellent for Cell Membranes

- Hydrophobic interaction is the driving force for their formation
- They self-assemble in water
- They tend to close on themselves
- They self-seal (holes unfavorable)
- They can be extensive: up to mm

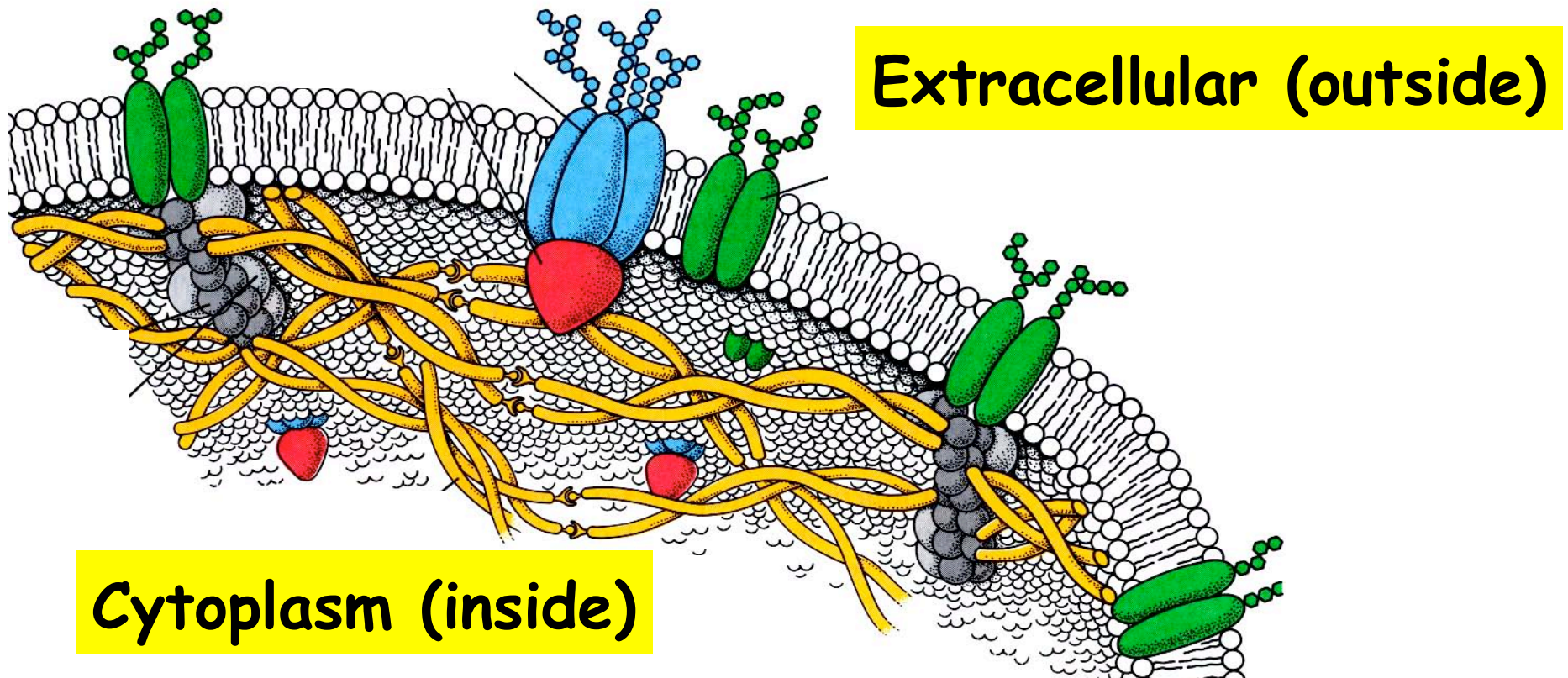




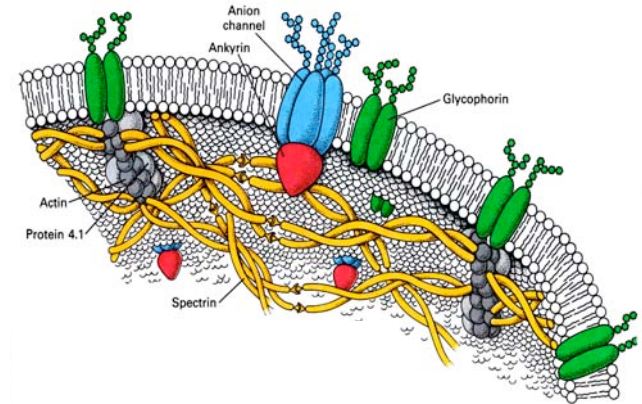
# Why Do Living Cells Need Membrane Channels (Proteins)?

- Living cells need to exchange materials and information with the outside world

... however, in a highly selective manner.



# Proteins in Membranes



- As receptors, detecting signals from outside:

Light

Odorant

Taste

Chemicals

Hormones

Neurotransmitters

Drugs

- As channels, forming gates and pumps
- As generators of electric/chemical potentials

Energy storage

Neurophysiology

- As energy transducers

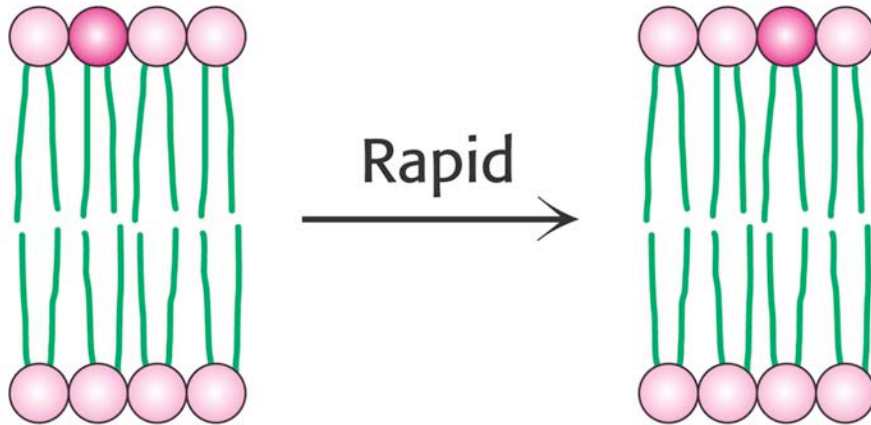
Photosynthesis

Oxidative phosphorylation

**A highly selective  
permeability  
barrier**

Internal  
membranes for  
organelles

# Lipid Diffusion in Membrane

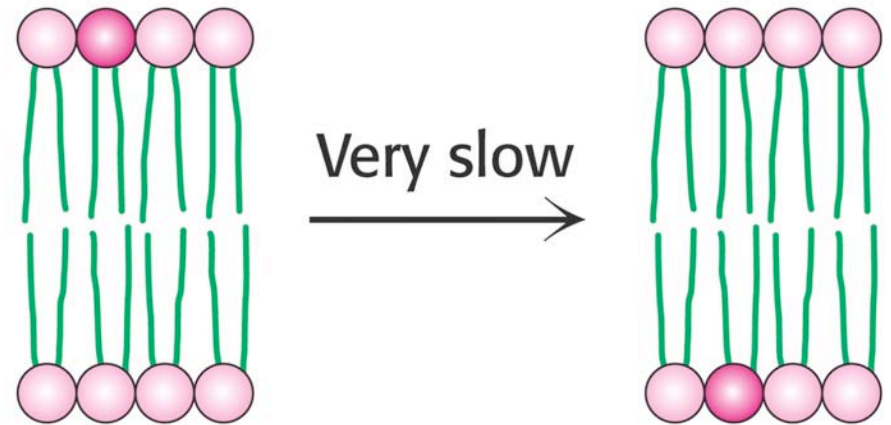


Lateral diffusion

$$D = 1 \mu\text{m}^2 \cdot \text{s}^{-1}$$

$$50 \text{ \AA} \text{ in } \sim 2.5 \times 10^{-5} \text{ s}$$

$$D_{\text{lip}} = 10^{-8} \text{ cm}^2 \cdot \text{s}^{-1}$$
$$D_{\text{wat}} = 2.5 \times 10^{-5} \text{ cm}^2 \cdot \text{s}^{-1}$$



Transverse diffusion  
(flip-flop)

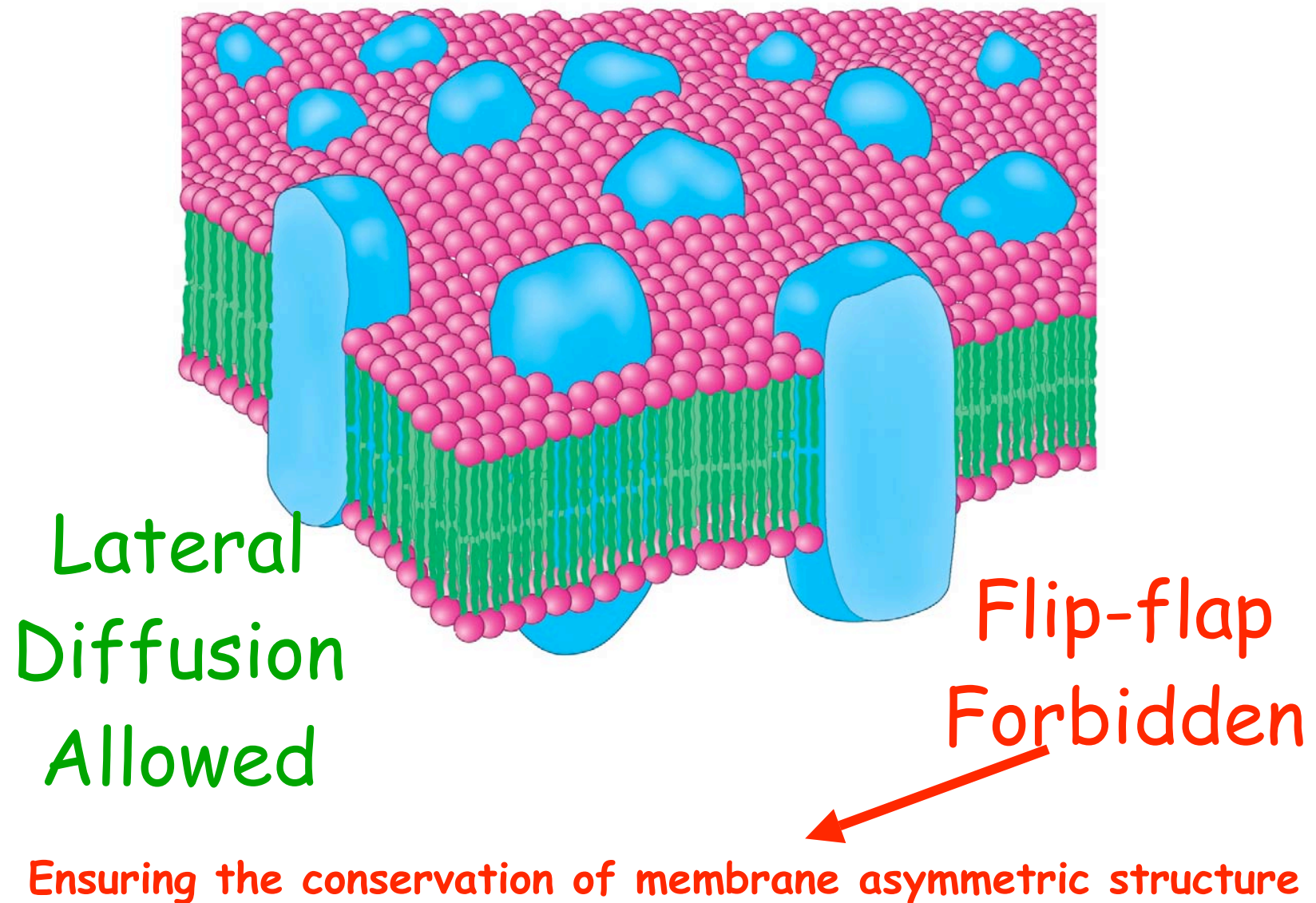
Once in several hours!

$$(10^4 \text{ s})$$

**~9 orders of magnitude  
difference**

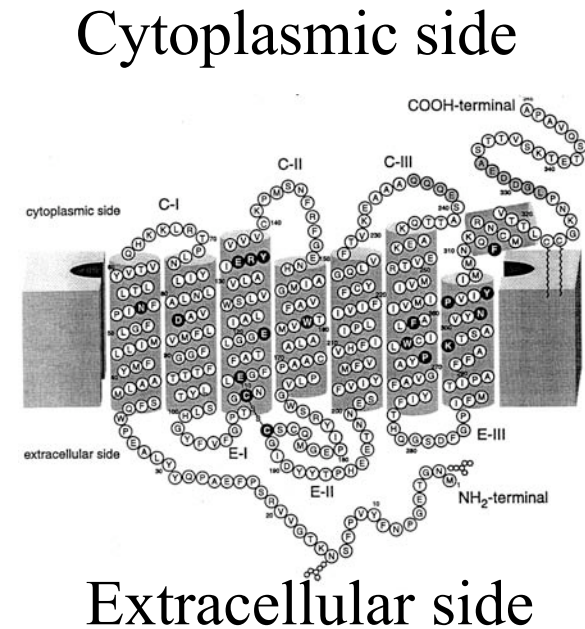
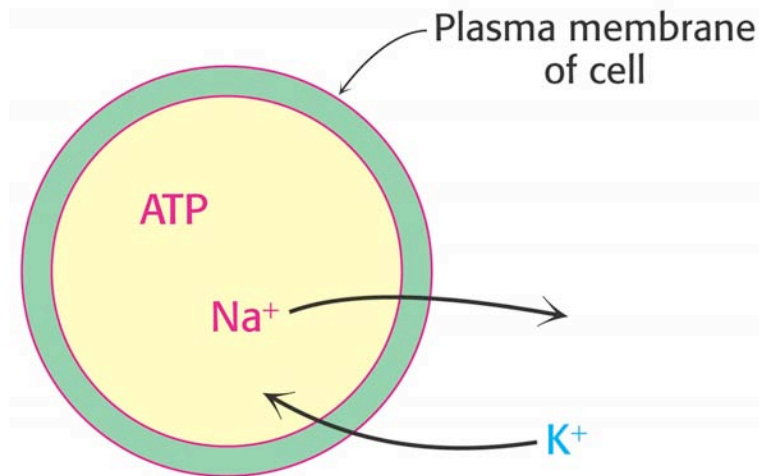


# Fluid Mosaic Model of Membrane



# Importance of Asymmetry

Apart from passive transport mechanisms, all membrane proteins function in a directed fashion, and their correct insertion into the cell membrane is essential for their biological function.

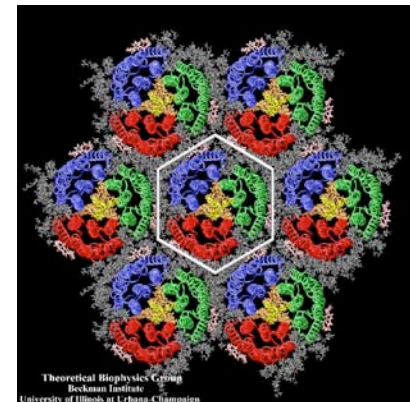


# Protein/Lipid ratio

- Pure lipid: insulation (neuronal cells)
- Other membranes: on average 50%
- Energy transduction membranes (75%)

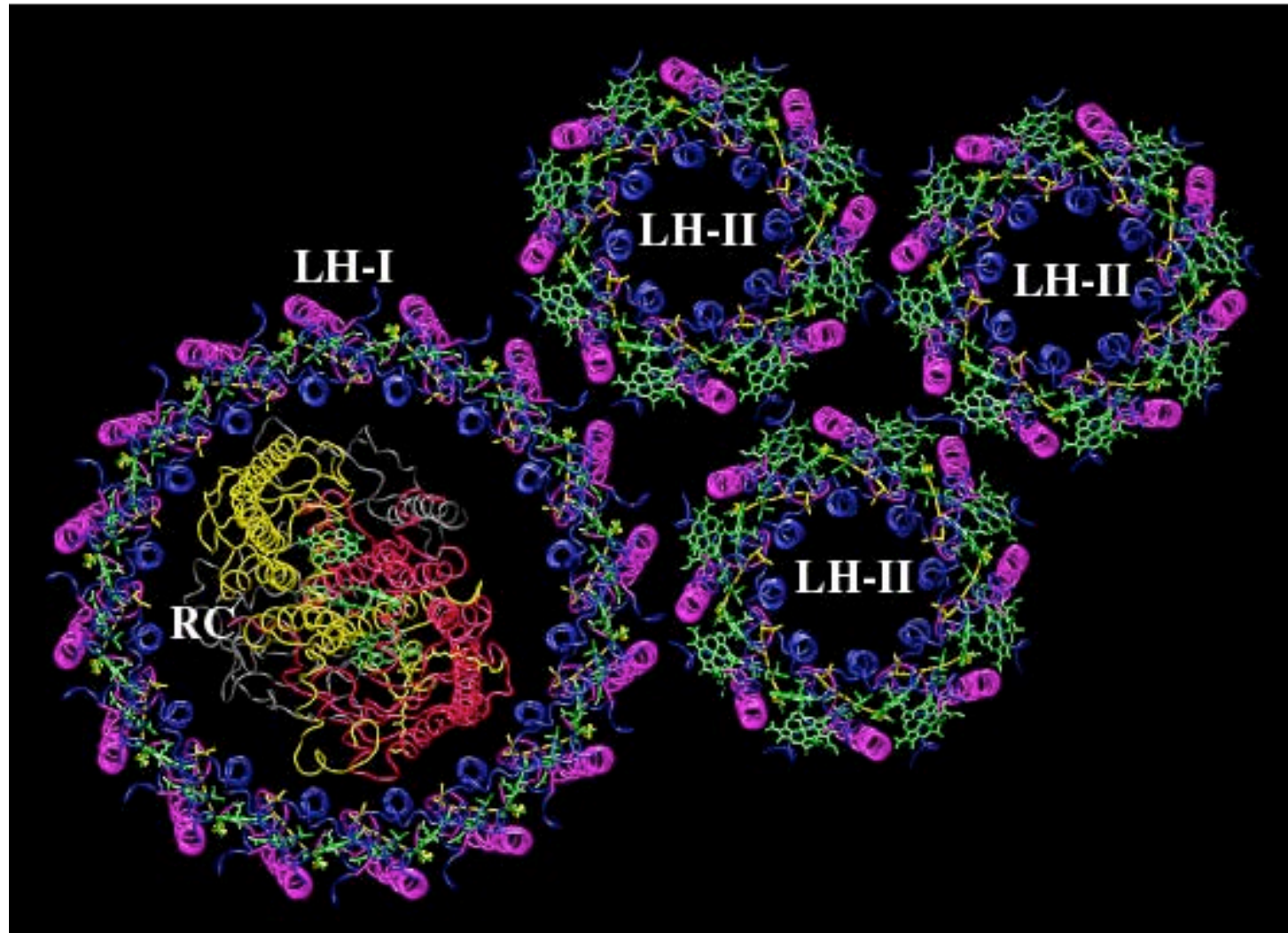
Membranes of mitochondria and chloroplast

Purple membrane of halobacteria



- Different functions = different protein composition

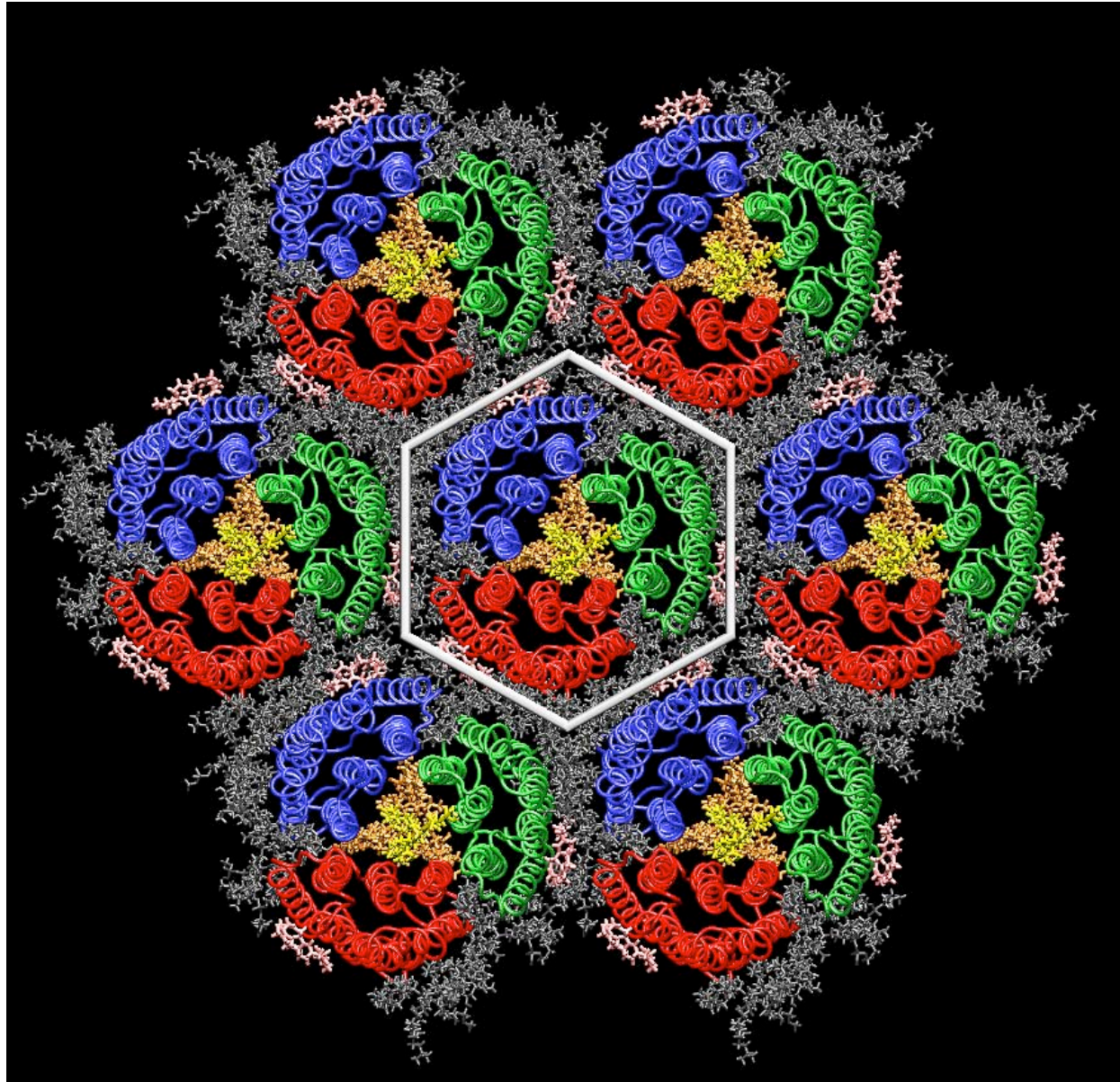
# Protein / Lipid Composition



Light harvesting complex of purple bacteria



# Protein / Lipid Composition

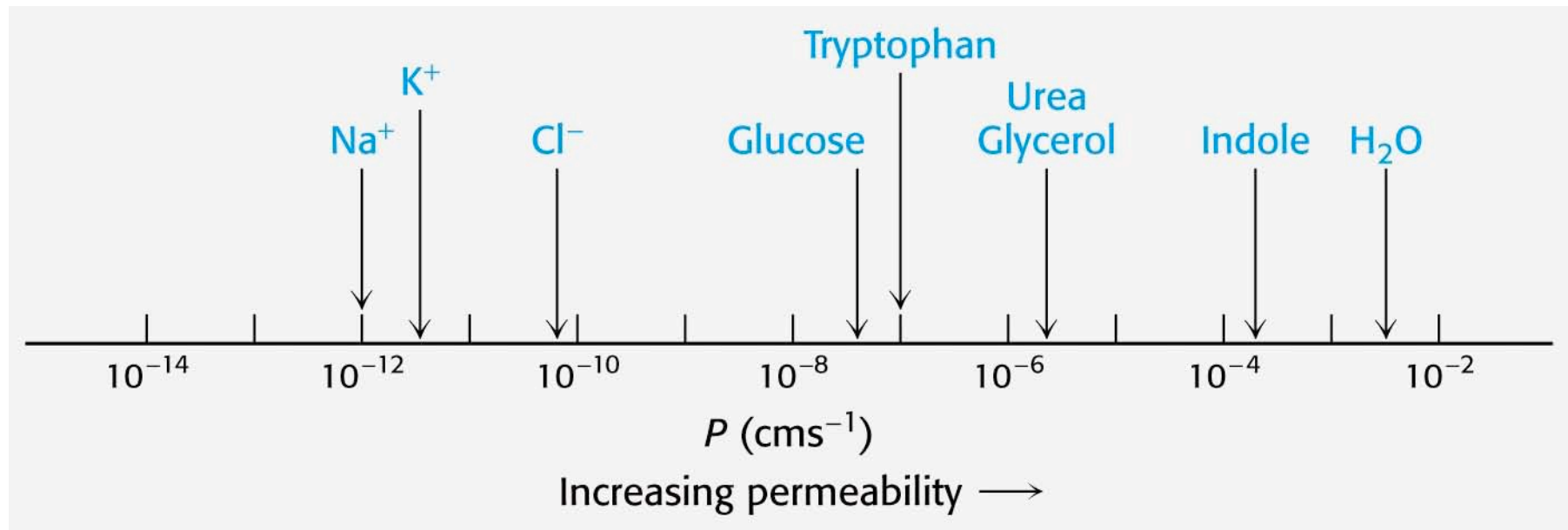


The purple membrane of halobacteria

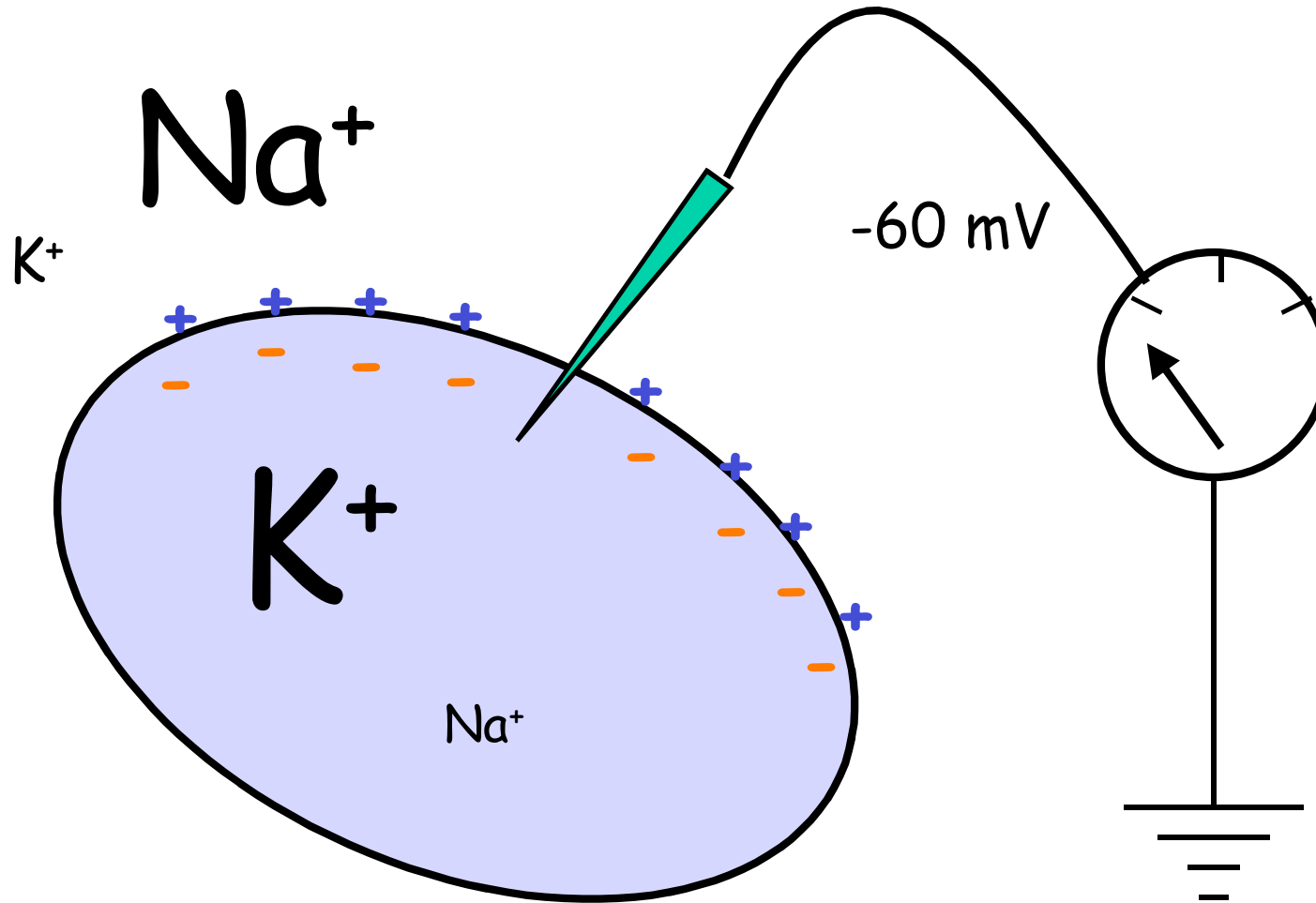


# Bilayer Permeability

- Low permeability to charged and polar substances
- **Water** is an exception: small size, lack of charge, and its high concentration
- **Desolvation of ions is very costly.**



# Membrane Electrical Potential



The ratio of ions is about 1 to 10

Action potential in excitable cells

# Properties of Ion Channels

Membrane-spanning protein

Hydrophilic ion conductive pathway

Water-filled

Traversing ion must lose hydration shell

Selective

charge screening and size

Gating properties

Exist in open and closed states

Substrate is charged and the conduction can be measured very precisely, as opposed to water channels.

# Control of conduction in ion channels

Gating mechanisms (open-closed transition) due to

Membrane potential change (voltage gated channels)

K channels

Binding of a molecule (ligand-gated channels)

Acetylcholine nicotinic receptor (Na channel)

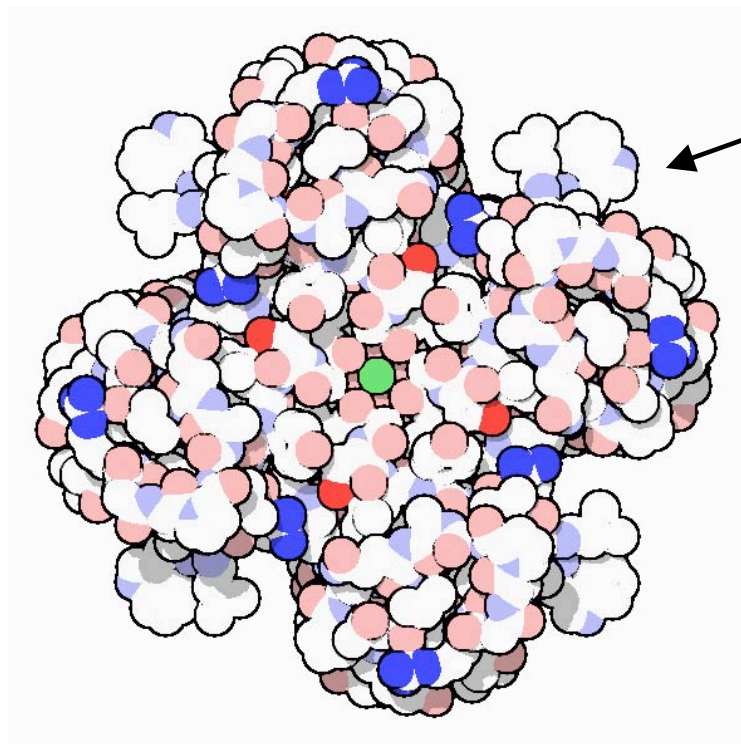
Glutamate receptor (Ca channel)

Voltage and ligand gating can co-exist

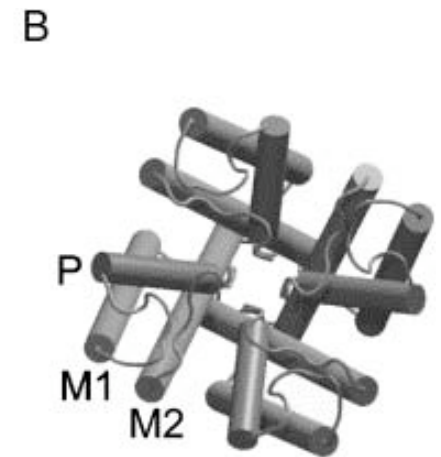
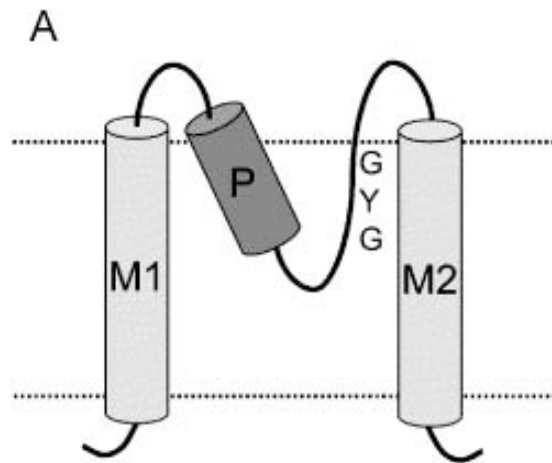
Mechanical gating (MscL)

# KcsA Potassium Channel

Under physiological conditions, the selectivity filter of the KcsA dehydrates, transfers, and rehydrates one  $K^+$  ion every 10 ns.

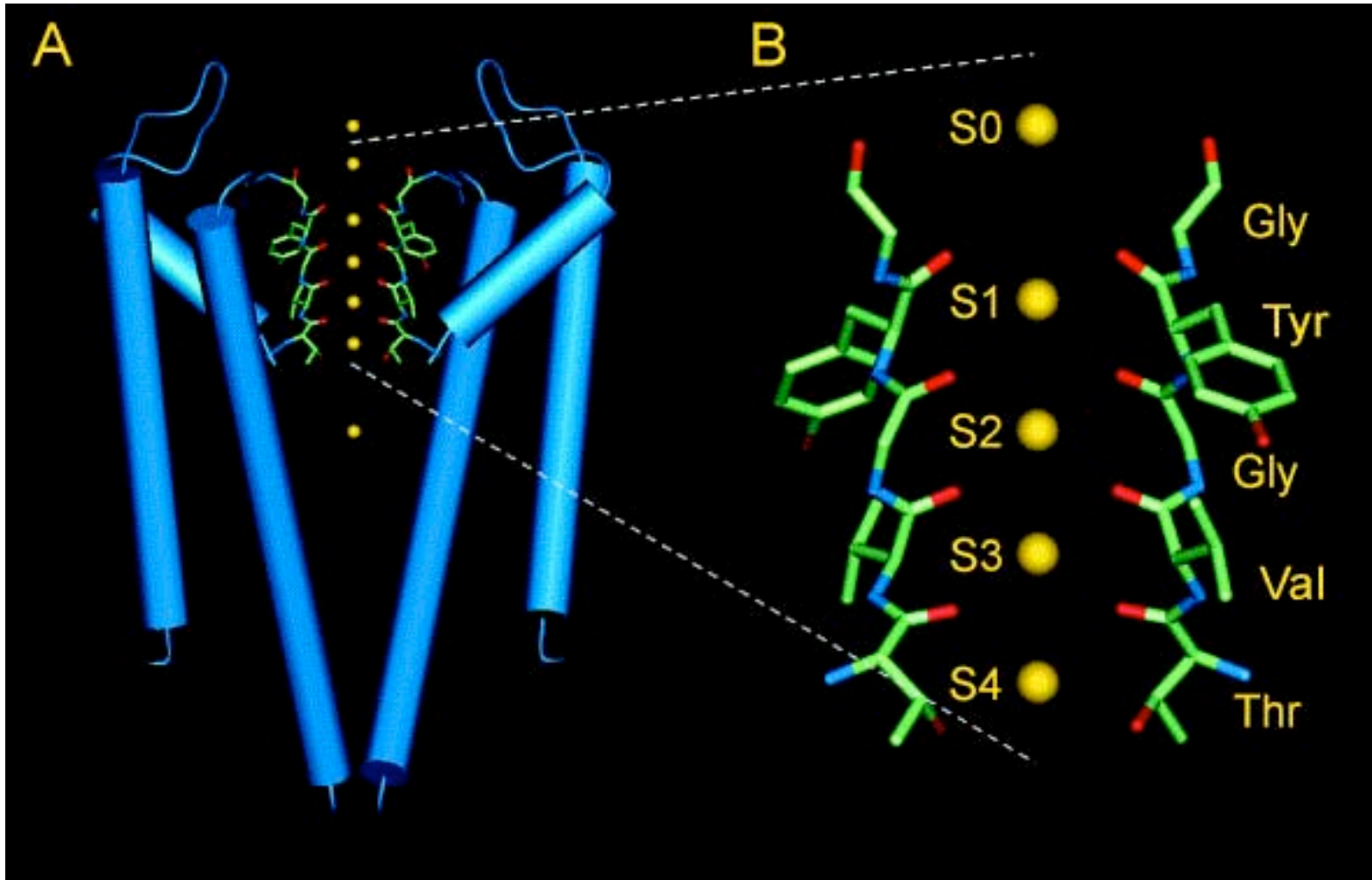


PDB Feb 2003  
molecule of the month

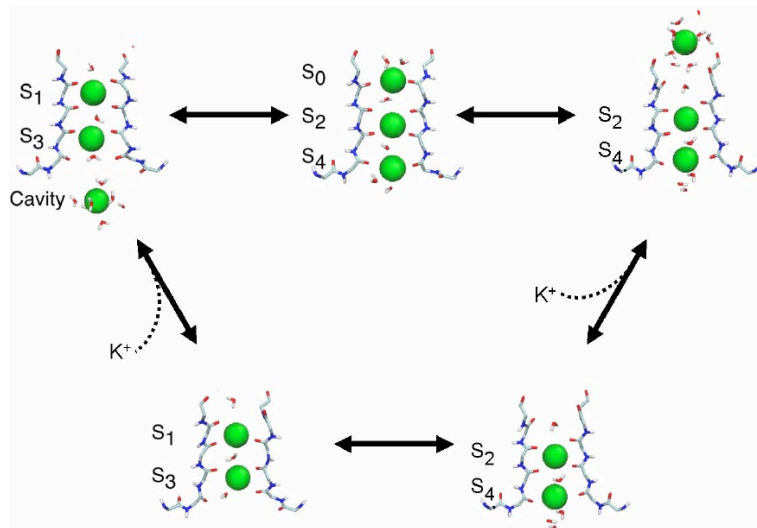
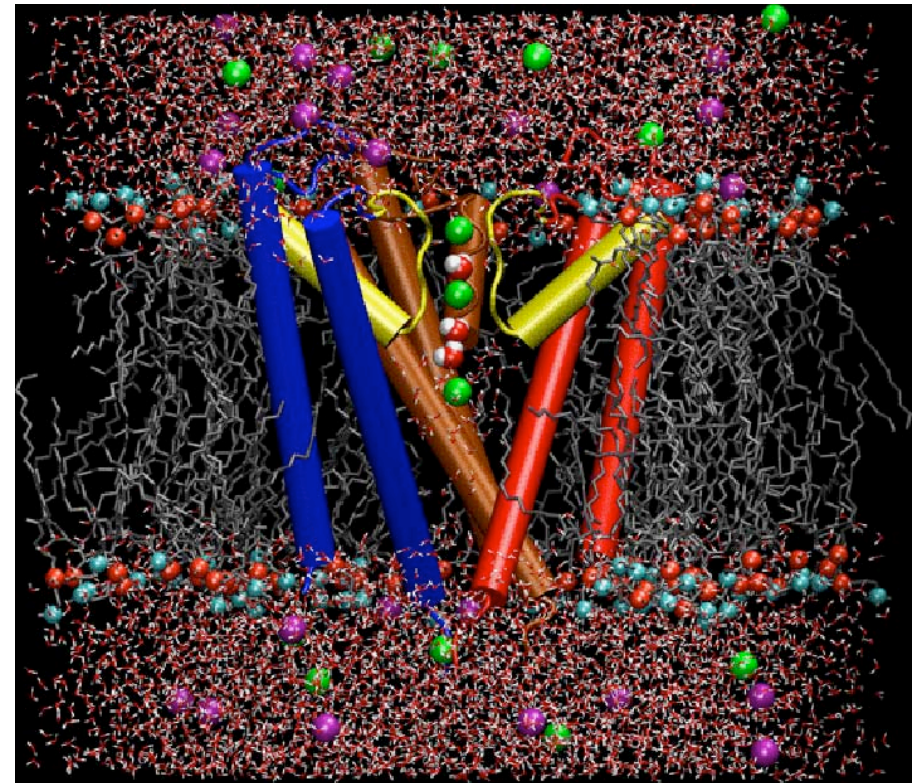
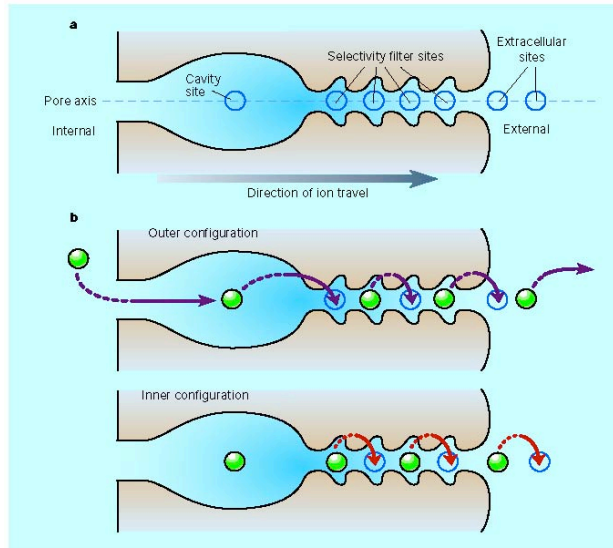




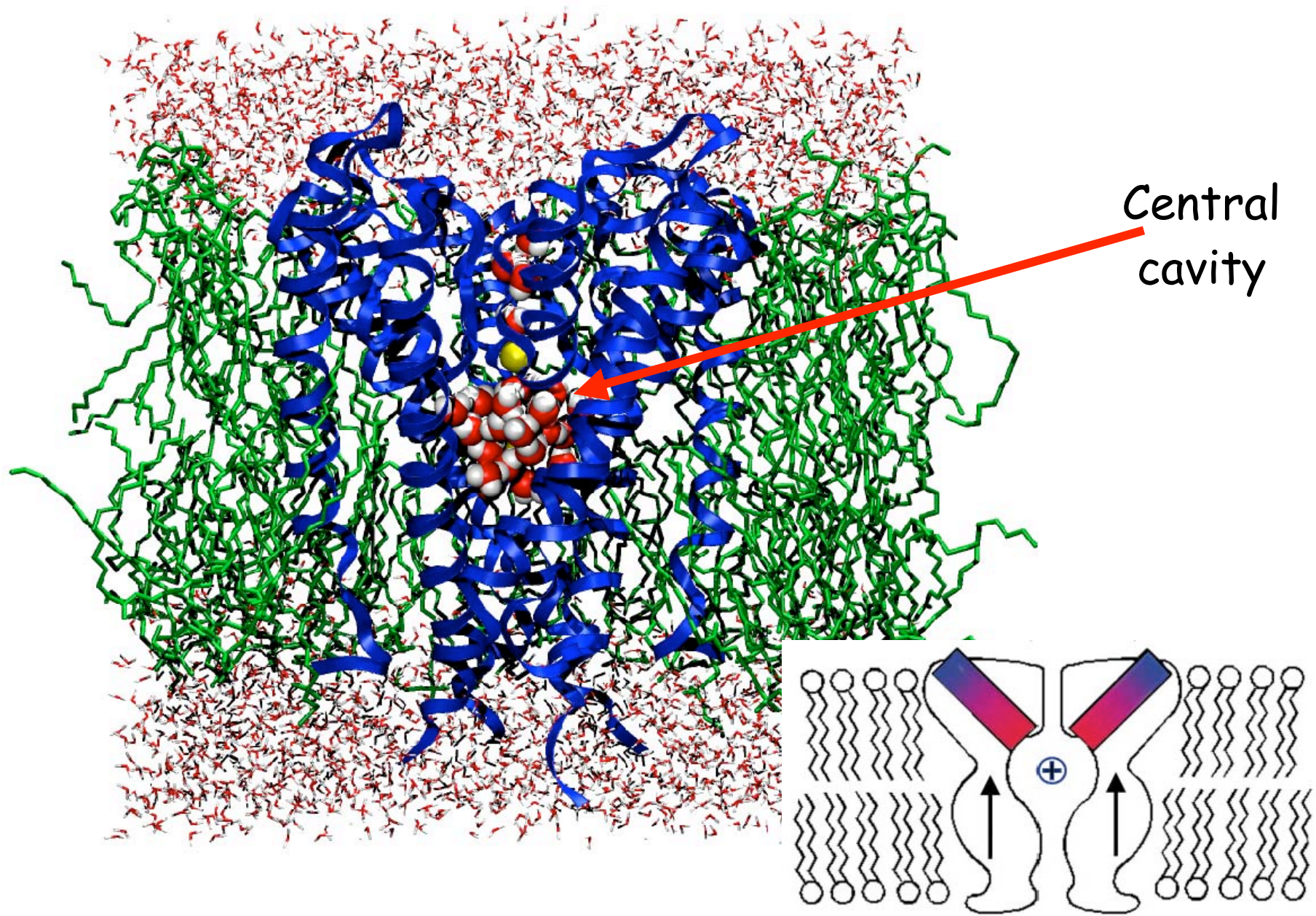
# K binding sites in the selectivity filter



# Overall Mechanism of Ion Conduction



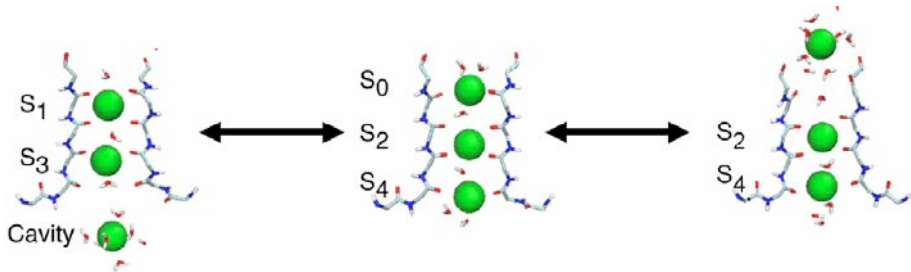
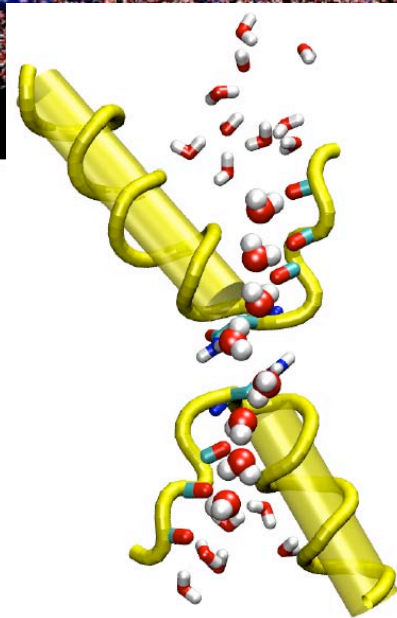
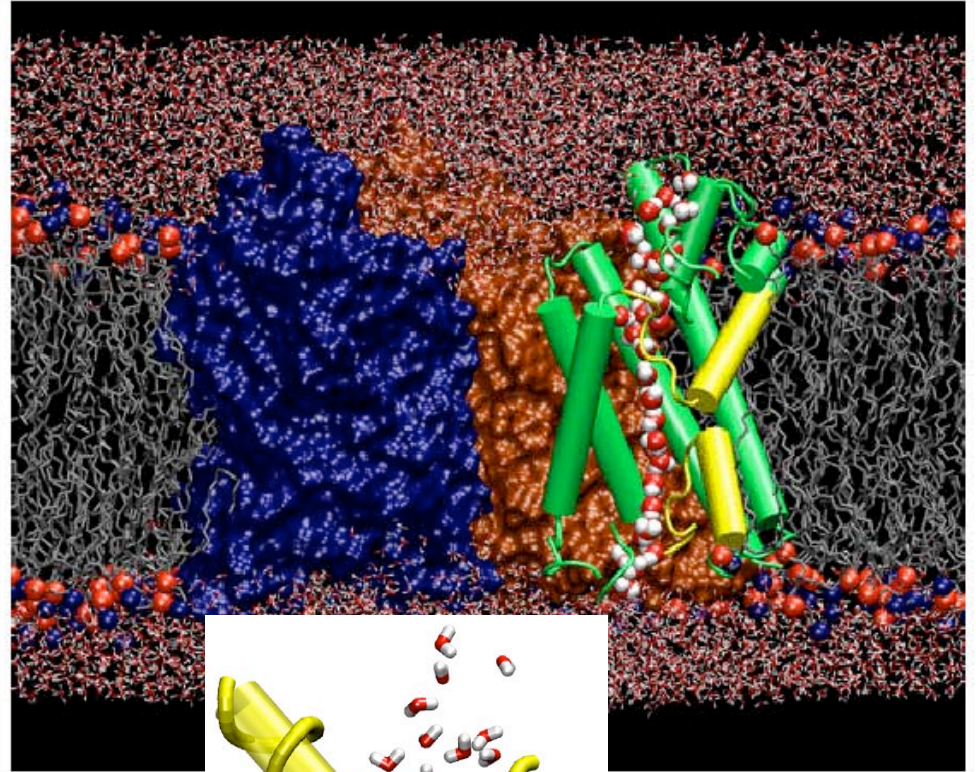
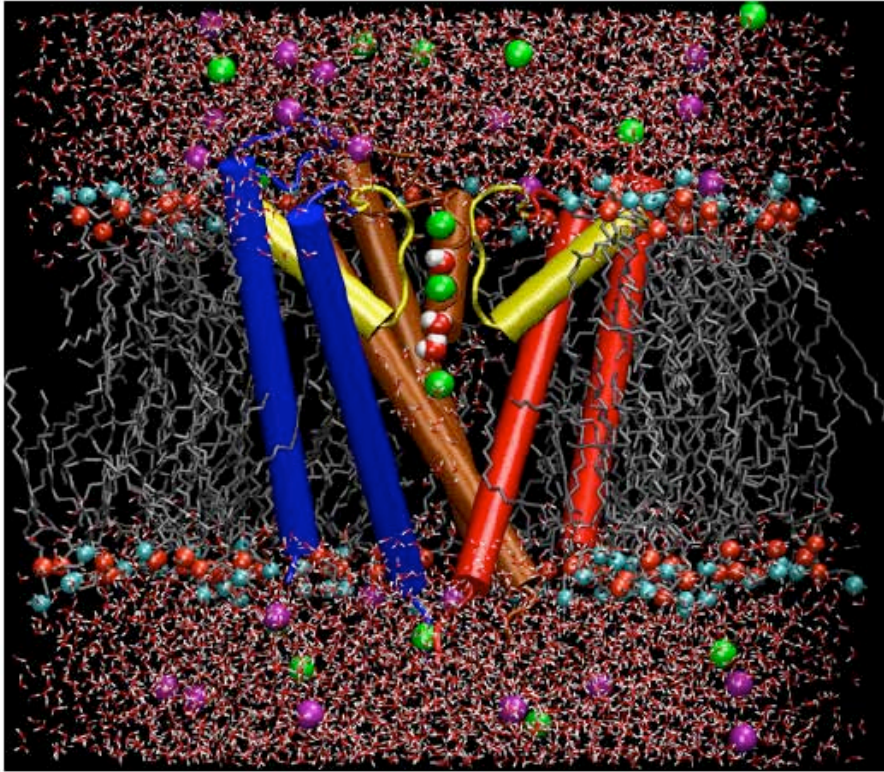




Central  
cavity

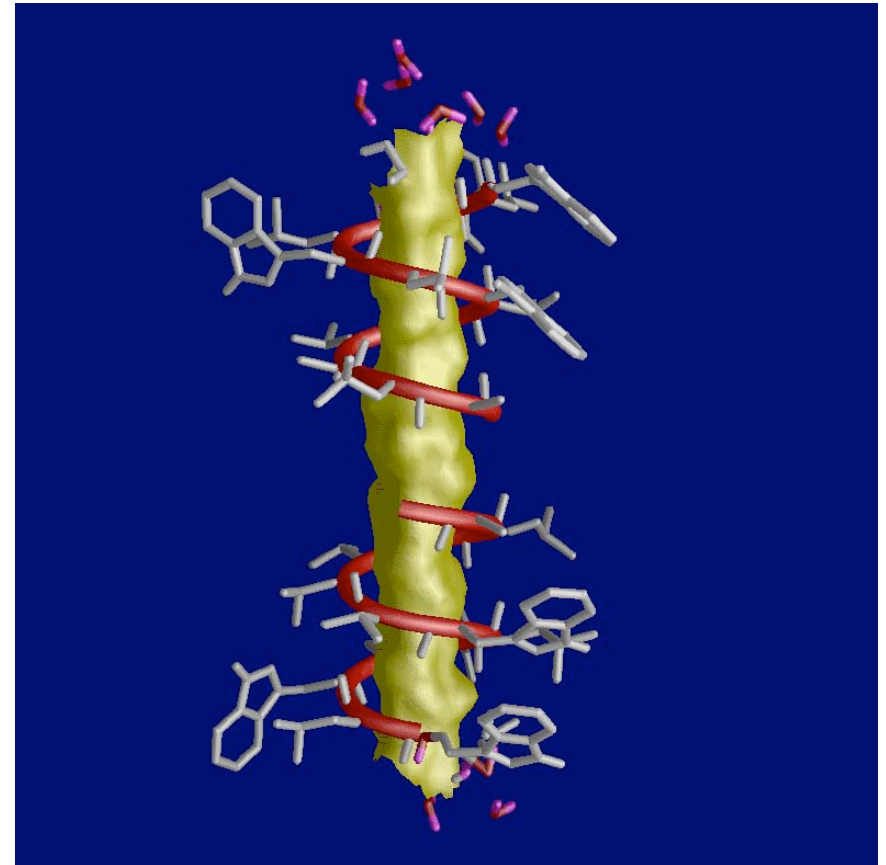
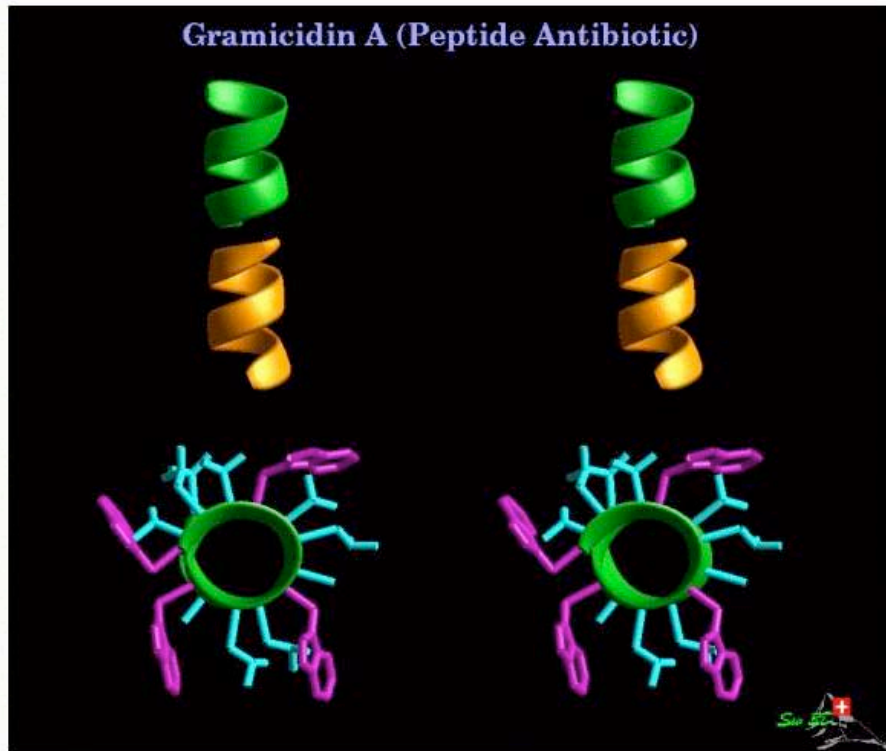


# Comparison of Potassium Channel and Water Channel



# Gramicidin A

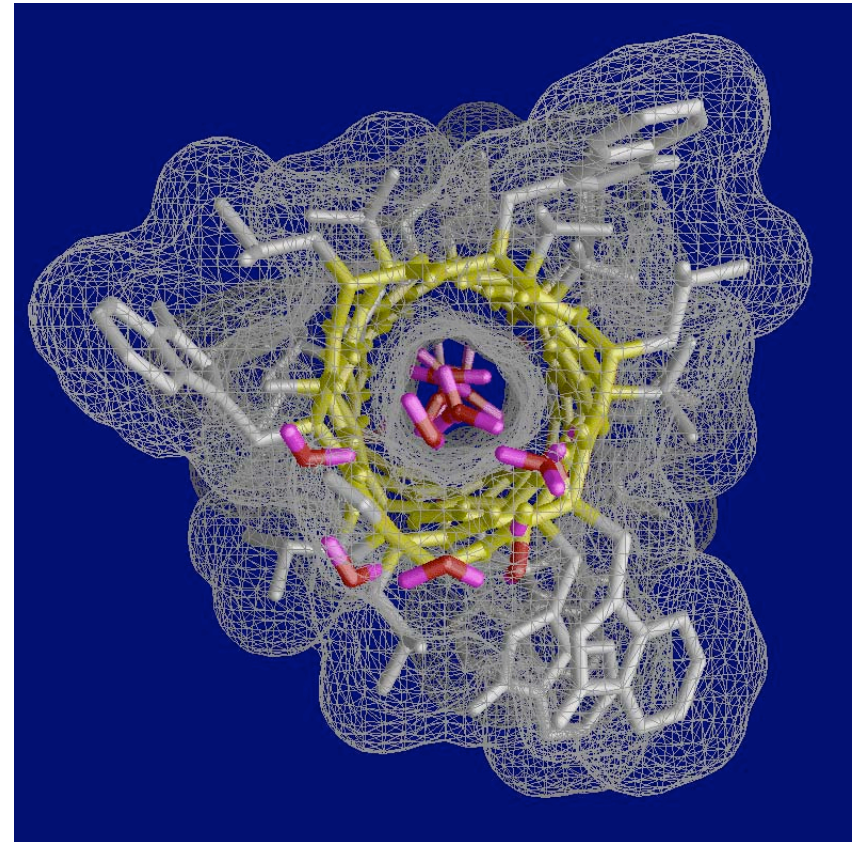
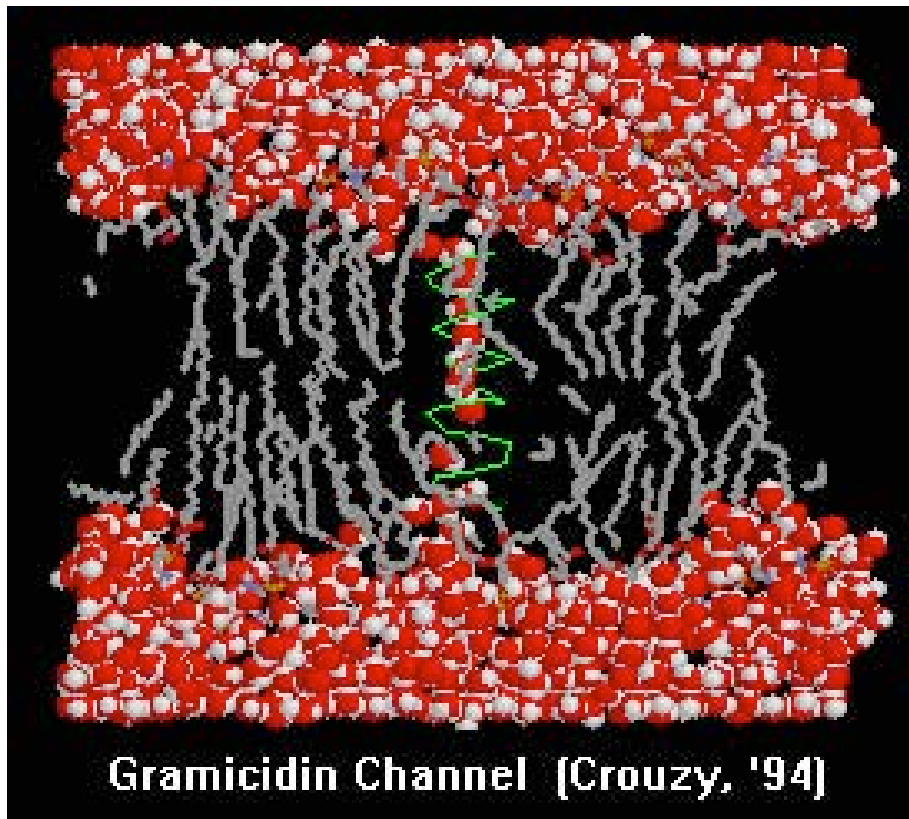
an ion leak inside the membrane



Through dissipating the electrochemical potential of membranes, gramicidin A acts as an antibiotic.



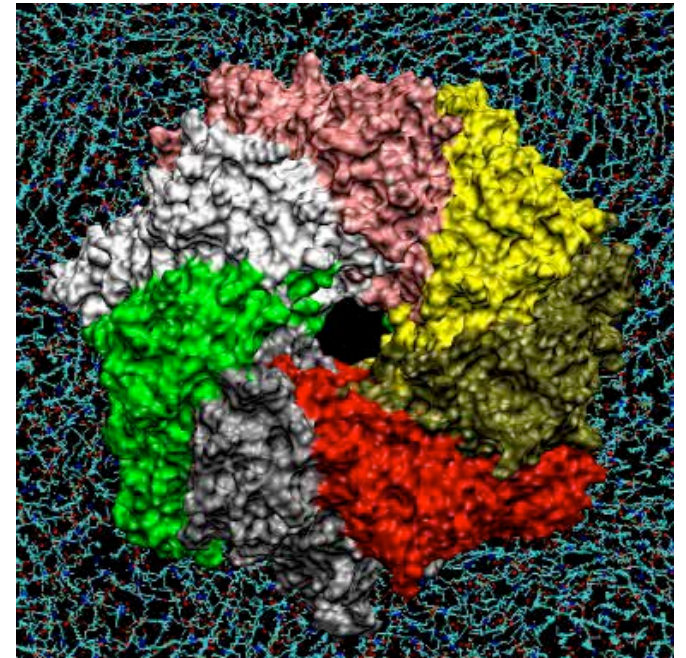
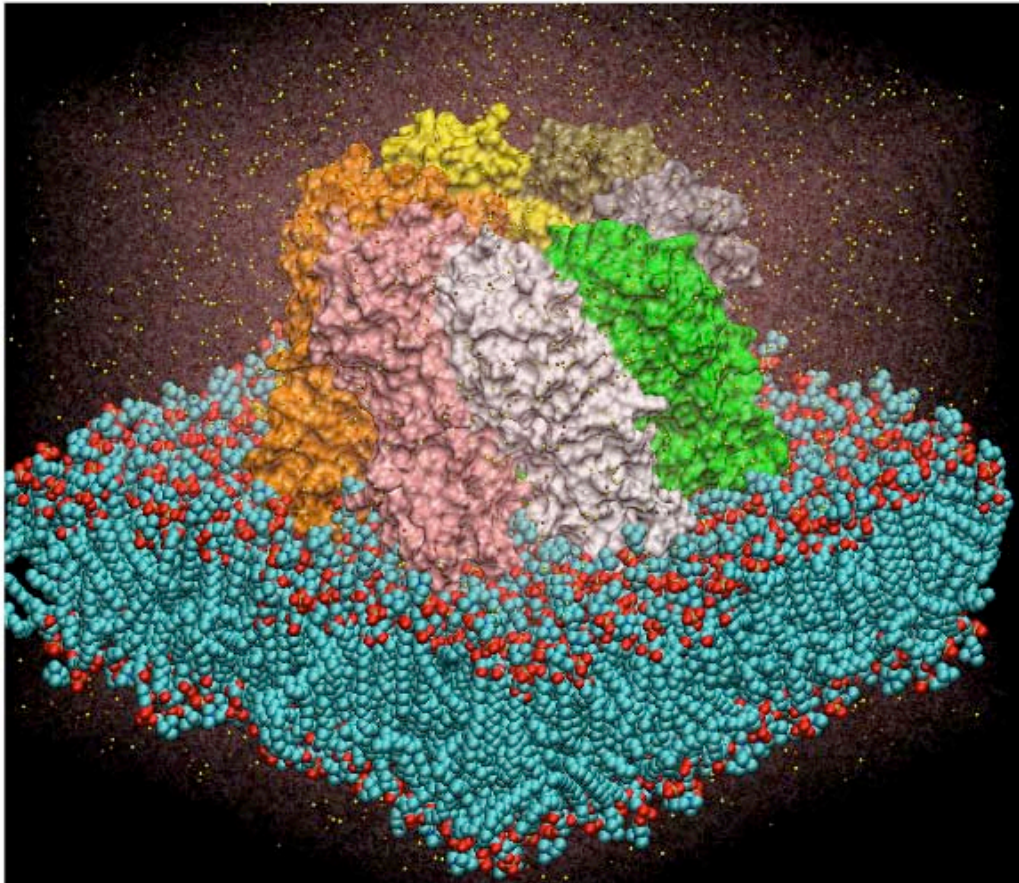
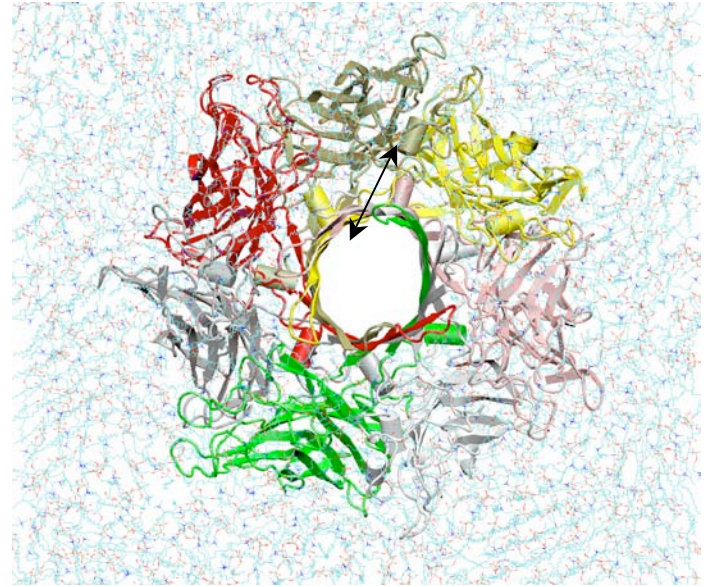
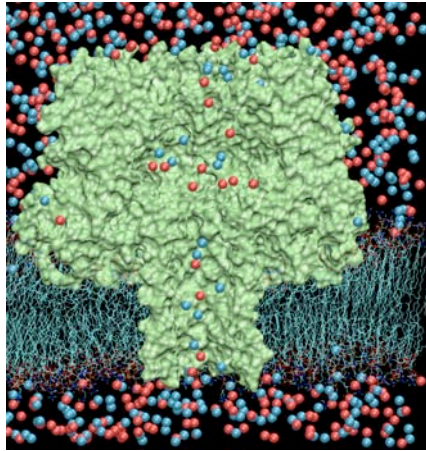
# Gramicidin can form a proton wire



It also provides a membrane channel with a simple structure which can be simulated for a long time.

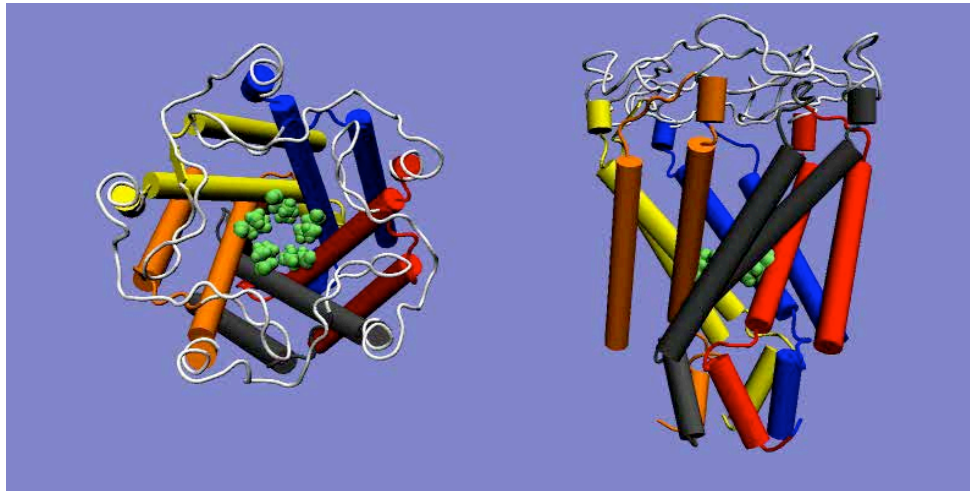


# $\alpha$ -hemolysin channel





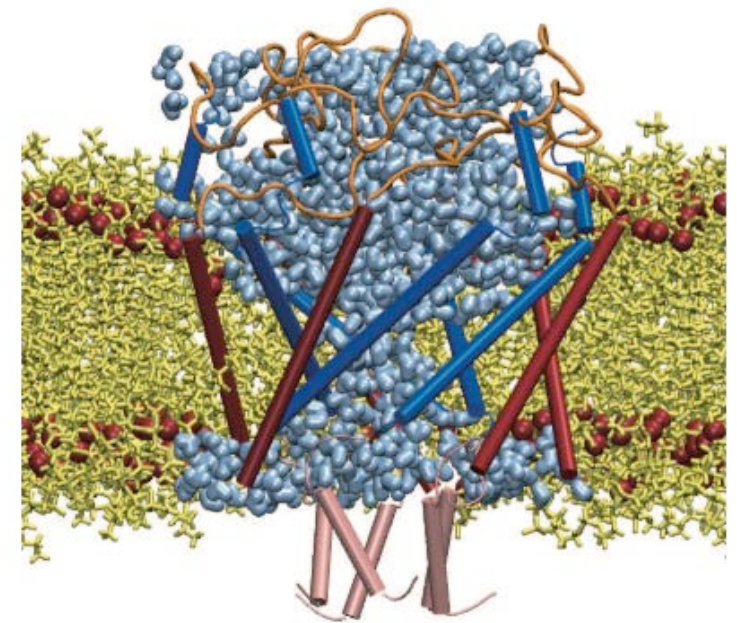
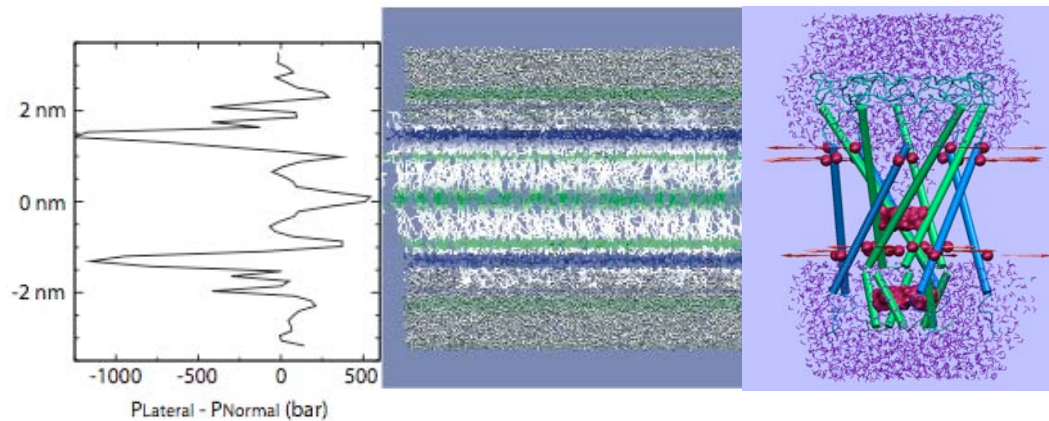
# Mechanosensitive Channel of Large Conductance (MscL)



## Steered MD simulation

- MscL from *E. coli* based on homology model
- Sufficient water for full hydration of loops and N-terminal helix bundle
- Constant radial force applied to residues at the ends of M1 and M2 (16, 17, 40, 78, 79, 98)
- 10 ns simulation time

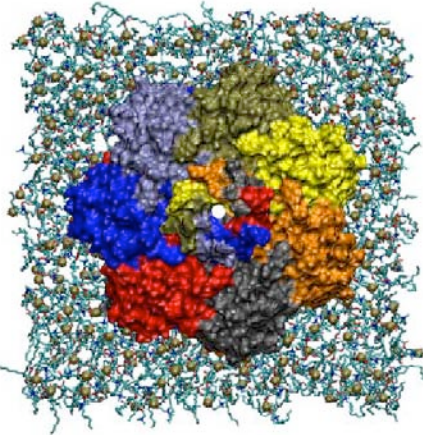
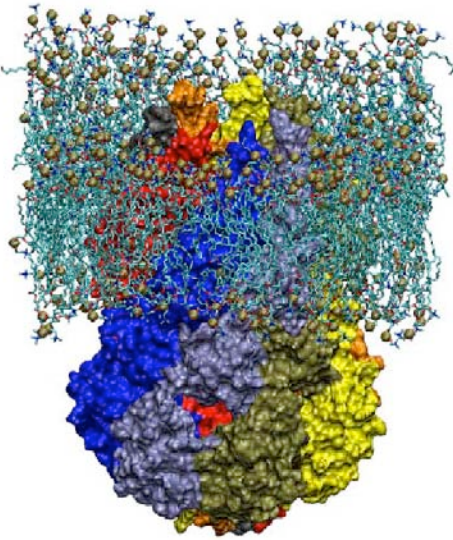
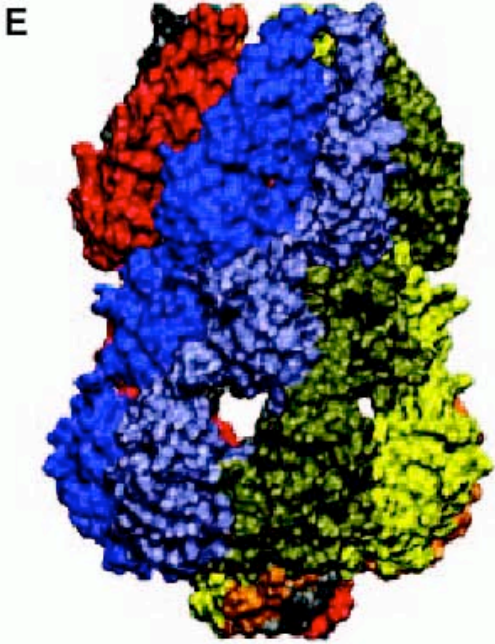
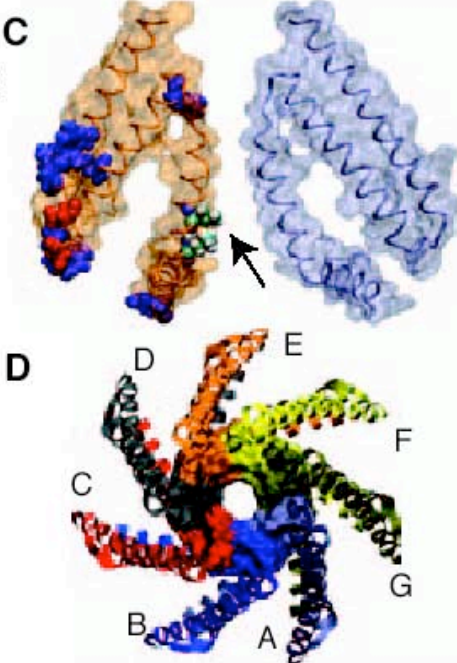
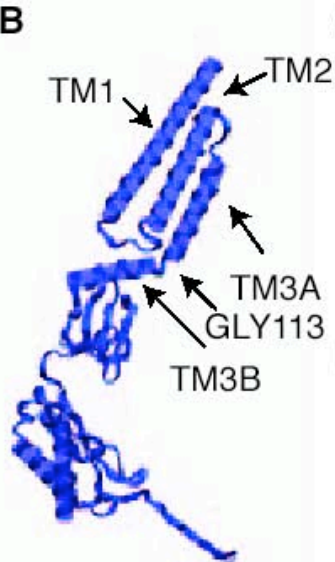
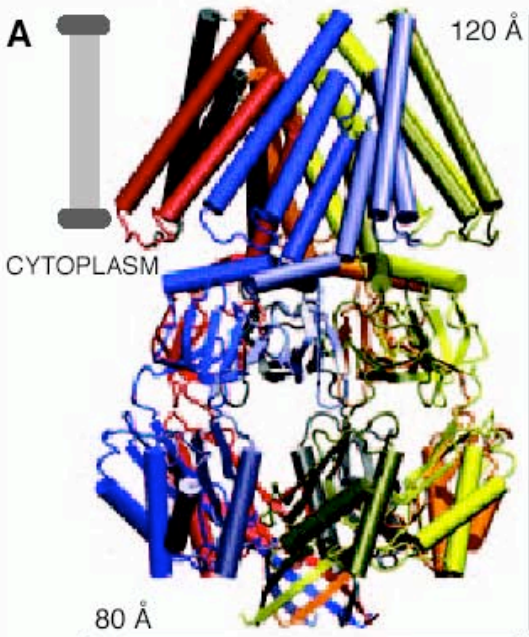
## Pressure profile in membrane guided simulation



J. Gullingsrud and K. Schulten, *Biophys. J.* **85**: 2087-2099 (2003)

Opening of channel

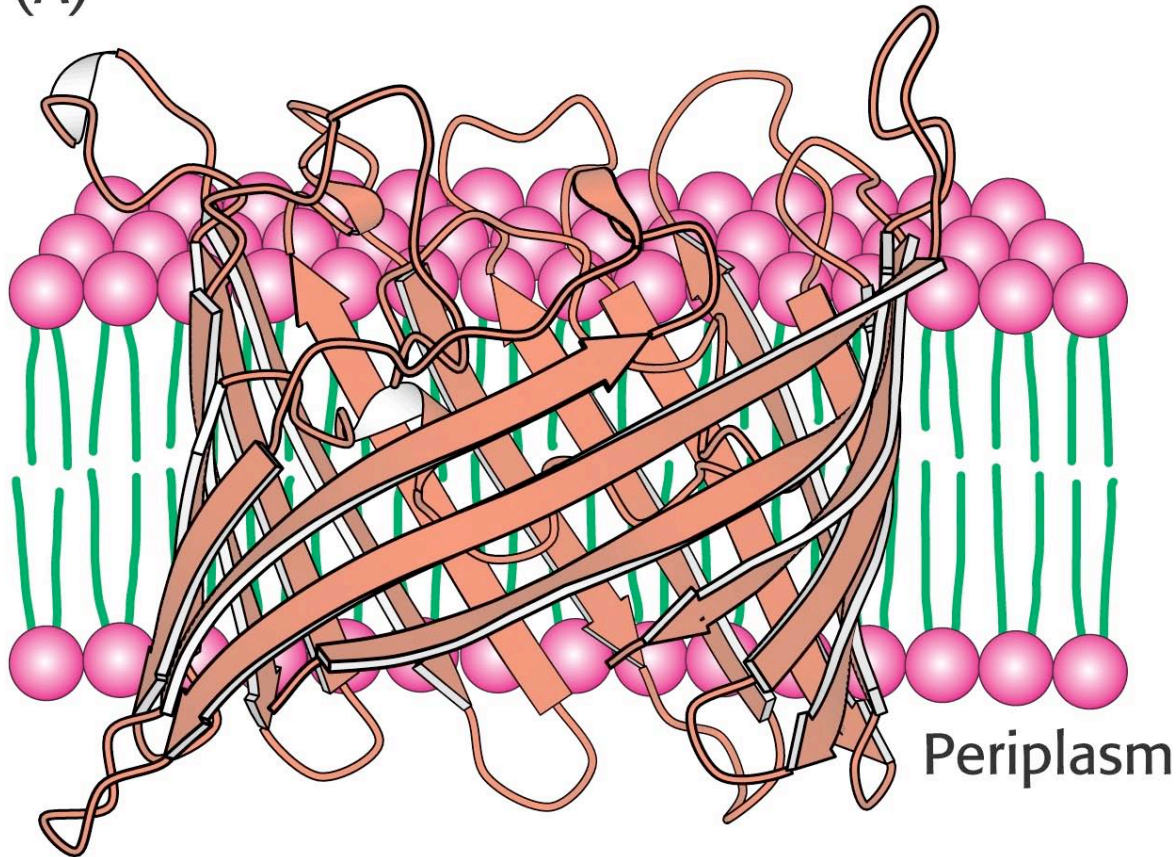
# Mechanosensitive Channel of Small Conductance



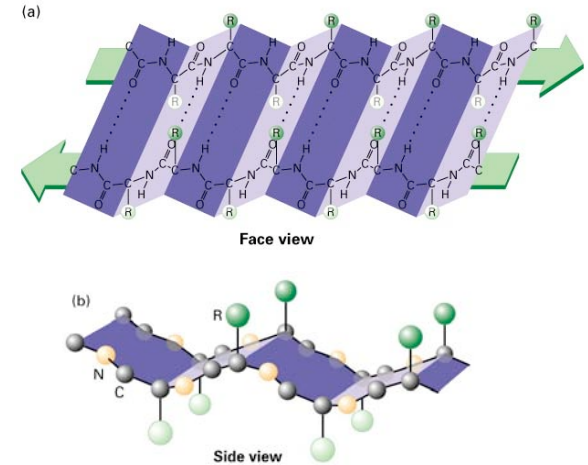


# Porins: An example of $\beta$ -barrel membrane proteins

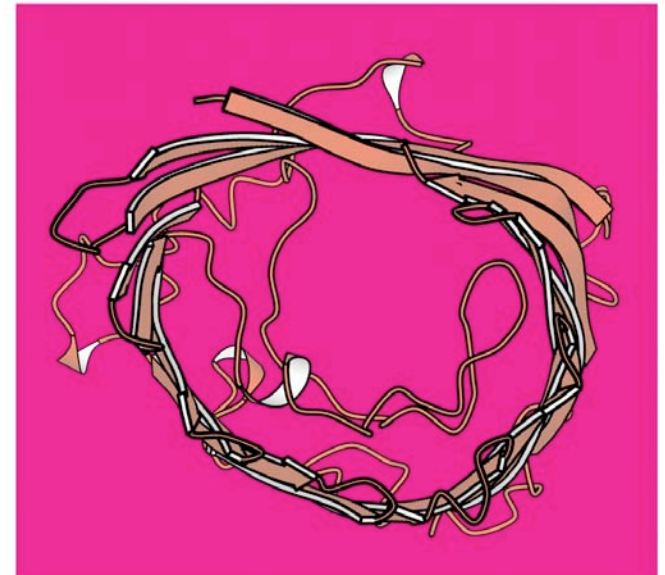
(A)



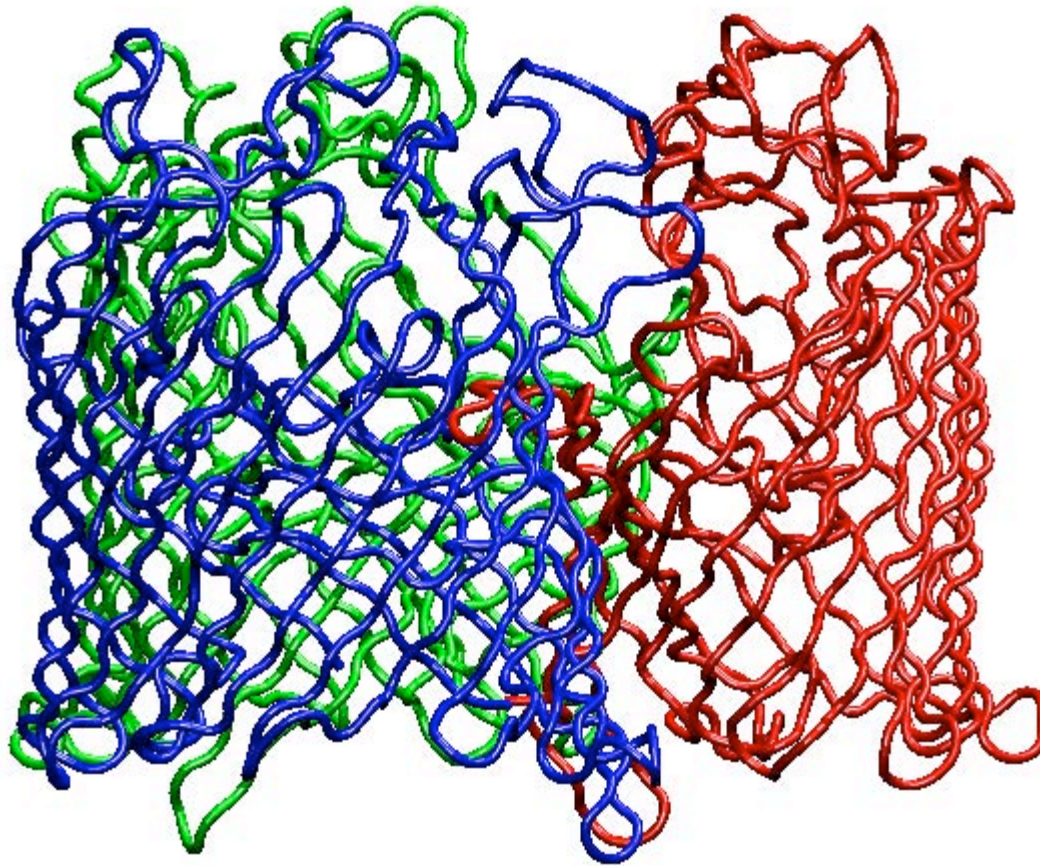
~18  $\beta$ -strands - found in outer membranes of  $G^-$  bacteria and mitochondria



(B)



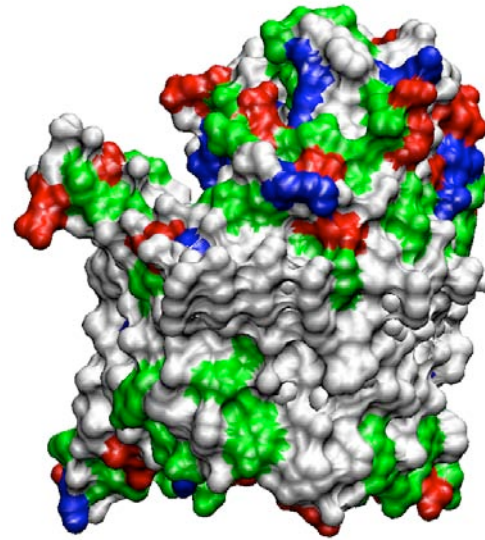
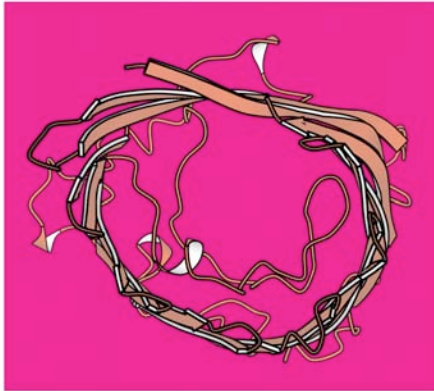
# Porins: An example of $\beta$ -barrel Membrane Proteins



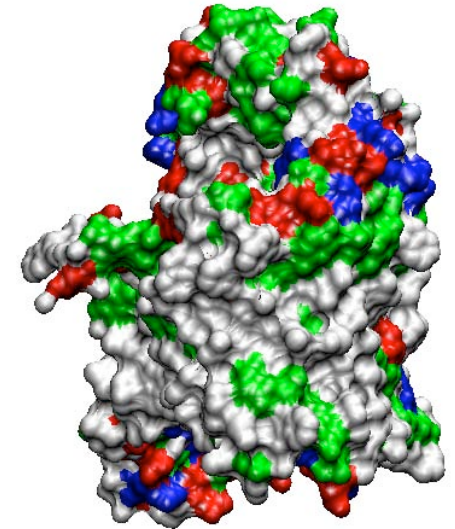
Usually form oligomers in the membrane.



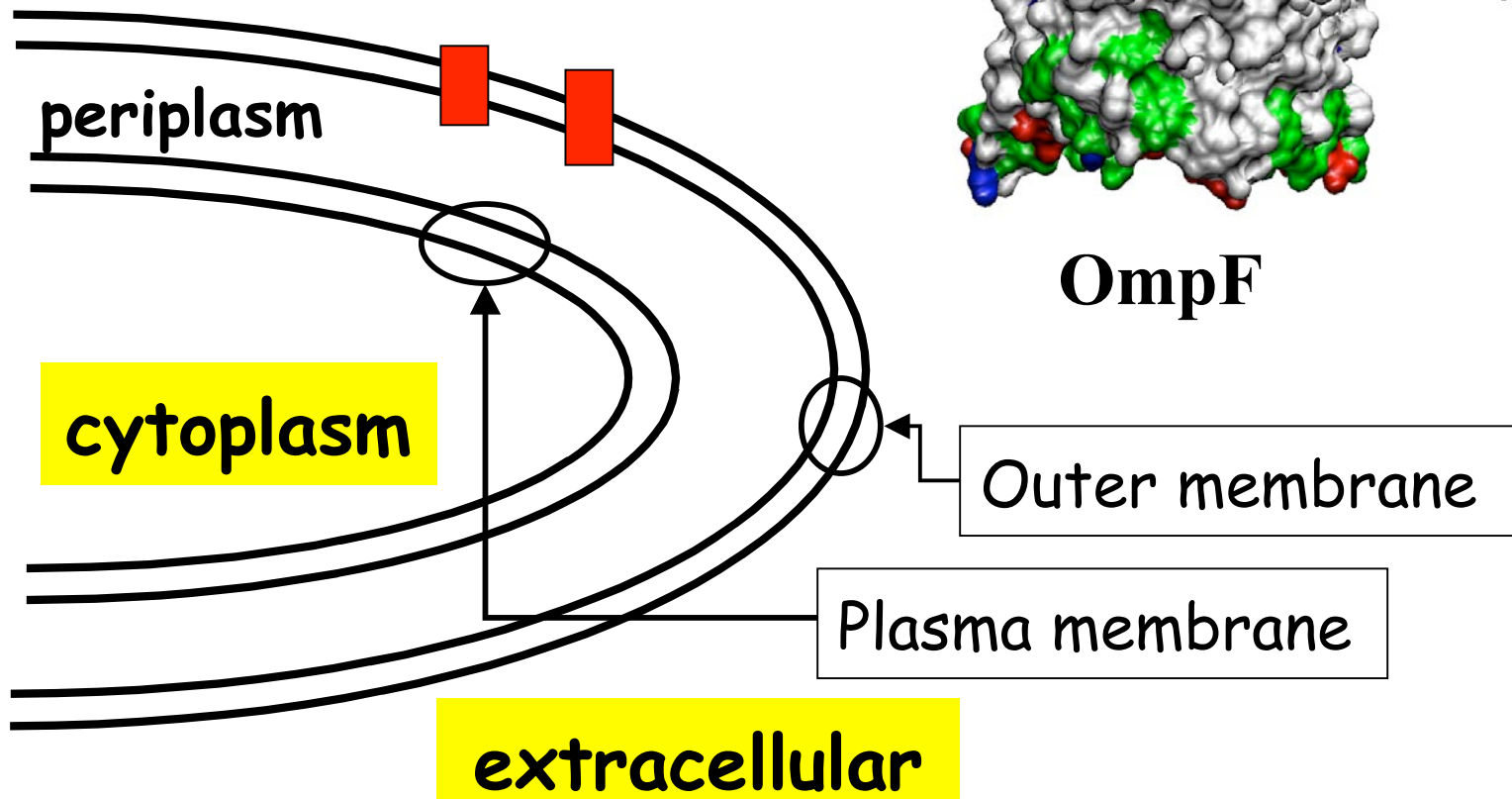
# Porins: Non-selective Pores of the Outer Membrane



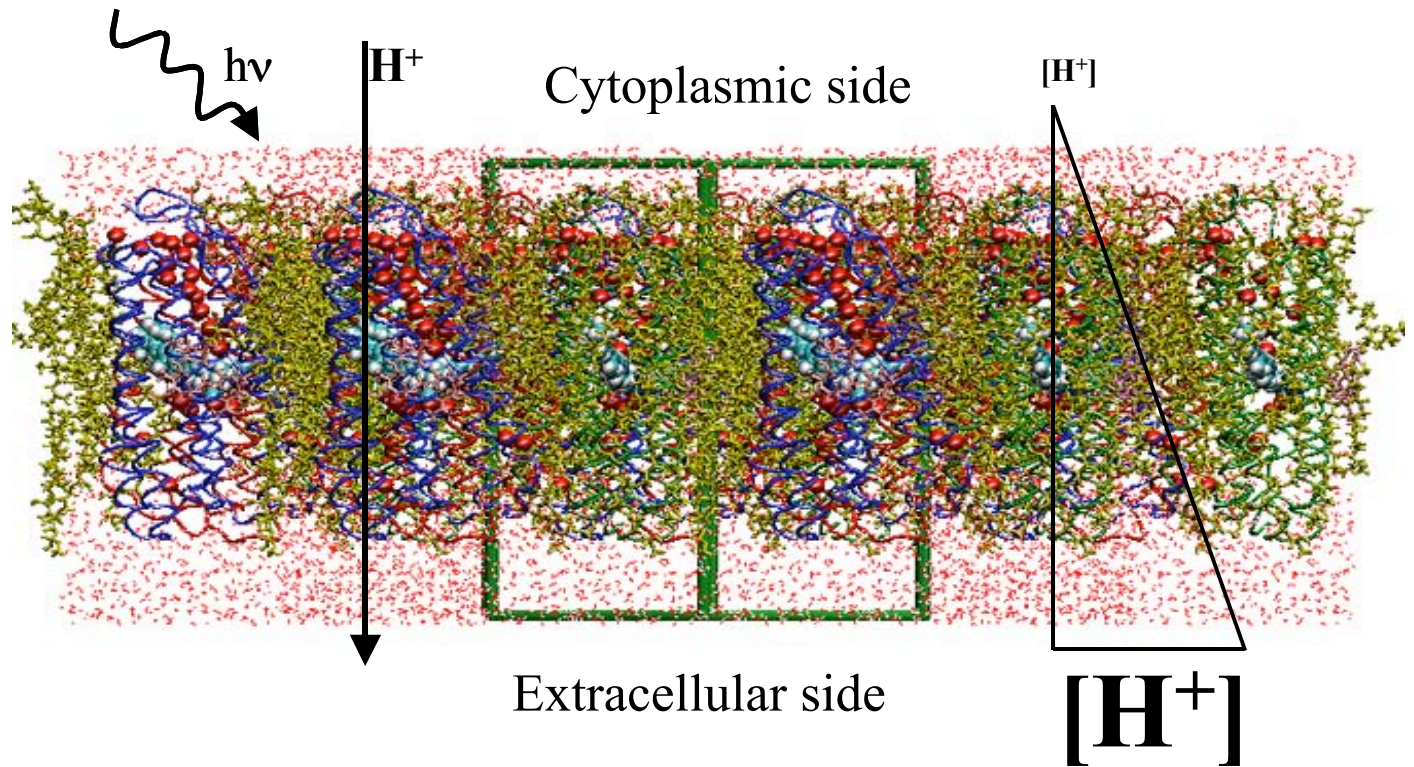
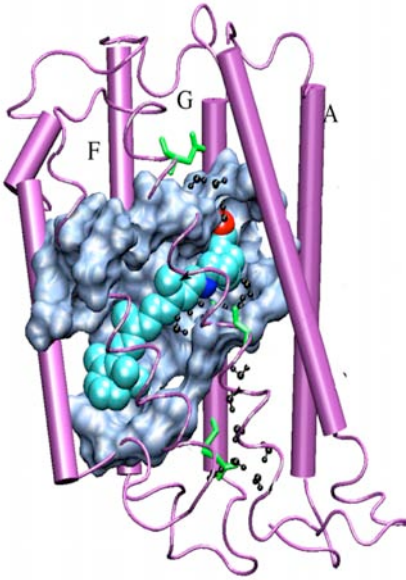
OmpF



Maltoporin



# Bacteriorhodopsin uses sunlight and generates a transmembrane proton gradient





# ATP synthase uses the proton gradient to produce ATP

